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(54) RUTHENIUM COMPLEX-BASED PHOTOSENSITIZER DYES FOR DYE-SENSITIZED SOLAR CELLS

(75) Inventors: Eric Wei-Guang Diau, Hsinchu County

(TW); Wei-Kai Huang, New Taipei

(TW)

(73) Assignee: National Chiao Tung University,

Hsinchu (TW)

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H01L 31/042 (2006.01)

(52) **U.S. Cl.**

USPC 546/2; 136/252

(58) Field of Classification Search

USPC 546/2; 136/252 See application file for complete search history.

(56) References Cited

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Primary Examiner — Charanjit Aulakh

(74) Attorney, Agent, or Firm — Muncy, Geissler, Olds & Lowe, P.C.

(57) ABSTRACT

This invention relates to ruthenium complex-based photosensitizer dyes for dye-sensitized solar cells, which have a general structural formula represented by formula (I).

OZ₂
ON Normalia (I)

Normalia (I)

Normalia (I)

9 Claims, 1 Drawing Sheet

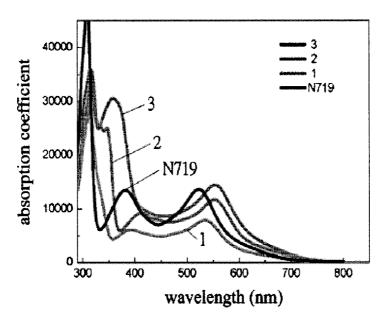


FIG. 1

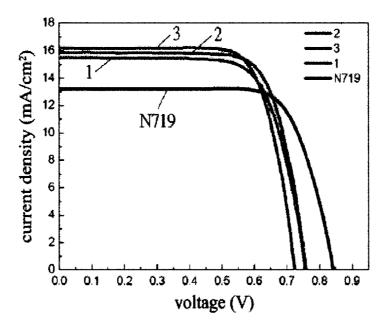


FIG. 2

40

45

1

RUTHENIUM COMPLEX-BASED PHOTOSENSITIZER DYES FOR DYE-SENSITIZED SOLAR CELLS

CLAIM OF PRIORITY

This application claims the priority benefit of Taiwanese Application Serial Number 100117109, filed on May 16, 2011. All disclosure of the Taiwanese application is incorporated herein by reference.

BACKGROUND OF THE INVENTION

A. Field of this Invention

This invention relates to a type of ruthenium complex photosensitizer dyes. More specifically, this invention relates to the type of ruthenium complex photosensitizer dyes for dye-sensitized solar cells.

B. Description of the Relevant Art

In conventional dye-sensitized solar cells, N719 photosensitizer dye is usually employed. However, since the N719 photosensitizer dye has a relatively large molecular structure, the sun light is not easy to be adsorbed onto thin titanium dioxide photoanodes of solar cells, leading to low photoelectric current. Therefore, the conventional dye-sensitized solar cells employing the N719 photosensitizer dye have low current density, and the overall solar elements have low efficiency. In this situation, in order to improve photoelectric current, the number of layers of the titanium dioxide photoanode element must be increased, which, in turn, increases the cost for manufacturing.

SUMMARY OF THE INVENTION

This invention relates to a type of ruthenium complex photosensitizer dyes for dye-sensitized solar cells, which are represented by the general formula (I):

where Z_1 and Z_2 individually represent H, Li, Na, or tetraalkyl ammonium group represented by the formula (a), 2

$$X_2$$
 X_3
 N
 X_1
 X_4
 X_4

where X_1 to X_4 individually represent C_mH_{2m+1} (m is an integer from 1 to 6),

In formula (I), A is one of the followings: H or C_mH_{2m+1} (m is an integer from 1 to 15), or $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15);

$$X;$$
 X

$$X$$
 X
 X
 X
 X

117

$$\begin{array}{c} R_1 \\ \downarrow \\ N \\ R_1; \end{array}$$

$$X$$
 R_4 ;

$$R_{7}$$
;

-continued

$$\begin{array}{c}
R_{11} \\
\downarrow \\
N
\end{array}$$

$$\begin{array}{c}
R_{13} \\
N \\
\end{array}$$
130

$$OR_{14}$$
 N
 S

In formulae (110) to (132), n is zero or an integer from 1 to 15.

In formula (I), B, C, and D are individually selected from the followings: H, or C_mH_{2m+1} (m is an integer from 1 to 15), or $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15);

60

$$R_{17}$$
;

$$R_{18}$$
;

$$OR_{20}$$
;

$$R_{21};$$

$$R_{24}$$
;

154

-continued

 $\begin{array}{c} & & & & 149 \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & &$

$$\begin{array}{c} & & & 150 \\ \hline & & & \\ X & & & \\ X & & & \\ \end{array}$$

$$R_{28}$$
 R_{28} R

$$R_{29}$$
 R_{29}
 R_{29}
 R_{29}
 R_{29}
 R_{29}
 R_{29}

-continued

158

$$X$$
 X
 X
 X

$$\begin{array}{c} X \\ X \\ X \\ X \end{array}$$

OR₃₃;

R₃₄;

R₃₅;

R₃₅

167

168

20 N 25

 OR_{37} ; OR_{37} ; OR_{37} OR_{37} OR_{37} OR_{37} OR_{37}

170

R₃₉;

X R_{40} ;

-continued

 R_{41} ;

 $\begin{array}{c} X \\ \hline \\ X \\ \hline \\ X \end{array}$ R₄₃;

R₄₄ R₄₄;

 R_{45}

178 N

OR₄₆

R₄₇ , ,

15

20

50

60

182

where R_3 , R_4 , R_7 , R_9 , R_{10} , R_{21} , R_{22} , R_{24} , R_{26} , R_{27} , and R_{39} to R_{43} represent $C_m H_{2m+1}$ (m is zero or an integer from 1 to 15);

R, R_{12} to R_{15} , R_{19} , R_{28} to R_{33} , R_{37} , and R_{44} to R_{48} represent C_mH_{2m+1} (m is an integer from 1 to 15);

 R_1, R_{11}, R_{17} , and R_{35} represent $C_m H_{2m+1}$ (m is an integer from 35 integer from 1 to 6), 1 to 15) or phenyl;

 R_2 , R_{16} , and R_{34} represent $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15);

 R_{18}, R_{20}, R_{36} , and R_{38} represent $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ or $_{40}$ C_mH_{2m+1} (p is an integer from 1 to 30, m is an integer from 1 to 15).

In formulae 123, 125, 147, 148, 150, 170, 172, 173 and 175, X represents Se, S or O,

In formulae 110 to 115 and 134 to 138, X represents F, Cl, Br, I or C_mH_{2m+1} (m is an integer from 1 to 15).

Other aspects and advantages of this invention will become apparent from the following detailed descriptions.

BRIEF DESCRIPTION OF THE FIGURES

FIG. 1 shows the UV-visible spectra of ruthenium complex photosensitizer dyes 1, 2 and 3 of this invention and conventional N719 dye.

FIG. 2 shows the IV curves of the ruthenium complex photosensitizer dyes 1, 2 and 3 of this invention and the conventional N719 dye.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

According to one embodiment of this invention, a ruthenium complex photosens tizer dye is represented by the following general formula (I):

where Z_1 and Z_2 individually represent hydrogen atom (H), lithium (Li), sodium (Na), or tetra-alkyl ammonium group represented by the following formula (a),

$$X_3$$
 X_2
 X_3
 X_4
 X_4
 X_4
 X_4

where X_1 to X_4 individually represent C_mH_{2m+1} (m is an integer from 1 to 6),

In formula (I), A is one of the followings: H, or C_mH_{2m+1} (m is an integer from 1 to 15), or $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15);

$$X;$$
 X

30

-continued

$$X$$
 X
 X
 X
 X
 X
 X

$$R_1$$
 120 45 N R_1 ;

$$R_2$$
;

$$X$$
 R_4 ;

$$R_{7}$$
;

$$R_9$$
;

$$R_{II}$$
 R_{II}
 R_{II}
 R_{II}
 R_{II}

$$\begin{array}{c} R_{13} \\ \downarrow \\ N \\ \downarrow \\ S \end{array};$$

5 N N 10

$$OR_{14}$$
132
15
 N
20

In the formulae (110) to (132), n is zero or an integer from 1 to 15,

In formula (I), B, C, and D are individually one of the followings: H, or C_mH_{2m+1} (m is an integer from 1 to 15), or 30 $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15);

$$\begin{array}{c} X \\ X \\ X \\ X \end{array}$$

$$R_{18}$$
;

$$OR_{19}$$
;

 OR_{20} ; 145 OR_{20} OR_{20} OR_{20}

$$\begin{array}{c} X \\ \end{array} \begin{array}{c} R_{22}; \end{array} \hspace{1cm} 25$$

$$R_{24}$$
; R_{24} ; R_{2

$$R_{26}$$
; R_{26} ; R_{2

$$\begin{array}{c} R_{29} \\ \\ \end{array}$$

$$\begin{array}{c} R_{31} \\ \downarrow \\ N \\ \downarrow \\ S \end{array}$$

$$X$$
 X
 X
 X
 X
 X
 X

40

$$R_{36}$$
; R_{36} ; R_{36} R_{36}

$$R_{38}$$
;

$$X$$
 R_{40} ;

$$R_{41}$$
;

$$\begin{array}{c} X \\ \hline \\ X \\ X \end{array}$$
 R₄₃;

-continued

OR46

N

R47

N

N

181 40 N N S

OR₄₈
50
N
55

where R_3 , R_4 , R_7 , R_9 , R_{10} , R_{21} , R_{22} , R_{24} , R_{26} , R_{27} , and R_{39} to R_{43} represent C_mH_{2m+1} (m is zero or an integer from 1 to 15); R, R_{12} to R_{15} , R_{19} , R_{28} to R_{33} , R_{37} , and R_{44} to R_{48} represent C_mH_{2m+1} (m is an integer from 1 to 15);

 R_1, R_{11}, R_{17} , and R_{35} represent $C_m H_{2m+1}$ (m is an integer from 1 to 15) or phenyl;

 R_2 , R_{16} , and R_{34} represent $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15);

 R_{18}, R_{20}, R_{36} , and R_{38} represent $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ or C_mH_{2m+1} (p is an integer from 1 to 30, m is an integer from 1 to 15).

In formulae 123, 125, 147, 148, 150, 170, 172, 173 and 175, X represents Se, S or O.

In formulae 110 to 115 and 134 to 138, X represents F, Cl, Br, $_{10}$ I or C_mH_{2m+1} (m is an integer from 1 to 15).

Several exemplary embodiments will be described below to illustrate the processes of synthesis of the ruthenium complex photosensitizer dyes 1, 2 and 3 of this invention. The structures of the ruthenium complex photosensitizer dyes 1, 2 and 3 are showed below. It must be understood that the exemplary embodiments should be regarded as illustrative rather than restrictive.

HO O S S S F F

1. The Synthesis of Ligands of Ruthenium Complex Photosensitizer Dye 1 Step 1

Embodiment Number One

$$\begin{array}{c|c} & & & \\ & & \\ & & \\ NH_2 & + & \\ & & \\ NH_2 & + \\ & \\ NH_2 & + \\$$

Put 20 mL of polyphosphoric acid (PPA) into a double-necked flask, stir and preheat the solution to about 90 $^{\circ}$ C. Then, slowly add about 2.46 g (about 0.02 mol) of 2-picolinic acid into the double-necked flask, and keep stirring the mixture evenly for about 30 minutes. Afterwards add about 2.16 g (about 0.02 mol) of o-phenylenediamine, then raise temperature to about 150 $^{\circ}$ C. and keep stirring evenly the mixture for about 4 hours. When the reaction ends, lower the temperature of the mixture to about 100 $^{\circ}$ C. Then, carefully and rapidly pour the reactant into iced water, and neutralize the mixture to weak alkalinity (about pH 9) with 1M sodium

hydroxide (NaOH) solution. A pinkish purple color solid product is precipitated now. The solid is filtered out by suction, then it is dried by heat and purified by column chromatograph using a hexane/ethyl acetate (EA) solution (1:2) as eluent. About 2.21 g of white color solid is obtained with the yield of about 56.7%. The white solid product is 2-(pyridin-2-yl)-1H-benzimidazole.

Step 2

10

20

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Mix about 0.195 g (about 1 mmol) of the white solid product obtained in Step 1, about 0.276 g (about 2 mmol) of ³⁰ K₂CO₃, and about 15 mL of N,N'-dimethylformamide (DMF) in a reaction vessel and keep stirring for about 10 minutes. Then, carefully inject about 0.385 mL (about 3 mmol) of 2,4-difluorobenzyl bromide into the mixture by a syringe and let the mixture react for about 3 hours at room temperature. When the reaction ends, pour the reactant into iced water and extract using EA. Then collect and dehydrate the organic layer using anhydrous sodium sulfate. After filtration and concentration, a yellow color viscous liquid is obtained. Then purify the yellow viscous liquid by column chromatograph using a hexane/EA solution (3:1) as eluent. Then, dry it by vacuum suction and collect about 0.242 g of beige-white color solid (i.e. the ligands L1). The yield is about 75.3%.

2. The Synthesis of Ruthenium Complex Photosensitizer Dye 1 $\,$

Step 2

-continued

The steps of synthesis of ruthenium complex photosensitizer dye 1 are shown below.

- (1) In the environment of DMF and argon (Ar), two equivalents of the ligand L1 (about 0.257 g, about 0.8 mmol) $_{40}$ are reacted with one equivalent of [RuCl₂(p-cymene)]₂ (about 0.244 g, about 0.4 mmol) at a temperature of about 70° C. for about 4 hours to form [Ru(L1)(pcymene) C1 coordination from broken dichloridebridged structure.
- (2) Then, add two equivalents of 4,4'-dicarboxy-2,2'-bipyridine (dcbpy, L) (about 0.195 g, about 0.8 mmol) and raise the temperature to about 140° C. for reaction for about 4 hours, resulting in [Ru(L)(L1)(C1)₂].
- (3) Finally, add excess amount of potassium thiocyanate (KSCN) and let the mixture react at about 140° C. for about 5 hours. After that, remove residual DMF from the reaction vessel using a distillation apparatus and KSCN residuals by water (H_2O) . After suction filtration, about 55 0.563 g of the ruthenium complex photosensitizer dye 1 is obtained. The yield is about 90%.

Embodiment Number Two

1. The Synthesis of Ligands of Ruthenium Complex Photosensitizer Dye 2

Step 1

Br
$$NH_2$$
 Δ PPA NH_2 N

Put 20 mL of polyphosphoric acid (PPA) into a doublenecked flask, stir and preheat to about 90° C. Then, slowly add about 2.00 g (about 0.02 mol) of 4-bromopyridine-2-carboxylic into the flask, and keep stirring the mixture evenly for about 30 minutes. Afterwards, add about 1.08 g (about 0.01 mol) of o-phenylenediamine and raise the temperature to about 150° C. and keep stirring the mixture evenly for about 4 hours. When the reaction ends, lower the temperature to about 100° C. Then, carefully and rapidly pour the reactant into iced water, and neutralize it to weak alkalinity (about pH 9) with 1M sodium hydroxide solution. A pinkish purple color solid product is precipitated. The solid product is filtered by suction and dried by heat, then purified by column chromatograph using a hexane/EA solution (1:3) as eluent. About 1.47 g of white color solid is obtained with the yield of 65 about 53.8%. The white solid product is 2-(4-bromopyridine-2-yl)-1H-benzimidazole.

Mix about 0.273 g (about 1 mmol) of the white solid 20 product obtained in Step 1, about 0.276 g (about 2 mmol) of K₂CO₃, and about 15 mL of DMF in a reaction vessel and stir for about 10 minutes. Then, carefully inject about 0.385 mL (about 3 mmol) of 2,4-difluorobenzyl bromide into the mixture by a syringe and keep at room temperature for about 3 hours. When the reaction ends, pour the reactant into iced water and extract using EA. Collect the organic layer and dehydrate it using anhydrous sodium sulfate. After filtration and concentration, a yellow color viscous liquid is obtained. Then purify the yellow viscous liquid by column chromatograph using a hexane/EA solution (2:1) as eluent. Then, after vacuum suction, about 0.273 g of beige-white color solid is obtained with the yield of about 68.3%. The beige-white solid product is 1-(2,4-difluorobenzyl)-2-(4-bromopyridin-2-yl)benzimidazole.

Step 3

$$Br + 45$$

Mix about 0.24 g (about 0.6 mmol) of the beige-white solid product obtained in Step 2, about 0.092 g (about 0.72 mmol) of thiophen-2-yl-2-boronic acid, about 0.0462 g (about 0.04 mmol) of Pd(PPh₃)₄, and about 20 mL of tetrahydrofuran (THF) in a reaction vessel. In the environment of nitrogen gas, add about 2 mL of 2M K₂CO₃, then the mixture is heated and refluxed for about 8 hours. When the reaction ends, pour the reactant into iced water and extracte using CH₂Cl₂. Collect and dehydrate the organic layer using anhydrous magnesium sulfate. After filtration and concentration, a light yellow color viscous liquid is obtained. Purify the light yellow viscous liquid by column chromatograph using a hexane/EA solution (10:1) as eluent. Then, about 0.22 g of beige-white color solid (i.e. the ligands L2) is obtained with the yield of about 91%.

2. The Synthesis of Ruthenium Complex Photosensitizer Dye 2

The steps of synthesis of ruthenium complex photosensi- $_{60}$ tizer dye 2 are shown below.

- (1) In the environment of DMF and argon, two equivalents of the ligand L2 (about 0.322 g, about 0.8 mmol) are reacted with one equivalent of [RuCl₂(p-cymene)]₂ (about 0.244 g, about 0.4 mmol) at a temperature of 65 about 70° C. for 4 hours to fog in [Ru(L2)(p-cymene)]Cl coordination from broken dichloride-bridged structure.
- (2) Then add two equivalents of 4,4'-dicarboxy-2,2'-bipyridine (dcbpy, L) (about 0.195 g, about 0.8 mmol) and raise the temperature to about 140° C. for reaction for about 4 hours, resulting in [Ru(L)(L2)(C1)₂].
- (3) Finally, add excessive amount of potassium thiocyanate (KSCN) into the mixture and let it react at about 140° C. for about 5 hours. After that, remove residual DMF in the reaction vessel using a distillation apparatus and KSCN

residuals using water . After suction filtration, about 0.612 g of ruthenium complex photosensitizer dye 2 is obtained with the yield of about 88.5%.

Embodiments Number Three

1. The Synthesis of Ligands of Ruthenium Complex Photosensitizer Dye 3 Step 1

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$$

Mix about 0.96 g (about 0.03 mol) of sulfur (S) powder, about 1.07 g (about 0.01 mol) of 2,4-lutidine, and about 1.08 g (about 0.01 mol) of o-phenylenediamine in a single-necked flask. Then, raise the reaction temperature to about 160° C. and keep stirring evenly for about 6 hours till the yellow 30 brown liquid turn into yellow solid. Finally, the reaction is terminated by addition of methanol. After filtering out the sulfur powder, the collected liquid is evaporated by a rotary concentrator. About 1.3 g of light yellow color solid is obtained with the yield of about 62.2%. The light yellow solid 35 is 2-(4-methylpyridin-2-yl)benzimidazole. Step 2

$$\begin{array}{c} N \\ N \\ N \\ \end{array}$$

$$\begin{array}{c} N \\ \end{array}$$

$$\begin{array}{c} K_2CO_3, Br \\ \end{array}$$

$$\begin{array}{c} N \\ \end{array}$$

$$\begin{array}{c} N \\ \end{array}$$

$$\begin{array}{c} N \\ \end{array}$$

$$\begin{array}{c} N \\ \end{array}$$

Mix about 0.209 g (about 1 mmol) of the light yellow solid product obtained in Step 1, about 0.276 g (about 2 mmol) of K_2CO_3 , and about 15 mL of DMF in a reaction vessel and keep stirring for about 10 minutes. Then, carefully inject about 0.385 mL (about 3 mmol) of 2,4-difluorobenzyl bromide into the mixture by a syringe, and let it react for about 3 hours at room temperature. When the reaction ends, pour the mixture into iced water then extracte using EA. Collect and

dehydrate the organic layer using anhydrous sodium sulfate. After filtration and concentration, a yellow color viscous liquid is obtained. Purify the yellow viscous liquid by column chromatograph using a hexane/EA solution (4:1) as eluent. Then, after vacuum suction, about 0.23 g of beige-white color solid is obtained with the yield of about 68.7%. The beige-white solid product is 1-(2,4-difluorobenzyl)-2-(4-methylpyridin-2-yl)benzimidazole. Step 3

Put about 0.335 g (about 0.001 mmol) of the beige-white solid product obtained in Step 2 in a 100 mL double-necked flask. Then, the flask is alternately evacuated then filled with nitrogen gas for three times. After that, add 10 mL of anhydrous THF into the flask. Then slowly add 2M lithium diisopropylamide (LDA, about 0.0012 mol, about 0.6 mL) at a temperature of -20° C. Keep stirring the mixture for about 30 minutes at above temperature. Add about 0.17 g (about 0.0015 mol) of thiophene-2-carbaldehyde dissolved in anhydrous THF into the flask. Keep stirring the mixture for about 10 minutes at above temperature then move to room temperature for reaction for about 2 hours. The reaction is terminated by addition of methanol (MeOH). Next, after THF is removed, the residue was extracted three times using CH₂Cl₂. Collect and dehydrate the organic layer using anhydrous magnesium sulfate. After filtration and concentration, about 0.172 g of light yellow color viscous liquid is obtained. Place the light yellow viscous liquid in a 100 mL single-necked flask then add about 10 mL of pyridine. Under ice bath condition, add POC1₃ (about 0.0012 mol/0.11 mL). Then put the mixture in room temperature for reaction for about 10 minutes. Terminate the reaction by adding MeOH. Next, after pyridine was removed, the residue was extracted three times using CH₂Cl₂ (about 10 mL) and saline water. Collect and dehydrate the organic layer using anhydrous magnesium sulfate. Then, after vacuum suction, a yellow color liquid is obtained. Purify the yellow liquid by column chromatograph using a hexane/EA solution (2:1) as eluent. About 0.17 g of beige-white color solid (i.e. the ligands L3) is obtained with the yield of about 39.6%.

2. The Synthesis of Ruthenium Complex Photosensitizer Dye 3

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The steps of synthesis of ruthenium complex photosensitizer dye 3 are shown below.

- (1) In the environment of DMF and argon, two equivalents of the ligand L3 (about 0.343 g, about 0.8 mmol) is reacted with one equivalent of $[RuCl_2(p\text{-cymene})]_2$ (about 0.244 g, about 0.4 mmol) at a temperature of about 70° C. for about 4 hours to form [Ru(L3)(p-cymene)]Cl coordination . from the broken dichloride-bridged structure.
- (2) Then add two equivalents of 4,4'-dicarboxy-2,2'-bipyridine (dcbpy, L) (about 0.195 g, about 0.8 mmol) to the mixture and raise the temperature to 140° C. for reaction for about 4 hours, resulting in [Ru(L)(L3)(Cl)₂].
- (3) Finally, add excessive amount of KSCN to the mixture and let react at about 140° C. for about 5 hours. After that, remove residual DMF in the reaction vessel using a distillation apparatus and residuals of KSCN using water. After suction filtration, about 0.586 g of ruthenium complex photosensitizer dye 3 is obtained with the yield of about 82.3%.
- FIG. 1 shows the comparison of UV-visible spectra of ruthenium complex photosensitizer dyes 1, 2 and 3 of this invention and conventional N719 dye.
- FIG. 2 shows the comparison of IV curves of the ruthenium complex photosensitizer dyes 1, 2 and 3 of this invention and conventional N719 dye.

Table 1 shows the comparison of photoelectric conversion efficiency of solar cell devices employing above-mentioned ruthenium complex photosensitizer dyes 1, 2 and 3 and conventional N719 dye. As shown the efficiency of the ruthenium complex photosensitizer dyes 1, 2 and 3 are better than that of the conventional N719 dye. The photoelectric conversion efficiency (η) is obtained by the equation below,

$$\eta = \frac{P_{mp}}{P_{in}} = \frac{J_{mp} * V_{mp}}{P_{in}} = \frac{J_{SC} * V_{OC} * FF}{P_{in}}$$

where P_{in} is the input radiation power, and P_{mp} is the maximum output power (= $J_{mp} \times V_{mp}$), and FF stands for the fill factor defined as

$$FF = \frac{J_{mp} * V_{mp}}{J_{SC} * V_{OC}}$$

$$45$$

where $L_{\it sc}$ is the short circuit current, and $V_{\rm \infty}$ is the open circuit voltage.

TABLE 1

Dye	V_{oc} (V)	J_{sc} (mA·cm ⁻²)	FF	η (%)	
1	0.760	15.490	71.84	8.46	55
2	0.762	15.774	74.37	8.94	
3	0.730	16.116	74.31	8.73	
N719	0.842	13.193	74.64	8.29	

Since the ruthenium complex photosensitizer dyes of this invention have smaller molecular structures than the conventional N719 photosensitizer dye, they can be adsorbed, in greater amount, on thin titanium dioxide photoanodes of solar cells, leading to larger photoelectric current. Therefore, the layers of the titanium dioxide photoanodes can be reduced 65 thus simplifying the processes and lowering the cost of manufacturing of the elements. Since the dye-sensitized solar cells

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employing the ruthenium complex photosensitizer dyes of this invention have higher photocurrent density than those using conventional N719 photosensitizer dye, the former has superior overall solar elements efficiency than the latter.

Although some embodiments of this invention are described in details above, it is intended that the scope of this invention may not be limited by the descriptions above, but rather by the claims appended hereto. Also, it is intended that the following appended claims be interpreted as that all possible alterations, pemiutations, and equivalents fall within the true spirit and scope of this invention.

What is claimed is:

1. A ruthenium complex photosensitizer dye for dye-sensitized solar cells, the ruthenium complex photosensitizer dye having the following general formula (I):

where Z_1 and Z_2 individually represent H, Li, Na, or tetraalkyl ammonium group represented by formula (a),

$$X_{2}$$
 X_{3}
 N
 X_{1}
 X_{2}
 X_{3}

In formula (a), X_1 to X_4 individually represent $C_m H_{2m+1}$ (m is an integer from 1 to 6),

In formula (I), A is one of the followings:

H, or C_mH_{2m+1} (m is an integer from 1 to 15), or CH_2 [OC_2H_4] $_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15);

$$R_2$$
;

$$R_4$$
;

$$R_7$$
;

$$R_9$$
;

$$R_{10} R_{10}$$

$$R_{II}$$
 R_{II}
 R_{II}
 R_{II}
 R_{II}
 R_{II}

$$\begin{array}{c} R_1 \\ N \\ N \\ R_1; \end{array}$$

129 OR_{12} 5 10

In formulae (110) to (132), n is zero or an integer from 1 to 45

In formula (1), B, C, and D are individually selected from the followings: H, or C_mH_{2m+1} (m is an integer from 1 to 15), or $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15);

$$X$$
 X
 X
 X
 X
 X
 X
 X
 X

$$R_{17}$$
;

OR₁₉;

OR₁₉

$$OR_{20}$$
; 145 15 OR_{20} ; OR_{20}

$$\begin{array}{c} 146 \\ \hline \\ 0 \\ \hline \end{array}$$

$$R_{22}$$
;

$$\begin{array}{c} & & & 149 \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\$$

$$R_{35}$$
; 167 R_{35} ; R_{35} R_{3

$$R_{36}$$
 OR_{37} ;

 169
 35

$$X$$
 R_{40} ;

$$R_{41}$$
;

$$\begin{array}{c}
X \\
X \\
X
\end{array}$$
R₄₃;

181

$$OR_{48}$$

where R_3 , R_4 , R_7 , R_9 , R_{10} , R_{21} , R_{22} , R_{24} , R_{26} , R_{27} , and R_{39} to R_{43} represent C_mH_{2m+1} (m is zero or an integer from 1

46 R, R_{12} to R_{15}, R_{19}, R_{28} to $R_{33}, R_{37},$ and R_{44} to R_{48} represent C_mH_{2m+1} (m is an integer from 1 to 15),

 R_1, R_{11}, R_{17} , and R_{35} represent $C_m H_{2m+1}$ (m is an integer from 1 to 15) or phenyl,

 R_2 , R_{16} , and R_{34} represent $CH_2[OC_2H_4]_pOC_mH_{2m+1}$ (p is an integer from 1 to 30, m is an integer from 1 to 15),

 R_{18} , R_{36} , and R_{38} represent $CH_2[OC_2H_4]_pOC_m$, H_{m2+1} or C_mH_{2m+1} (p is an integer from 1 to 30, in is an integer from 1 to 15);

10 In formulae 123, 125, 147, 148, 150, 170, 172, 173 and 175, X represents Se, S or O; in formulae 110 to 112, 114, 115 and 134 to 138, X represents F, Cl, Br, I or C_mH_{2m+1} (m is an integer from 1 to 15); and in formulae 113, X represents F, Cl, Br, or I;

15 with the provision that at least one of Z_1 , Z_2 , B, C and D is not H.

2. The ruthenium complex photosensitizer dye of claim 1, where Z_1 and Z_2 Represent H. 182

3. The ruthenium complex photosensitizer dye of claim 2, ²⁰ where A is represented by the formula (111).

4. The ruthenium complex photosensitizer dye of claim 3, in formula (111), X represents F and n is equal to 1.

5. The ruthenium complex photosensitizer dye of claim 4, where C and D represent H.

6. The ruthenium complex photosensitizer dye of claim 5, where B is represented by the formula (147).

7. The ruthenium complex photosensitizer dye of claim 6, in formula (147), X represents S and R₂₂ represents H.

8. The ruthenium complex photosensitizer dye of claim 5, where B is represented by the formula (172).

9. The ruthenium complex photosensitizer dye of claim 8, in formula (172), X represents S and R₄₀ represents H.