

White-Light-Emitting Diodes Using Red-Emitting LiEu(WO₄)_{2-x}(MoO₄)_x Phosphors

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The fundamental luminescence and chromaticity properties of a class of red-emitting phosphors $\text{LiEu}(WO_4)_{2-x}(MoO_4)_x$ under near-UV excitation have been investigated. The photoluminescence (PL) and photoluminescence excitation (PLE) spectra reveal that the absorption is mainly attributed to $O \to Mo(W)$ charge transfer at 300 nm and the Eu^{3+} transitions in near-UV and visible regions, and red emissions are assigned to intraconfigurational 4f-4f transitions of Eu^{3+} . Both PL and PLE intensity were found to increase with increasing doped Mo content. The intensity of the $^5D_0 \to ^7F_2$ emission of Eu^{3+} activated at 394 nm was found to reach a maximum when the relative ratio of Mo/W was 2:0. The CIE chromaticity coordinates were found to be (0.66, 0.33) for $\text{LiEu}(MoO_4)_2$ and it reached the same level as the commodity phosphor $Y_2O_2S:Eu^{3+}$. A white-light-emitting diode fabricated using the composition-optimized $\text{LiEu}(MoO_4)_2$ and the matching green- and blue-emitting phosphors shows that the quality of red color reproduction based on $\text{LiEu}(MoO_4)_2$ was found to be much higher and improved than that using $\text{La}_2O_2S:Eu^{3+}$. © 2007 The Electrochemical Society. [DOI: 10.1149/1.2731042] All rights reserved.

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Solid-state lighting (SSL) using light-emitting diode (LED) and phosphor materials to generate white light is the current research focus in the lighting industry. SSL technology has several advantages over conventional fluorescent lamps such as reduced power consumption, compactness, efficient light output, and longer lifetime. The availability of white light (WL)-LEDs should open up a great number of exciting new application fields: white light sources to replace traditional incandescent and fluorescent lamps, and backlights for portable electronics, medical, and architecture lightings, etc.1 In the first WL-LED produced in 1996, an excellent method using a blue-LED chip in combination with a phosphor that exhibits yellow emission under blue excitation was proposed.² The yellowemitting phosphor is trivalent-cerium-activated yttrium aluminum garnet (YAG), $(Y,Gd)_3(Al,Ga)_5O_{12}$: Ce^{3+} . The luminous efficacy (η_I) of commercial WL-LEDs is above 25 lm/W at a forward-bias current (I_f) of 20 mA at room temperature. This value is higher than the efficacy of an incandescent lamp (15 lm/W) but lower than that of fluorescent tubes (75 lm/W).4 Nowadays, the products of WL-LEDs have a general color-rendering index (Ra) of 75-80, which is enough for universal illumination.⁵ However, the WL-LEDs based on blue LED plus yellow phosphor have several disadvantages, including the halo effect of blue/yellow color separation and poor color-rendering properties caused by lack of red component in the spectrum.

There are two solutions proposed to solve the problems. One method is to excite multiphosphors using near-ultraviolet (n-UV) LED sources. And The other method employing a blue-LED chip and yellow-emitting YAG and sulfide phosphors was proposed in order to improve its color-rendering property. However, the sulfide-based phosphors are inherently chemically unstable and the lifetime of the sulfide materials is deficient under excitation with ultraviolet irradiation. It is regarded to be very difficult to use sulfide phosphors in WL-LEDs for general illumination. Hence, there has been a widespread and growing interest in developing new families of redemitting phosphors with high absorption in the near-UV to blue region. In this work we report a class of new red-emitting phosphors LiEu(WO₄) $_{2-x}$ (MoO₄) $_x$, based on which along with experimentally optimized green- and blue-emitting phosphors a WL-LED was fabricated and its performance was investigated.

Experimental

The samples of $LiEu(WO_4)_{2-x}(MoO_4)_x$ phosphors were prepared by first ballmilling the reactant mixture consisting of a stoichiometric amount of Li₂CO₃ (99.9%), Eu₂O₃ (99.99%), MoO₃ (99.9%), and WO₃ (99.9%) (all from Aldrich Chemicals, Milwaukee, WI, USA.) for 20-30 min and then fired at 600-800°C for several hours. The photoluminescence (PL) and photoluminescence excitation (PLE) spectra of LiEu(WO₄)_{2-x}(MoO₄)_x ($0 \le x \le 2.0$) were measured by using a Spex Fluorolog-3 spectrofluorometer (Instruments S.A., NJ, USA) equipped with a 450 W Xe light source and double excitation monochromators. The Commission International de l'Eclairage (CIE) chromaticity coordinates for all samples were determined by a Laiko DT-100 color analyzer equipped with a charge-coupled device (CCD) detector (Laiko Co., Tokyo, Japan). To compare the performance of $LiEu(MoO_4)_2$ and the commodity of La₂O₂S:Eu³⁺ (Kasei Optonix KX-681B) phosphors in the application of SSL, phosphor-conversion WL-LEDs were fabricated. Based on standard LED technology, the first WL-LED was fabricated by using an n-UV LED chip (average $\lambda_{em} = 392$ nm Cree, Catalog no. C395MB290-S0100) in pumping $BaMgAl_{10}O_{17}:Eu^{2+}$ (Kasei optonix KX-661), (Ba,Sr)–Si–Al–O:Eu $^{2+}$,Dy $^{3+}$ (Nantex RU-G534) and LiEu(MoO₄)₂ phosphors simultaneously. The above-mentioned BaMgAl₁₀O₁₇:Eu²⁺ and Eu²⁺,Dy³⁺-codoped (Ba,Sr)-Si-Al-O phosphors have been tested to be excellent commodities because of their nontoxicity, thermal stability, and intense luminescence properties. The phosphors were encapsulated in a transparent epoxy resin (KBIN A2015). The second WL-LED based on La₂O₂S:Eu³⁺ and the same green- and blue-emitting phosphors was also fabricated at the same collocation for comparison. The optimized weight ratio of $R[LiEu(MoO_4)_2]$:G:B phosphors is 0.92:0.06:0.02, whereas that for the R(La₂O₂S:Eu³⁺):G:B phosphors combination is 0.69:0.23:0.08. This ratio change depends on the target color of the white region. The relative emission spectra of n-UV chip WL-LED based on LiEu(MoO₄)₂ and La₂O₂S:Eu³⁺ operated at room temperature and at I_f of 20 mA were measured using a 50 cm single-grating monochrometer.

Results and Discussion

Rietveld structural refinement was carried out by using the GSAS software package and reported elsewhere. Briefly, both LiEu(WO₄)₂ and LiEu(MoO₄)₂ are found to be isostructural with KEu(MoO₄)₂, which are crystallized in the triclinic space group P1 with a = 10.4287(3) Å, b = 5.2079(1) Å, c = 6.7350(2) Å,

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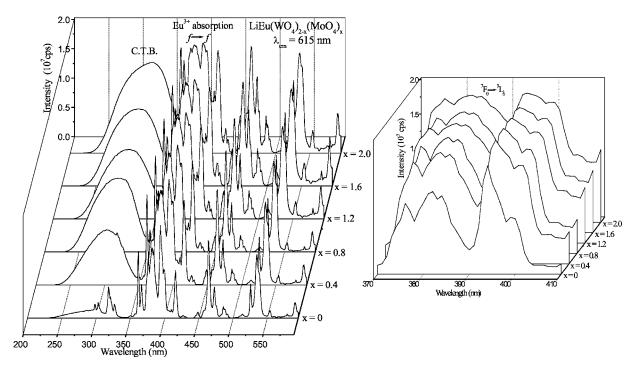


Figure 1. PLE spectra for $LiEu(WO_4)_{2-x}(MoO_4)_x$ phosphors with $x = (a) \ 0$, (b) 0.4, (c) 0.8, (d) 1.2, (e) 1.6, and (f) 2.0.

 $\alpha = 112.764(5)^{\circ}, \qquad \beta = 112.716(3)^{\circ}, \qquad \gamma = 89.930(3)^{\circ}, \\ V = 306.372(2) \text{ Å}^3, Z = 2 \text{ for LiEu}(WO_4)_2, \text{ and } a = 10.4094(3) \text{ Å}, \\ b = 5.1989(2) \text{ Å}, \qquad c = 6.7540(2) \text{ Å}, \qquad \alpha = 112.686(3)^{\circ}, \\ \beta = 112.575(2)^{\circ}, \quad \gamma = 90.043(4)^{\circ}, \quad V = 306.604(4) \text{ Å}^3, \quad Z = 2 \text{ for LiEu}(MOO_4)_2. \\ \text{9 The final weighted-profile } R \text{ values } (R_{\text{wp}}) \text{ of } 6.69\% \\ \text{and } 4.29\% \text{ were reported by Chiu et al. for LiEu}(WO_4)_2 \text{ and LiEu}(MOO_4)_2, \text{ respectively.} \\ \text{9}$

Figure 1 shows the PLE spectra of six samples with selected compositions of $\text{LiEu}(WO_4)_{2-x}(MoO_4)_x$ (x = 0, 0.4, 0.8, 1.2, 1.6, 2.0) monitored at 615 nm that is attributed to ${}^5D_0 \rightarrow {}^5F_2$ emission of Eu³⁺ ions. The intense broad band at near 300 nm is assigned as the charge-transfer (CT) transition originated from $O \rightarrow W$ or O

Mo (i.e., ligand to metal charge transfer). However, the CT band of Eu^{3+} – O^{2-} was not clearly observed in the excitation spectra, which could be due to possible overlap of the CT band with that of tungstate or molybdate group. In the range from 350 to 550 nm, all samples show characteristic intraconfigurational 4f-4f absorption transitions of Eu³+: sharp $^7F_0 \rightarrow ^5L_6$ transition for 394 nm, $^7F_0 \rightarrow ^5D_2$ transition for 465 nm, and $^7F_1 \rightarrow ^5D_1$ transition for 535 nm. As compared with the O \rightarrow Mo(W) CT band, remarkable changes were observed in the intensity of characteristic absorptions of Eu³⁺ ion in the PLE spectra shown in Fig. 1, respectively; the maximum absorption of the Eu³+ $^7F_0 \rightarrow ^5L_6$ peak becomes stronger with increasing doped molybdenum content. A three-dimensional (3D) spectral representation at the right of Fig. 1 shows the expanded PLE spectra in the spectral range of 370-410 nm for comparison in detail.

Figure 2 shows the Mo-content dependence of PL spectra for LiEu(WO₄)_{2-x}(MoO₄)_x (x = 0, 0.4, 0.8, 1.2, 1.6, 2.0) phosphors excited by n-UV radiation at 394 nm. The spectra essentially consist of sharp emissions with wavelengths ranging from 580 to 720 nm, which are associated with the $^5D_0 \rightarrow ^7F_J$ (J = 1, 2, 3, 4) multiplet transitions from the excited levels of Eu³⁺ to the ground state, although no emission corresponding to tungstate or molybdate is observed. However, the presence of an absorption band due to a tungstate or molybdate group in the excitation spectrum of Eu³⁺, when monitored for Eu³⁺ emission (at 615 nm), very likely suggests that

the energy absorbed by the WO_4^{2-}/MoO_4^{2-} group is transferred to Eu³⁺ levels nonradiatively. That is to say, the emission corresponding to Eu3+ ions has been observed under excitation of the CT band of either the WO_4^{2-} or MoO_4^{2-} group. This occurrence is known as "host-sensitized" energy transfer.¹⁰ However, the intensity of Eu³⁺ emission is weaker with CT band excitation when compared to that due to Eu3+ excitation. This observation reveals that the energy transfer from the MoO_4^{2-}/MoO_4^{2-} group to Eu^{3+} is inefficient, which was proposed to be strongly dependent on the bond angles of O–Mo(W)–O and Mo(W)–O–Eu $^{3+}$ by Blasse and Bril. 11 The presence of multiplets in the emission spectra is attributed to the (2J + 1) Stark components of J-degeneracy splitting. The ⁵D₀ is an unsplitted singlet band, simplifying in a significant way the application of the group theory and of electronic transition selection rules. In general, the Eu³⁺ emission lines are hypersensitive, that is to say, they are highly sensitive to the environment. The electric-dipole transition would be dominant when Eu³⁺ occupied the lattice site of a non-centrosymmetric environment in the scheelite phases. The

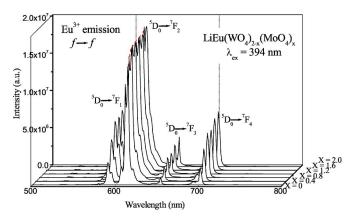


Figure 2. PL spectra of LiEu(WO₄) $_{2-x}$ (MoO₄) $_x$ (x = 0, 0.4, 0.8, 1.2, 1.6, 2.0) under 394 nm near-UV excitation.

Table I. Comparison of excitation $(^{7}F_{0} \rightarrow {^{5}L_{6}})$ and emission $(^{5}D_{0} \rightarrow {^{7}F_{2}})$ intensity of LiEu(WO₄)_{2-x}(MoO₄)_x (x = 0–2 in steps of 0.4)

	Intensity	ntensity (10 ⁷ cps)	
Compositions	$\lambda_{em} = 615 \text{ nm}$	$\lambda_{ex} = 394 \text{ nm}$	
LiEu(WO ₄) ₂	1.582	1.215	
$LiEu(WO_4)_{1.6}(MoO_4)_{0.4}$	1.803	1.612	
$LiEu(WO_4)_{1.2}(MoO_4)_{0.8}$	1.817	1.742	
$LiEu(WO_4)_{0.8}(MoO_4)_{1.2}$	1.820	1.746	
$LiEu(WO_4)_{0.4}(MoO_4)_{1.6}$	1.824	1.794	
$LiEu(MoO_4)_2$	1.885	1.889	

sharp and intense lines all have 5D_0 as the emitting level, though transitions from 5D_1 and 5D_2 levels are also present with a much lower intensity. This feature, common to all molybdates previously studied, $^{12\text{-}14}$ is probably related to the high-energy phonons of the molybdate (or tungstate) groups, which favor nonradiative deexcitation 5D_0 of toward 5D_0 . The intensity of $^5D_0 \rightarrow ^7F_{2,4}$ (electric-dipole transition) was found to be much stronger than that of $^5D_0 \rightarrow ^7F_{1,3}$ (magnetic-dipole transition), although the intensity of $^5D_0 \rightarrow ^7F_0$ was not clearly observed. The major emission of LiEu(WO₄)_{2-x}(MoO₄)_x was found at 615 nm ($^5D_0 \rightarrow ^7F_2$), which corresponds to red emission. Other transitions of Eu³⁺ from the 5D_1 excited levels to 7F_1 ground states, for instance, $^5D_0 \rightarrow ^7F_1$ located at 570–720 nm and $^5D_1 \rightarrow ^7F_1$ located at 520–570 nm, are both very weak; therefore, the more saturated CIE chromaticity benefited greatly by the reasoning.

When the molybdenum content is increased, the lithium europium double tungsto-molybdate shows stronger red emission at 615 nm by exciting at 394 nm. As indicated in Fig. 2, the intensity of ${}^5D_0 \rightarrow {}^7F_{1,2,3,4}$ transition was all found to increase with the increasing Mo/W content; however, the relative ratio of ${}^5D_0 \rightarrow {}^7F_1/{}^5D_0 \rightarrow {}^7F_2$ was found to change insensibly with Mo/W content. The I- λ curve of ${}^5D_0 \rightarrow {}^7F_2$ reaches a maximum when the relative ratio of Mo/W is 2:0, as revealed in Table I. The reason for this observation may be due to the advent of ion pair interaction between Eu³⁺ ions, which is expected to be much stronger in molybdate than in tungstate crystal because of the differences in the Eu³⁺ - Eu³⁺ distance in molybdate [average distance 3.86(6) Å] and that in tungstate [average distance 3.9(1) Å] phosphors. One of the possibilities is that the distance between two Eu³⁺ ions can affect the energy transfer between the two ions.

The CIE color coordinates (x,y) and relative luminance of the red-emitting phosphors investigated in this work are tabulated in Table II. For the series of $\text{LiEu}(WO_4)_{2-x}(MoO_4)_x$ phosphors with different Mo/W ratios, the experimental CIE (x,y) coordinates indicate that as the concentration of Mo increases, the x value increases

Table II. Comparison of CIE chromaticity coordinates of phosphors investigated in the work $[\lambda_{ex}=394 \text{ nm}]$ for LiEu(WO₄)_{2-x}(MoO₄)_x; $\lambda_{ex}=342 \text{ nm}$ for the commodity of Y₂O₂S:Eu³⁺].

Phosphor compositions	CIE color coordinates (x,y)	Relative luminance (a.u.)
LiEu(WO ₄) ₂	(0.65, 0.35)	0.8
$LiEu(WO_4)_{1.6}(MoO_4)_{0.4}$	(0.65, 0.34)	0.9
$LiEu(WO_4)_{1,2}(MoO_4)_{0,8}$	(0.65, 0.34)	1.0
$LiEu(WO_4)_{0.8}(MoO_4)_{1.2}$	(0.65, 0.34)	1.1
$LiEu(WO_4)_{0.4}(MoO_4)_{1.6}$	(0.65, 0.34)	1.2
$LiEu(MoO_4)_2$	(0.66, 0.33)	1.4
$Y_2O_2S:Eu^{3+}$	(0.66, 0.33)	1.0

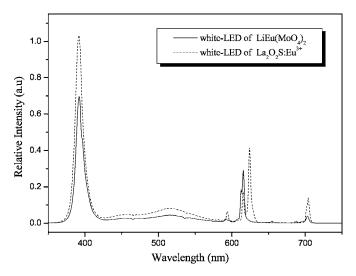


Figure 3. The EL spectra of n-UV chip-based WL-LED with LiEu(MoO₄)₂ and La₂O₂S:Eu³⁺ phosphors and a forward-bias current of 20 mA, respectively.

while y value decreases slightly, and the chromaticity was found to shift away from red toward deep-red. A similar observation to change relative Mo/W ratios has also been witnessed by Sivakumar and Varadaraju in the $\mathrm{AgLa_{0.95}Eu_{0.05}(WO_4)_{2-x}(MoO_4)_x}$ (x=0-2) system, ¹⁶ and they argued that minor distortions in the crystal structure play a crucial role in determining the luminescence properties in these systems. In addition, the absorption of the sample is considered to be one of the decisive factors in evaluating the relative luminance of LED phosphors. In our work the phosphor samples and commodity were prepared by solid-state reactions and their particle sizes were estimated to be in the micrometer range, as we measure the relative luminance of the samples whose size was reasonably assumed to be microsized.

We also found that with increasing doped Mo content the relative luminance increases from 0.8 (x = 0) to 1.4 (x = 2.0) monotonically. The CIE chromaticity coordinates were found to be (0.66, 0.33) for LiEu(MoO₄)₂, and it reached the same level as the commodity phosphor Y₂O₂S:Eu³⁺ (Kasei Optonix P22-RE3, the optimal excitation $\lambda_{ex} = 342 \text{ nm}$). $Y_2O_2S:Eu^{3+}$ phosphor was traditionally used in color television and it is a red-emitting candidate for SSL application at present. We have also observed that the CIE chromaticity coordinates of LiEu(MoO₄)₂ approach that [i.e., (0.67,0.33)] for the NTSC red. Consequently, our investigation results indicate that LiEu(WO₄)_{2-r}(MoO₄)_r can be used as a better near-UV conversion phosphor as compared to Y_2O_2S :Eu³⁺ in the application as a red-emitting phosphor for WL-LEDs. In addition, $LiEu(WO_4)_{2-r}(MoO_4)_r$ also emits light in the red region under 465 nm blue light excitation. Therefore, the new phosphor can be combined with a blue chip and YAG phosphor to WL-LEDs with higher Ra.

Furthermore, phosphor-conversion LED lamps were fabricated by precoating blue/green/red phosphors onto n-UV LED chips prior to packaging into LED lamps. Figure 3 shows the comparison of LED emission spectra of the WL-LED based on an n-UV chip, LiEu(MoO₄)₂, and La₂O₂S:Eu³⁺ at a I_f of 20 mA, respectively, where La₂O₂S:Eu³⁺ is currently a popular red-emitting phosphor for WL-LED application. We observed three bands in the LED emission spectra and the shorter-wavelength peak located at 392 nm originated directly from the n-UV LED chip. The broad emission band located at 430–580 nm originated from the blue and green phosphors, and the sharp emission located at 590–720 nm, respectively, originated from the red-emitting phosphor. The observation of a short-wavelength emission indicates that n-UV light emitted from

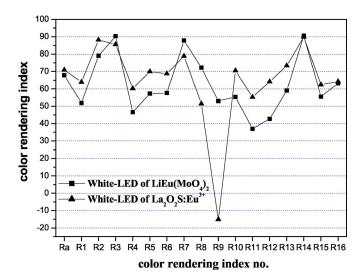


Figure 4. CRI variation of WL-LEDs based on an n-UV chip, LiEu(MoO₄)₂, and the commodity of $La_2O_2S:Eu^{3+}$ at an I_f of 20 mA, respectively.

the LED chips was not completely absorbed by the precoated phosphors in our LED lamps. It is also well known that down-conversion efficiency depends strongly on phosphor composition and phosphor Therefore, it is necessary to develop high-quality phosphors and a well-controlled packaging process to improve the output intensity of phosphor-converted LED lamps.

When the two WL-LEDs were operated at a I_f of 20 mA at room temperature, color temperature ($T_{\rm cp}$) and the luminous efficacy ($\eta_{\rm L}$) were found to be 5551 K and 6 lm/W for the WL-LED based on LiEu(MoO₄)₂, and 6782 K and 10 lm/W for the other WL-LED based on La₂O₂S:Eu³⁺, respectively. The CIE chromaticity of WL-LED based on LiEu(MoO₄)₂ and that based on La₂O₂S:Eu³⁺ were found to be (0.33, 0.33) and (0.31, 0.34), respectively. However, compared to commercial WL-LEDs (>25 lm/W) at a forward-bias current (I_f) of 20 mA at room temperature, the luminous efficacy (η_L) observed for our n-UV + blue/green/red white LED lamps was still insufficient, probably due to the inherently weak f-f absorption of Eu³⁺ and the lower efficacy of the LED chip (about 8–10 mW). Figure 4 shows the color-rendering index variation for WL-LEDs based on an n-UV chip, LiEu(MoO₄)₂, and La₂O₂S:Eu³⁺ operated at an I_f of 20 mA, respectively. The CRI value of the WL-LED of $LiEu(MoO_4)_2$ ($Ra \approx 67.8$) is close to the other WL-LED fabricated with a commodity of $La_2O_2S:Eu^{3+}$ ($Ra \approx 70.8$). The performance of the former is still deficient in the proportions of visible emission spectrum in the WL-LED of LiEu(MoO₄)₂. Above all, some of the CRI values for WL-LED based on LiEu(MoO₄)₂ are clearly larger than those of that based on La₂O₂S:Eu³⁺. In particular, the CRI no. 9 value (that shows color reproduction quality in the red region) of $LiEu(MoO_4)_2$ is 53.05, which is much higher and improved than that using La $_2\mathrm{O}_2\mathrm{S:Eu^{3+}}$ (-15.20). Therefore, LiEu(MoO $_4\mathrm{)}_2$ can be used as a red-emitting component in fabrication of phosphorconversion WL-LEDs.

Formerly, Narukawa et al. have reported the fabrication of n-UV + blue/yellow WL-LED lamps. 18 It has been shown that the optical properties of such n-UV + blue/yellow WL-LED lamps were excellent with a high color-rendering index, Ra, of 85. Compared to their n-UV + blue/yellow WL-LED lamps, the color-rendering index, Ra, of 67.8 observed for our n-UV + blue/green/red WL-LED

lamps was lower, possibly imputed to inherently weak absorption and emission attributed to Eu³⁺ f-f transition. The selection of blue or green phosphor still demands improvement (R1-R12 are both low). Our investigation results suggest that although LiEu(MoO₄)₂ was demonstrated to show potential for general illumination approaching the commodity of La₂O₂S:Eu³⁺, further study is needed to improve our n-UV + blue/green/red WL-LED efficiency. Furthermore, the new sulfide-free phosphor of $LiEu(WO_4)_{2-x}(MoO_4)_x$ can be combined with blue chip and YAG phosphor to form WL-LEDs with higher Ra, a commodity of which La₂O₂S:Eu³⁺ is unable to achieve.

Conclusions

We have successfully synthesized a class of novel red-emitting $LiEu(WO_4)_{2-x}(MoO_4)_x$ phosphors by solid-state reaction technique at high temperature. The luminescence properties under near-UV excitation of the double tungsto-molybdate phosphors have been investigated. Chromaticity characteristics and comparative luminance for the $LiEu(WO_4)_{2-x}(MoO_4)_x$ phosphors have also been measured and compared with a commodity phosphor Y₂O₂S:Eu³⁺. We also fabricated WL-LEDs by integrating an n-UV LED chip, the LiEu(MoO₄)₂ phosphor, and optimized matching green- and blueemitting phosphors. The Ra value of the WL-LED based on $LiEu(MoO_4)_2$ was found to be 67.8, approaching that of the other WL-LED ($Ra \approx 70.8$) based on the commodity of La₂O₂S:Eu³⁺ when operated at a forward-bias current (I_f) of 20 mA. Notably, the CRI no. 9 value showing color reproduction in the red region has been improved from -15.20 to 53.05 for the WL-LEDs using $LiEu(MoO_4)_2$ as the red-emissive component.

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