

Available online at www.sciencedirect.com



Tetrahedron Letters

Tetrahedron Letters 47 (2006) 7179-7183

A novel photoinduced ring opening and isomerization of adamantane-2-spiro isoxazolines using Mo(CO)₆

Annamalai Senthilvelan,^a Gene-Hsiang Lee^b and Wen-Sheng Chung^{a,*}

^aDepartment of Applied Chemistry, National Chiao-Tung University, Hsinchu 30050, Taiwan, ROC ^bInstrumentation Center, National Taiwan University, Taipei 106, Taiwan, ROC

> Received 28 February 2006; revised 21 July 2006; accepted 31 July 2006 Available online 17 August 2006

Abstract—The Mo(CO)₆-mediated photoinduced ring-opening reactions of adamantane isoxazolines involve novel rearrangement that provide enaminoketones as major products and β -hydroxy ketones as minor ones; in contrast, only β -hydroxy ketones and α , β -unsaturated ketones were obtained under thermal condition. © 2006 Elsevier Ltd. All rights reserved.

Isoxazolines are an interesting heterocyclic family because of their diverse biological applications. When treated with proper reducing reagents, isoxazolines can yield γ -amino alcohols, β -hydroxy ketones, α , β -unsaturated ketones, and β -hydroxy nitriles;¹ therefore, they are frequently used as precursors in the synthesis of various acyclic compounds.

Photoinduced ring-opening reactions of five-membered heterocycles and their subsequent rearrangements have received much attention both as synthetic intermediates and in the study of their reaction mechanisms.^{2–4} The photoreactions of 2-isoxazolines were studied by Schmid,⁵ Matsuura,⁶ and Mukai,⁷ who showed that N–O bond fission occurs on irradiation. It has also been reported⁸ that irradiation of 3-phenyl-2-isoxazoline gave isomeric 4-phenyl-2-oxazoline, ring opened β -amino aldehyde and benzonitrile. Nitta et al.⁹ reported that the iron carbonyls induced an isoxazoline ring cleavage under thermal and photochemical conditions.

Based on these findings and our interest in the studies of adamantane containing five-membered heterocyclic ring systems,¹⁰ we hereby report the ring-opening reactions of adamantane-2-spiro isoxazolines under thermal and photochemical conditions using Mo(CO)₆. Interestingly, a novel photoisomerization reaction of adamantane

isoxazolines to enaminoketones was observed (vide infra).

The adamantane-isoxazolines **1a**–**d** were synthesized in moderate yields (Table 1) by refluxing a mixture of methyleneadamantane and the corresponding nitrile oxides in dry THF using the reported method of 1,3-dipolar addition reactions.¹¹ The structures of each adamantane-isoxazolines **1a**–**d** were confirmed by spectral data, in which **1a** has been reported previously.¹¹

We performed the ring-opening reaction of **1c** initially under direct photolysis conditions; after irradiation for 16 h, however, only complex mixture of products along with starting material were observed; therefore, we sought an alternative approach. In addition to direct

Table 1. Synthesis of substituted adamantane-isoxazolines 1a-d

+ 0	R Cl Et ₃ N Dry THF 24 h	R N 1a-d	
Compound	R	Yield (%)	
1a	Phenyl	59	
1b	<i>p</i> -Tolyl	55	
1c	<i>p</i> -Anisyl	52	
1d	5-Chlorofuran-2-yl	58	

Keywords: Adamantane isoxazolines; Photorearrangement; Isomerization; Enaminoketones.

^{*} Corresponding author. Tel.: +886 3 5131517; fax: +886 3 5723764; e-mail: wschung@cc.nctu.edu.tw

^{0040-4039/\$ -} see front matter © 2006 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2006.07.156

photolysis, several other methods are known for achieving N–O bond cleavage in an isoxazoline ring, including reduction with Raney Ni,¹² LiAlH₄,¹³ H₂/Pd–C,¹⁴ TiCl₃,¹⁵ SmI₂,¹⁶ and Mo(CO)₆.¹⁷ Among these methods, the Mo(CO)₆-mediated ring opening of isoxazolines appears to be the most efficient one. To the best of our knowledge, Mo(CO)₆-mediated ring-opening reactions of adamantane isoxazolines have not been reported previously.

Table 2 presents the products and yields of the Mo(CO)₆-mediated photochemical cleavages of the adamantane isoxazolines **1a**–**d**. The expected β -hydroxyketones **2b**–**d** and the α , β -unsaturated ketone **3a** were obtained as minor products (<18%); surprisingly, unexpected novel photorearrangements occurred to give the ring-expanded enaminoketones **4a**–**d** as major products (30–59% yields). For example, the irradiation (254 nm, Rayonet) of **1a** (R = Ph) in the presence of 1.5 equiv of Mo(CO)₆ in acetonitrile/water gave enaminoketone **4a** in 58% yield and α , β -unsaturated ketone **3a** in 10% yield (entry 1, Table 2).¹⁸

Although we eventually confirmed the structures of the photoproducts 2b-d, 3a, and 4a-d spectroscopically, the determination of the major products was not straightforward. For example, in the ¹H NMR spectrum of 4c, a broad singlet appears at δ 11.8, which is quite unusual for phenyl or adamantyl protons; we assign this signal to the enamine NH proton. In addition, a vinylic CH proton appears as a broad singlet at δ 5.6. It is noteworthy that these signals for the vinyl CH and enamine NH protons disappeared upon the addition of D_2O . The ¹³C NMR spectrum of 4c displays the characteristic signals of a carbonyl carbon atom at δ 187.4, a quaternary enamino carbon atom [NH(CHR₂)C=CH-] at δ 174.6, and a vinylic CH carbon atom at δ 89.3. The mass spectrum of 4c exhibits the molecular ion at m/z 297, the same mass as that of its starting material **1c.** This finding implies that product **4c** is a geometrical isomer of 1c, that is, no net addition or fragmentation occurred during photolysis. Taken together, these spectral data of 4c are consistent with the enaminoketone structure. Gratifyingly, 4b is a crystalline compound

Table 2. $Mo(CO)_6$ -Mediated photochemical ring-opening reactions of
adamantane-isoxazolines

P

1a-d Cł	hv H ₃ CN/H ₂ O Mo(CO) ₆	H + +	`О н… + Г	
	2b-d	3a		4a-d
Compound	R	2 Yield (%) ^a	3 Yield (%) ^a	4 Yield (%) ^a
1a	Phenyl	_	10	58
1b	<i>p</i> -Tolyl	12	_	59
1c	p-Anisyl	18		30
1d	5-Chlorofuran-2-yl	15	_	30

^a Isolated yields based on recovered starting materials.

and we confirmed its structure through X-ray crystallography (Fig. 1).¹⁹

It is noteworthy that the enaminoketone 4a has been reported previously by Eguchi and co-workers to arise through an interesting rearrangement that occurred from the unstable adduct of aldonitrone and phenylacetylene.²⁰ Our spectral data of 4a-d support the previous structural assignment. We also performed the ring-opening reactions of 1a-d with Mo(CO)₆ under thermal conditions to determine whether the enaminoketones 4a-d are also obtained thermally. When we reacted 1a with Mo(CO)₆ (3.5 equiv) in CH₃CN/H₂O at 80 °C, we obtained only the α , β -unsaturated ketone **3a** in 29% yield. Under the same reaction conditions, 1b gave 3b in 49% yield; in contrast, cycloadduct 1c afforded both the β hydroxyketone 2c and the α,β -unsaturated ketone 3c in 44% and 31% yields, respectively. Cycloadduct 1d gave 2d in 23% yield (Table 3).²¹ In none of these cases did we observe even a trace amount of the corresponding enaminoketone.

In order to extend the scope of this novel photorearrangement with other sprio systems, the ring-opening reaction of isoxazoline-5-spiro cyclobutane 5^{22} was carried out. Irradiation of 5 with Mo(CO)₆ (1.5 equiv) in wet acetonitrile (Scheme 1) using 254 nm for 3 h gave the expected enaminoketone 6 and β -hydroxy ketone 7 in 32% and 15% yield, respectively. Whereas thermal reaction of 5 in wet acetonitrile containing Mo(CO)₆ (3.5 equiv) for 24 h gave only 7 (30%) with recovery of 5 (60%). No trace of spiro ring opened or rearranged product was observed in this condition (Scheme 1).

The structures of compounds **6** and **7** were confirmed by spectral data (see Supplementary data). From the formation of enaminoketone **6** it is clear that, in addition to the N–O bond cleavage of the isoxazoline ring, C₁– C₂ bond cleavage of spiro cyclobutane ring was also involved in the photolysis of **5** mediated by Mo(CO)₆. The rearrangement of isoxazoline-5-spiro cyclobutanes²² and cyclopropanes,²³ via N–O bond cleavage, under flash vacuum pyrolysis conditions and irradiation conditions has already been reported; however, their observed products are different from ours. Mo(CO)₆ mediated ring-opening reaction of isoxazoline-5-spiro cyclopropane to enaminoketone has also been reported.^{18a}



Figure 1. ORTEP plot of the solid state structure of 4b.

Table 3. Mo(CO)₆-Mediated thermal ring-opening reactions of adamantane-isoxazolines



^a Isolated yields based on recovered starting material.



Scheme 1. Mo(CO)₆-Mediated ring-opening reactions of isoxazoline **5**. (^aIsolated yield based on recovered starting material.)

Further, we investigated the ring-opening reaction of 3,5-disubstituted isoxazoline **8** (containing no 5-spiro group) using Mo(CO)₆. Compound **8** was synthesized by 1,3-dipolar cycloaddition reaction of 1-allyloxy-4tert-butyl benzene with phenyl nitrile oxide, using the reported method.¹¹ The reaction of **8** with Mo(CO)₆ (1 equiv) in CH₃CN/H₂O, under photochemical condition for 2 h gave the expected β -hydroxy ketone **9** in 47% yield, in addition to cleaved *t*-butyl phenol **10** (19% yield) (Scheme 2). Thermal ring-opening reaction of **8** with Mo(CO)₆ (3 equiv) under similar condition for 36 h also gave β -hydroxy ketone **9** in 62% yield with recovered **10** (34% yield). The structures of compounds **8–10** were confirmed by spectral data. Furthermore, compound **10** was compared with the authentic sample.

From these observations, it is clear that the ring-opening reactions of isoxazolines 1a-d, 5 and 8 were mediated by $Mo(CO)_6$ under both the reaction conditions. The readily occurred rearrangement of isoxazolines 1a-d and 5 to the corresponding enaminoketones 4a-d and 6 under photolysis is mainly due to the 5-spiro substituents: adamantane and cyclobutane moieties.

Based on our experimental results and literature evidences, we proposed a plausible mechanism for the formation of β -hydroxyketones 2 and the enaminoketones 4 (Scheme 3). Complex A, formed initially between the isoxazoline and Mo(CO)₆, undergoes N–O bond cleavage to provide a nitrene complex B. In the presence of



Scheme 2. Mo(CO)₆-Mediated ring-opening reactions of isoxazoline **8**. (^aIsolated yield based on recovered starting material.)



Scheme 3. A plausible mechanism of the Mo(CO)₆-Mediated ring opening reactions of 1.

water complex **B** gave β -iminoalcohol **E**, which hydrolyzes subsequently to form β -hydroxyketone **2**. Similar mechanisms for the formation of β -hydroxyketones from isoxazolines under thermal^{9,24} and photochemical⁹ reaction conditions have been reported previously. The photochemical formation of **4** came from complex \mathbf{A}^{25} via the aziridine intermediates **D** (itself obtained from complex **B** via \mathbf{C}^{26}) and was followed by a homologous Beckmann rearrangement.²⁷ Although such a rearrangement may seem strange, a related but different rearrangement from an adamantane oxaziridine intermediate to an amide has been reported.²⁸

In summary, we report here a convenient method for the $Mo(CO)_6$ -mediated ring-opening reactions of adamantane-2-spiro isoxazolines under photochemical and thermal conditions. This cycloaddition-cleavage protocol provides a novel isomerization pathway for the synthesis of substituted enaminoketones under photochemical conditions.²⁹

Acknowledgements

We thank the National Science Council (NSC) of Taiwan, ROC for financial support.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet. 2006.07.156.

References and notes

- (a) Kozikowski, A. P. Acc. Chem. Res. 1984, 17, 410–416;
 (b) Easton, C.; Hughes, C. M. M.; Savage, G. P.; Simpson, G. W. Adv. Heterocycl. Chem. 1994, 60, 261–327.
- Lablache-Combier, A. In *Photochemistry of Heterocyclic* Compounds; Buchardt, O., Ed.; Wiley: New York, 1976.
- 3. Padwa, A. In *Rearrangements in Ground and Excited States*; de Mayo, P., Ed.; Academic Press: New York, 1980; Vol. 3, p 501.
- (a) Lablache-Combier, A. In CRC Handbook of Organic Photochemistry and Photobiology; Horspool, W., Song, P.-S., Eds.; CRC: Boca Raton, FL, 1995; pp 1063–1120; (b) Buscemi, S.; Pace, A.; Pibiri, I.; Vivona, N. J. Org. Chem. 2002, 67, 6253–6255; (c) Buscemi, S.; D'Auria, M.; Pace, A.; Pibiri, I.; Vivona, N. Tetrahedron 2004, 60, 3243–3249.
- Claus, P.; Gilgen, P.; Hansen, H.-J.; Heimgartner, H.; Jackson, B.; Schmid, H. *Helv. Chim. Acta* 1974, *57*, 2173– 2192.
- 6. Itoh, Y.; Matsuura, T. Tetrahedron 1975, 31, 1373-1380.
- (a) Mukai, T.; Sukawa, H. *Tetrahedron Lett.* **1973**, 1835– 1838; (b) Seshimoto, O.; Kumagai, T.; Shimizu, K.; Mukai, T. *Chem. Lett.* **1977**, 1195–1198; (c) Saiki, H.; Miyashi, T.; Mukai, T.; Ikegami, Y. *Tetrahedron Lett.* **1977**, 4619–4622.
- 8. Kumagai, T.; Shimizu, K.; Kawamura, Y.; Mukai, T. *Tetrahedron* **1981**, *37*, 3365–3376.
- 9. Nitta, M.; Kobayashi, T. J. Chem. Soc., Perkin Trans 1 1984, 2103–2108, and references cited therein.
- (a) Tsai, T.-L.; Chen, W.-C.; Yu, C.-H.; le Noble, W. J.; Chung, W.-S. J. Org. Chem. **1999**, 64, 1099–1107; (b) Chu, J.-H.; Li, W.-S.; Chao, I.; Chung, W.-S. Tetrahedron **2004**, 60, 9493–9501.
- Chung, W.-S.; Tsai, T.-L.; Ho, C.-C.; Chiang, M. Y. N.; le Noble, W. J. J. Org. Chem. 1997, 62, 4672–4676.
- 12. Curran, D. P. J. Am. Chem. Soc. 1983, 105, 5826-5833.
- Jäger, V.; Colinas, P. A. In *The Chemistry of Heterocyclic Compounds*; John Wiley & Sons: New York, 2002; Vol. 59, Chapter 6.
- 14. Asaoka, M.; Mukuta, T.; Takei, H. Tetrahedron Lett. 1981, 22, 735–738.
- 15. Andersen, S. H.; Sharma, K. K.; Torssell, K. B. G. *Tetrahedron* 1983, 39, 2241–2245.
- 16. Bode, J. W.; Carreira, E. M. Org. Lett. 2001, 3, 1587-1590.
- (a) Tranmer, G. K.; Tam, W. Org. Lett. 2002, 4, 4101– 4104; (b) Gelabert, L. I.; Fascio, M. L.; D'Accorso, N. B. J. Heterocycl. Chem. 2003, 40, 341–344.
- The recovery of starting material is commonly observed in the ring-opening reaction of isoxazolines using Mo(CO)₆, see: (a) Guarna, A.; Guidi, A.; Goti, A.; Brandi, A.; De Sarlo, F. *Synthesis* **1989**, 175–178; (b) Bode, J. W.; Carreira, E. M. *J. Org. Chem.* **2001**, *66*, 6410–6424.
- 19. Crystal structure data for **4b**: C₁₉H₂₃NO, M = 281.38, triclinic, a = 9.8122(1) Å, $\alpha = 106.8226(9)^{\circ}$, b = 12.9988(2) Å, $\beta = 101.9080(8)^{\circ}$, c = 13.1727(2) Å, $\gamma = 94.8232(9)^{\circ}$, V = 1554.97(4) Å³, T = 295(2) K, space group $P\overline{\tau}$, Z = 4, $\mu = 0.073$ mm⁻¹, 29,260 reflections collected (RI = 0.0495, wR2 = 0.1329), 7122 independent reflections (R(int) = 0.0393, RI = 0.0879, wR2 = 0.00879, wR2 = 0.0879, wR2 = 0.090, wR2 = 0.0879, wR2, wR2 = 0.0879, wR2, wR2 = 0.090, wR2, wR2 = 0.090, wR2, w

0.1572). Crystallographic data for the structure in this letter have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 299771. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [fax: +44 (0) 1223 336033 or e-mail: data_request@ccdc.cam.ac.uk].

- Yu, Y.; Ohno, M.; Eguchi, S. Tetrahedron 1993, 49, 823– 832.
- 21. In addition to $Mo(CO)_6$ -mediated ring-cleavage, we also attempted to cleave cycloadduct **1a** using Zn/AcOH, LiAlH₄ and Raney-Ni. In each of these cases, only complex mixtures of products were observed without any major product.
- 22. Goti, A.; Brandi, A.; De Sarlo, F.; Guarna, A. *Tetrahedron* **1992**, *48*, 5283–5300, and references cited therein.
- Guarna, A.; Brandi, A.; De Sarlo, F.; Goti, A.; Pericciuoli, F. J. Org. Chem. 1988, 53, 2426–2429.
- Baraldi, P. G.; Barco, A.; Benetti, S.; Manfredini, S.; Simoni, D. Synthesis 1987, 276–278.
- Nitta, M.; Kobayashi, T. J. Chem. Soc., Perkin Trans 1 1985, 1401–1406.
- 26. Combier, A. L.; Pollet, A. Tetrahedron 1972, 28, 3141-3151.
- Cunningham, M.; Lim, L. S. N.; Just, G. Can. J. Chem. 1971, 49, 2891–2896.
- 28. Okada, K.; Saito, Y.; Oda, M. J. Chem. Soc., Chem. Commun. 1992, 1731–1732.
- 29. Procedures for the photochemical ring-opening reaction of 1b: An acetonitrile (18 mL) solution of isoxazoline 1b (0.09 mmol) and molybdenum hexacarbonyl (0.135 mmol) containing water (three drops) was irradiated by using a Rayonet multilamp reactor (254 nm) for 4 h. The solvent was removed under reduced pressure and the residue was purified by column chromatography to give 1b (20%), in addition to 2b and 4b (yields are summarized in Table 2). Procedures for the thermal ring-opening reaction of 1c: A mixture of 1c (0.105 mmol) and molybdenum hexacarbonyl (0.105 mmol) in acetonitrile (18 mL) containing water (three drops) was refluxed at 80 °C for 48 h. After removing the solvent, the residue was purified over silica gel column to give 1c (43%), in addition to 2c and 3c. Spectral data of some selected compounds: For 2b: colorless solid; mp 88–90 °C; $R_f = 0.60$ (hexane/EtOAc = 5/1); ¹H NMR (CDCl₃, 300 MHz) δ 1.48–1.91 (m, 12H), 2.29–2.35 (m, 2H), 2.42 (s, 3H, CH₃), 3.36 (s, 2H, 4'-CH₂), 4.42 (s, 1H, OH), 7.26-7.29 (m, 2H, ArH), 7.85-7.89 (m, 2H, ArH); ¹³C NMR (CDCl₃, 125 MHz) δ 22.3 (CH₃), 28.0 (CH), 28.6 (CH), 33.2 (CH₂), 35.4 (CH₂), 37.5 (CH), 40.0 (CH₂), 44.3 (CH₂), 75.6 (C_q), 129.1 (CH), 130.0 (CH), 135.8 (C_q), 145.1 (C_q), 202.7 (C_q); MS (EI) m/z (%): 284 (M⁺, 5), 266 (40), 265 (50), 251 (20), 134 (50), 119 (100), 91 (50); HRMS calcd for C₁₉H₂₄O₂ 284.1776; found 284.1780. For 3c: colorless solid; mp 97–99 °C; $R_f = 0.63$ (hexane/ EtOAc = 5/1); ¹H NMR (CDCl₃, 300 MHz) δ 1.88–2.05 (m, 12H), 2.54 (br s, 1H), 3.87 (s, 4H), 6.56 (s, 1H), 6.91-6.95 (m, 2H, ArH), 7.94–7.98 (m, 2H, ArH); ¹³C NMR (CDCl₃, 75.4 MHz) & 28.0 (CH), 33.6 (CH), 36.9 (CH₂), 39.3 (CH₂), 40.4 (CH₂), 41.8 (CH), 55.4 (CH₃), 113.5 (CH), 114.3 (CH), 130.5 (CH), 132.4 (Cq), 162.9 (Cq), 170.4 (C_g), 191.2 (C_g); MS (EI) m/z (%): 283 (M⁺+1, 5), 282 (M^{+} , 40), 254 (10), 251 (30); HRMS calcd for $C_{19}H_{22}O_2$ 282.1620; found 282.1628. For 4c: colorless solid; mp 96–98 °C; $R_{\rm f} = 0.25$ (hexane/

For 4c: colorless solid; mp 96–98 °C; $R_f = 0.25$ (hexane/ EtOAc = 5/1); ¹H NMR (CDCl₃, 300 MHz) δ 1.72–1.97 (m, 10H), 2.09 (br s, 2H), 2.55 (br s, 1H), 3.66 (br s, 1H), 3.81 (s, 3H, OCH₃), 5.59 (s, 1H), 6.88 (d, J = 8.7 Hz, 2H, ArH), 7.84 (d, J = 8.7 Hz, 2H, ArH), 11.83 (br s, 1H, NH); ¹³C NMR (CDCl₃, 75.4 MHz) δ 27.0 (CH), 32.9 $\begin{array}{l} ({\rm CH_2}),\, 34.9 \; ({\rm CH_2}),\, 36.0 \; ({\rm CH_2}),\, 40.4 \; ({\rm CH}),\, 48.9 \; ({\rm CH}),\, 55.3 \\ ({\rm CH_3}),\, 89.3 \; ({\rm CH}),\,\, 113.3 \; ({\rm CH}),\,\, 128.6 \; ({\rm CH}),\,\, 133.4 \; ({\rm C_q}),\\ 161.4 \; ({\rm C_q}),\, 174.6 \; ({\rm C_q}),\, 187.4 \; ({\rm C_q});\, {\rm MS} \; ({\rm EI}) \; \textit{m/z} \; (\%): \; 297 \end{array}$

 $(M^{+}, 15), 296\,(100), 280\,(10), 190\,(10), 162\,(5), 135\,(18), 91\,(7); HRMS calcd for <math display="inline">C_{19}H_{23}NO_2$ 297.1729; found 297.1715.