

## Orbital symmetry and electron correlation in $\text{Na}_x\text{CoO}_2$

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### Abstract

We present measurements of polarization-dependent soft X-ray absorption on  $\text{Na}_x\text{CoO}_2$ . The results reveal that the electronic states determining the low-energy excitations of  $\text{Na}_x\text{CoO}_2$  have predominantly  $a_{1g}$  symmetry with significant O 2p character. A large transfer of spectral weight observed in O 1s X-ray absorption provides spectral evidence for strong electron correlations in  $\text{Na}_x\text{CoO}_2$  with the Na doping higher than half. For the Na concentration less than half, our measurements of O 1s absorption suggest that  $\text{Na}_x\text{CoO}_2$  exhibits band-like electronic character.

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Sodium cobalt oxides ( $\text{Na}_x\text{CoO}_2$ ) have attracted renewed interest because of their exceptionally large thermoelectric power [1] and the discovery of superconductivity in their hydrated counterparts [2]. Despite intensive experimental and theoretical works, there remain many unresolved issues concerning the electronic structure of  $\text{Na}_x\text{CoO}_2$ , including the orbital character of the valence electrons responsible for low-energy excitations and the Coulomb correlations of Co 3d electrons [3].

The lattice of  $\text{Na}_x\text{CoO}_2$  exhibits a trigonal distortion, leading to a splitting of  $t_{2g}$  states into  $e'_g$  and  $a_{1g}$  states. The  $e'_g$  states spread over the  $ab$  plane, whereas the  $a_{1g}$  state extends to the  $c$ -axis [4]. Band-structure calculations in the local-density approximation (LDA) show that the  $a_{1g}$  state has a one-particle energy higher than that of  $e'_g$  and is most relevant to low-energy excitations [5]. These calculations are however, different from a crystal-field approach in which the compressed trigonal distortion stabilizes the  $a_{1g}$  state [6].

Many microscopic models with strong electron correlations explicitly included have been proposed to explain the

spectacular properties of  $\text{Na}_x\text{CoO}_2$  [6–9]. To comprehend the effect of electron correlations is imperative for an understanding of the electronic structure of  $\text{Na}_x\text{CoO}_2$ . On the other hand, a recent LDA +  $U$  study (LDA including the on-site Coulomb energy  $U$ ) [10] explains the Fermi surface measured by angle-resolved photoemission [11,12] and concludes that  $\text{Na}_x\text{CoO}_2$  is a moderately correlated system. One, therefore, requires further spectral evidence for strong electron correlations to justify microscopic models of correlated electrons for  $\text{Na}_x\text{CoO}_2$ .

Here, we report measurements of soft X-ray absorption spectroscopy (XAS) on  $\text{Na}_x\text{CoO}_2$  pertinent to its orbital character of the electronic states determining the low-energy physics. We discuss the spectral character of strongly correlated electrons of  $\text{Na}_x\text{CoO}_2$  with various Na concentrations. We measured XAS of  $\text{Na}_x\text{CoO}_2$  with the Dragon beamline at the National Synchrotron Radiation Research Center in Taiwan. Details of XAS measurements and sample preparation are discussed elsewhere [13,14].

In order to determine the symmetry of electronic states in the low-energy excitations, we resorted to measurements of polarization-dependent O 1s XAS of  $\text{Na}_{0.5}\text{CoO}_2$ , as plotted in Fig. 1. The O 1s XAS shows that

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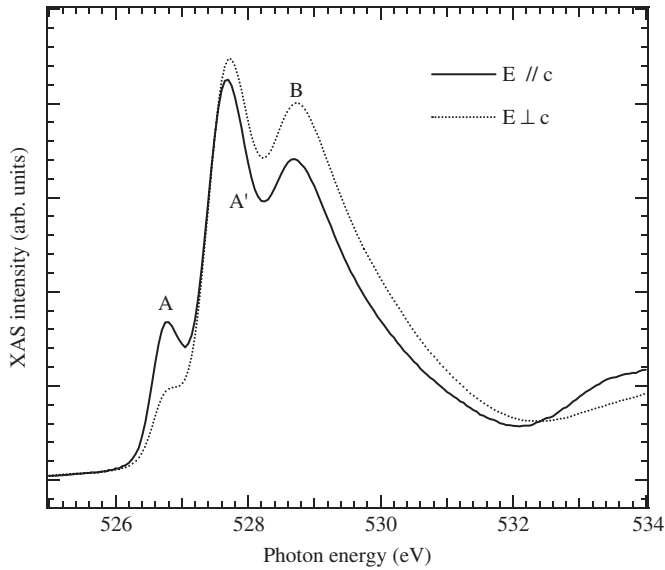


Fig. 1. Polarization-dependent O 1s XAS of  $\text{Na}_{0.5}\text{CoO}_2$  with the E vector of the light perpendicular (dotted line) and parallel (solid line) to the  $c$ -axis.

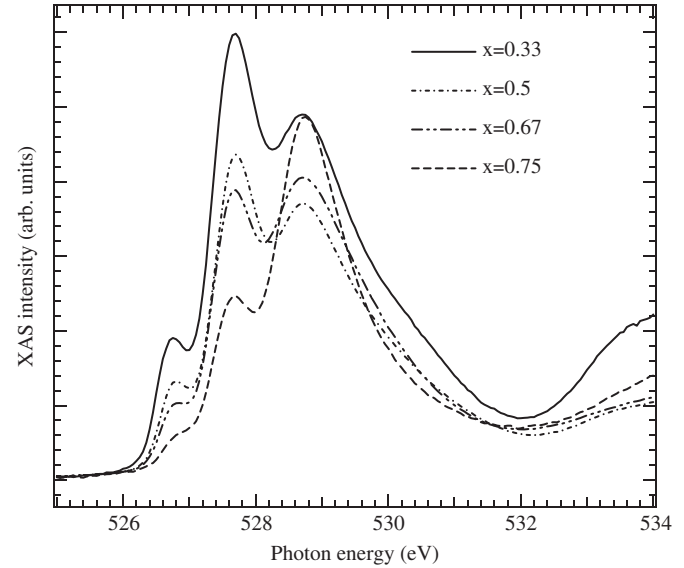


Fig. 2. Doping-dependent isotropic O 1s XAS spectrum of  $\text{Na}_{0.5}\text{CoO}_2$ , i.e.,  $(I_{\parallel} + I_{\perp})/2$ .

the lowest-energy peak at 526.8 eV (labelled as A) has a strong  $z$  component, indicating an out-of-plane electronic state. The in-plane components of two other peaks at 527.6 and 528.7 eV (labelled as A' and B, respectively) are slightly larger than their corresponding out-of-plane components. Because peak A and A' (or B) have opposite polarizations, peak A results predominantly from adding an electron to a state of  $a_{1g}$  symmetry, whereas peaks A' and B correspond to adding electrons to states of  $e_g$  symmetry. In other words, the symmetries of the transitions associated with peaks A, A' and B correspond mainly to  $(a_{1g})^1 \rightarrow (a_{1g})^2$ ,  $(a_{1g})^1 \rightarrow (a_{1g})^1(e_g)^1$ , and  $(a_{1g})^2 \rightarrow (a_{1g})^2(e_g)^1$ , respectively. O 1s XAS of  $\text{Na}_{0.5}\text{CoO}_2$  with final states of  $a_{1g}$  symmetry has a large out-of-plane polarization, whereas that with  $e_g$  symmetry has an in-plane polarization.

We plot doping-dependent isotropic O 1s XAS of  $\text{Na}_x\text{CoO}_2$  in Fig. 2. As the doping  $x$  increases from 0.5 to 0.67 and 0.75, the intensities of peaks A and A' decrease, but peak B increases in intensity. These variations of XAS peaks are derived from the change in the relative concentration of  $\text{Co}^{4+}$  and  $\text{Co}^{3+}$ , because a fraction  $x$  of  $\text{Co}^{4+}$  changes to  $\text{Co}^{3+}$  when the mother compound  $\text{CoO}_2$  is doped with Na. The peaks A and A' (peak B) originate from O 2p hybridized with  $\text{Co}^{4+}$  ( $\text{Co}^{3+}$ ) 3d states and correspond to adding one electron to the  $\text{Co}^{4+}$  states of  $a_{1g}$  and  $e_g$  symmetries ( $\text{Co}^{3+}$  of  $e_g$  symmetries), respectively. Such a spectral-weight transfer of the one-electron addition observed in  $\text{Na}_x\text{CoO}_2$  is a general feature of strongly correlated systems [15]. In contrast, the doping-dependent XAS spectra of  $\text{Na}_x\text{CoO}_2$  when Na concentration less than 0.5 do not exhibit a spectral weight transfer as observed in  $\text{Na}_x\text{CoO}_2$  with high Na doping and other correlated

oxides. We found that the unoccupied O 2p density of states in the vicinity of the Fermi level obtained from LDA + U calculations [16] are consistent with the measured XAS spectra of  $\text{Na}_x\text{CoO}_2$  with low Na doping, e.g.,  $x = 0.3$  and 0.33, suggesting a band-like electronic character.

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