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Received 00th January 20xx, Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

Synthesis and study of hybrid hydrogen-bonded bent-core liquid crystal complexes containing C₆₀- and Si-based proton donors[†]

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Novel hydrogen-bonded (H-bonded) C_{60^-} and Si-based bent-core liquid crystal (LC) complexes containing a bent-shap proton acceptor and C_{60^-} and Si-based single/double armed proton donors were synthesized and mingled to produce hybrid H-bonded LC complexes. The SmCP phases were able to be introduced in the hybrid LC complexes with a very hign C_{60} -based moiety up to 90 mol%. Among our H-bonded LC complexes, the hybrid LC complexes with low contents of the C_{60} -based moiety (≤ 24 mol%) possessed the highest spontaneous polarization value (ca. 494 nC/cm²), the widest mesophasic range (40.8°C) and the lowest saturated electric field (8.8 Vpp/µm). Under electric fields of modified triangle waveforms, both series of hybrid H-bonded LC complexes displayed an anti-ferroelectric (AF) to a ferroelectric (FE) polar switching as the C_{60} -based moiety increased up to 24-50 mol%. Therefore, the hybrid H-bonded LC complexes containing a broad molar ratio (0-90 mol%) of C_{60} -based moiety reveled a tunable route for the electro-optical applications of C_{60} -based H-bonded liquid crystals.

were luxuriant (including some cost-effective ways) to enhance the arrangements of buckyballs.^{31,37-50} With the advantages of easy-

manipulating and flexible ratio-control (also reported as an

effective conduit for charge transfer),⁵¹ hydrogen-bonds (H-bonds)

were our aim to be introduced to the C_{60} -based liquid crystal (L

systems.⁵² The electro-optical properties of fullerene-based LCs

have been determined in the smectic⁵³ and columnar⁴⁹ phases

accompanied with the molecular arrangements as well as the

orientations of π -conjugated cores in mesophases to provide

essential information for practical appearances. 54a-c However, the

Introduction

Buckyball (also named as buckminsterfullerene or [60]fullerene) is a structure of truncated icosahedron with formula C_{60} , which was first synthesized by Kroto et al. in 1985.1 The buckyball structure displays various special properties, such as observable waveparticle duality,² superconductivity,³⁻⁶ non-linear optical property⁷ and high electronic affinity.^{8,9} Moreover, with proper surface modification, its derivatives showed extensive applications, e.g., hydration,¹⁰⁻¹² hydrogenation,^{13,14} halogenation,¹⁴ oxygenation,^{15,16} cycloaddition,¹⁷⁻¹⁹ free radical reaction,²⁰ photoreaction,^{21,22} metallofullerene, 23,24 catalyst,²⁵ endohedral biomedical sensor/therapy^{26,27} and electron acceptor.^{28,29} The arrangements of charge transporting materials (e.g., buckyballs) were drawn much attention recently due to the improving of charge transfer in molecular electronics.³⁰⁻³⁶ By attaching to mesogens, [60]fullerene could be aligned via liquid crystalline mesophases, and the methods

electro-optical performance of C₆₀-based FLCs are rare,^{54d} especially for those with H-bonds. Therefore, the combination of H-bonding, bent-core LCs, and fullerenes in this study are required for further investigations on their mesomorphic and electro-optical aspects. However, to our best knowledge, no hybrid H-bonded C_{60} -basec liquid crystals have been synthesized up to date, which may be resulted from the weak H-bonding between proton donors and acceptors as well the strong aggregation of C₆₀ nanoparticles (NPs) to disturb the arrangements of mesogens.⁵⁵ Therefore, an efficient blending method was applied to induce and stabilize the mesophases of H-bonded C60-based LCs in our research. Moreover, LCs with ferro- and anti-ferroelectricities were utilized to reinforce the control of the applied electric fields as well as strengthen t e photovoltaic effect on charge transporting materials.⁵⁶⁻⁶⁰ According to the previous research of Ros et al.,³⁷ covalent-bonded bent-co LCs based on [60]fullerene displayed broad ranges of SmCP phas. and novel physical properties. Hence, as shown in Figs. 1 and

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⁺ Electronic Supplementary Information (ESI) available: The details of synthetic steps and the elemental characterization of components **FIA**, **FIIA**, **IIA**, **SIA** and **NBF14**; the setup of P_s value measurement; the blending ratios of hybrid complexes; POM images of **FIA24** and **FIIA24**; crystallographic parameters of synthesized compounds and hybrid complexes; P_s values vs. applied electric fields of **FIA2** and **FIIA24**. See DOI: 10.1039/x0xx00000x

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various H-bonded C_{60} -based bent-core complexes, i.e., **FIA100** and **FIIA100** (analogue **IIA100** without C_{60} as a comparison), were synthesized and investigated in this study, where a bent-shaped proton acceptor **NBF14** and the corresponding proton donors **FIA**, **FIIA** and **IIA** were prepared. In addition, the previously reported H-bonded Si-based bent-core complex **SIA100** containing the proton donor **SIA** was utilized as a LC host to mix with Au-based covalent-bonded bent-core dopant in order to facilitate the smectic alignment of Au-nanoparticles by electric fields.⁶¹ Furthermore, the Si-based bent-core LC complex **SIA100** in this survey was also mingled with both H-bonded C₆₀-based bent-core complexes **FIA100** and **FIIA100** to induce and even extend the mesophasic ranges of the hybrid H-bonded bent-core LCs.



Fig. 1 Molecular structures of proton donors FIA, FIIA, SiA and IIA as well as a proton acceptor NBF14.



Fig. 2 (a) H-bonded bent-core complexes FIA100, FIIA100, SiA100 and IIA100; (b) hybrid LC complexes FIA50 and FIIA50 containing 50% molar ratio of SiA100 and 50% molar ratio of FIA100 and FIIA100, respectively.

Notably, with a minor molar ratio of **SiA100**_A(<u>i.e.on10</u> mole%), the hybrid H-bonded bent- $Co^{10}LC^{39}$ (Complex B containing both Si-based **SiA100** and C₆₀-based **FIA100** (or **FIIA100**) were able to display SmCP phases, and the electro-optical as well as mesophasic properties could be manipulat d and optimized via the blending ratio of the Si- and C₆₀-baseo complexes, which could be more easily controlled than the covalent-bonded C₆₀-based LC systems.

Experimental

Synthesis. Synthetic steps of components FIA, FIIA, IIA, SiA and NBF14 are shown in Scheme S1 in the Supporting Information. Chemical characterization data of FIA, FIIA, IIA, SiA and NBF14 are provided below: 4-((12-(4-(1-methylfulleropyrrolidin-2- yl)phenoxy)dodecyl)oxv) benzoic acid, (FIA). ¹H NMR (300 MHz, CDCl₃) δ (ppm): 8.00 (u, J = 9.0 Hz, 2H), 7.40 (m, 4H), 6.88 (d, 2H), 4.96 (d, J = 9.0 H 1H), 4.87 (s, 1H), 4.22 (d, J = 9.0 Hz, 1H), 4.01-3.92 (m, 4H), 2.78 (s, 3H), 1.75 (m, 4H), 1.36-1.23 (m, 16H). ¹³C NMR (MHz, CDCl₃) δ (ppm) = 26.26, 29.38, 29.84, 30.60, 34.49, 40.29, 66.61, 68.23, 68.48, 70.22, 114.37, 114.81, 122.50, 124.73 128.35, 128.95, 136.08, 139.83, 140.16, 140.41, 141.79, 141.94, 142.08, 142.25, 143.4, 144.65, 144.87, 144.97, 145.39, 145.5 145.59, 145.76, 146.03, 146.38, 146.46, 146.55, 147.07, 147.33 147.55, 159.44, 163.32, 166.49. Anal. Calcd for C₈₈H₃₉NO₄: C, 90.01; H, 3.35; N, 1.19. Found: C, 90.58; H, 2.86; N, 0.99. MS (FAB-) m/z: 1173.3 (exact mass); found, 1174.6.

4,4'-(((malonylbis(oxy)))bis(dodecane-12,1-diyl))bis(oxy))(1,2methanofullerene C₆₀)-61,61-dibenzoic acid, (**FIIA**). ¹H NMF (300 MHz, CDCl₃) δ (ppm):7.96 (d, *J* = 9.0 Hz, 4H), 6.90 (d, *J* = 8.8 Hz, 4H), 4.47 (m, *J* = 6.6 Hz, 4H), 4.01 (t, *J* = 6.6 Hz, 4H 1.89 (m, 8H), 1.43–1.29 (m, 32H). ¹³C NMR (75 MHz, CDCl₃) δ (ppm) = 26.01, 28.52, 29.13, 29.21, 29.40, 29.58, 67.31, 68.18, 114.05, 122.30, 131.55, 138.50, 138.59, 138.67, 139.12, 139.13 139.18, 140.13, 140.34, 141.02, 141.20, 141.54, 141.84, 142.05, 142.32, 142.51, 142.85, 143.02, 143.36, 143.57, 144.01, 144.34 144.58, 145.09, 145.11, 145.28, 145.37, 145.54, 146.02, 146.15 146.45, 146.93, 147.21, 147.68, 162.93, 166.89. Anal. Calcd for C₁₀₁H₅₈O₁₀: C, 84.74; H, 4.08. Found: C, 85.33; H, 4.55. MS (FAB+) *m/z*: 1430.4 (exact mass); found, 1431.1.

4,4'-(((malonylbis(oxy))bis(dodecane-12,1-diyl))bis(oxy)) dibenzoic acid, (**IIA**). ¹H NMR (300 MHz, DMSO-d₆) δ (ppm) = 7.85 (d, *J* = 9.0 Hz, 4H), 6.97 (d, *J* = 8.7 Hz, 4H), 4.05–3.97 (m, 8H), 3.46 (s, 2H), 2.11–2.00 (m, 8H). 1.80–1.50 (m, 8H), 1.45– 1.30 (m, 24H) Anal. Calcd for C₄₁H₆₀O₁₀: C, 69.07; H, 8.48. Found: C, 68.77; H, 8.41.

4-(Dimethylethylsiloxyl) undecyloxybenzoic acid, (SiA). The Hdonor SiA was synthesized and reported in the reference,⁶¹ , NMR (300 MHz, CDC₁₃) δ (ppm): 8.05 (d, *J* = 9.0 Hz, 2H), 6.95 (*J* = 9.0 Hz, 2H), 4.05 (t, *J* = 6.6 Hz, 2H), 1.81 (m, 2H),1.52–1.32 (m, 16H), 0.92 (t, *J* = 8.1 Hz, 3H), 0.48 (t, *J* = 7.8 Hz, 4H), 0.01 6H). Anal. Calcd for C₂₂H₃₈O₃Si : C, 69.79; H, 10.12. Found: C

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69.62; H, 10.05. MS ($^{*})$ m/z: 378.26 (exact mass); found, 378 ($M^{*}).$

3'-(4-(2,3-Difluoro-4-(tetradecyloxy)benzoyloxy)benzoyloxy) biphenyl-4-yl isonicotinate, (**NBF14**). ¹H NMR (300 MHz, CDCl₃) δ (ppm): 8.87 (d, 2H), 8.29 (d, *J* = 8.3 Hz, 2H), 8.02 (d, *J* = 8.4 Hz, 2H), 7.86 (m, 1H), 7.67 (d, *J* = 8.2 Hz,2H), 7.54 (d, *J* = 4.7 Hz, 2H), 7.46 (s, 1H), 7.42 (d, *J* = 8.5 Hz, 2H), 7.31 (d, *J* = 8.7 Hz, 2H), 7.23–7.19 (m, 1H), 6.89–6.82 (m, 1H), 4.10 (t, *J* = 6.3 Hz, 2H), 1.80–1.75 (m, 2H), 1.43–1.28 (m, 22H), 0.86 (t, *J* = 6.3 Hz, 3H). Anal. Calcd for C₄₆H₄₇F₂NO₇: C, 72.33; H, 6.20; N, 1.83. Found: C, 72.15; H, 6.42; N, 1.83.

Sample preparation. The H-bonded complexes were prepared by dissolving proton acceptor NBF14 and proton donors FIA, FIIA, SiA or IIA with equimolar amounts of pyridyl and benzoic acid groups in anhydrous tetrahydrofuran (THF). After each LC mixture was slowly evaporated at room temperature, the THF was completely removed in vacuum for 24 h, and then the hybrid LC complexes were obtained.

Characterization. ¹H and ¹³C NMR spectra were verified with Varian Unity 300 MHz spectrometer. Mass spectra and elemental analyses were determined with Micromass TRIO-2000 GC-MS and Perkin-Elmer 240C elemental analyzer, respectively. MALDI-TOF measurements were carried out on AutoFlex III MALDI mass spectrometer (Bruker Daltonics, Breman, Germany). Fourier transform infrared spectra (FTIR) were recorded on a Perkin-Elmer Spectrum 100 Series with pressed KBr pellets. Mesophasic patterns, enthalpies of mesophase transitions and X-ray diffraction (XRD) were measured with polarizing optical microscope (POM, Leica DMLP; equipped with a hot stage, Linkam TMS-94/LTS350), differential scanning calorimeter (DSC, Perkin Elmer Diamond; rate 5°C/min of heating or cooling) and synchrotron X-ray radiation (at beamlines BL13A1, BL17A1 and BL01C2 at National Synchrotron Radiation Research Center, NSRRC, Taiwan), respectively. Typical X-ray scattering patterns were obtained with detectors (marCCD165 or mar345 image plates) for 1 to 30 seconds. The scattering angles were calibrated with two standard samples of silver behenate and silicon with beam diameter 0.5 mm for the 10-keV beam ($\lambda = 0.124$ nm⁻¹). The electro-optical properties were measured with prepared samples injected in commercially non-rubbed indium-tin-oxide (ITO) cells (mesophase state; thickness 9 μ m; active area 0.25 cm²). With a digital oscilloscope (Tektronix TDS-3012B) connected to a high-power amplifier (Gwinstek) and a function generator (Tektronix AFG 3021), measurements of spontaneous polarization were made with a modified triangular-wave method at a frequency of 50 Hz (see Fig. S1 in the Supporting Information).⁶²

Results and discussion

Molecular Structures and Phase Behavior of Synthesized Complexes. As shown in Fig. 2a, single-armed H-bonded bentcore complexes **FIA100** and **SiA100**⁶¹ are composed of proton donor FIA (or SiA) and acceptor NBF14 with 1:1/molar_ratio; while double-armed H-bonded bent-core Complexes Filate and IIA100 consist of proton donor FIIA (or IIA) and acceptor NBF14 with 1:2 molar ratio. The existence of H-bonded in the mesophases can be proved by temperature-various FT R spectroscopy (see the Supporting Information). Among these H-bonded bent-core complexes, upon cooling only SiA100 and IIA100 processed the smectic CP (SmCP) and smectic C (SmC) phases, respectively. However, the C60-based H-bonded bentcore complexes FIA100 and FIIA100 did not display any mesophases in contrast to their C60-based covalent-bonded analogues,³⁷ because the mesophases were hindered by the stronger aggregation tendency of C₆₀⁵⁵ with weaker H-bonds ir the H-bonded complexes (i.e., FIA100 and FIIA100). Therefore, SiA100 was introduced to the H-bonded C60-based bent-core complexes (FIA100 and FIIA100) to extend the mesophasic ranges of the hybrid H-bonded bent-core complexes. illustrated in Fig. 2b, binary complex FIA50 was prepared frc FIA, SiA and NBF14 with 1:1:2 mole ratio and binary complex FIIA50 was produced by FIIA, SiA and NBF14 with 1:2:4 mc1a ratio. In general, binary complexes FIAx and FIIAx were denoted as **x** % molar ratio of C₆₀-based bent-core proton donor (i.e., x = 4, 24, 50, 76, 90 and 100). For example, since a double-armed proton donor FIIA bearing two -COC functional groups would be H-bonded with two pyridyl moieties of proton acceptor NBF14 (see Fig. 2b and Table S1 of the Supporting Information). In addition, the relationship for x values and weight percents (wt%) of C₆₀-based bent-core complexes are also presented in Table S1 in order to compare with our previous research.⁶¹ As shown in Table 1 and Fig. 3, the broadest mesophasic ranges (upon cooling) of binary complexes FIAx and FIIAx were both extended from 33.4°C (SiA100) to 40.8 (at x = 24) and 37.6°C (at x = 50), respective'. The POM images of typical broken-fan textures of the SmCP phase for x = 24 were shown in Fig. S2 of the Supporting Information.

The single-armed binary complexes FIAx displayed lower crystallization temperatures (SmCP-Cr) than their doublearmed analogues FIIAx, which could be attributed to the colligative property of C60-based complexes. Notably, although most binary complexes possessed monotropic mesophase, only binary complexes FIA90 and FIIA90 containing a very high content of C₆₀-based H-bonded bent-core complexes demonstrated narrow mesophasic temperature ranges 11.4 and 10.8°C, respectively (see Fig. 3b). Importantly, higher uptaken loads of C60-based component can be blended into the binary complexes FIAx and FIIAx (i.e., x = 50, 76 and برو without phase separation) than our previous Au-based covalent-bonded bent-core dopant (phase separation occurred at larger than 20 wt%) in SiA100. Therefore, an effective hybrid method for high contents of H-bonded C₆₀-based bentcore complexes was introduced in this study. In addition, http://www.complexes.com/ increasing the content of C60-based moiety to moderate values. the polarity of the SmCP phase varied sequentially from antiferroelectricity to ferroelectricity in the binary complexes FI x and FIIAx (at x = 24 ~ 50, see Fig. 3, which was also discussed in the section "Electro-Optical Properties").

Table 1Phase transition temperatures and enthalpies ofsynthesized compounds and hybrid complexes	
Compounds and complexes	Phase transition temperature/°C [enthalpy/J/g] ^a heating (top)/ cooling (bottom)
FIA100	Cr 89.3[21.0] Iso Iso 72.3[-18.8] Cr
SiA100	Cr 97.5[17.1] SmCP _A 116.4[23.6] Iso Iso 124.0[-26.5] SmCP _A 90.6[-13.8] Cr
FIIA100	Cr 87.0[25.6] Iso Iso 82.8[-23.6] Cr
IIA100	Cr 115.5[53.2] Iso Iso 127.0 ^b SmC 126.0 ^b Cr
FIA4	Cr 97.1[69.2] Iso Iso 117.5[-30.9] SmCP _A 87.7[-19.0] Cr
FIA24	Cr 95.3[36.7] Iso Iso 120.1[-16.8] SmCP _A 79.3[-14.4] Cr
FIA50	Cr 90.5[33.5] Iso Iso 106.8[-9.1] SmCP _F 72.8[-15.0] Cr
FIA76	Cr 87.0[26.9] Iso Iso 106.1[-0.2] SmCP _F 73.4[-24.9] Cr
FIA90	Cr 87.0 [29.5] Iso Iso 80.4[-1.7] SmCP _F 69.0[-20.7] Cr
FIIA4	Cr 98.5[31.8] Iso Iso 121.8[-17.7] SmCP _A 88.7[-11.8] Cr
FIIA24	Cr 98.7[16.8] SmCP _A 117.0[14.1] Iso Iso 123.5[-17.0] SmCP _A 86.8[-17.9] Cr
FIIA50	Cr 97.5 [36.2] Iso Iso 115.4[-11.3] SmCP _F 77.8[-20.1] Cr
FIIA76	Cr 91.5[33.1] Iso Iso 100.5[-3.2] SmCP _F 78.9[-22.8] Cr
FIIA90	Cr 90.1[31.1] Iso Iso 88.4[-1.5] SmCP _F 77.6[-24.0] Cr

 a The phase transitions were measured by DSC at the 2^{nd} heating and 1^{st} cooling scans with a cooling rate of 5°C/min with peak onset., Cr = crystal solid; SmC = smectic C phase; SmCP_F = ferroelectric smectic C phase; SmCP_A = anti-ferroelectric smectic C phase; Iso = isotropic phase.

^b Mesophases obtained by POM and XRD measurements. (The data of DSC and POM were furnished in Figs. S3 and S4, respectively.)



Fig. 3 (a) Mesophasic behaviors and (b) mesophasic ranges of FIAx (black), FIIAx (white) and SiA100 (gray).

Molecular structures based on X-ray analysis. Vilo Agrderato decipher the molecular structures of the C60184828 (68At16868 complexes, FIAx and FIIAx were probed with powder XRD measurements (see Table S2 of the Supporting Information) The C60-based bent-core complexes FIAx and FIIAx possess c one set of sharp peaks in small angle and broad diffuse scattering signals in wide angle regions indicating a smectic lamellar order of uniformly mixed systems coupled with similar orders of lateral intermolecular distances, respectively (e.g., x = 50 shown in Figs. 4a-4c). Sharp peaks indexed as (01) in the small-angle region at the associated d-spacing values of d₁ with 46 ~ 50 Å for hybrid complexes FIAx and FIIAx, which were shorter than the calculated molecular lengths of FIA100 FIIA100 and SiA100 (MM+ method, 65 ~ 76 Å) indicating the existence of tilted angles examined in bent-core hybrid complexes (see Table S2). Furthermore, no interdigitated structures are revealed due to the observation of one su ordered diffraction pattern only, and the d-spacing values FIAx and FIIAx in Fig. 4c reflected a sequentially descending intensity from (01) to (02) and (03), which indicated simple monolayer organizations instead of C₆₀ aggregation-induced bilayer structures with the strongest intensity indexed at (02).^{37,63,64} Both hybrid complexes FIAx and FIIAx possessed similar mesophases of SmCP because of the monolay structure, rather than a bilayer structure induced by the competitions between Si-based nanosegregations and H-bonds



Fig. 4 Proposed molecular orders and XRD patterns of (a) FIA50 and (b) FIIA50 as well as (c) both 1-D XRD integrations; (d) the d-spacing values of FIAx (black), FIIAx (white) and SiA100 (gray) at T = T_{Iso-SmCP} - 10°C upon cooling.

The d-spacing values of hybrid complexes FIAx and FIIAx with same x values were almost the same, except for x = 90 with the largest d-spacing difference possibly due to different measured temperatures (see Fig. 4d and Table 1). Therefore, inferred from the monolayer structures characterized by the XRD measurements of FIAx and FIIAx, the decreased d-spaci g

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value of SmCP with increasing C_{60} molar ratio might be mainly attributed to the reduced measured temperature in each hybrid complex, which also suggested that no NP-induced nanosegregations occurred in **FIAx** and **FIIAx**.

Electro-Optical Properties. The electro-optical properties were investigated from the saturated spontaneous polarization (P_s) values obtained via non-rubbed ITO-sandwiched cells to avoid the influence of polyimide layers to ferroelectricity.⁶⁵ Under electric fields of modified triangle waveforms, both FIAx and FIIAx displayed an anti-ferroelectric (AF) to a ferroelectric (FE) polar switching as C₆₀ increased:⁶¹ the responsive current signals at the zero-voltage position declined and even vanished as the molar ratio of C_{60} -based moiety with $\mathbf{x} \ge 50$ (see the orange rectangle area marked in Figs. 5a-5b), indicating a ferroelectric polar switching, where the SmCP range was maximized at x = 24 (see Table 1). According to the previous research,^{65,66} bent-core complexes processed anti-ferroelectric polar switching behavior due to the easy-stacking of bentshaped molecules with anti-polar order from neighboring layers, but the anti-polar packing could be diminished via NPs aggregated at the smectic layer interfaces. Hence, the switching from anti-ferroelectric to ferroelectric polar order along with C₆₀ ratio can be realized from the aggregation of NPs at the interfaces of smectic layers. The trends of P_s values and the saturated P_s values at certain electric fields of hybrid complexes FIAx and FIIAx are described in Fig. S7. As shown in Fig. 5c, as the C₆₀-based moiety increased, both P_s values of



Fig. 5 The saturated P_s values under modified triangular waveforms of (a) **FIAx** and (b) **FIIAx** under SmCP phases; (c) saturated P_s and (d) E_{sat} values of **FIAx** (black), **FIIAx** (white) and **SiA100** (gray) upon cooling. Blue dashed lines indicated applied electric field (33.3 V/µm at 50 Hz), and the orange rectangular was marked as the evidence for AF-FE switching depending on C₆₀ molar ratio for **FIA4** and **FIIA24**, respectively. (The detailed switching current responses and POM images to the applied triangular waveform in the mesomorphic phases of **FIA24** (SmCP_A) and **FIA76** (SmCP_F) are illustrated in Figs. S5 and S6, respectively.)

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hybrid complexes FIAx and FIIAx increased to be saturated at minor C_{60} doped ranges of x = 4 ~ 24, which are highle in that the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of x = 4 ~ 24, which are highle in the minor C_{60} doped ranges of (356 nC/cm²) of **SiA100** (at $\mathbf{x} = 0$) and reached the largest P values of 494 and 490 nC/cm² (both at $\mathbf{x} = 4$) for FIAx and FIIAx series, respectively. In addition, the P_s values of FIAx and FIL[®] dropped continuously to become smaller at higher C₆₀ dopeo ranges of $x = 50 \sim 90$. Similarly, in contrast to FIIAx with the same **x** values, **FIAx** (at $\mathbf{x} \ge 24$) possessed lower P_s values due to higher C_{60} -based moieties. The minor doping effect of C_{60} based moiety on promoted P_s values could be attributed to the reinforced separation of induced dipoles by the NP dopants under electric fields.⁶⁷ The saturated electric field (E_{sat}) was defined as the electric field at 90% of saturated $P_{\rm s}$ values (see Fig. S3 of the Supporting Information), and the lowest E_{sat} values of FIAx and FIIAx series were obtained at 8.8 and 12.0 Vpp/µm for FIA4 and FIIA24, respectively (see Fig. 5d), which were possibly due to the easy driving tendencies of the decreased packing order in the mesophases with minor C based contents.⁶⁸ By further increasing the ratio of C₆₀-based moiety, the field induced dipole reorientation was restricted by larger viscosities of the mesophases with higher C₆₀-basec contents and led to low P_s values and high E_{sat} values.⁶⁵ Notably, the saturated P_s value of SiA100 decayed to hali values for **FIAx** and **FIIAx** series at relatively high ratios of C_6^{-1} based moiety, i.e., **FIAx** at $\mathbf{x} \ge 76$ (i.e., 84 wt%, see Table S1) and FIIAx at x \geq 90 (i.e., 92 wt%), which happened in our previous report on Au-based LC composite at 5 wt% of NPs (a relatively low ratio of surface-modified Au NPs).⁶¹ Moreover, in contrast to a low wt% (5 wt%) of surface-modified Au NPs ir the previous Au-based LC composite,⁶¹ the AF-FE switching behavior happened at a higher content of surface-modified C_{60} (x = 50, i.e., 63 wt% and 56 wt% for FIAx and FIIAx, respectively) in both hybrid LC complexes. Finally, unlike the phase separation induced in the previous Au-based LC composite containing ~ 20 wt% content of surface-modified Au NPs, much higher contents of surface-modified C₆₀ still could be miscible in both FIAx and FIIAx series at x = 90 (i.e., 94 and 92 wt%, respectively), which could be useful to utilize hybric LC complexes with high contents of surface-modified C₆₀, Therefore, the anti-ferroelectric (AF) to a ferroelectric (FE) polar switching behavior as well as optimized P_{s} and E_{sat} values could be introduced by adjusting the contents of surfacemodified C₆₀ in the hybrid LC complexes.

Conclusion

Single- and double-armed C_{60} -based H-bonded bent-core complexes (**FIA100** and **FIIA100**) were mixed with Si-based Hbonded bent-core LC **SiA100** successfully to induferroelectric and anti-ferroelectric mesophases, where the hybrid C_{60} -based H-bonded LC complexes are synthesized and reported for the first time. Adjusting various molar ratios surface-modified C_{60} in the hybrid H-bonded bent-core LC complexes, the transition temperatures and ranges of the SmCP phases could be reduced and extended, respective *v* Compared with Au-based LC composites, much higher contents of surface-modified C_{60} could be miscible in bo h

hybrid LC complexes **FIAx** and **FIIAx** series at $\mathbf{x} = 90$ (i.e., 94 and 92 wt%, respectively). The anti-ferroelectric (AF) to a ferroelectric (FE) polar switching behavior occurred in the hybrid LC complexes with moderate molar ratios of C_{60} -based moiety. The electro-optical properties, such as P_s and E_{sat} values, could be manipulated and optimized via the blending ratio of the Si- and C_{60} -based complexes, which could be more easily controlled than the covalent-bonded C_{60} -based LC systems.

Acknowledgements

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The financial support of this project is provided by the Ministry of Science and Technology (MOST) in Taiwan through MOST 103-2113-M-009-018-MY3 and 103-2221-E-009-215-MY3. We also thank Drs. Ming-Tao Lee and Hwo-Shuenn Sheu for the assistance of measuring XRD at 13A1 and 01C2 in National Synchrotron Radiation Research Center (NSRRC) in Taiwan.

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Published on 11 November 2015. Downloaded by Monash University on 12/11/2015 06:32:52.

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