

## Reducing Ni residues of Metal Induced Crystallization Poly-Si with a Simple Chemical Oxide Layer

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The high leakage current is the most important issue of MIC-TFT. Ni residues in the MIC-TFT is the major cause of the leakage. In this study, a chemical oxide layer was used to avoid excess of Ni atoms into a-Si layer during MIC process. The process is simple and without extra thermal annealing. It was found that the Ni concentration of poly-Si film with chemical oxide layer was much less than that of conventional MIC poly-Si film. As a result, the leakage current was improved.

### Introduction

Low-temperature polycrystalline silicon thin-film transistors (LTPS-TFTs) have attracted considerable interest for their use in active-matrix liquid crystal displays (AMLCD). For commercial manufacturing, Ni-metal-induced crystallization (MIC) is optimum technology of LTPS due to low cost, large area, low temperature ( $\sim 500^\circ\text{C}$ ) and short time for crystallization (0.5 - 5 h). However, Ni and  $\text{NiSi}_2$  residues are easy to increase the leakage current and shift the threshold voltage [1]-[2]. Recently, several studies have demonstrated the reducing of Ni contamination; however, the on current of poly-Si films decreased [3].

In this study, a chemical oxide layer was used to avoid excess of Ni atoms into a-Si layer during MIC process. The manufacture processes were very simple and without extra thermal annealing.

### Experimental

N-type self-alignment poly-Si TFTs were investigated in this study. A 100-nm-thick undoped  $\alpha$ -Si layer was deposited onto a 500-nm-thick oxide-coated Si wafer by low pressure chemical vapor deposition (LPCVD) system. The 5-nm-thick cap oxide layer was formed by a simple chemical method [4]. A 5-nm-thick Ni film was then deposited and subsequently annealed at  $500^\circ\text{C}$  for 6 h in  $\text{N}_2$  to form poly-Si films. To reduce the Ni contamination, the remained Ni film and chemical oxide layer were then removed by wet etching. The islands of poly-Si regions were defined by Reactive ion etching (RIE). Next, a 100-nm-thick gate insulator was deposited by PECVD. Then a 100-nm-thick poly-Si film was deposited as the gate electrode by LPCVD. After defining the gate, self-aligned 30 keV phosphorous ions were implanted at a dose of  $5 \times 10^{15} \text{ cm}^{-2}$  to form the source/drain and gate. Dopant activation was performed at  $600^\circ\text{C}$  in  $\text{N}_2$  ambient for 12 h. It is worthy to note that manufacture processes were very simple and without extra thermal annealing. This method is very compatible with MIC processes. It just added a simple step that a-Si coated sample was immersed in chemical solution for a short time before depositing the Ni film.

## RESULTS AND DISCUSSION

Figure 1 displays the silicide etching holes after MIC. Note that silicide-etching solutions only remove the Ni silicide not the  $\alpha$ -Si. These holes were residues of the Ni silicides that had been etched away by a silicide-etching solution. As shown in Fig. 1, Ni concentration of poly-Si film with chemical oxide layer was much less than that of conventional MIC poly-Si film. Figure 2 shows the  $I_D$ - $V_G$  transfer characteristics at a drain bias of 5 V for  $10 \times 10 \mu\text{m}^2$  devices. The measured and extracted key device parameters are summarized in Table I. The threshold voltage ( $V_{th}$ ) is defined at a normalized drain current of  $I_{DS} = (W/L) \times 100\text{nA}$  at  $V_{DS} = 5\text{V}$ . The field-effect mobility ( $\mu_{FE}$ ) is extracted from the maximum value of transconductance at  $V_D = 0.1\text{V}$ . The electrical characters of MIC-TFTs with chemical oxide were significantly improved. It was attributed to lower Ni concentration in the MIC poly-Si films with chemical oxide layer, as shown in Fig. 1. It is known that Ni-related defects might degrade electric performance because the trap states introduced include dangling bonds and strain bonds [5]. Chemical oxide layer reduced content of Ni into channel layer during MIC annealing process to decrease Ni-related defects. Therefore, the carrier mobility increases due to lower impurity scattering of Ni-related defects.

### Summary

The chemical oxide layer was used to avoid excess of Ni atoms into a-Si layer during MIC process which was very simple and without extra thermal annealing. The only extra step was immersed a-Si sample in chemical solution before depositing the Ni film. The electrical performance of MIC-TFTs with chemical oxide layer was significantly improved. It was attributed to lower nickel concentration in the MIC poly-Si films with chemical oxide layer.

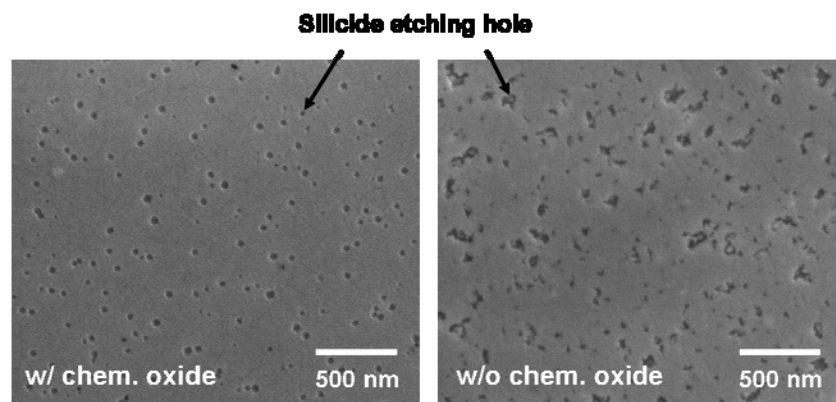


Fig. 1 The silicide etching holes after metal induced crystallization of poly-Si with and without chemical oxide layer.

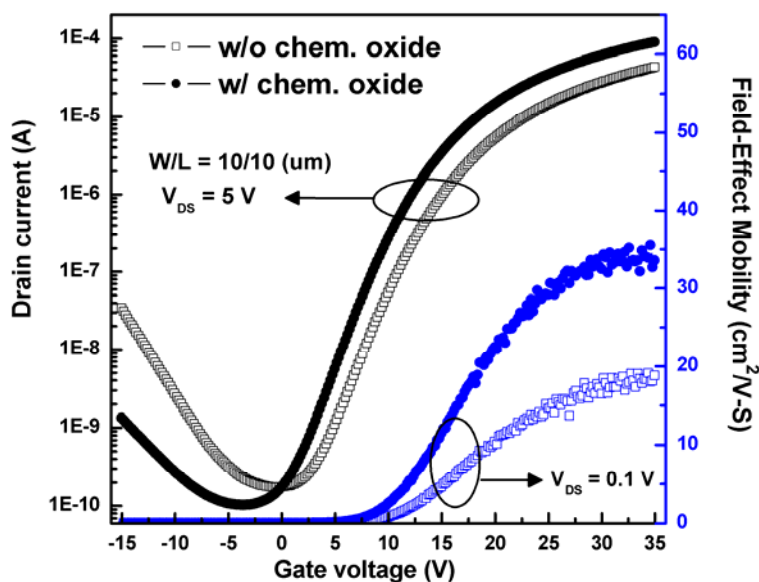


Fig. 2 Typical  $I_{DS}$ - $V_{GS}$  transfer characteristics and field-effect mobility.

**TABLE I.** Device characteristics of the MIC-TFTs and FODIC-TFTs

Device Parameters	w/o chem. oxide	w/ chem. oxide
Field-Effect Mobility $\mu_{FE}$ ( $\text{cm}^2 / \text{V-s}$ )	19.1	35.5
Threshold Voltage $V_{th}$ (V)	10.9	8.7
Subthreshold Slope S.S (V / dec)	2.67	2.66
$I_{min}$ (pA/ $\mu\text{m}$ )	17.2	10.2
Max on/off ratio ( $\times 10^5$ )	2.47	8.74

### Acknowledgments

This project was funded by Sino American Silicon Products Incorporation and the NSC of the ROC under Grant No. 98-2221-E-009-041-MY3. Technical supports from the National Nano Device Laboratory, Center for Nano Science and Technology and the Nano Facility Center of the National Chiao Tung University are also acknowledged.

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