# Improving the Electrical Integrity of  $Cu-CoSi<sub>2</sub>$ Contacted  $n^+p$  Junction Diodes Using Nitrogen-Incorporated Ta Films as a Diffusion Barrier

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*Abstract—***The study on improving the electrical integrity of Cu–CoSi**<sup>2</sup> **contacted-junction diodes by using the reactively** sputtered  $\text{TaN}_x$  as a diffusion barrier is presented in this paper. In this study, the Cu(300 nm)– $\cos i_2(50 \text{ nm})/n^+$  p junction diodes **were intact with respect to metallurgical reaction up to a 350 C thermal annealing while the electrical characteristics started to degrade after annealing at 300 C in N**<sup>2</sup> **ambient for 30 min.** With the addition of a 50-nm-thick  $\text{TaN}_x$  diffusion barrier **between Cu and CoSi**2**, the junction diodes were able to sustain annealing up to 600 C without losing the basic integrity of the device characteristics, and no metallurgical reaction could be observed even after a 750 C annealing in furnace. In addition, the** structure of TaN<sub> $x$ </sub> layers deposited on  $\cos i_2$  at various nitrogen flow rates has been investigated. The  $\text{TaN}_x$  film with small grain **sizes deposited at nitrogen flow ratios exceeding 10% shows better barrier capability against Cu diffusion than the others.**

*Index Terms—***Cobalt, copper, diffusion barrier, junction, nitrogen, tantalum.**

# I. INTRODUCTION

**F** OR A long time, metal silicides had proven valuable to both ohmic and Schottley contains ohmic and Schottky contacts. Aside from applications as ohmic or Schottky contacts, high conductivity metal silicides were proposed in 1979 as contacts to doped polysilicon or polycide structure to reduce resistance and give extra interconnection capability [1]. In 1981, the concept was extended to diffusion areas by forming self-aligned silicide at both polysilicon and diffusion areas simultaneously for improving both contact and interconnects [2]. Silicide-related technologies have therefore become an integral part of submicron devices in recent years.

The current trend in shrinking integrated circuits to improve the performance of chips requires substituting copper (Cu) for aluminum as interconnects in the deep submicron-integrated circuits [3], [4]. Recently, copper has been used as the global interconnect material for ultralarge-scale integration (ULSI) cir-

Manuscript received April 24, 2002. This work was supported in part by the National Science Council, Taiwan, R.O.C., under Contract NSC-90-2215-E-035-007 and the Feng Chia University (FCU-RD-88-01). The review of this paper was arranged by Editor R. Shrivastava.

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Digital Object Identifier 10.1109/TED.2002.804692

cuits as a replacement for aluminum due to its very low electrical resistivity and excellent electromigration resistance [5]. In the near future, copper will also be used in short-distance local interconnects for high-speed ULSI circuits.

It is well known that Cu forms Cu–Si compounds at very low temperatures (200 $\degree$ C) [6] and also introduces deep level traps in silicon, a phenomenon that deteriorates device performance [7]. Therefore, the realization of thermally stable Cu-contacted systems suppressing mass transport over interface due to the high diffusivity and reactivity of Cu is an essential issue in the use of Cu as a local interconnect material. Since copper is directly in contact with the source–drain area of MOSFETs in the local interconnects, it is expected that direct contact between copper and metal-silicided junction is inevitable. In recent years, CoSi has received a lot of attention due to its superior characteristics, compared to  $TiSi<sub>2</sub>$ , on issues of phase transformation, bridging effect, and chemical stability [8]. Unfortunately, Cu was found to diffuse across the silicide layer and form the Cu–Si compound below the silicide layer in the Cu–CoSi<sub>2</sub>–Si systems [9], [10]. Because the reaction in the Cu–silicide–Si structure occurs at a relatively low temperature, an effective diffusion barrier layer is required for the application of the Cu–silicide–Si structure.

The diffusion barriers employed directly in between the Cu and Si have been widely investigated in recent years. Various barrier layers for the Cu diffusion, such as refractory metals (Ta, W, Mo, Ti, and Cr) and their nitrides, have been developed [11]–[16]. Among these materials, tantalum (Ta) has been extensively investigated as a diffusion barrier for Cu since it not only shows relatively high melting temperature of 3087  $\mathrm{^{\circ}C}$  but is also shown to be thermal stable with respect to Cu. In addition, it was reported that the barrier capability of Ta films against Cu diffusion could be improved by incorporating nitrogen into Ta films using the reactive sputtering technique [15], [16]. In the TaN $<sub>x</sub>$ –Si system, several researchers have investigated that the</sub> phase of TaN $<sub>x</sub>$  films sequentially formed by sputtering Ta under</sub> increasing amounts of nitrogen partial flow includes the tetragonal metastable phase Ta  $(\beta$ -Ta), nitrogen-incorporated cubic Ta  $[\alpha$ -Ta(-N)], hexagonal Ta<sub>2</sub>N, and NaCl-type TaN [16]. In the Cu–TaN<sub>x</sub>/n<sup>+</sup>p junction diodes, the  $\alpha$ –Ta(–N) structure with reduced grains functioned as an effective barrier against Cu diffusion [16]. In this study, we used the TaN<sub>x</sub> barrier to improve the electrical integrity of the  $Cu-CoSi<sub>2</sub>$  contacted junction diodes. Besides, with increasing the nitrogen flow ratio, the evolution of structure of the TaN<sub>x</sub> films reactively sputtered on  $CoSi<sub>2</sub>$  films were investigated.



Fig. 1. Cross-sectional view of the Cu–TaN<sub>x</sub>–CoSi<sub>2</sub>/n<sup>+</sup>–p junction diodes.

#### II. EXPERIMENTAL PROCEDURE

The substrates used in the experiments were 6-in, p-type, (100)-oriented silicon wafers with a resistivity of  $6\sim 9$   $\Omega$ -cm. After standard RCA cleaning, the wafers were administered to the LOCOS process to define active region. The  $n^+$ -p junctions were formed by  $As<sup>+</sup>$  implantation at 60 keV with a dose of  $5 \times 10^{15}$  cm<sup>-2</sup> followed by the rapid thermal annealing (RTA) at 1050 °C for 30 s in  $N_2$  ambient. After the junctions were formed, a layer of cobalt (Co) 15 nm thick was deposited first and then followed by deposition of a TiN film (30 nm) that was used to prevent oxidation of cobalt. A two-step annealing process was used to form  $CoSi<sub>2</sub>$ . The first annealing step was carried out at 500 °C for 30 s in  $N_2$ . The silicide-forming reaction occurs whenever the silicon and metal are in contact, that is, on the contact window. The unreacted cobalt and TiN films were selectively removed in a piranha solution  $(H_2SO_4: H_2O_2: H_2O$  in the 1:1:5 volume ration). The second step of annealing was performed at  $750\,^{\circ}\text{C}$  for 30 s to achieve low resistivity  $CoSi<sub>2</sub>$  films. After the  $CoSi<sub>2</sub>$  film was formed, a 50-nm-thick  $\text{TaN}_x$  film was deposited onto the contact windows by reactively sputtering tantalum target in  $N_2$ Ar mixture. In this experiment, the nitrogen flow ratio is defined as the ratio of  $N_2$  partial flow to the total gas flow ( $N_2$ ) + Ar) and the deposited TaN<sub>x</sub> film is denoted as TaN<sub>x</sub>( $\times$ %). For example, TaN<sub>x</sub>(10%) is the TaN<sub>x</sub> film reactively sputtered by 10% nitrogen flow ratio. Then a Cu film 300-nm-thick was subsequently deposited in the same sputtering system without breaking vacuum. During sputtering, the pressure was maintained at 6 mtorr, and the sputtering power was 500 W and 1500 W for TaN $<sub>x</sub>$  and Cu, respectively. The Cu pattern was defined</sub> and etched using dilute  $HNO<sub>3</sub>$  solution, while the TaN<sub>x</sub> film was etched using  $Cl_2$  plasma to form  $Cu$ -TaN<sub>x</sub>-CoSi<sub>2</sub>/n<sup>+</sup>-p junction diodes. The cross-sectional view of the fabricated diodes is shown in Fig. 1. Finally, the samples were thermally annealed at a temperature ranging form 200  $\rm{^{\circ}C}$  to 800  $\rm{^{\circ}C}$  for 30 min to investigate the electrical characteristics.

#### III. RESULTS AND DISCUSSION

# *A. Cu–CoSi –Si Contact System*

The sheet resistance  $R_s$  of the Cu–CoSi<sub>2</sub>–Si structure after thermal annealing in  $N_2$  for 30 min at various temperatures



Fig. 2. Sheet resistance versus annealing temperature for the  $Cu-CoSi<sub>2</sub>-Si$ sample. The inset shows the XRD spectra of the  $Cu-CoSi<sub>2</sub>-Si$  structure.

is shown in Fig. 2. Several results have indicated that Cu diffuses fast in Si and forms Cu–Si compounds at temperatures as low as 200  $\rm{^{\circ}C}$  [6], [7]. The formation of Cu–Si compounds results in the increase of the measured sheet resistance values. The sheet resistance of  $Cu(300 \text{ nm})$ – $CoSi<sub>2</sub>(50 \text{ nm})$ –Si structure remains stable for anneal at temperatures up to  $350^{\circ}$ C. (The AES depth profile also indicates that no apparent structure intermixing occurred in the Cu–CoSi<sub>2</sub>–Si structure up to 350  $\rm{^{\circ}C}$ annealing.) Just like Al, Cu reacts at relatively low temperatures with silicides and Si [9], [10]. A sharp increase of sheet resistance was observed after annealing at 400 °C, indicating a structural change had occurred after this annealing. The inset of Fig. 2 shows the glancing angle  $(3^{\circ})$  X-ray diffraction (XRD) results. It reveals that  $Cu<sub>3</sub>Si$  phase has formed after annealing at 400 °C. The formation of high-resistivity  $Cu<sub>3</sub>Si$  and related in decrease Cu result in the drastic increase in sheet resistance, as shown in Fig. 2.

The reverse leakage current density of  $Cu-CoSi<sub>2</sub>/n<sup>+</sup> –p$  junction diodes measured at  $-5$  V for 20 randomly chosen samples is illustrated in Fig. 3. It is seen that although the Cu film in the Cu–CoSi<sub>2</sub>–Si structure keeps intact up to 350 °C thermal annealing from the sheet resistance, XRD, and AES measurements, the junction starts to degrade after 300  $^{\circ}$ C annealing from the diode characterization. The leakage current increases further with increased annealing temperature. SIMS (secondary ion mass spectroscopy) measurement was used to investigate the failure mechanism of the fabricated diodes. Results showed that a small amount of interdiffusion of Cu and Si is found for the sample annealed at 300  $\mathrm{^{\circ}C}$ . At this stage, the junction has failed and the leakage current densities were on the order of  $10^{-4}$ A/cm<sup>2</sup>. On the contrary, the volume fraction of the Cu<sub>3</sub>Si phase is too small to be detected by the XRD measurement. Thus, the sheet resistance of  $Cu-CoSi<sub>2</sub>-Si$  sample is not significantly affected by the relatively small interdiffusion as shown in Fig. 2. Therefore, it implies that the sheet resistance measurement is only valid for the global interconnections and the junction leakage evaluation is a suitable method for the local interconnections of Cu-metallization scheme.



Fig. 3. Histograms showing the distributions of leakage current density of the  $Cu-CoSi<sub>2</sub>/n<sup>+</sup>$ -p junction diodes under various annealing temperatures.



Fig. 4. Sheet resistance versus annealing temperature for the Cu–TaN<sub>x</sub>–CoSi<sub>2</sub>–Si sample. The TaN<sub>x</sub> layer was reactively sputtered at the nitrogen flow ration ranging from 0% to 20%. The inset is the cross-sectional view of the multilevel interconnection for integrated circuits.

## *B. Cu–TaN –CoSi –Si Contact System*

Although Fig. 2 shows that with a 50-nm thick film of  $CoSi<sub>2</sub>$  between Cu and Si, the electrical properties of junction diodes not affected after annealing at  $350^{\circ}$ C. However, this temperature is not high enough for the back-end requirement of ULSI processing. We therefore, used the TaN<sub>x</sub> diffusion barrier to improve the electrical integrity of the Cu–CoSi<sub>2</sub> contacted junction diodes. Fig. 4 shows the sheet resistance versus the annealing temperature for the  $Cu$ –TaN $_{x}$ –CoSi $_{2}$ –Si samples, where the TaN<sub>x</sub> barrier layer was deposited by the nitrogen flow ratio ranging from 0% to 20%. It is seen that with the addition of the diffusion barrier reactively sputtered at nitrogen flow ratios of 0%, 3%, and 5% in that order between Cu and CoSi<sub>2</sub>, the sheet resistance of Cu(300 nm)/TaN<sub>x</sub>(50 nm)– $CoSi<sub>2</sub>(50 nm)$ – $Si$  structures does not change up to 600 C furnace annealing. However, a drastic increase in sheet resistance was found after annealing above  $700\degree\text{C}$ . The drastic increase in sheet resistance is attributed to the formation of  $Cu<sub>3</sub>Si$  precipitates from the XRD measurement (see Fig. 6). Besides, as the nitrogen flow ratio exceeds 10%, the sheet resistance of  $Cu-TaN_x-CoSi_2-Si$  samples was stable up to  $700\degree$ C furnace annealing. A sharp increase of sheet resistance was observed after annealing at  $750\,^{\circ}\text{C}$ , indicating a structure change had occurred at this temperature. The sheet resistance measurement results indicate that  $\text{TaN}_x$  layers with  $N_2$  flow ratios above 10% on CoSi<sub>2</sub> are more effective barriers against Cu diffusion. As seen the inset of Fig. 4, it should be noticed that current flows in the Cu–barrier–dielectric system (long-distance global interconnection) is a horizontal type and the current flow path in the barrier layer is in parallel with the Cu path, while for the Cu–barrier–Si system (short distance local interconnection), the current path is a vertical type and the  $\text{TaN}_x$  resistance is in series to Cu. It is reported that the resistivity of reactively sputtered  $\text{TaN}_x$  films initially decreased with increasing nitrogen flow ratio and reached a minimum value of 159  $\mu\Omega$ -cm for 5% nitrogen flow ratio, then increased slightly between the nitrogen flow ratios of 5% and 10% and increased dramatically as the nitrogen flow ratio exceeding 10% [16]. Increasing the nitrogen ratio  $(>10%)$  in the reactively sputtered  $\text{TaN}_x$  films may lead to a higher barrier capability against Cu diffusion while the higher nitrogen flow rate will result in the highly resistivity of the TaN $_{x}$  films.

The SEM micrographs for the CoSi<sub>2</sub> surface after removal of the Cu and TaN<sub>x</sub> layers by the  $HNO<sub>3</sub> + H<sub>2</sub>O<sub>2</sub> (1 : 1)$  solution are shown in Fig. 5. After  $700^{\circ}$ C annealing, precipitates were found on the CoSi<sub>2</sub> surface for the Cu–TaN<sub>x</sub>(0% and 5%)–CoSi<sub>2</sub>–Si samples as shown in Fig.  $5(a)$  and (b), while the surface of CoSi<sub>2</sub> remains intact for the Cu–TaN<sub>x</sub>(10%)–CoSi<sub>2</sub>–Si sample as shown in Fig. 5(c). This implies that the barrier capability of TaN<sub>x</sub> layers against Cu diffusion in the Cu–TaN<sub>x</sub>–CoSi<sub>2</sub>–Si structure increased with increasing the  $N_2$  flow rate during the reactive sputtering. However, the 800  $\,^{\circ}$ C thermal annealing [Fig.  $5(d)$ ] reveals precipitate formation on the  $CoSi<sub>2</sub>$  surface for the Cu–TaN<sub>x</sub>(10%)–CoSi<sub>2</sub>–Si sample which is consistent with the results of sheet resistance measurement as shown in Fig. 4.

Fig. 6 shows the XRD spectra for the  $Cu$ –TaN<sub>x</sub>–CoSi<sub>2</sub>–Si samples subjected to anneal in  $N_2$  for 30 minute at various temperatures. For the TaN $<sub>x</sub>$  films deposited at the nitrogen</sub> flow ratio of 0% and 5%, the diffraction patterns reveal that the samples remain unchanged after anneal at temperatures up to 600  $\rm{^{\circ}C}$ , while different sets of peaks belonging to Cu<sub>3</sub>Si, TaSi<sub>2</sub>, and Ta<sub>5</sub>Si<sub>3</sub> are found after 700  $\mathrm{^{\circ}C}$  annealing as shown in Fig. 6(a) and (b). The cross-sectional TEM micrographs have showed that most of the Cu transported across the barrier and  $CoSi<sub>2</sub>$  layers and reacted with Si substrate, resulting in the drastic increase of sheet resistance as shown in Fig. 4. In the meanwhile, the cross-sectional TEM measurement indicates that the  $CoSi<sub>2</sub>$  layer did not dissociate during the process of the high-temperature annealing, implying the Si of  $Cu<sub>3</sub>Si$ , TaSi<sub>2</sub>, and  $Ta_5Si_3$  comes from the Si substrate. Another series of XRD patterns presented in Fig. 6(c) and (d) indicates the samples (TaN<sub>x</sub> films deposited at the nitrogen flow ratio of 10% and 20%) remain unchanged after anneal at a temperature up to 700 °C. But intermetallic compounds were also observed on the structure of Cu–TaN<sub>x</sub>(10% $\sim$ 20%)–CoSi<sub>2</sub>–Si samples after



Fig. 5. SEM micrographs of CoSi<sub>2</sub> after removal of the Cu and TaN<sub>x</sub> films from the Cu–TaN<sub>x</sub>–CoSi<sub>2</sub>–Si samples. The sputtered TaN<sub>x</sub> films and the annealing temperatures are (a) TaN<sub>x</sub>(0%), 700 °C; (b) TaN<sub>x</sub>(5%), 700 °C; (c) TaN<sub>x</sub>(10%), 700 °C; and (d) TaN<sub>x</sub>(10%), 800 °C, respectively.

800 $\degree$ C annealing. The annealing temperatures resulted in the abrupt increase of sheet resistance as shown in Fig. 4, consistent with the results of SEM in Fig. 5. The results of Figs. 4 and 6 show that the annealing temperature before the  $Cu-CoSi<sub>2</sub>-Si$ samples become unstable increased from 350  $\degree$ C to 600  $\degree$ C with the addition of a 50 nm thickness of TaN<sub>x</sub> (0% $\sim$ 5%) as a barrier between Cu and  $\cos i_2$ . Increasing the nitrogen flow rate during the sputtering of  $\text{TaN}_x$  films will lead to a higher barrier capability of the TaN<sub>x</sub> films and the failure annealing temperature for the Cu–TaN<sub>x</sub>(10% $\sim$ 20%)–CoSi<sub>2</sub>–Si samples is further increased to 700  $\mathrm{^{\circ}C}$ .

In our previous investigation [16], the XRD pattern for different Ta and TaN<sub>x</sub> films deposited on (100) silicon substrate under different nitrogen flow rates follows the zone model, which progressively changed from voided columnar film of Ta, through fibrous of reduced grains  $[\alpha$ -Ta(-N)], featureless structure (Ta<sub>2</sub>N), and finally to columnar structure (TaN). The quasi-amorphous structure of  $\alpha$ -Ta(-N) films lengthen the diffusion path of Cu before it can react with Si, hence the TaN $_{x}$ (3 $\sim$ 5%) films provided a better barrier capability against Cu diffusion than those of TaN<sub>x</sub>(0% and 10 $\sim$ 30%) films [16]. Fig. 7 shows the XRD spectra of the TaN<sub>x</sub> films deposited at the various nitrogen flow ratios on the  $CoSi<sub>2</sub>$  film. Just like the TaN $_{x}$  deposited on the Si, the crystallographic structure of TaN<sub>x</sub> film deposited on the CoSi<sub>2</sub> was also affected by

the nitrogen flow ratio during the reactive sputtering. In this figure, tetragonal  $\beta$ -Ta is observed without any nitrogen flow. By incorporating nitrogen into the Ta films, the relatively low resistivity as measured from the four-point probe along with the phase identified by XRD patterns clearly indicates that the deposited films of nitrogen flow ratios of 3% and 5% are the low-resistivity  $\alpha$ -Ta phase and nitrogen-doped body-centered cubic (bcc)  $\alpha$ -Ta(-N) phases, respectively. The diffraction pattern for the films deposited at the nitrogen flow ration of 10, 15, and 20% respectively contains the high-resistivity face-centered cubic (fcc) TaN<sub>x</sub> phase. In contrast to the TaN<sub>x</sub> films deposited on the Si substrate, these patterns indicate that all the deposited films are not composed of an amorphous-like structure.

The bright-field image and selective area diffraction indicated that grain sizes of the TaN<sub>x</sub> films deposited on the CoSi<sub>2</sub> film decreased with the increased nitrogen flow ratio. Fig. 8(a) displays a transmission electron bright-field micrograph, taken from the diffuse ring of the diffraction pattern (inset) for  $\beta$ –Ta films with grain size of  $20 \sim 30$  nm deposited without any nitrogen flow. On the contrary, the matrix exhibits a speckle feature of size  $\sim$ 2 nm for TaN<sub>x</sub> films deposited at 20% nitrogen flow ratio [Fig. 8(b)]. The activation energies of the Cu diffusion in the TaN were measured to be 1.3 eV–2.7 eV, which indicated that the Cu diffusion in the TaN would be controlled



Fig. 6. XRD spectra for (a) Cu–TaN<sub>x</sub>(0%)–CoSi<sub>2</sub>–Si; (b) Cu–TaN<sub>x</sub>(5%)–CoSi<sub>2</sub>–Si; (c) Cu–TaN<sub>x</sub>(10%)–CoSi<sub>2</sub>–Si; and (d) Cu–TaN<sub>x</sub>(20%)–CoSi<sub>2</sub>–Si samples subjected to anneal at various temperatures.



Fig. 7. XRD spectra of the TaN<sub>x</sub> deposited at the various nitrogen flow ratios on CoSi<sub>2</sub>. The phase identified by XRD patterns indicates that the deposited films of nitrogen flow ratios of 0, 3, and 5% are the  $\beta$ -Ta phase,  $\alpha$ -Ta phase, and nitrogen-doped bcc- $\alpha$ –Ta(–N) phases, respectively. The diffraction pattern for the films deposited at the nitrogen flow ratios of 10, 15, and 20% respectively contains the fcc-TaN $_x$  phase.

by grain boundaries and lattice diffusion [13]. Since the barrier layer allows Cu to preferentially penetrate through the grain

boundaries, a barrier structure with small grains increases the diffusion path of Cu to react with Si. Hence the TaN<sub>x</sub> films deposited at the nitrogen flow ratio of  $10\% \sim 20\%$  yield the better barrier properties than those of  $0\% \sim 5\%$ .

As discussed in the Cu–CoSi<sub>2</sub>/n<sup>+</sup>–p junction diodes, the electrical characteristics are usually more sensitive than the metallurgical properties for the analysis of the reaction that occurred within the structure. Fig. 9 illustrates the reverse-biased current densities of the Cu–TaN<sub>x</sub>–CoSi<sub>2</sub>/n<sup>+</sup>–p junction diodes annealed at various temperatures. For the diodes without any heat treatment (as-deposited), the leakage current densities remain stable on the order of  $10^{-8}$  A/cm<sup>2</sup>. Nevertheless, the diode leakage increased with increasing the annealing temperature and all of diodes failed after annealing at  $650$  °C. The abrupt increase of leakage current density in the  $650$   $\degree$ C-annealed diodes implies that the junctions were markedly destroyed even though no evidence of reaction can be found by the material analysis. As seen in Fig. 9, it is interesting that in the 600 C-annealed diode the barrier capability of TaN<sub>x</sub> layer against Cu diffusion increased with increasing the nitrogen flow ratio. In other words, a 50-nm  $\text{TaN}_x$  (nitrogen flow ratio  $>10\%$ ) layer is observed to be an effective diffusion barrier up to 600 °C between Cu and  $\cos i_2$  for maintaining the electrical integrity of the junction diodes. While at the





 $(b)$ 

Fig. 8. Bright-field images of  $\text{TaN}_x$  films deposited at nitrogen flow ratio of (a) 0% and (b) 20%.

same annealing temperature, the diodes failed with the  $\text{TaN}_x$ diffusion barrier sputtered at the nitrogen flow ratios ranging from 0% to 5%. SIMS measurement was used to investigate the failure mechanism of the fabricated diodes. Results showed that an interdiffusion of Cu and Si across the barrier film is found for the Cu–Ta–CoSi<sub>2</sub>–Si sample annealed at 600 °C. The electrical evaluation reveals that this interdiffusion destroyed the junction characteristics as shown in Fig. 9. In addition, the SIMS analysis for the Cu–TaN<sub>x</sub>(5%)–CoSi<sub>2</sub>–Si sample annealed at  $600\text{ °C}$  showed that the relatively small amount of Cu atoms diffuses into the Si substrate as compared to the  $Cu$ –Ta– $CoSi<sub>2</sub>$ –Si sample. Although the small amount of Cu atoms diffused into Si substrate did not affect the Cu sheet resistance significantly (see Fig. 4), the junction leakage showed a three-order of magnitude increase in the leakage current



Fig. 9. Leakage current density of the Cu–TaN<sub>x</sub>–CoSi<sub>2</sub>/n<sup>+</sup>–p junction diodes versus the annealing temperature. The TaN<sub>x</sub> layer was reactively sputtered at the different nitrogen flow ratios.

measurement (from  $10^{-8}$  to  $10^{-5}$  A/cm<sup>2</sup>, as seen in Fig. 9). Furthermore, SIMS measurement was also used to investigate the interdiffusion for the Cu–TaN<sub>x</sub>( $>$ 10%)–CoSi<sub>2</sub>–Si samples annealed at  $600\degree$ C. In the measurement, an insignificant penetration of Cu atoms through the barrier film resulted in the integrity of junction characteristics. The results of SIMS analysis are consistent with the electrical evaluations. They suggest that the barrier capability of TaN<sub>x</sub> films against Cu diffusion increased with increased nitrogen-flow rate during the reactive sputtering. Again, it should be noticed that although increasing the nitrogen flow rate will lead to a higher barrier capability of the TaN<sub>x</sub> layer against Cu diffusion, the high nitrogen ratio results in the high resistivity of the TaN<sub>x</sub> films. From the XRD measurement and junction characterization, results show that the crystallographic structure and the barrier capability against Cu diffusion of TaN<sub>x</sub> films are not only dependent on the nitrogen ratio but also on the substrate type.

## IV. CONCLUSION

Metallurgical reaction in  $Cu-CoSi<sub>2</sub>-Si$  structure was observed after a 400  $^{\circ}$ C furnace annealing, while the electrical degradation of  $Cu-CoSi<sub>2</sub>/n<sup>+</sup>-p$  junction diodes was observed after annealing at 300 °C. By using a 50 nm thickness of TaN<sub>x</sub> layer as diffusion barrier, the Cu–TaN<sub>x</sub>–CoSi<sub>2</sub>/n<sup>+</sup>–p junction diodes were able to sustain the 30 min thermal annealing up to 600 C without degrading the basic electrical characteristics. With a higher temperature annealing of  $750^{\circ}$ C, metallurgical reaction occurred in local spots with the formation of  $Cu<sub>3</sub>Si$ , TaSi<sub>2</sub>, and Ta<sub>5</sub>Si<sub>3</sub>. The high-temperature failure of barrier capability for the TaN<sub>x</sub> layers is due to interdiffusion of Cu and Si across the TaN $_x$  film structure and the interdiffusion is affected by the microstructure of the TaN $<sub>x</sub>$  films. By increasing</sub> the nitrogen flow ratio, the evolution of structure of the  $\text{TaN}_x$ films changes from pure  $\beta$ -Ta through low-resistivity  $\alpha$ -Ta phase, nitrogen-doped body-centered cubic  $\alpha$ -Ta(-N) phase, and finally to small grain size face-centered cubic  $\text{TaN}_x$  phase. Among them, the small grain size face-centered cubic  $\text{TaN}_x$  structure deposited at the nitrogen flow ratio exceeding 10% showed the better barrier capability against Cu diffusion than the others. Although increasing the nitrogen ratio  $( >10\%)$ in the reactively sputtered TaN<sub>x</sub> films may lead to a better diffusion barrier against Cu, the higher nitrogen ratio will result in the highly resistivity of the TaN $<sub>x</sub>$  films. Hence, when the</sub> TaN $<sub>x</sub>$  films are reactively sputtered at the different nitrogen</sub> flow rate as the barrier against Cu diffusion, a tradeoff between the barrier capability and resistivity of the diffusion barrier is very important for copper metallization in short distance local interconnects for the high-speed ULSI circuits. Furthermore, the crystallographic structure and the barrier property of  $\text{TaN}_x$ films are not only dependent on the nitrogen flow rate but also on the infrastructure during the deposition process.

#### ACKNOWLEDGMENT

The authors would like to thank the National Nano Device Laboratory, Hsinchu, Taiwan, R.O.C., for providing an excellent processing environment.

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