

High Reliability Polyoxide Fabricated by Using TEOS Oxide Deposited on Disilane Polysilicon Film

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Abstract—This paper reports the TEOS oxide deposited on the polysilicon film which was prepared by using the disilane chemical vapor deposition. Compared to the thermally grown oxide or TEOS (tetra-ethyl-ortho-silicate) oxide deposited on the conventional silane polysilicon film, it had symmetrical J - E characteristics that had lower leakage currents but much higher breakdown field, a lower electron trapping rate, and a much larger charge to breakdown. These good properties are attributed to the smoother surface of the deposited disilane poly-I film and the more incorporation of nitrogen during the rapid thermal annealing (RTA) in N_2O ambient. It is suitable to be as the inter-polyoxide of the electrically-erasable programmable read only memory (EEPROM).

I. INTRODUCTION

IN ORDER to have good data retention characteristics for the deep submicron electrically-erasable programmable read only memory (EEPROM), a thin polyoxide with a low leakage current, a high breakdown field (E_{bd}), a large charge to breakdown (Q_{bd}) and a low electron trapping rate is required [1]–[4]. Recently, TEOS vapor deposited polyoxide with rapid-thermal-annealing (RTA) in N_2O has been reported to have a higher reliability due to its smoother interface after oxidation [5], [6]. The quality of the TEOS (tetra-ethyl-ortho-silicate) polyoxide is strongly related to the surface roughness and the doping concentration of polyI. Unfortunately, for the conventional polysilicon film, the surface roughness increases as the doping concentration of polyI decreases [7], reducing the advantage obtained by the TEOS polyoxide.

Disilane polysilicon film was widely used in fabrication of the low-temperature thin-film transistor (TFT) for its lower deposition temperature and larger grain [8], [9]. It was also used as HSG (hemispherical grain) poly-Si films in dynamically random accessible memories (DRAM's) for its wide transition temperature [10]. It has a much smoother surface in the low doping concentration regime. However, little study was done on growing oxides on it, especially, on depositing TEOS oxide on it. This paper reports, for the first time, the results on investigation on low pressure chemical vapor deposition (LPCVD) TEOS oxide deposited on the disilane polysilicon followed with RTA in N_2O . It is found that the fabricated polyoxide

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TABLE I
DETAILED PROCESS STEPS FOR EACH SAMPLE

sample	Silane Poly-I	Disilane Poly-I	Thermal O ₂	Thermal N ₂ O	TEOS	TEOS + RT N ₂ O	TEOS + RT O ₂	TEOS + RT N ₂
D-TH-O ₂		▲	▲					
D-TH-N ₂ O		▲		▲				
D-T		▲			▲			
D-T-RT-N ₂ O		▲				▲		
D-T-RT-O ₂		▲					▲	
D-T-RT-N ₂		▲						▲
S-TH-N ₂ O	▲			▲				
S-T-RT-N ₂ O	▲					▲		

has a very high Q_{bd} , a low electron trapping rate, and a high E_{bd} . The polyoxide, when used in EEPROM applications, can improve performance and reliability of the EEPROM device.

II. EXPERIMENTS

In this study, n^+ -polysilicon/polyoxide/ n^+ -polysilicon capacitors were fabricated and used in the measurement. At first, p-type wafers were thermally oxidized in dry O_2 at 950 °C to have a 100 Å oxide. Then a 200 nm disilane polysilicon film (poly-I) was deposited at 460 °C. The poly-I film was then phosphorous-implanted at an energy of 30 keV of a dose of $5 \times 10^{15} \text{ cm}^{-2}$. After rapid-thermal (RT) annealed at 950 °C for 30 s in an N_2 ambient to obtain a sheet resistance of 70 Ω/cm^2 for poly-I, the wafers were then deposited with a TEOS oxide of a thickness of 130 Å at 700 °C. The oxide was then RT annealed in an N_2O ambient at 950 °C for 30 s. For comparison, the same oxide but RT annealed in O_2 or N_2 ambient at 950 °C for 30 s were also prepared. A second silane polysilicon (poly-II) of a thickness of 300 nm was then deposited again. The poly-II was doped by $POCl_3$ and driven-in at 850 °C for 1 h in an N_2 ambient to obtain a sheet resistance of 40 Ω/cm^2 . It was then patterned and grown a 100 nm passivation oxide. After contact hole opening, Al film was deposited, patterned, and sintered at 350 °C for 40 min in an N_2 ambient to be gates and contacts of the capacitors. In the above, also for comparison, similar capacitors but with their poly-I grown with silane at 620 °C, of a sheet resistance of 140 Ω/cm^2 , were prepared. Also, for comparison, similar capacitors but their oxides grown in O_2 and N_2O at 850 °C by the conventional thermal oxidation method were prepared. The process steps for each sample are compiled in Table I.

The thickness (equivalent oxide thickness E_{ox}) of the oxide was measured by CV measurement and verified by TEM, and Q_{bds} and E_{bds} were measured by an HP4145b semiconductor analyzer.

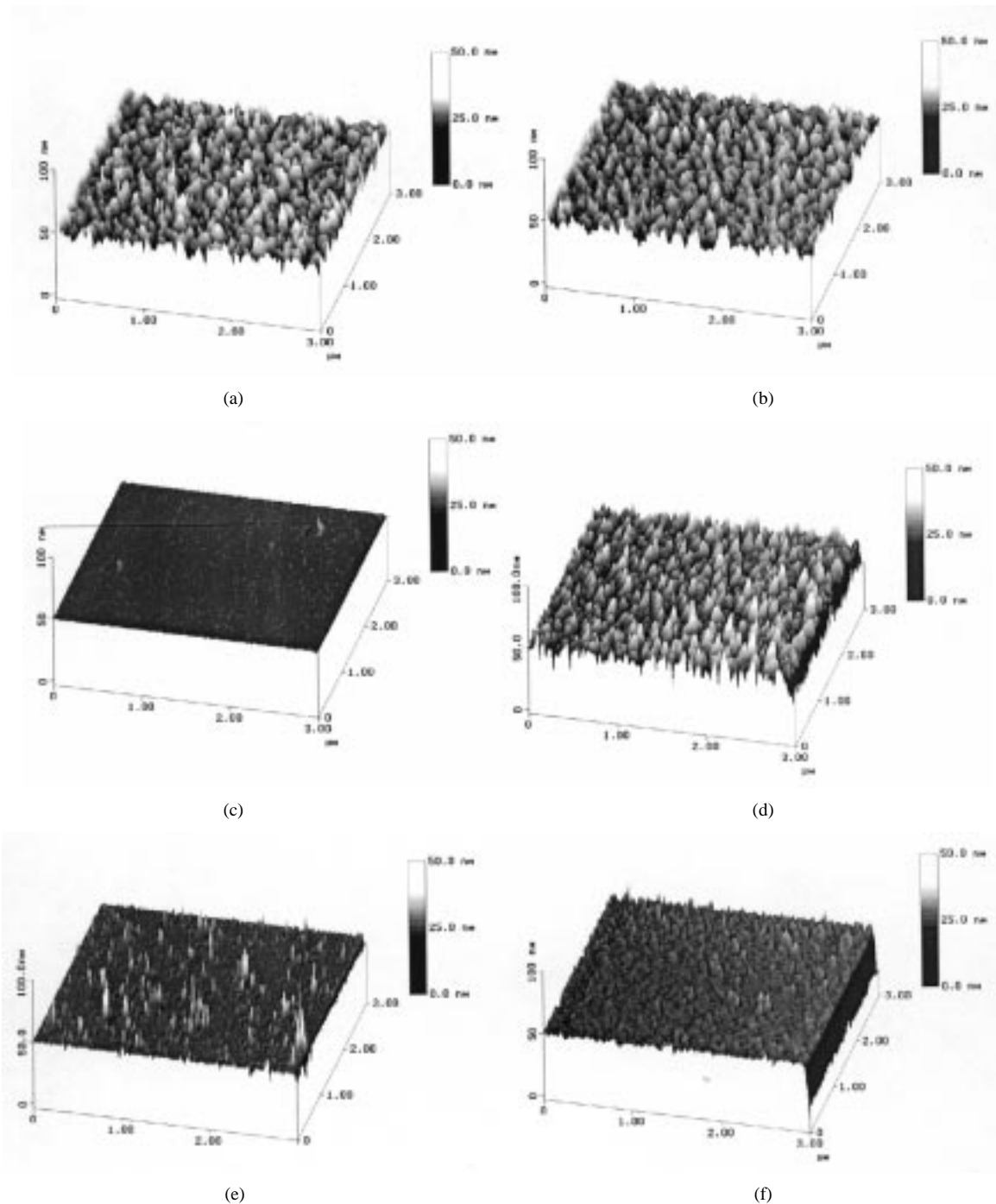


Fig. 1. AFM images of polyI surface of (a) D-TH-O₂; (b) D-TH-N₂O; (c) D-T-RT-N₂O; and (d) S-T-RT-N₂O, (e) D-T, and (f) D-T-RT-O₂ samples with their poly-II and polyoxide removed. The surface roughness is 44 Å, 31 Å, 3 Å, 49 Å, 12 Å, and 34 Å, respectively.

III. RESULTS AND DISCUSSION

Fig. 1 shows the atomic force microscope (AFM) images of the poly-I surfaces of the following samples: (a) D-TH-O₂: the O₂ thermal oxide grown on the disilane poly-I; (b) D-TH-N₂O: the N₂O thermal oxide grown on the disilane poly-I; (c) D-T-RT-N₂O: the TEOS oxide deposited on the disilane poly-I with RTN₂O annealing; (d) S-T-RT-N₂O: the TEOS oxide deposited on the silane poly-I with RTN₂O annealing; (e) D-T: the TEOS oxide deposited on the disilane poly-I; and (f) D-T-RT-O₂: the TEOS oxide deposited on the disilane poly-I

with RTO₂ annealing, respectively. In the above, all images are the surface images of poly-I's of the capacitors with their poly-II's and polyoxides removed. The roughness obtained from the AFM measurement for each sample are D-TH-O₂: 44 Å, D-TH-N₂O: 31 Å, D-T-RT-N₂O: 3 Å, S-T-RT-N₂O: 49 Å, D-T: 12 Å, and D-T-RT-O₂: 34 Å, respectively. From these pictures and data, it can be found that

- 1) the TEOS with the RTN₂O sample had the smoothest surface;
- 2) the disilane samples had smoother surface than that of the silane samples;

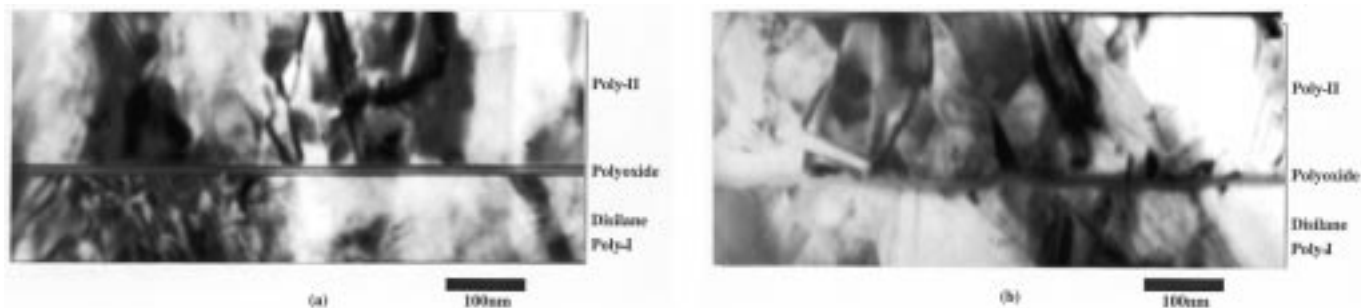


Fig. 2. TEM images of the (a) D-T-RT-N₂O and (b) S-T-RT-N₂O sample. The poly-I thickness is 2000 Å, polyoxide thickness is 150 Å.

3) the polyoxides treated in N₂O ambient improved the poly-I/oxide interface roughness.

In addition, for disilane samples, the thermally grown polyoxide had a rougher interface than that of the TEOS oxide. This indicates that thermal oxidation enhanced interface roughness. Also, the oxides grown or annealed in the N₂O ambient had a smoother interface than those treated in the O₂ ambient. This was probably caused by the fact that N₂O oxidation or annealing provided nitrogen which passivated grain boundaries of the deposited poly-I layer and reduced the grain boundary enhancement effect [1].

Fig. 2 shows the cross section TEM images of the D-T-RT-N₂O and S-T-RT-N₂O samples. These pictures show that the D-T-RT-N₂O sample had a smoother interface than that of the S-T-RT-N₂O sample. Two samples had a similar grain size on the poly-I film. However, it is noticed that, for the D-T-RT-N₂O sample, both its polysilicon-II/oxide/polysilicon-I interfaces had similar smoothness. In addition, from the TEM picture of the D-T-RT-N₂O sample, the oxide thickness can be estimated to be 150 Å, which is the same as the value derived from the CV characteristic of the sample. The dielectric constant of the polyoxide was 3.9.

Fig. 3 shows the (a) positive and (b) negative *J-E* characteristics of the six samples. The D-T-RT-N₂O sample, i.e., the TEOS oxide deposited on the disilane polysilicon film with RTN₂O annealing, had the lowest leakage current and the highest breakdown field. This could be attributed to the much smoother poly-I of this sample. In the figure, the D-T sample, i.e., the D-T-RT-N₂O sample without the RT N₂O annealing, had a poorer characteristics than that of the D-T-RT-N₂O sample, although it had a relatively smooth poly-I/oxide interface. Hence, the RT annealing in N₂O is an important step in improving the quality of the oxide. It annealed the deposited oxide, created an additional 20 Å thermal oxide at the poly-I/oxide interface, and introduced nitrogen into the oxide. The latter will be addressed further next by the support of SIMS data. In addition, it is seen that besides the D-T-RT-N₂O sample, all other samples had asymmetrical *J-E* characteristics. That the D-T-RT-N₂O sample had the symmetrical *J-E* characteristics is believed due to the fact that it had a symmetrical smoothness on its both polysilicon-II/oxide and oxide/polysilicon-I interfaces, as revealed by the previous TEM pictures.

Comparing the *J-E* characteristics of the D-T sample with those of the thermally grown polyoxides grown on either the disilane poly-I, i.e., D-TH-O₂ and D-TH-N₂O, or on the silane

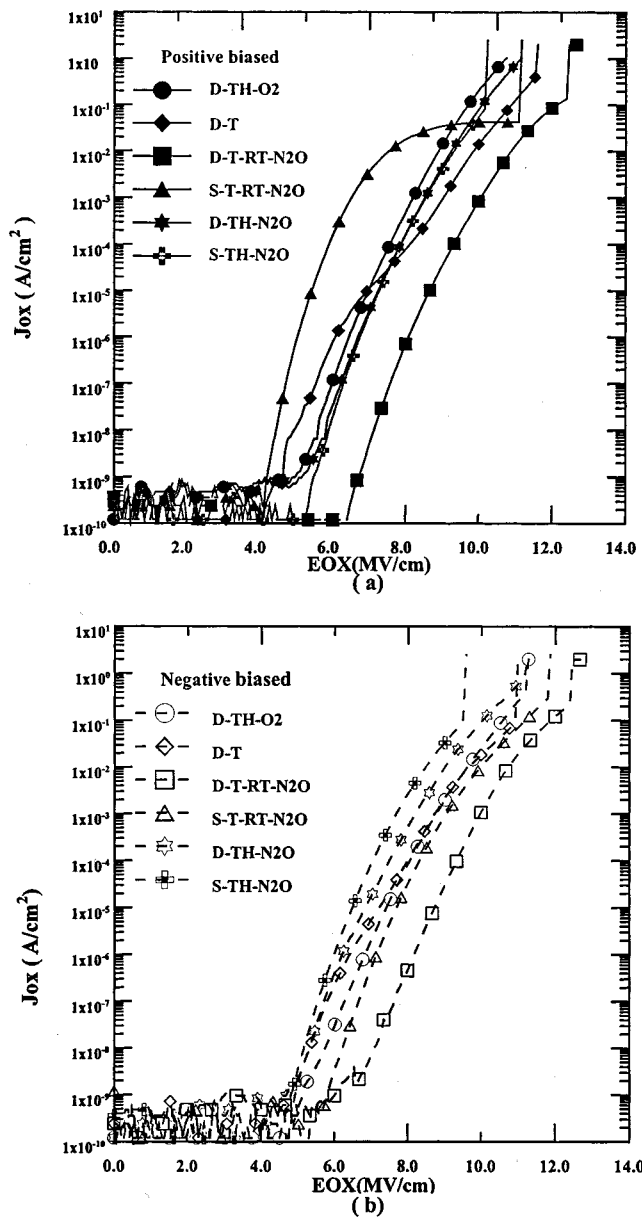


Fig. 3. *J-E* characteristics of the D-TH-O₂, D-TH-N₂O, S-TH-N₂O, D-T-RT-N₂O, S-T-RT-N₂O, and D-T six samples with (a) poly-II positive biased and (b) poly-II negative biased.

poly-I, i.e., D-TH-N₂O and S-TH-N₂O, we find that the D-T sample, even though it had relatively smooth poly-II/oxide and oxide/poly-II interfaces, had a higher leakage current at the low

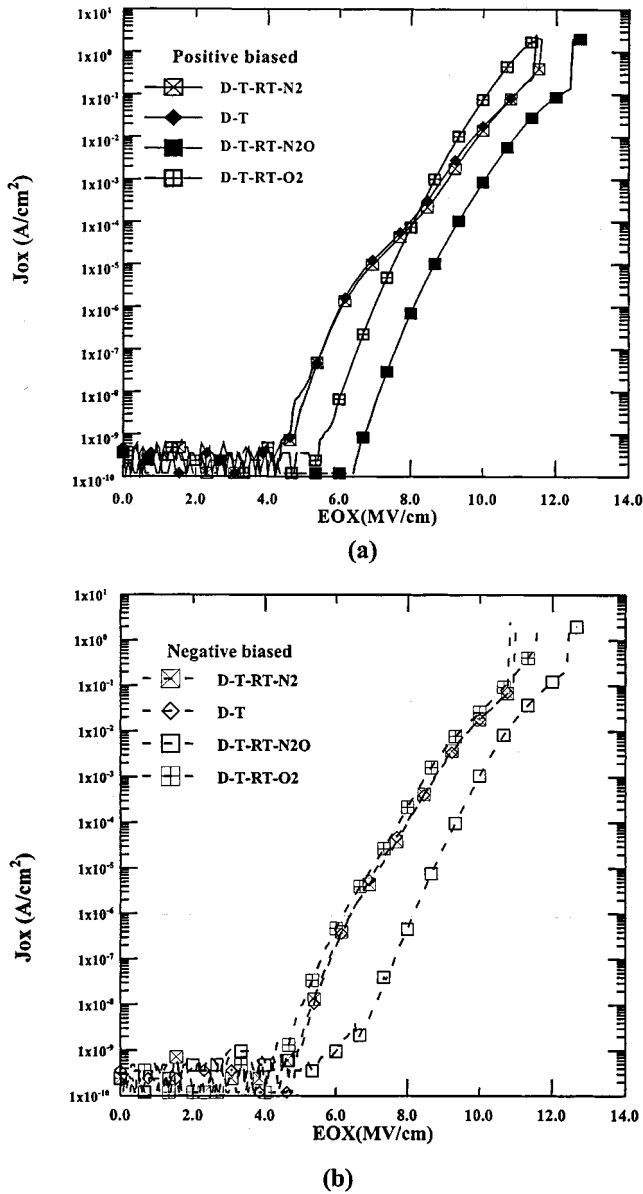


Fig. 4. J - E characteristics of the D-T-RT-N₂O, D-T, D-T-N₂, and D-T-RT-O₂ four samples with (a) poly-II positive biased and (b) poly-II negative biased.

electric field regime. It is because, the TEOS oxide, deposited at a low temperature (700 °C), had a porous structure before RT annealing. This led a high trap density existing in the oxide, providing conduction paths for injected electrons under the applied field.

Fig. 4 shows the (a) positive J - E characteristics and (b) the negative J - E characteristics of the TEOS oxide deposited on the disilane films with rapid thermal annealing at 950 °C 30 s in N₂, O₂, and N₂O ambients, respectively. We can see that the RTN₂O annealed sample had the lowest leakage current. This can be simply explained by the fact that it had a relatively smoother interface than those of other samples. The RTN₂ sample did not have any improvement as compared with the as-deposited TEOS sample. In addition, we see that the RTO₂ improved the J - E characteristics of the TEOS oxide only in the positive bias. The improvement could be due

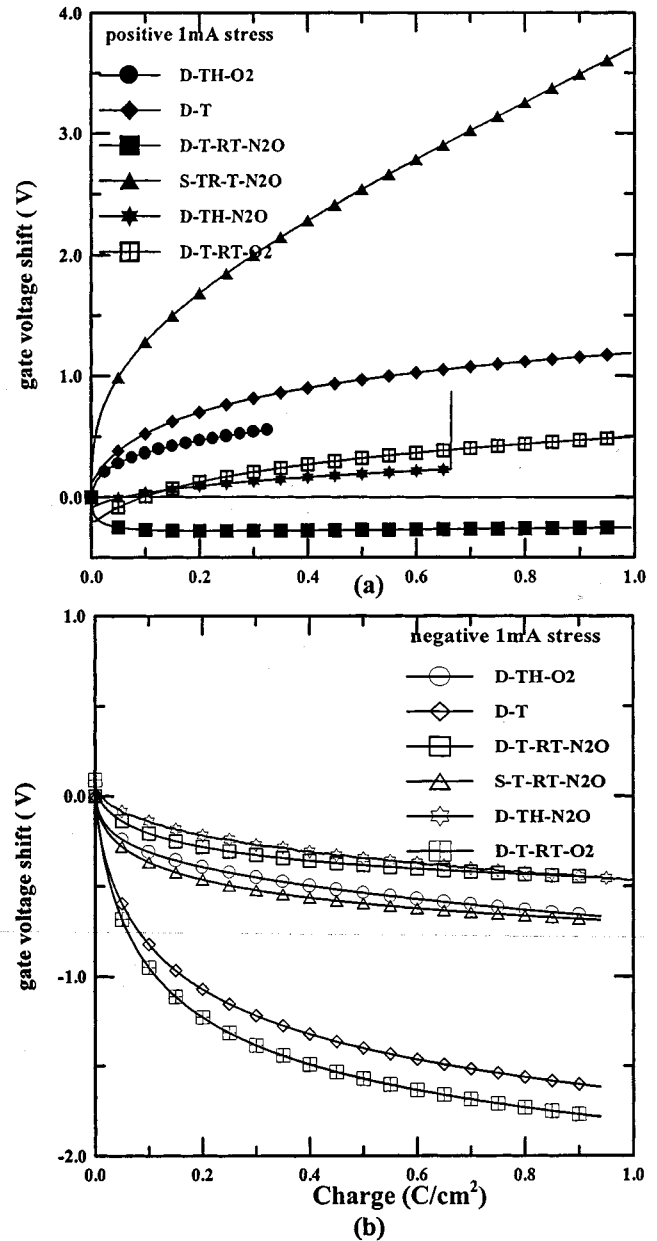
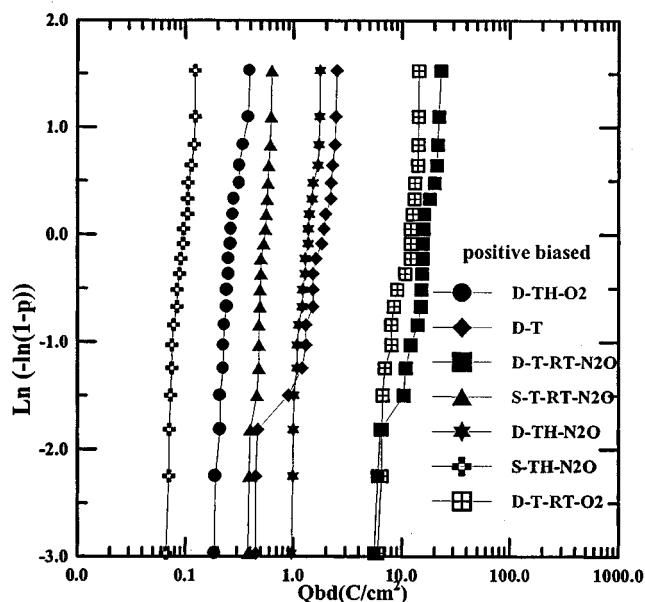


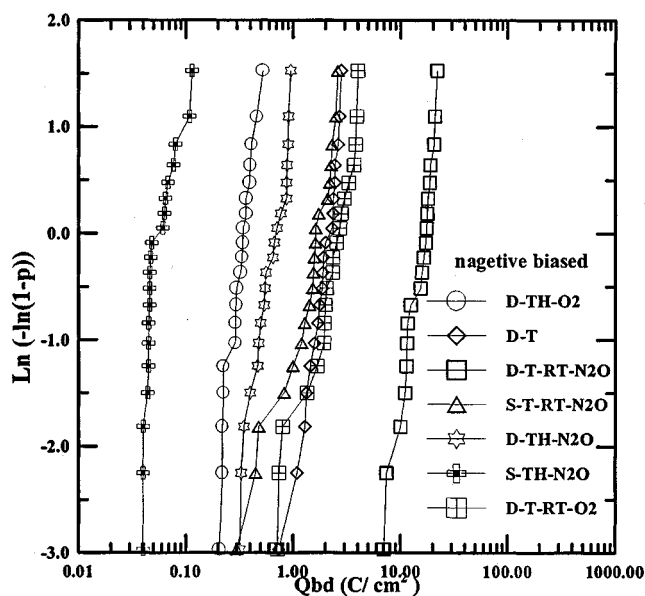
Fig. 5. Electron trapping characteristics of the D-TH-O₂, D-TH-N₂O, D-T-RT-N₂O, S-T-RT-N₂O, D-T-RT-O₂, and D-T six samples with (a) poly-II positive biased and (b) poly-II negative biased.

to the additional thermal oxide grown during the annealing process. However, this process also enhanced the roughness of the polyoxide/polysilicon-I interface, consequently, that of the polysilicon-II/polyoxide interface. This might be the reason that the RTO₂ sample had the worse J - E characteristic than that of the as-deposited TEOS sample in the negative bias. Hence, in one word, owing to the smoothest interface and an additional oxide grown during annealing, the TEOS oxide deposited on the disilane polysilicon film with RTN₂O annealing is the best way to fabricate low leakage current polyoxides in our experiment.

Fig. 5 shows electron trapping characteristics of both polarities for the six samples under the 1 mA/cm² constant current stress. The electron trapping characteristics under stress are



(a)



(b)

Fig. 6. Weibull plots of Q_{bd} of the D-TH-O₂, D-TH-N₂O, D-T-RT-N₂O, S-T-RT-N₂O, D-T-RT-O₂, and D-T six samples with (a) poly-II positive biased and (b) poly-II negative biased where the oxide thickness of the D-TH-O₂, D-TH-N₂O, and D-T is 130 Å and 150 Å of D-T-RT-N₂O, D-T-RT-O₂, and S-T-RT-N₂O, respectively.

determined by a combination of factors such as the injecting surface roughness, the intrinsic quality of the oxide, and the nitrogen incorporation at the interface. In general, a rougher injecting surface leads to a nonuniform electric field distribution, consequently, nonuniform injected electron distribution both at the injection interface and in the bulk of the polyoxide, leading to a higher trapping rate. In the figure, for the positive stress, where the poly-I was the injecting surface, the S-T-RT-N₂O sample had the highest trapping rate, reflecting its interface roughness. While the D-T-RT-N₂O sample had the lowest trapping rate, which is consistent with the fact that it had the smoothest interface. In addition, for the D-T-RT-N₂O sample,

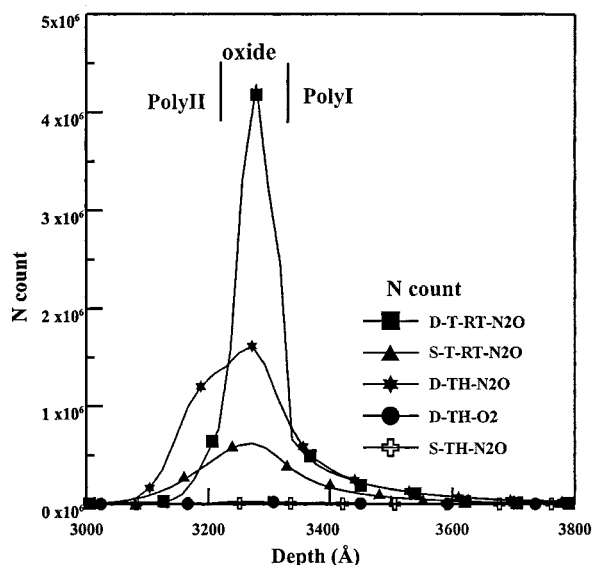


Fig. 7. SIMS profiles of Nitrogen count of the D-TH-O₂, D-TH-N₂O, S-T-RT-N₂O, D-T-RT-N₂O, and S-T-RT-N₂O samples, respectively.

it had asymmetric trapping characteristics, and for the positive polarity stress, it had a net hole trapping characteristic at the beginning stage of the stress. This might be due to the more nitrogen incorporation within the oxide since similar phenomenon were observed for the reported nitrogen-rich oxides prepared by N₂O or NO oxidation [11], [12]. Comparing the positive stress electron trapping characteristics of the variously prepared oxide samples on the disilane poly-I, the N₂O treated samples had smaller electron trapping rates than those of samples without the N₂O treatment. This indicates that N incorporation suppresses electron trapping [1]. For the negative stress electron trapping characteristics, however, since they are less relevant to the interface roughness of the polyoxide/poly-I, the N incorporation and the oxide intrinsic quality are the dominant factors in determining the characteristics. In the figure, it is still seen that the D-T-RT-N₂O sample had the smallest electron trapping rate, which reflected fact that it had the smoothest polysilicon-II/oxide injecting interface and the best oxide quality.

Fig. 6 shows Weibull plots of Q_{bds} of our experimental samples in both polarities, respectively. Also, it is seen that the D-T-RT-N₂O sample had the best Q_{bd} distributions for both polarities. Also, the D-T-RT-O₂ sample had the next better Q_{bd} distributions and the D-T was the third one. That is, for the Q_{bd} distribution consideration, the TEOS deposited on disilane film is a superior process then the thermal or the silane poly-I processes. Furthermore, among all the distributions, the D-T-RT-N₂O sample had the most symmetrical distributions. This is also believed to be due to the fact that the D-T-RT-N₂O had symmetrically smooth surfaces at both poly-I and poly-II interfaces. In addition, the N incorporation might also contribute to the result.

Fig. 7 shows the SIMS nitrogen profiles of D-T-RT-N₂O, S-T-RT-N₂O, D-TH-O₂, D-TH-N₂O, and S-TH-N₂O samples, respectively. The profiles show that the D-T-RT-N₂O samples had the highest nitrogen distribution among all the samples.

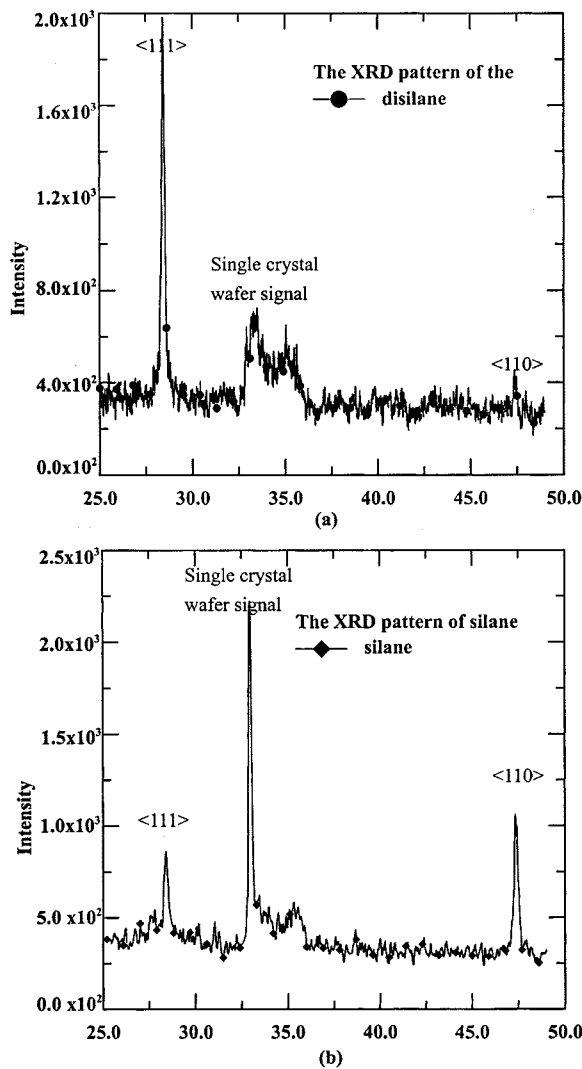


Fig. 8. XRD profiles of (a) disilane and (b) silane samples.

It is believed that it was this higher concentration of nitrogen incorporating reduces the dangling or strained bonds, consequently traps, in oxides. With fewer trappings, also the stronger nitrogen silicon bonds, the oxide had a lower leakage, a lower electron trapping rate, higher E_{bd} and Q_{bd} . In the figure, it is seen that the RT- N_2O samples had higher nitrogen contents than those of the thermal N_2O samples. Also, the disilane samples with the N_2O treatment had higher nitrogen content than those of the silane samples. That is, the nitrogen content in the D-T-RT- N_2O and the D-TH- N_2O samples were higher than that of the S-T-RT- N_2O sample. Fig. 8 shows the XRD (X-ray diffraction) patterns for (a) disilane polysilicon film and (b) silane polysilicon film, respectively. The patterns show that the disilane film had a strong peak at $\langle 111 \rangle$ orientation but weak at $\langle 110 \rangle$ orientations while the silane film had the strong $\langle 110 \rangle$ orientations but relatively weak $\langle 111 \rangle$ orientation. This means that the nitrogen incorporation was heavily dependent on the microstructure of poly-I film. This microstructure variations for polysilicons prepared by different methods had been discussed in [13] and a surface energy dependence model, that is, the

microstructure difference will affect the surface energy of the polysilicon, was also proposed. From the SIMS and XRD data, it may be deduced that the disilane polysilicon surface may have a lower barrier in forming Si-N bonds. In our experiment, we also found that the resistivity of the disilane poly-I film was two times lower than that of the silane film. It may also due to this microstructure difference between these two polysilicon films.

In conclusion, with the smoothest interface and the highest nitrogen incorporation, the D-T-RT- N_2O had the smallest electron trapping rate characteristics and also the largest Q_{bd} distribution.

IV. CONCLUSION

In this work, the TEOS oxide deposited on the disilane polysilicon and then RT annealed in N_2O had been investigated along with other oxides such as those deposited on the silane polysilicon or directly thermally oxidized on polysilicon films. It was found that, due to the smoother surface of the disilane poly-I polysilicon and the higher nitrogen incorporation during the RT N_2O annealing process, the TEOS oxide had symmetrical positive and negative $J-E$ characteristics, a much lower electron trapping rate, and a high Q_{bd} than all other oxides. The oxide is very suitable for the application as the inter-polyoxide for EEPROM.

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