國立交通大學

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博士論文

全域組成氮化銦鎵薄膜之 磊晶成長與光學特性分析

Growths and Optical Properties of Entire Composition of In_xGa_{1-x}N Epilayers

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中華民國九十九年七月

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摘要

在本論文中,我們利用在有機金屬化學氣相磊晶 (metalorganic chemical vapor deposition, MOCVD)系統上另行安裝一種氨預熱的裝置以增加熱分解效率,藉由改變成長溫度與改變三甲基銦 (Trimethylindium, TMIn)之莫耳流量, 成功地成長全域組成氮化銦鎵(In,Ga1,N)薄膜。首先,薄膜成長溫度由 750 降低至 650 °C,固相銦組成可由 14 %增加至 40 %,由光激螢光光譜顯示,我們成功地調變氮化銦鎵薄膜發光波長,並深入深紅光 738 奈米(nm)的波段,其譜線寬度為 180 meV。另外,當氣相比增加至 73 %時能將固相銦組成提高至 44 %,其發光波長可進一步延伸至 950 nm,譜線寬度為 235 meV。我們認為以另行安裝 一種五族預熱裝置以提高氣熱分解效率的方式,已能突破目前以 MOCVD 成長氮化錮鎵薄膜無法達到全域組成皆能發光的瓶頸,特別是 650-1100 奈米波段, 遠超過目前商業 MOCVD 的極限。

此外,透過光激發光光譜分析結果顯示,高溫成長的樣品(>700°C)會出現 高能量與低能量兩個發光譜峰,隨著成長溫度從750-700°C,高能量從2.94至 2.58 eV,低能量從2.44至2.07 eV,高低發光譜峰能量差約500 meV。由倒置空 間圖譜(reciprocal space map, RSM)與陰極螢激發螢光譜(cathodoluminescence spectroscopy, CL)顯示高低發光譜峰分別來自薄膜下、上應力層(strained)與鬆弛 層(relaxed)所致。700°C以下成長之樣品,由於銦組成較高與底層氮化鎵緩衝層 不匹配程度亦較大,所以並無應力層存在,故發光光譜只存在一譜峰。

為了進一步獲得更長波長之氮化銦鎵薄膜,我們並在 650 °C 固定成長溫度

下,改變三甲基銦 (Trimethylindium,TMIn)之莫耳流量,探討固、氣相 In/III 族比值相互關係。實驗結果顯示,過高的氣相 In/III 族比會產生大量金屬銦顆粒 析出於表面,直徑約 3-5 微米不等,同時不利於高銦組成氮化銦鎵薄膜之成長。 此外,透過低溫 14-K 光激發光光譜分析結果顯示,氮化銦鎵薄膜發光峰值可由 2.75 eV 至 1.29 eV,所涵蓋波長由藍光,綠光,紅光至近紅外波段。在這些樣品 中,發光之光子能量相對於吸收能隙其能量差(稱史塔克位移,Stokes shift),由 120 meV 增加至 570 meV 隨著銦組成的增加由 16 至 44 %,我們認為較大的史 塔克位移的現象歸因於銦含量分佈不均勻所造成。



Growths and Optical Properties of Entire Composition of In_xGa_{1-x}N Epilayers

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ABSTRACT

In this dissertation, the use of preheating ammonia installation can be used to increase the thermal decomposition efficiency of ammonia, we have demonstrated that the entire composition of $In_xGa_{1,x}N$ epilayers prepared by MOCVD can be achieved merely by varying the growth temperature and In vapor mole fraction. First, the In solid composition, as anticipated, was increased from 0.14 to 0.40 as the growth temperature decreased from 750 to 650 °C, which corresponds to a wavelength can be extends to the 738 nm in the deep red region with linewidth 180 meV. Further, as input In vapor mole fraction raise to 73 %, we could estimate that the In solid composition of the $In_xG_{1-x}N$ films increases to 0.44, which corresponds to a wavelength range extend to near infrared 950 nm with linewidth 235 meV. Therefore, we have surmounted the technique bottleneck of the absence of entire composition of InGaN epilayers, especially the emission wavelength 650-1100 nm.

For samples grown at temperature > 700 $^{\circ}$ C, separated by about 500 meV, two emission peaks are observable. The corresponding emission peaks, namely high and low emission peaks, are shift from 2.94 to 2.58 eV and from 2.44 to 2.07 as the growth temperature decrease from 750 to 700 $^{\circ}$ C. The high peak energy originates

from strained layer closer to GaN buffer and low energy from relaxed layer near the surface, as revealed by the results of high-resolution x-ray reciprocal space mapping (RSM) and cathodoluminescence (CL) measurements. For samples grown at temperature < 700 °C, high In content epilayers having a large lattice mismatch with under GaN buffer layer, the nearly absence of strained layer resulting in the feature of single emission peaks in both PL and CL spectra. In order to further extend the wavelength of the InGaN epilayer, we grew the InGaN sample at a growth temperature of 650 °C, in an attempt to investigate the dependence of InGaN solid composition on input In reactant flow rate. For the In solid composition, we thought that too high the TMIn flow rate will lead to decrease of In concentration solid, unfavorable to the high In content InGaN growth. Besides, we also reveals that the 14-K photoluminescence peak energy of InGaN epilayers exhibit a wide emission tunability from 2.75 to 1.29 eV, covering a wavelength ranges from blue, green, red and even reaches infrared 960 nm spectrum region. It shown that the bowing parameter of $b \sim 2.3$ eV for our results and the literature data for the band gap of $In_xGa_{1-x}N$ over the entire composition. Our observation provides conclusive evidence that the $In_xGa_{1-x}N$ epilayers exhibits a larger Stokes shift, showing that the alloy's inhomogeneity.

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Chapter 1 Introduction

In last decades, the wide-bandgap nitride semiconductors including indium nitride (InN), gallium nitride (GaN), aluminum nitride (AlN) and their alloys have played animportant role in optoelectronics devices. Due to their strong chemical bounds, high thermal conductivity and mechanical characteristic of III-V nitrides, they are superior materials for high temperature and high power devices [1-3]. Recently, InN band gap of 0.65 eV [4] has been reported by A. A. Klochikhin et. al., rather than 1.9 eV value accepted in the earlier days [5-7]. Therefore, their quaternary AlInGaN alloys exhibit a quite wide bandgap tunability from 0.65 eV for InN to 6.3 eV for AlN, covering a wavelength range from infrared, entire visible, to ultraviolet spectral regions. Among them, , the In-rich InGaN alloys are of potential importance for applications in many fields including high-brightness blue, green [8], and red light emitting diodes (LEDs) [9], full-spectrum multi-junction solar cells [10] and phosphor-free solid state lighting (SSL) [11].

Most of LED applications based on wide-bandgap InGaN semiconductors are related to single-color light applications, such as traffic signals, indicator light and signage display, the continual development of LEDs has advanced the technology to a stage beginning to replace the traditional light sources due to its longer lifetime, low consuming power and maintains cost. As far as the light illumination is concerned, the traditional light sources including incandescent, fluorescent (includes compact fluorescent and linear fluorescent) and high intensity discharge light sources have dominated the last century of lighting. For incandescent light bulb, it consumes more than 95 % of the supplied energy as heat, leaving only a small fraction (~ 5 %) of the input energy converting into visible light. Consequently, the incandescent light bulb exhibits as quite low luminous efficacy, lying in the range from 3 to 20 lm/W [12]. Recent efforts have demonstrated that the luminous efficacy of incandescent source can be further advance to a value of 80 lm/W [13]. On the other hand, the fluorescent lamp, having the efficacy from 25 to 111 lm/W [12], is typically more efficient than incandescent lamp. Linear and compact fluorescent lamps through multi-photon phosphors can raise the efficacy to 200 lm/W [13]. Among all traditional light sources, high intensity discharge lamps (which include mercury vapor, high-pressure sodium lamps and metal halide) are currently the most efficacious commercial lamps, with efficacies range from 25 to 150 lm/W [12].

The history of LED development can be traced back to 1950s, in which the first semiconductor light emitting diode was demonstrated using gallium arsenide (GaAs) materials. In 1962, General Electric's Advanced Semiconductor Laboratory invented the first visible spectrum LEDs. The red GaAsP-based LEDs, Green GaP-based LEDs were subsequently commercialized in the late of 1960s. However, the blue LEDs based on gallium nitride (GaN) material was not emerged in the market until 1990s, owing to the difficulties in preparing high-quality InGaN film. At present, the white light sources made from LEDs are primarily utilizing either blue or UV LEDs to stimulate fluorescent or phosphorescent to create white light, with efficacies superior to fluorescent and HID lighting sources [14].

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Solid state lighting (SSL) is referred to lighting that uses semiconductor LEDs, organic light emitting diodes (OLEDs), or polymer light emitting diodes (PLEDs) as illumination sources. The SSL allow the energy directly convert into light via the manipulation of electrons in a solid, as opposed to incandescent bulbs use thermal radiation or fluorescent tubes. Figure 1-1 shows the historical and predicted efficacy for different light sources [15].



Fig. 1-1 Historical and Predicted Efficacy of Light Sources [15]

Although the efficiency of SSL based on LEDs is presently an emerging technology with potential to greatly exceed that of traditional light sources, and have already replaced traditional lamps in many applications, the greatest impact of SSL will likely be in general illumination applications, which necessitate high energy conversion efficiency as well as high luminous efficiency of white light sources.

Fig. 1-2(a) shows a curve of the luminous efficacy of the human eye response. Fig 1-2(b) shows a plot of the spectral power distributions (SPDs) for three light sources including: a fluorescent lamp, an incandescent lamp and a RYGB SSL white light source. The human eye is sensitive to light with a wavelength between 400 to 700 nm, with a sensitivity peak at 555 nm. However, the radiation spectrum from the incandescent lamp is very broad, moreover, the filaments mitted in the infrared range, in such light bulbs convert more than 95 % of the supplied energy into heat, and with only a small fraction (~ 5 %) is converted to visible light. As shown at the bottom of Fig. 1-2, the spectrum of the fluorescent lamp is made up of blue, green and red illumination. Compare with incandescent lamp the fluorescent lamp achieved relatively high luminous efficacy. Nevertheless, the luminous efficacy of the 100 %-efficient four-color SSL (417 lm/W) is about 4 times greater than that of the fluorescent lamp and about 30 times greater than that of the incandescent lamp.

To achieve such SSL, significant challenges on LED technology must be overcome. One of the challenges for SSL is that high optical efficiency must be achieved across the visible spectrum to match the white light sources. When the light is generated from inorganic LEDs, much higher brightness than indicator optoelectronics devices is needed.



Fig. 1-2 (Top figure) A plot of the luminous efficacy for the human eye response as a function of wavelength. (Bottom figure) A plot of the spectral power distributions (SPDs) for three sources including: a fluorescent lamp, an incandescent lamp and a RYGB SSL white light source.

As shown in Fig. 1-3 the state-of-the-art external quantum efficiency (EQE) for LEDs measured at I_f = 350 mA is shown as a function of various wavelength [16]. The InGaN based high-power LEDs exhibit high EQEs in the wavelength between violet and blue regions, but rather poor EQE at longer wavelengths. At longer wavelength devices, achieved by increasing indium concentration in the InGaN active region, degraded InGaN material quality is resulted. This is attributed primarily to either the increased strain between GaN and InGaN layer or lower growth temperature associated with the MOCVD growth.

In regard with red LEDs, the current most efficient materials are $(Al_xGa_{1-x})_yIn_{1-y}P$ - based alloys. Its direct bandgap is available from x = 0 (InGaP), at ~ 1.9 eV (~ 650 nm) to 0.53, at ~ 2.2 eV (~ 560 nm, green-yellow), beyond which the $(Al_xGa_{1-x})_yIn_{1-y}P$ becomes an indirect bandgap semiconductor, so that limiting practical radiative wavelength to green-yellow ranges [17].



Fig. 1-3 External quantum efficiency vs. peak wavelength for high-power LEDs based on InGaN LEDs and III-phosphide (InGaAlP light emitting layer) material systems [16].

Although remarkable progress has been made in InGaN material growth, many efforts of InGaN study are focused on Ga-rich alloys, which are used as active layers in blue and green light emitting diodes. Relatively little attention has been paid to the growth of In-rich InGaN alloys in spite of its potential importance in the fabrication of LEDs in the range from yellow to red wavelengths.

The difficulties in growing high In-content InGaN arise not only from the high vapor pressure of N₂ over InN [18], high In desorption rate, but also from the low solubility of InN in GaN solid miscibility gap, which usually results in the formation of phase separation in InGaN to deteriorate the film quality. I. The situation is getting even worse when MOCVD growth technique is employed for sample preparation since ammonia (NH₃) is commonly used as N source precursor. Furthermore, in order to reduce the volatility of indium, the growth temperature of In-rich InGaN epilayers is limited to the 550-650 °C. Unfortunately, this lower growth temperature results in poor crystalline quality due to the lower decomposition rate of NH₃. In addition to the effect of the lower growth temperature, we think that the low decomposition rate of NH₃ resulting in the short supply of the active nitrogen atoms is one main reason resulting in the poor quality of the MOCVD InGaN film grown at the low temperature. Although InGaN films seem to contain entire solid indium compositions, MOCVD-grown In-rich InGaN alloys still remains a matter of subject, especially for emission wavelengths in the range of 650-1100 nm, where no luminescent spectrum has yet been reported in the literature. In this dissertation, we will cover research in the fields of InN, Ga-rich and In-rich InGaN epilayers, and the incorporation behaviors of In on various MOCVD-growth parameters.

This dissertation is divided into six main chapters. A brief introduction is present in this chapter. In chapter 2, we give detailed descriptions for our experiment set up, including MOCVD system and the characterization equipments, such as the x-ray diffraction, photoluminescence and atomic force microscopy measurement systems. The epitaxial growths of InGaN films as a function of growth temperature, together with the discussion of their optical propertied investigated by photoluminescence, x-ray reciprocal mapping, and absorption and photoluminescence excitation were described in chapter 4. In chapter 5, we discussed the incorporation behaviors of In on InGaN epilayers grown at 650 °C under various In flow rates. The anomalous temperature-dependent photoluminescence spectra were fitted well using thermal transfer model. Finally a general conclusion is presented in chapter 6.



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Chapter 2 – Experimental Details

This chapter briefly describes the experimental setup of deposition system and characterization tools such as x-ray diffraction, photoluminescence, and atomic force microscopy employed for studying material crystalline quality, surface morphology, and optical properties.

Owing to the different growth parameters required for depositions of InN and InGaN materials, two metalorganic chemical vapor deposition (MOCVD) systems were employed in this study. A home-made reactor system was utilized to grow InGaN epilayers, while a commercial system, Aixtron 200 RF-S horizontal reactor, was used to grow InN nano-dot samples.

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2.1 Metalorganic chemical vapor deposition system

The home-built MOCVD system, equipped with a horizontal-reactor, mainly consists of six parts: a computer control system, a gas handling and gas-mixing system, a reactor chamber, a vacuum and exhaust system and the heating system. The corresponding schematic diagram of gas handling and gas-mixing system and the reactor design are shown in Fig. 2-1 and Fig. 2-2, respectively.

In this system, there are four metal-organic sources lines: trimethylgallium (TMGa), trimethylindium (TMIn), trimethylallium (TMAl) and bicyclopntadinarylmagnesium (CP_2Mg) used for as the precursors of Ga, In, Al and

p-dopant, and three hydride lines, high-flow NH₃, low-flow NH₃ and SiH₄, in order to match the growth requirements needed in growing different In-content InGaN alloys. The MO sources are held in stainless steel bubblers immersed in thermostated baths with accurately controlled total vapor pressures to ensure steady supplies of all the source precursors. High-purity ammonia (NH₃, 5N5, Air Porducts & Chem. Inc.) was used as the N sources. The detailed parameters of source lines, such as, saturation vapor pressure, bubbler temperature, maximum flow rate of mass-flow controller are illustrated in Table 2-1. The switches and flow rate of each source precursor are controlled by main computer program via pneumatic valves and mass flow controllers (Bronkhorst-MFC) through PCI-1739U and PCI-1620A cards.



	Source	Bubbling Temperature (°C)	Mass Flow Controller (sccm)	Vapor Pressure (mmHg)	Diluted Conc. In H ₂
	TMGa	5	2 20	88.59	
MO	TMAl	17	50	7.35	
source	TMIn	17	500	1.35	
	Cp ₂ Mg	17	500	0.02	
			200		
Hydride source	NH ₃		5000		
			10000		
	SiH_4		25		2%

Table 2-1 The source line details of MOCVD system for III-nitride growth

For MOCVD nitride growth, the susceptor, heated either by resistance heater or RF coil, provides not only a temperature needed for film deposition, but also the thermal energy to crack source precursors and generate reactive radicals to conduct chemical reaction on the growing surface. Since only a few percentage of NH3 can be decomposed at InGaN growth temperatures, approximately 600-800oC,in order to increase decomposition efficiency of NH₃, many methods have been attempted, including thermally pre-cracked ion-supplied method [1] and laser-assisted deposition[2-3].

For the former, the pre-heating device was commonly connected directly with NH₃ gas tube right in front of reactor; nevertheless, even though the NH₃ gases are totally decomposed, the active nitrogen radicals still have to travel at least a distance of entrance length in the reactor before taking reactions with other reactants on the heated substrate surface. This weakens the activity of nitrogen radicals. Consequently, although the thermal precracking method can enhance the ammonia cracking to a certain extent, it does not bring a significant improvement of film quality for nitride deposition, in particular InGaN growth.

The special feature regarding our home-build MOCVD system is equipped with two heating units, as shown in Fig. 2-2. One is the lower graphite susceptor, on which the substrate is placed and thermally controlled at a growth temperature to perform the deposition of various epitaxial layers; the other is the upper graphite plate, disposed face-to-face to the lower one in the reactor, thermostated normally at a higher temperature for the purpose of pyrolyzing the hardly decomposed source reactants, such as NH₃.

Both heating plates are made of graphite, coated by high-purity silicon carbon (SiC) (POCO Inc.) to prevent or minimize any etching effects that could be brought about during the deposition. The upper graphite is inductively heated by a 30 kW, 30-80 kHz radio frequency (RF) generator; whereas the lower graphite is heated by a 50 kW, 30-100 kHz radio frequency (RF) generator. Both of graphite temperatures were monitored by R type thermocouples (Platinum/Rhodium alloy) surrounded by quartz sheaths, which were feedback controlled by Eurotherm 818P15 controllers.

High-purity hydrogen, generated by a palladium hydrogen purifier (Johnson Matthey HP-B-8549), was used as the main carrier gas. For InGaN deposition, nitrogen purified by a getter purifier (Japan Pionics Co., LTD) was instead used as carrier gas in order to suppress reevaporation of nitrogen atoms from the growing surface. Separated lines were designed in the gas-handling system so that the growth can be carried out in a complete H_2 or N_2 ambiant.

2.2 Characterization Analysis

2.2.1 High Resolution X-Ray Diffraction (HRXRD) System

X-ray diffraction is a non-destructive method of analyzing the structural properties of a material. Double crystal x-ray diffraction is particularly adequate for the examination of materials with high crystalline quality, such as single crystal epitaxial films. Double crystal diffraction is high-resolution x-ray diffraction (HRXRD) technique, featured with a resolution as narrow as few arcseconds. Because of high-resolution capability, many fine multi-layered structures, even with layer thickness of 50 Å, can be analyzed with monolayer accuracy at interfaces. Furthermore, by taking experimental data into the simulation progarm, detailed information regarding the film structure, such as layer thickness and composition as well as degree of strain or relaxation can be well retrieved.

Since the wavelength of X-ray is comparable to the interatomic distances of crystal, when x-ray is incident into crystal, interference pattern is produced. Constructive interference occurs when the optical path of the x-ray scattered from neighboring planes of atoms is an integral number of wavelengths : $n\lambda$, where *n* is an integer. Under such a circumstance, i.e., Bragg diffraction condition, the diffracted intensity is maximum, as illustrated in Fig. 2-3 (a), (b). The equation is given below:

$2d\sin\theta = n\lambda$,

Where λ is the wavelength of the incident x-ray, a copper (Cu) with the wavelength of

0.154051 nm (Cu K_{α 1}), was used as the target of the x-ray source, *d* is the spacing of adjacent parallel planes of atoms, θ is the Bragg angle for diffraction, *n* is an integer, representing the order number of the diffraction.

2.2.2 Photoluminescence (PL) and Photoluminescence Excitation (PLE) systems

Photoluminescence spectroscopy is a fundamental tool to characterize the optical properties of materials, in which light is illuminated directly onto a sample, being absorbed to excite electrons from ground states to higher excited states. This process is called "photo-excitation". When the excess energy is dissipated through the emission of light, such a luminescence process, is called photoluminescence. The **1896** intensity and spectral content of this photoluminescence is a direct measure of various material properties, such as GaN. In addition to radiative luminescence, there is alternative way to release the excess energy, which is referred to as nonradiative processes.

Fig. 2-4 illustrates several basic types of optical transitions in a semiconductor. [4] These transitions are classified as follows: The first one (1) is the interband transition, including (a) intrinsic emission correlated light emission with energy closed to bandgap, where excitons may play a dominant role under a low temperature environment, and (b) higher-energy emission involving energetic or hot carriers, sometimes related to avalanche emission. The second classification (2) is the transition involving impurities or point defects; namely (a) conduction band to acceptor, (b) donor to valence band, (c) donor to acceptor (D-A pair emission), and (d) deep levels. The third classification (3) is the intraband transition involving hot carriers, also called deceleration emission. Not all transitions can occur during the photoluminescence measurement, and not all transitions are types of radiative processes. For an efficient luminescent material, its radiative transitions have to predominate over non-radiative ones (such as the Auger nonradiative recombination) [5].

In our laboratory, a He-Cd laser (Kimmon IK5552R-F) operating at 325 nm line (3.815 eV) was used as the excitation source for the photoluminescence measurement. The PL measurement system is shown in Fig. 2-5. The schematic diagram of PL and PLE detection systems includes the reflection mirrors, focusing and collecting lenses, 1896 sample holder and cooling system. The samples were mounted on the flat surface of copper holder cryostated at a temperature near 12 K. The typical excitation power density is of ~ 900 W/cm², with a spot size of ~ 100 μ m, as estimated from reflecting microscope objective with the magnification of 25. We used a single-grating monochromator (ARC Spectro Pro-500) equipped with an either photomultiplier tube (Hamamatsu R955 PMT) having high efficiency in the UV range or photon counter (Hamamatsu C1230) for PL detection. The normal applied voltage of PMT is 1000 V as determined from pulse height distribution (PHD) analysis. According to the reciprocal linear dispersion (1.67 nm/mm with 1200 grooves/mm grating), the entrance and exit slits are both opened to about 50 µm, the resolution is about 0.0835

nm at 325 nm. Moreover, because of the wide PL spectral range needed for III-nitride materials, an irradiance standard (Model 200C Optronic Laboratories) method was used to calibrate the overall spectral response (including collection lens, spectrometer and detector). The PL scan was performed by exposing the sample under excitation for about 0.1 to 10 second at a step of 0.1 to 1nm, and the data were transmitted through an 8255 I/O card and recorded by a computer.

The PLE detection system (PTI model 101) is very similar to that of PL detection system except the light source, where d a 150 W Xenon lamp (A-1010B arc lamp housing with LPS-220B lamp power supply) is used as a light source. The light is dispersed by a monochromator with SID-101 controller. The light entrance and exit slits were both opened to about 500 µm. The wavelength resolution is from 2 to 4 nm for PLE detection.

2.2.3 Atomic force microscopy (AFM)

The atomic force microscope (AFM), invented by Gerd Binning and Calvin F. Quate consisted of a flexible, elastic cantilever and sharp probe tip, is a powerful tool that is capable of measuring interactive between atoms of the tip and sample surface by using scanning probe. Fig. 2-6 shows the configuration of AFM used in this study. The sharp tip is attached to a flexible cantilever, controlled by AC and DC voltages. When the AC signal was set at the resonance frequency, the cantilever will have maximum amplitude. The scanner controls movements of the cantilever along the

sample surface (X-Y plane) and (Z-axis), respectively. As the distance between the tip and sample surface is changed, the corresponding attractive and repulsive forces were also changed, which in turn bring a change of resonance amplitude and deflection of the cantilever. The degree of deflection is then sent to the control system and is used as the AFM signal. During the measurement, the microprobe is feedback controlled by a by piezo-electric crystal to bring the parameters back to its original values. From the deflection signals, we could analyze the sample surface morphology. There are three operation modes that can be used to scan surface morphology. In contact mode, the tip touches the surface while scanning over the sample. Although high resolution can be obtained in this way, the deformation of tips or samples often happens. Usually, non-contact is preferred to avoid the probe deformation since it utilizes the long rang Van der Waal's force between the tip and sample. However, the sensibility and resolution is getting poorer because of the interference. For the surface potential microscopy it is necessary to use conductive cantilevers with a relatively small resonance frequency for operation in non-contact mode. An alternative is tapping mode. Tapping mode is a combination of contact mode and non-contact mode. Detection of tapping mode is more sensitive than that of non-contact mode and less destructive on probes than that in contact mode.

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Reactor

Susceptor



Fig. 2-1 The schematic diagram of reactor system



Fig. 2-2 The schematic diagram of piping equipment of MOCVD system



Fig. 2-3 Schematic diagram showing Bragg diffraction from planes of atoms in a crystal (b) shows part of (a) in detail.


Fig. 2-4 illustrates several basic types transitions in a semiconductor



Fig. 2-5 Photoluminescence and Photoluminescence excitation detection system block diagram.



Fig. 2-6 Schematic diagram of AFM system

Chapter 3 - Growth and Characterization of InN nanodots

Recently, indium nitride has received considerable attention because of its superior intrinsic properties, such as narrow direct band gap, low effective mass, high electron mobility, and large drift velocity. [1-2] In addition, the use of InN and its alloys with GaN and AlN can extend its emission from the UV to the near-infrared wavelength, making it very suitable for the fabrication of light-emitting devices emitting at different wavelengths. Nevertheless, compared with other nitride semiconductors, InN binary epilayer remains one of the least studied materials owing to the low dissociation temperature, high saturation vapor pressure of nitrogen, and high In escaping rate associated with this type of material [3-4]. Even fewer efforts have been made on the study of its nano-scale physical structures. The first emission properties of InN dots were reported in 2005 in a study by Ruffenach et al. involving metalorganic vapor phase epitaxy (MOVPE), [5] in which the photoluminescence (PL) peak energy of InN dots encapsulated by SiO2 was found to be almost invariant to the measured temperature. The size tunable emission properties of InN dots, indicative of quantum size effect, have recently been realized by Ke et al. in 2006. [6] The associated peak energy was observed to blueshift systematically from 0.78 to 1.07 eV as the average dot height was tuned from 32.4 to 6.5 nm. On the other hand, the optical behaviors of InN nanodots directly exposed to air, i.e., without any capping layer, have not yet been reported. The uncapped surface quantum dots usually exihibit much weaker and broader PL properties than the capped ones, owing to the existence of a high density of trap states on the surface. In this work, we thus devote ourselves

to the study of uncapped InN nanodots by focusing on the dependence of their optical properties on growth temperature. Temperature dependent PL measurement reveals that good optical quality InN nanodots can be achieved in a wide growth temperature range, even at temperatures as high as 725 °C.

3.1 Experimental details of InN nanodots

In this study, the uncapped InN dot samples were grown on 1-um-thick GaN buffer layer/sapphire (0001) substrates by AIX 200 RF-S horizontal-reactor MOCVD system. The InN dots were deposited by flow-rate modulated epitaxy (FME) technique with six growth cycles at temperatures varied from 550 to 725 °C. The gas flow sequence for one growth cycle consists of four steps: 20 s trimethylindium(TMIn) 896 step, 20 s NH₃ step, intervened with a 10 s nitrogen carrier gas purge in between. The TMIn and NH3 flow rates were 150 and 18000 sccm for the In and N steps, respectively. During the TMIn step, a small amount of NH₃ at a flow rate of 500 sccm was also provided to suppress the re-evaporation of In atoms in this step. The TMIn step is also referred to as the growth step because of the deterministic nature of group-III elements during deposition which governs the growth, whereas the NH₃ step is referred to as the annealing step, in which abundant NH₃ is supplied to convert the unreacted In atoms from the previous growth step into an InN form. The other detail grown conditions were described in Table 3-1. The PL measurements were performed using the 488 nm line of an argon-ion laser as an excitation source. The PL signals

analyzed by a 0.5m monochromator were detected by an InGaAs photodiode with a cutoff wavelength of 2.05 mm. The surface structures were examined by an NT-MDT atomic force microscopy (AFM) system, and images were taken using the noncontact tapping mode with a silicon cantilever.

The theory of the periodic FME growth according to this study is shown in Fig. 3-1. First, substrate 1 is provides (Fig. 3-1 (a)), and buffer layer 2 is grown on the substrate (Fig. 3-1 (b)). Purge gas 3 is turned on and the first reactant 4 is modulated to a range below the first molar flow rate (Fig. 3-1 (c)), so that the second reactant 5 turned on thereafter forms metal or metal-rich compound islands 6 on the buffer layer (Fig. 3-1 (d)). After the formation of said island, purge gas is turned on to clean remaining second reactant which does not form islands (Fig. 3-1 (e)). Subsequently, the first reactant 4 is modulated to a range above the second molar flow rate (Fig. 3-1 (f), so that said islands form high quality nanoparticles 7 with excellent structure under sufficient first reactant molar flow rate, and a growth cycle is completed. The geometric shape and size can be controlled through modulating the first and second reactant molar flow rates and the turn-on time.

The time chart of reactant precursor flow sequence is shown in Fig. 3-2. In the first purge step (as shown in Fig. 3-2 (a), purge gas (hydrogen 3) was turned on to clean remaining excess NH₃ reactant, and NH₃ reactant 4 was modulated to the lower first molar flow rate range (2.23×10^{-2} mole/min), so that TMIn reactant turned-on in next step was able to grow metal indium or In-rich islands on GaN buffer layer below the first molar flow rate without the formation of 2D grown InN film. Further, NH₃

reactant was modulated to the lower first molar flow rate to avoid re-evaporation of nitrogen atoms on the surface of GaN buffer layer.

	Time	Temperature	Pressure	NH ₃	TMGa	
	(min)	(°C)	(mbar)	(mole/min)	(mole/min)	
Desorption	10	1150	200	-	-	
Nucleation	4	520	500	$1.79 \mathrm{x10^{-1}}$	2.95x10 ⁻⁵	
Annealing	1	1120	200	8.93x10 ⁻²		
GaN	60	1120	200	1.79x10 ⁻¹	1.18x10 ⁻⁴	
InN dots (In step)	20 (sec)	500-725	200	2.23x10 ⁻²	-	
Purge	10 (sec)	500-725	200	2.23x10 ⁻¹ -8.04x10 ⁻¹	-	
InN dots (NH3 step)	20 (sec)	500-725	200	8.04x10 ⁻¹	-	
Purge	10 (sec)	500-725	200	2.23x10 ⁻¹ -8.04x10 ⁻¹	-	

Table 3-1 The growth conditions of InN dots on GaN



In TMIn reactant turn-on stage (as shown in Fig. 3-2 (b)), TMIn reactant 5 was turned on with molar flow rate setting of 1.20×10^{-5} mole/min for 20 seconds, to form metal indium of In-rich islands on GaN buffer layer. In the second purge stage (as shown in Fig. 3-2 (c)), TMIn reactant was turned off, and purge gas 3 was turned on for 5 seconds to clean remaining TMIn reactant which did not form metal indium or In-rich islands. In NH₃ reactant turn-on stage (as shown in Fig. 3-2 (d)), the molar flow rate of NH₃ reactant 4, which was modulated from below the first molar flow rate to above the second molar flow rate (8.04×10^{-1} mole/min), was turned on for 10 seconds, so that islands formed InN nanoparticles 7 with excellent quality at sufficient NH₃ reactant molar flow rate.

3.2 Growth temperature effect of InN nanodots

There is no doubt that growth temperature is one of the key parameters in determining the epitaxial properties of grown films and their nanostructures. We thus grew InN dots by FME at temperatures from 550 to 725 °C. The resulting AFM micrographs are shown in Fig. 3-3. As can be seen in the figure, truncated hexagonal InN nanodots, faceted by {10-12} or {10-13} planes, spread on the surfaces of all the samples. In droplets, formed either directly on the GaN surface or piled on the top surfaces of InN islands, occur only in samples grown at temperatures \leq 575 °C. The corresponding structural properties, determined from AFM micrographs, and electron concentrations, estimated by analyzing the 20K PL spectra using a line-shaped model,[7] are summarized in Table 3-2. The average height for these InN dots ranges from 19 to 38 nm and their average diameter lies between 180 and 263 nm. Much 896 larger heights were attained for metallic In droplets. Their heights and diameters are 138/311 and 107/386 nm/nm for samples grown at 550 and 575 °C, respectively. The variation in electron concentration seemingly coincides well with the aforementioned growth regions. Electron concentrations as high as $(4.0-4.2) \times 10^{18}$ cm⁻³ are observed for those droplet-containing samples prepared at low temperatures. The electron concentration decreases to $(2.2-2.3) \times 10^{18}$ cm⁻³ in the mid-temperature growth region and further decreases to $\sim 1.7 \times 10^{18}$ cm⁻³ when samples were prepared at 725 °C.

Table 3-2. Average height, diameter, and electron concentration of the InN dots grown at temperatures from 550 to 725 °C. The number in brackets represents the standard deviation of the size distribution in this measurement.

Tg (°C)	550	575	600	625	650	675	700	725		
Electron concentration										
$n_{op} (10^{18} \text{cm}^{-3})$	4.0	4.2	2.2	2.3	2.3	2.2	1.9	1.7		
InN dots										
H (nm)	25(11)	19(9)	30(14)	26(12)	31(11)	32(12)	33(11)	38(11)		
D (nm)	211(50)	180(52)	263(60)	200(65)	233(61)	208(61)	205(56)	196(52)		
In droplets										
H (nm)	138(56)	107(48)								
D (nm)	311(82)	386(126)								

. Miller

Fig. 3-4 shows an Arrhenius plot of InN dot density (solid circles) as a function of reciprocal temperature. For comparison, the In growth efficiency (open circles), **1896** defined as the ratio of growth rate to input In molar flow rate, is also included in the figure. Three distinct regimes can be clearly observed, namely, low-, mid-, and high-temperature grown regimes, separated by dividing temperatures of ~600 and 650 °C. It can be seen that the dot density increases markedly from 7×10^7 to 8×10^8 cm⁻² as the growth temperature increases from 550 to 600 °C. It tends to increase slightly in the temperature range of 600-650 °C, starts to decrease rapidly at ~675 °C, and eventually disappears for temperatures >730 °C. The sharp decrease in dot density at high temperatures is considered to relate to the thermal etching or fast evaporation of adsorbed In atoms on the surface, as manifested by the observation of the onset of declining growth efficiency at these temperatures, also shown in Fig. 3-4.

Additionally, both the AFM micrographs and X-ray diffractions indicate clearly that high densities $5-9 \times 10^7$ cm⁻² of cone-shaped In droplets cover the entire surface for those low-temperature samples. (denoted by half-filled grey circles in Fig. 3-4) As shown in Fig. 3-5, signals of metallic indium (101) and (110) diffraction peaks (32.9 and 31.3°) are clearly observed in these low-temperature samples, in addition to those observed in the X-ray spectra of sapphire (006), 41.6°, GaN (001), 34.5°, and InN (002), 31.3°. The formation of droplets at low growth temperatures indicates an insufficient supply of active nitrogen radicals during deposition, stemming from the poor cracking efficiency of NH3 for temperatures <575 °C. Regarding the high-temperature growth regime (> 650 °C), one might also expect the reappearance of In droplets on the surface as observed in conventional InN MOCVD growth, which has been proposed to be caused by the high desorption rate of N atoms from InN material itself at these temperatures. [8] However, this is not observed in our FME-grown InN samples. Neither In droplets nor their X-ray signals were detected. Our results suggest that the use of an FME scheme in InN dot growth, particularly the NH₃ step, can suppress to a certain extent the decomposition of InN at high growth temperatures and concurrently provide a way for unreacted In adatoms to change into the InN form.

The 20 K PL spectra for the above-described InN nanodots are shown in Fig. 3-6(a). The peak energies and full widths at half maximum (FWHMs) are shown in Fig. 3-6(b). Inferior optical properties were obtained for low-temperature-grown samples (550 and 575 °C), whose peak energies are located at approximately 0.80 eV

with FWHMs as high as ~100 meV. These emission energies are considerably higher than the reported band gap energy of 0.69 eV,[8] indicative of a strong Burstein-Moss effect due to the high electron concentration in these samples. Such a high electron concentration in these low-temperature-grown samples presumably originates from structural defects, such as stacking faults, voids, pits, or point defects, induced by the formation of In droplets on the surface, similar to the cases of GaAs and GaN systems.[9,10]

When the growth temperature is increased to 600 °C, these dots show marked improvement in their optical properties. Not only the peak energy and FWHM are improved to 0.77 eV and 71 meV, respectively, but also the PL intensity is markedly increased, almost by a tenfold increase in magnitude. The fairly good-optical-quality growth region remains not only in the temperatures of 600- 650 °C, but also, beyond 896 our expectations, extends into the high-temperature region, where fast In desorption begins to occur. For those samples grown at temperatures > 650oC, the PL peak energy is shifted slightly toward a lower value, accompanied by a slow linewidth broadening in their PL spectra. Note that even for InN nanodots grown at 725 °C, the upper limit of our growth temperatures, reasonably good optical properties can still be obtained. The corresponding PL peak energy and FWHM are of 0.75 eV and 74 meV, respectively, comparable to typical results, 0.70-0.83 eV and 100-150 meV, for InN bulk films prepared by conventional MOVPE [11-13]. The results for these uncapped InN nanodots are quite encouraging. Although the linewidths (71-74 meV) are still large, they can already compete with those of the uncapped InAs and InGaAs surface quantum dots (54-150 meV) [14-16]. In accord with study of InAs dots, the luminescent properties can be further improved significantly when dots are well capped by a suitable material, this sheds light onto future device applications of InN nanodots.

To gain more insight into the emission properties of these InN dot samples, we subsequently conducted temperature-dependent PL measurements. The results are plotted in Fig. 3-7 and compared with those for a 500-nm-thick InN bulk film grown at 625 °C. For the bulk film, the measured peak energy fits well with the Varshini-type band-gap shrinkage feature, $E(T)=E_0-\alpha T^2/(\beta+T)$ with α =5.71×10⁻⁴ eV/K, β =900 K, and a PL redshift of ~40 meV. With regard to InN nanodots, small blue shifts or slight shifts are observed for those low- and mid-temperature-grown samples. In contrast, clear redshifts of 15-20 meV are observed for samples grown at high temperatures. The corresponding Varshini's parameters are in the ranges of (1.3-2.2) ×10⁻⁴ eV/K and 400-500 K for α and β , respectively.

It has been calculated that the band-gap shrinkage for InN from 0 °C to room temperature due to the effects of lattice dilation and electron-phonon interaction is ~60 meV[17]. Thus, the large deviations of temperature-induced PL shifts from this value strongly suggest that there are blue-shift mechanisms to counterbalance the fundamental band-gap shrinkage in these InN nanodots. The corresponding displacements of peak energy in comparison with the band edge E_0 are approximately 63-68, 60-62, and 40-45 meV for our low-, mid-, and high-temperature-grown InN dot samples, respectively. Such anomalous blue shifts were usually interpreted as the

successive filling of localized states [18] and/or band tail states [19] as well as the Coulomb screening of the piezoelectric electric field [20] by photogenerated or thermally released carriers.

The causes of the temperature-induced blue-shift mechanisms of InN nanodots studied here are still unclear at this stage. Aside from the combined effects of the above-mentioned mechanisms, we believe the surface electron accumulation may be the major cause responsible for the blue-shift in our nano-scale InN dots. This is described as follows. Owing to the high surface state density ($\sim 2.4 \times 10^{13}$ cm⁻²), Cimalla et al. [21] reported that the free electron concentration for InN crystal thinner than 300 nm mainly comes from the surface electrons, rather than the bulk electrons. A recent study by Swartz et al [22] further pointed out that the surface electron concentration tends to increase significantly when the temperature is increased from 25 to 250 K, whereas bulk electron concentration remains almost unchanged in this temperature range. Because the surface-area to volume ratio of dot is much higher than bulk material, much stronger surface electron accumulation is expected to be in InN dot structure. Consequently, as the measured temperature is increased, large amounts of trapped surface electrons of our InN nonodot samples are thermally agitated to the conduction band. This gives rise to an alleviated Fermi energy and hence an observed blue-shifted PL energy in the temperature-dependent PL spectra for these InN dot samples.

Regarding the reduced blue-shift effect observed in the high-temperature grown samples, it can be attributed to the carrier compensation effect due to generation of

large densities of In vacancies arising from the fast evaporation of adsorbed In atoms during InN deposition. These In vacancies act as acceptors [23] in InN compensate the thermally stimulated surface electrons in this type of samples.

3.3 Conclusions of InN nanodots growth

We have investigated the PL properties of uncapped InN nanodots, prepared by flow-rate modulation epitaxy at growth temperatures from 550 to 725 °C. Our results indicate that the presence of In droplets on the surfaces of the low-temperature-grown samples indeed causes the deterioration of the optical properties. As for the high-temperature-grown samples (600-725 °C), where no droplets are formed, reasonably good luminescence results are attained. The corresponding PL peak energy shifts slowly from 0.77 to 0.75 eV with linewidth varying gradually from 71 to 74 meV as the growth temperature increases. Moreover, unlike the small blue shift or absence of variation in the samples grown at temperature < 650 °C, clear redshifts of 15-20 meV from 20 to 280 K were observed in our high-temperature InN dot samples, which are rarely seen in nanoscale InN structures.

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Fig. 3-1 A chart showing the principle of the periodic flow rate modulation epitaxy growth InN dots on GaN buffer layer.



Fig. 3-2 A time chart showing the modulation of reactant molar flow rate in the periodic flow rate modulation epitaxy.



Fig. 3-3 AFM images of InN dots grown at (a) 550, (b) 575, (c) 600, (d) 625, (e) 650, (f) 675, (g) 700, and (h) 725 °C.



Fig. 3-4 Arrhenius plots of InN dot density (solid circles) and growth efficiency (open circles) as a function of growth temperature. The In droplet density (gray cirles) is also shown in the figure.



Fig. 3-5 Double-crystal XRD data ($\theta/2\theta$ scan) of InN dots grown at various temperatures.



Fig. 3-6 (a) 20-K PL spectra of InN nanodot samples grown by FME from 550 to 725 °C. The corresponding variations of peak energy and FWHM against growth temperature are shown in (b).



Fig. 3-7 Temperature dependences of the PL peak energies for our InN nanodot and bulk samples.

Chapter 4 - Growth and Characterization of $In_xGa_{1-x}N$ films Grown at Various Temperatures

The direct-bandgap $In_xGa_{1-x}N$ alloys have been recently investigated intensively because of the superior property of their wide spectral tuning capability, which can be changed continuously from ultraviolet to entire visible region and extended into near infrared region [1-3]. This offers many possibilities for this material in varieties of device applications, including high-brightness blue [4-6], green [7] and red light emitting diodes (LEDs) [8], full-spectrum multi-junction solar cells [9] and phosphor-free solid state lighting [10].

Despite of these peculiar features, up-to-date many efforts of InGaN study have been directed toward the Garrich alloys, which are used as active layers in blue and green light emitting diodes, few attempts have been made on In-rich InGaN (x>0.30). This is mainly due to the difficulties in obtaining high-quality InGaN films, attributed to the low solubility between InN and GaN, low dissociation temperature of InN as well as high equilibrium vapor pressure of nitrogen [11] associated with this material. The situation is getting even worse when metalorganic chemical vapor deposition (MOCVD) growth technique is employed for sample preparation since NH₃ is commonly used as N source precursor. Due to poor cracking efficiency, the pyrolysis of NH₃ is necessitated to proceed at a higher temperature that imposes conflicting temperature requirement of InN dissociation. Consequently, the MOCVD-grown InGaN film, particularly when the In concentration higher than ~25%, often accompanied with the appearances of In segregation [12], phase separation [13-15], compositional fluctuation, In metal droplets and/or their complex point defects, which give rise to a drastic drop in luminescence efficiency. Even though the entire solid composition of $In_xGa_{I-x}N$ epilayers without compositional inhomogeneity, as examined by x-ray data, can be achieved by this method, the realization of high optical-quality, MOCVD-grown In-rich bulk InGaN alloys still remains a matter of subject, especially for emission wavelengths in the range of 635-1100 nm, where no photoluminescence data has yet been reported in the literature.

In this study, we have successfully grown $In_xGa_{1-x}N$ epilayers with clear PL signals by using MOCVD deposition method utilizing using by further investigate (with $0.18 \le x \le 0.4$) grown at various temperatures by MOCVD. X-ray diffraction (XRD), photoluminescence (PL) measurements are employed to study the alloy compositions and PL emission bands of these $In_xGa_{1-x}N$ films. Further, photoluminescence excitation (PLE) measurements clearly indicates that the absorption from InGaN related band edge emissions even for higher In concentrations (>30%).

4.1 Experimental details of $In_xGa_{1-x}N$ films

The In-rich In_xGa_{1-x}N epi-films employed here were grown on ~400 nm-thick GaN/(0001) sapphire substrates using a home-made low pressure metalorganic chemical vapor deposition (MOCVD) system, equipped with a horizontal quartz reactor. Trimethylgallium (TMGa), trimethylindium (TMIn), and ammonia (NH₃) were used as Ga, In and N sources, respectively, and nitrogen was used as carrier gas. In order to enhance the decomposition efficiency, NH₃ was thermally pyrolyzed before conducting the chemical reactions with other reactants so that the growth can be performed at a lower temperature to minimize the effects caused by InN dissociation. The detailed description of NH₃ pre-cracking method and reactor design will be published elsewhere. During the deposition, the flow rates of TMGa, TMIn and NH₃ were kept at 5.89, 8 µmol/min and 4.8 L/min, repectively, while the growth temperature was tentatively varied from 750 to 650 °C with the aim of obtaining different solid composition of epilayers. The resulted layer thickness was in the range of 400 nm. The as-grown epilayers were then examined by x-ray diffractometer (XRD) in θ -2 θ scan mode of (002) reflection peak to estimate the In content in In_xGa_{1-x}N, assuming Vegard's law is valid.

PL measurements were carried out at T = 14 K using the 325-nm line of a He-Cd laser as an excitation source and a photomultiplier tube as a detector. Photoluminescence excitation (PLE) detection system (PTI model 101) is the same as the PL one except the excitation source that is provided by a 150W Xe lamp focused into a 1/4 m spectrometer (SID-101).

4.2 Growth temperature effect of $In_xGa_{1-x}N$ films

Fig. 4-1(a) shows XRD (002) θ -2 θ patterns of the investigated In_xGa_{1-x}N films, scanned from InN to GaN positions (31.3-34.5°). All samples appear to be single diffraction peaks. The lack of side diffraction peaks indicates no large variations in In contents or macroscopic phase separation in these samples. The measured In concentration, as anticipated, was increased from 0.14 to 0.40 as the growth temperature decreased from 750 to 650 °C, owing to the reduced In desorption rate at lower growth temperatures. The full widths at half maximum (FWHM) of x-ray curve is usually used as a figure of merit for crystalline quality. Normally a broader spectrum in the curve is resulted as the In content x closes to 0.5, because the synthesis of crystal has already enter the highly immiscible region in accord with the InGaN phase diagram predicted by thermodynamics, where separate crystals are easily formed to precipitate in the matrix. Nevertheless, we do observe a FWHM as narrow as ~700 arcsec for x=0.40 sample grow at 650 °C, which is compatible with that of low In-content InGaN films, suggesting the high crystalline quality of the layer.

Fig. 4-1(b) shows PL spectra of above $In_xGa_{1-x}N$ films taken at 14K. For samples grown at temperatures ≥ 700 °C, two emission peaks, separated by about 500 meV, are observable. The corresponding emission peaks, namely high and low emission peaks, are shifted from 2.94 to 2.58 eV and from 2.44 to 2.07 eV as the In content is increased from 0.18 to 0.30. More delighted results were attained for those samples grown at lower temperatures, in which regardless a declining in PL intensity is obtained for x=0.38 and 0.40 samples, attributed to the onset of generation of large quantity of non-radiative recombination centers, single but clear emission peaks were observed. Their emission wavelengths lie at 681 (1.82) and 738 nm (1.68 eV), respectively, readily in the range of deep red color area. To our knowledge, these are the maximum attainable emission wavelengths in the visible region ever reported for thick InGaN epilayers grown by MOCVD method. The variation of their linewidths as a function of solid composition is depicted in the inset of Fig. 4-1(b). The linewidths of above two low-temperature samples fitted by two overlapped Gaussian functions are determined to be 210 and 180 meV, respectively.

The presence of double luminescence peaks in our InGaN epilayer can be mainly ascribed to the effects of strain gradient along the growth direction, as revealed by the of high-resolution x-ray reciprocal results space mapping (RSM) and cathodoluminescence (CL) measurements, conducted on 725 (x=0.22) and 675 °C (x=0.38) samples. The RSM by high-resolution x-ray diffraction is using to detail the crystal structure and strain of an epitaxial film. The in-plane a and put-of-plane clattice constants of the hexagonal structure can be measurement, whereas the position of GaN and InGaN diffraction peaks in the reciprocal space can be performed. Fig. 4-2 (a)-(b) show the RSM of the (105) reflections of these two samples. The vertical line indicates growth coherent to GaN, whereas the dashed line represents the relaxation line. The full line connecting the relaxed, InGaN (R) and strained, InGaN **(S)**.

These two samples are thought to be exempla to representing the general optical

behaviors for films grown at temperatures above and below 700 °C, respectively. As shown in Fig. 4-2 (a), the RSM of 725 °C sample shows two distinct peaks, indicative of co-existence of both relaxed and strained sublayers in the deposited film. Depth-resolving cathodoluminescence spectra performed further ascertain the origins of double luminescence peaks. In Fig. 4-2 (c), by varying the electron probe beam energy, it clearly shows that the low energy peak 2.45 eV is originating from the relaxed region near the surface and high energy component 2.82 eV is coming from the deeper, strained region, close to the GaN underlayer. This finding is in good agreement with the results of S. Pereira *et al* [16]. This interprets well the observation of narrower linewidth of high energy peak with respect to that of low energy one in PL spectrum, accredited to the effective suppression of misfit defect formations in the strain layer.

In contrast, as shown in Fig. 4-2 (b) the entire film of 675 °C sample (x=0.38) seems to be fully relaxed as disclosed by RSM. This is not beyond the expectation. For the 450 nm-thick In_{0.38}Ga_{0.62}N on GaN film, the calculated critical layer thickness is so thin (~3 nm) that the growth is subjected to almost an immediate strain relief once InGaN is deposited on GaN, leaving only a small portion of epilayer being kept under elastic strain status, which gives negligible signals in the measurements. The nearly absence of strained layer leads to the light emissions from nowhere but from the relaxed region for these low-temperature films, resulting in the feature of single emission peaks in both PL and CL spectra, are shown in Fig. 4-2 (d).

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To gain more insights on the emission properties of these InGaN films, we

subsequently conducted optical absorption and photoluminescence excitation (PLE) measurements. Fig. 4-3 (a) shows the absorption coefficient squared, measured at 14K, plotted as a function of excitation photon energy. All absorption curves exhibit a characteristic of linear dependence on the photon energy above the onset of absorption, which is a typical for interband absorption in direct semiconductor. It is apparent that the absorption curves of x=0.38 and 0.40 samples are not as sharp as that of InGaN films with smaller In concentrations. We attributed it in part to the underestimation of absorption coefficient due to the scattering loss in transmission measurement, aroused from three-dimensional layer structure and enlarged composition fluctuation in In-rich film

To derive the bandgap of the alloys, a sigmoidal equation is used which includes sub-bandgap absorption states due to defects, impurities and inhomogenious distribution of In atoms and is given by [17-18]

$$\alpha = \alpha_0 / \left[1 + \exp(\frac{E_g - E}{\Delta}) \right]$$

where E_g is the "effective bandgap" and Δ is the bandtail broadening parameter. The resulting bandgap and Δ values from the best fits are displayed in Fig. 4-3 (b) and its inset. For comparison, their PL peak energies are also illustrated in the figure. One can notice that large Stokes shift between absorption edge and PL peak energy is exhibited here. The difference remains nearly invariant and is kept at ~ 500 meV. Nevertheless, the broadening parameter Δ is increased continuously to higher values, from 220 to 470 meV, with the raised InN fraction. These values are considerably higher than that of InN (96 meV) [19] and GaN (~7 meV). This evidences the consequence of structure disorder in affecting the optical transition. It has been point out that for InGaN ternaries the influence of structure disorder under low temperature conditions outweighs over other major broadening parameters, such as carrier-impurity interaction, carrier-phonon interaction in InGaN [19]. With the increasing In content, the concurrent increased structure disorder, caused by an increase in the degree of random cation distribution, size fluctuation and structure defects, would indubitably produce wider spreading of absorption continuum states in these In-rich InGaN films.

Fig. 4-4 shows 14-K PLE spectra of x=0.38 sample monitored at 1.73, 1.86 and 2.00 eV. The PL spectrum excited by HeCd laser emitting at 325 nm is also shown in the figure for reference. Several absorption edges are clearly observable. The positions marked by the arrows correspond to the contributions from InGaN alloys and GaN underlayer. It is found that the InGaN absorption edges for various detection energies are not the same; instead, it tends to decrease with the reduced detection energy. This suggests that the measured PL spectrum is due to the superposition of light emissions from various micro-regions of different In contents. Furthermore, we do notice strong excitonic resonance absorption at the detection energy of 2.00 eV. The enhanced absorption peak indicates the excitons that responsible for the high energy shoulder of PL emission are well localized in regions with size closer to the Bohr radius of exciton, ~3.4 nm, indicative of the existence of quantum nanostructures like quantum dots or quantum disks with fractal dimensions in the film.

Finally, an additional absorption peak, located at 2.75 eV, is exhibited. This is likely due to the absorption involved with nanoscale structure or interface states adjacent to GaN layer because the absorption energy is well above InGaN absorption edge, and independent of detection energy.



4.3 Conclusions of InGaN films growth

In summary, by means of an ammonia precracking method we have demonstrated that the emission wavelength of $In_xGa_{1-x}N$ alloys prepared by MOCVD can be advanced for the first time to very deep red region, up to a value of 738 nm, merely by varying the growth temperature. For high In content sample (*x*=0.38 and 0.40), the resulted FWHMs of θ -2 θ *x*-ray and PL spectra can be as narrow as 700 arcsec and 180 meV, respectively, which can compete the values of InGaN samples with similar In content grown by molecular beam epitaxy. Although macroscopic phase separation is not observed, we do find that the single emission PL band for *x*=0.38 InGaN film is actually an overlap of light emissions from different regions with slight In fluctuation, as unveiled by PLE measurement.



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Fig. 4-1 (a) Double-crystal XRD data ($\theta/2\theta$ scan) of InGaN films grown from 650 to 725 °C, the inset shows the FWHM as function of In content (x_{In}^{s}), and (b) 14-K PL spectra of InGaN films. The corresponding variations of FWHM against xsIn are also shown in the inset figure.



Fig. 4-2 XRD reciprocal space mapping of the (105) reflections of the 725 ^oC-sample (a) and 675 ^oC-sample (b), and the corresponding room temperature CL spectra acquired at different electron beam energies are shown in (c) and (d).



Fig. 4-3 (a) 14-K absorption coefficient squared as a function photon energy. (b) 14-K PL peak energy and band edge determined by optical absorption as a function of x^{s}_{In} . Some data are also shown (Ref. 20 and 21). The dashed curve shows the fit to the band gap energies (absorption) using a bowing parameter *b*=1.43 eV. The solid curve shows the fit to the band gap energies (PL) using a bowing parameter *b*=2.8 eV.



Fig. 4-4 14-K Photoluminescence and photoluminescence excitation spectra of $In_{0.38}Ga_{0.62}N$ films.

Chapter 5 Growth and Characterization of In_xGa_{1-x}N films Grown at Various TMIn Flow Rates

In recently rears, group III-nitride based light-emitting diodes (LEDs) have attracted considerable attention for solid-state lighting and display applications. Ternary InGaN alloys is the most promising material because of its direct band gap a wide and continuous range of wavelengths spanning from ultraviolet to infrared. Though a lot of efforts in the area of blue-green optoelectronic devices based on InGaN have proved fairly abundant in recently years, the growth of InGaN layer with high indium composition remains an extremely challenging, especially for higher concentrations of In in the ternary (>30%). As a result, there exits a "green-yellow gap" in LEDs [1]. The difficulty arises not only from the high vapor pressure of N₂ over InN, but also from the low solubility of InN in GaN solid miscibility gap that usually results in phase separation in InGaN [2-4]. However, due to the high thermal instability of nitrogen over InN [5] and low dissociation rate of ammonia at low growth temperature [6], the growth temperature of InGaN typically at 700-850 °C ranges. Generally, using higher growth temperatures of which approximately above 750 °C, better luminescence properties could be obtained but the content of In in the solid is limited to low values because In starts to evaporate at high temperatures [7-8]. Briefly, either the growth temperature must be lowered to inhibit In desorption or be kept high to ensure the decomposition of ammonia. Recently, some studies to emphasize the entire composition of the $In_rGa_{1-r}N$ grown by molecular beam epitaxy (MBE), which was reported by T. Yamaguchi [9-10]. Yamaguchi et al. have

successfully suppressed the phase separation and reported the optical properties of $In_xGa_{1-x}N$ epilayers of x = 0.16 to 1 were grown at 550 °C [11]. In a previous work, A. Yamamoto *et al.* [12] have reported the single phase $In_xGa_{1-x}N$ epilavers of x = 0.5 to 1 can be grown by MOCVD. In more recent work, H. X. Jiang et al. [13] have also present the results of single phase $In_xGa_{1-x}N$ alloys for the In composition ranging from 25% to 63% synthesized by MOCVD. Although a number of authors reported the growth of $In_rGa_{1-r}N$ epilayers by MOCVD at temperature 700-800 °C, the optical properties of $In_xGa_{1-x}N$ epilavers inside the previously thought phase separation regime especially for higher concentrations of In in the ternary epilayers (>30%) are still uncertain. In this article, we present the results of the growth, optical properties and crystalline quality of $In_xGa_{1-x}N$ epilayers (with $0.16 \le x \le 0.44$) using optical absorption, X-ray diffraction (XRD) and photoluminescence (PL) measurements. We have successfully grown In_xGa_{1-x}N epilayers with luminescence with peak wavelengths up to 960 nm. We also showed that the $In_xGa_{1-x}N$ epilayers exhibit a larger Stokes-shift as a function of *x*.

5.1 Experimental details of $In_xGa_{1-x}N$ films

In this study, the InGaN films of about 400-450 nm thickness were grown on 0.6-um-thick GaN epilayers/buffer layer/sapphire (0001) substrates by home-made horizontal-reactor MOCVD system. Trimethylgallium (TMGa), trimethylindium (TMIn), and high purity ammonia were used as the source precursors for Ga, In, and N, respectively. The 600 nm thick GaN layer were grown at 1130 °C using a hydrogen carrier gas, while the InGaN layer were grown at 650 °C using a nitrogen carrier gas. In order to enhance the decomposition efficiency, NH₃ was thermally pyrolyzed before conducting the chemical reactions with other reactants so that the growth can be performed at a lower temperature to minimize the effects caused by InN dissociation, wherein the temperature of the upper graphite plate is higher than the temperature of the lower graphite susceptor was setting 940 °C.

In an attempt to investigate the dependence of InGaN solid composition on input In reactant flow rate, we grew the InGaN sample at a growth temperature of 650 °C with the NH₃ flow rate kept at 0.48 SLM. The growth details for samples of InGaN were described in Table 5-1. The In solid compositions (x_{In}^s) in InGaN films was estimated from peak value of x-ray diffraction (XRD) curves of GaN (0002) plane in θ -2 θ scan and applying Vegard's law. PL measurements were carried out at T = 14 K using the 325-nm line of a He-Cd laser as an excitation source. The PL signals were analyzed by a 0.5 m monochromator and detected by a photomultiplier tube and an InGaAs photodiode with a cutoff wavelength of 2.05 mm.

Step	Time	Temperature	Upper graphite	Thickness	NH ₃	TMGa	TMIn
	(min)	(°C)	(°C)		(SLM)	(mole/min)	(mole/min)
Desorption	10	1150	1	-	-	24 17	-
Nucleation	4	530	-	30nm	2.7	1.18x10 ⁻⁵	-
Annealing	1	1180	÷		2.7	8	-
GaN layer	60	1120	-	0.6µm	2	2.95x10 ⁻⁵	
InGaN layer	120	650	940	0.4~0.45µm	4.8	5.89x10 ⁻⁶	3.19~22.3x10 ⁻⁶

Table 5-1 The growth conditions of InGaN films.

5.2 Solid composition incorporation of $In_xGa_{1-x}N$ films

In Fig. 5-1 shows XRD (0001) θ -2 θ spectra of the investigated In_xGa_{1-x}N epilayers grown on GaN/sapphire. As can seen, a diffraction peak corresponding to ternary InGaN epilayers was found to decrease gradually from 33.99° to 33.03°, as input In vapor mole fraction (x^{v}_{In}) was increase from 35 to 73 %. Using Vegard's law, we could estimate that the x^{s}_{ln} of the $\ln_{x}G_{1-x}N$ epilayers increases from 0.16 to 0.44 as x^{v}_{ln} is increased from 35 to 73 %. However, XRD of the InGaN epilayers with the highest $x_{ln}^{v} = 88\%$ shows additional peaks, which correspond to pure InN (002) (31.3°) and metallic indium (101) (33.5°), which is further confirmed by SEM measurements (as shown in Fig. 5-2). The inset of Fig. 5-1 shows the full width at half maximum (FWHM) of θ -2 θ spectra as functions of x^{s}_{In} in InGaN epilayers. Exceeding our predictions, extend into the In-rich growth region (where as seen by other authors with higher $x_{ln}^{s} > 28\%$, implies that InGaN epilayers are phase separated [14-15] the FWMH is almost maintains about the same at x^{s}_{In} from 0.16 to 0.40. Note that for the InGaN epilayers with the highest $x_{In}^{v} = 88\%$, the FWHM of InGaN increased rapidly to 3100 arc sec. However, broadening of FWHM with $x_{In}^{s} = 0.44$ is except in inhomogeneity of the solid solution. The results indicate that once the In droplets are formed, the crystal quality of InGaN epilayers deteriorate markedly.

In regard to the dependence of the resultant In solid distribution behavior of InGaN on input TMIn molar flow rates are illustrated in Fig. 5-3. As can be seen in Fig. 5-3 (a), at lower TMIn flow rates, it can be seen that the x_{In}^{s} linear increasing from 16 to 40 % as the TMIn flow rate increasing from 3.19 to 9.97 µmol/min for InGaN under a fixed TMGa flow rate. As TMIn flow rate is further increase, the x_{ln}^{s} is tends to saturate at about 44%, starts to decrease rapidly when the TMIn flow rate is higher than 15.9 µmol/min. As can be seen that under high TMIn flow rate region (> 9.97 µmol/min), the total growth rate of InGaN is nearly constant and doesn't increase (Fig. 5-3 (a)). Note that the In solid composition in InGaN alloys is determined by both incorporation behavior of both In and Ga atoms. To investigate their individual incorporation behaviors in relation to various growth parameters, we introduce a so-called growth efficiency (µm/mol) in this study, which is defined as the ratio of growth rate (µm/min) to the input molar flow rate of individual group III elements (mol/min). Assuming that the overall growth rate of InGaN is a linear combination of GaN and InN growth rates, hence the γ_{GaN} and γ_{InN} employed denote the growth

efficiency of GaN and InN in InGaN, respectively. By employing the concept of γ_{GaN} and γ_{InN} , more detailed results can be obtained. As shown in Fig. 5-3 (b), a increase in γ_{InN} , together with a increase in γ_{GaN} , is observed at low TMIn flow rates, which accounting for the increase of In solid composition at this flow rate region. Concerning the In growth efficiency, a high γ_{InN} , of ~180 µm/mol, can be obtained at low TMIn flow rates. This value reduces to 70~130 µm/mol when the TMIn flow rate is higher than 15.9 μ mol/min, different from increase slightly of γ_{GaN} at this flow rate region. Since at this region, the total growth rate of InGaN is nearly constant (seen Fig. 5-3 (a)), it is certain that the growth rate effect may be not the major factor responsible for the decrease of In growth efficiency. Base on above observation, when the TMIn flow rate exceeds 22.3 µmol/min, the InGaN epilayer is covered by numerous In droplets, we consider that once the droplets were formed, the will allure more In to consolidate into the droplets, deplete the In surfactant available for the growth of InGaN [16]. Therefore, to obtain an InGaN layer with high In solid composition and high quality, we must paid to avoid the formation of In droplets during growth.

Fig. 5-4 shows 14K PL spectra for $In_xGa_{1-x}N$ epilayers containing In solid composition 16, 22, 35, 40 and 44 %, respectively. For $x^s_{In} = 16$ and 22 %-samples, two emission peaks, separated by about 170-180 meV, are observable. The

corresponding emission peaks fitted by two overlapped Gaussian functions, namely high and low emission peaks, are shifted from 2.92 to 2.64 eV and from 2.75 to 2.46 eV as the In content is increased from 0.16 to 0.22. Such the phenomenon of double luminescence peaks in our InGaN epilayers can be mainly ascribed to the effects of strain gradient along the growth direction [17], we note that luminescence energy splitting ΔE values of P_H-P_L are increased with x in good agreement with reported by S. Pereira *et al.* [18] About the strain-free $In_xGa_{1-x}N$ film, the PL emission peaks are shifted from 2.75 to 1.29 eV as the In content is increased from 0.16 to 0.44, which gives rise to wavelength tunability in this set of samples. Although the luminescence efficiency drops rapidly as the PL peak shifts further into the near infrared color area (950nm, 1.30 eV), to our knowledge these are the maximum attainable emission wavelengths from visible to in the near infrared luminescence region ever reported for thick InGaN epilayers grown by MOCVD method. Further, the variation of their linewidths as a function of solid composition is depicted in the inset of Fig. 5-4. The good qualities in term of linewidths slightly decrease with the In content is increased from 0.16 to 0.40, exceeding our predictions, extend into near infrared region with only a slight broadening of PL linewidth. The increase in In content up to 44% is consistent with the XRD θ -2 θ curve FWHM which was maximum for 44% In content, thus indicates that the generation of defects or inhomogeneous distribution of Ga and

In atoms in $In_xGa_{1-x}N$ epilayers.

To gain more insights on the emission properties of these $In_xGa_{1-x}N$ epilayer samples, we now compare our data with recent contributions to literature are shown in Fig 5-5. These band gaps were measured by photoluminescence and optical absorption, respectively. As shown by the solid curve in Fig. 5-5, the composition dependence of the 14-K PL peak position in the entire composition can be well fit by $E(x) = E_{GaN} - x(E_{GaN} - E_{InN}) - bx(1-x)$, the following well-known equation: where E_{GaN} , E_{InN} , x and b indicate energy of the PL peak or absorption edge for GaN, energy of the PL peak or absorption edge for InN, In content and bowing parameter, respectively. The curve best fit was achieved with b=2.3 eV, while the PL peak energies was carried out with $E_{Gan} = 3.49$ eV, $E_{InN} = 0.77$ eV as shown by the solid line in Fig. 5-5. As shown by the dashed curve, J. Wu et al. [19] studied the band gaps determined from the optical absorption measurement, and according their results showed a small bowing parameter of ~ 1.43 eV. In recent work, M. Moret et al. [20] utilized the X-ray diffraction estimate the composition of $In_xGa_{1-x}N$ ternary materials grown MOCVD from extrapolating from the endpoint binaries, where the bowing parameter b is 2.8. Our results show that the composition dependence on these MOCVD-grown samples exhibits a smaller bowing in the entire composition range. In addition, we demonstrates that the InGaN epilayers exhibit a wide emission

tunability from 2.75 to 1.29 eV, covering a wide spectral ranges from blue, green, red and even reaches infrared 960 nm spectrum region. One can notice that at higher In concentration, the PL peak energy is shifted towards lower energy as compared with the absorption edge. The Stokes shift is then defined as the difference in energy between the effective band gap (absorption) and the emission peak energy (PL) is increases with increasing In content and is as large as 570 meV for $x_{ln}^s = 0.44$. Many authors believed the Stokes shift originates from the strain-induced piezoelectric field or the localization of the excitons. [21-22] R. Zheng et al. [23] concluded that the origin of the localized states in InGaN ternary alloy system is primarily due to the effects of alloy composition fluctuations. Y. H. Huang et al. [24] demonstrated that the inhomogeneity and carrier localization in the InGaN alloys by the comparison of SEM image and cathodoluminescence spectra. To our knowledge, the PL emission spectrum reflects the distribution of localized states, while the absorption transition reflects the onset of the density of delocalized states. Therefore, the fact that the Stokes shift reaches the maximum at $x_{ln}^s = 0.44$ implies that the largest degree of composition fluctuation occurs at the same time.

5.3 Conclusions of $In_xGa_{1-x}N$ films growth

We have carried out a systematic study on the In solid incorporation of InGaN under different input TMIn flow rate using MOCVD technique. By introducing a concept of growth efficiency for reactant Ga and In, their individual incorporation behaviors behaviors in relation to growth condition. Our results indicate that once In droplets formed, the In growth efficiency reduced significantly which causing a decrease of In solid concentration at high TMIn flow rates, and the crystal quality of InGaN films deteriorate markedly. In addition, strong infrared PL signals (960nm, 1.29 eV) have been observed from high In content sample. The MOCVD-grown $In_xGa_{1-x}N$ alloys exhibited a relatively small bowing parameter. The Stokes shift increases with increasing In concentration, suggesting spatial composition fluctuation and strong carrier localization in which.

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Fig. 5-1 XRD θ -2 θ curves of (0002) planes of In_xGa_{1-x}N films grown at various TMIn flow rates. The inset shows the FWHM as functions of the In solid composition (x_{In}^{s}).



Fig. 5-2 The surface morphologies of InGaN films grown at the TMIn flow rates of (a) 3.19, (b) 4.78, (c) 7.97, (d) 9.91, (e) 15.95, (f) 22.30 µmol/min, respectively.



Fig. 5-3 (a) The In solid content (x^{s}_{In}) and growth rate, (b) the individual growth efficiencies of GaN (γ_{GaN}) and InN (γ_{InN}) in ternary In_xGa_{1-x}N as function of TMIn flow rate.



Fig. 5-4 The 12-K PL sepatra of $In_xGa_{1-x}N$ films as functions of the In solid content (x^s_{In}) . The corresponding variations of FWHM and intensity against x^s_{In} are shown insert figure.



Figure 5-5 14-K PL peak energy and band edge determined by optical absorption as a function of x_{In}^{s} . Some data are also shown (Ref. 19 and 20). The dashed curve shows the fit to the band gap energies (absorption) using a bowing parameter *b*=1.43 eV. The solid curve shows the fit to the band gap energies (PL) using a bowing parameter *b*=2.3 eV.

Chapter 6 Conclusions

We have carried out a systematic study on the epitaxial growths and optical properties of InN dots and $In_xGa_{1-x}N$ films grown by metalorganic vapor chemical deposition. First, Flow-rate modulation epitaxy is utilized here to fabricate InN nanostructures. We have investigated the PL properties of uncapped InN nanodots, prepared at growth temperatures from 550 to 725 °C. Our experimental data indicate that the formation of droplets at low growth temperature samples (< 575 °C) due to insufficient supply of active nitrogen radicals during deposition, stemming from the poor cracking efficiency of NH₃ indeed causes the deterioration of the optical properties. As for the high-temperature-grown samples (600-725 °C), where no droplets are formed, good luminescence properties are obtained.

In addition, to gain more insight into the emission properties of these InN dot samples, we subsequently conducted temperaturedependent PL measurements. A blue shifts or slight shifts are observed for those low- and mid-temperature-grown samples. In contrast, clear redshifts of 15-20 meV are observed for samples grown at high temperatures. The temperature-induced energy blueshift around 20-280 K is considered to be partially connected to the bandfilling effects of thermally stimulated surface electrons. Moreover, clear redshifts of 15-20 meV was observed in our high-temperature InN dot samples, which are rarely seen in nano-scale InN structures. Regarding the reduced blue-shift effect observed in the high-temperature samples, we believe that it arise from the fast evaporation of adsorbed In atoms during InN deposition in this temperature range, which produces large densities of In vacancies, acting as acceptors to compensate the thermally stimulated surface electrons in this type of sample.

For InGaN study, our experimental results indicate that the solid composition and characteristic of InGaN were found to be sensitive to growth temperature. As growth temperature was decreased from 750 to 650 °C, the In content increased from 18 to 40 %. XRD measurements showed InGaN films appear to be single diffraction peaks. The lack of side diffraction peaks indicates no large variations in In contents or macroscopic phase separation in these samples. Furthermore, we do observe a FWHM as narrow as ~700 arcsec for x=0.40 sample grow at 650 °C, which is compatible with that of low In-content InGaN films, suggesting the high crystalline quality of the low growth temperature sample. A detailed study of optical properties and structural characterizations of InGaN grown by varying the growth temperature was undertaken. PL spectra of In_xGa_{1-x}N films taken at 14K revealed, for samples grown at temperatures ≥ 700 °C, two emission peaks, are observable. We has suggested two components of the luminescence of these sample origin in regions of different strain, as discriminated by performing a reciprocal space map (RSM) by high-resolution XRD and depth-resolving cathodoluminescence (CL) spectroscopy. On the other hand, for high In content sample (x=0.38 and 0.40), the resulted FWHMs of θ -2 θ x-ray and PL spectra can be as narrow as 700 arcsec and 180 meV, respectively, which can compete the values of InGaN samples with similar In content grown by molecular beam epitaxy. Although macroscopic phase separation is not observed, we do find that the single emission PL band for x=0.38 InGaN film is actually an overlap of light

emissions from different regions with slight In fluctuation, as unveiled by PLE measurement.

Regarding to the InGaN growth, we consider that the solid composition and characteristic of InGaN are determined not only by the growth temperature, but also by the input reactant TMIn flow rates. By introducing a concept of growth efficiency in our study, the individual incorporation behaviors of Ga and In can be elucidated. Our experimental results indicate that too high the TMIn flow rates, resulting in In droplets formed on the surface. Once the droplets were formed, the will allure more In to consolidate into the droplets, deplete the In surfactant available for the growth of InGaN, and the crystal quality of InGaN films deteriorate markedly. Therefore, to obtain an InGaN layer with high In solid composition and high quality, we must paid to avoid the formation of In droplets during growth.

We have carried out a systematic study on the In solid incorporation of InGaN under different input TMIn flow rate using MOCVD technique. By introducing a concept of growth efficiency for reactant Ga and In, their individual incorporation behaviors behaviors in relation to growth condition. Our results indicate that once In droplets formed, the In growth efficiency reduced significantly which causing a decrease of In solid concentration at high TMIn flow rates, and the crystal quality of InGaN films deteriorate markedly. In addition, strong infrared PL signals (960nm, 1.29 eV) have been observed from high In content sample. The MOCVD-grown In_xGa_{1-x}N alloys exhibited a relatively small bowing parameter. The Stokes shift increases with increasing In concentration, suggesting spatial composition fluctuation and strong carrier localization in which.

