

1. Preface

The pseudopotential approach is based on the observation that most physical and chemical properties of atoms are determined by the structure and dynamics of the atomic valence states. It exploits this by removing the core electrons as well as by replacing them and the strong ionic potential by a weaker pseudopotential that acts on a set of nodeless pseudo-wavefunctions rather than the true valence wave functions. Using the pseudopotential approximation, it will be easy to handle the valence electrons of the system and to reduce the dimension of the Hamiltonian matrix. The construction of the pseudopotential is described below.

First we compute the total ground state energy of a many-electron system. For convenience, we use the atomic unit with $\hbar = m = e = 1$. It is a self-consistent calculation. The procedure requires an initial guess for the electronic density, from which the Hartree potential and the exchange-correlation potential can be calculated. After self-consistent calculations, we can obtain the Kohn-Sham eigenstates of the ground state wave functions of atoms. These eigenstates will normally generate the ground state charge density. Our computation results are close to nonrelativistic experimental results.

In the total energy computation of many-body electron system, it is desired to reduce the number of basis in the momentum space and to reduce the drastic charge near the nucleus. By an equivalent one with only the valence orbital and the corresponding pseudopotential, it is

easy to replace the whole system. Having the ground state density, we can construct the pseudopotential with some guiding criteria. In the end, we will draw the pseudo-wave functions and wave functions of atoms.

2. The Premise of Physics

In last several decades, the algorithmic advance and the development of high-speed supercomputer have made Ab Initio quantum mechanical simulations possible for a wide range of physical systems. We will introduce how to compute the total ground state energy of atoms.

2.1 The Schrödinger equation

The equation has been developed (1926) by the Austrian physicist Erwin Schrödinger. For a single particle in three dimensions, the equation is:

$$i\hbar \frac{\partial}{\partial t} \Psi(\mathbf{r}, t) = -\frac{\hbar^2}{2m} \nabla^2 \Psi(\mathbf{r}, t) + V(\mathbf{r}) \Psi(\mathbf{r}, t),$$

where $\Psi(\mathbf{r}, t)$ is the wave function, which is the amplitude for the particle to have a given position \mathbf{r} at any given time t , \hbar is the value of the reduced Planck constant, m is the mass of the particle, and $V(\mathbf{r})$ is the potential. This equation is a time-dependent equation. By introducing the separation of variables, $\Psi(\mathbf{r}, t)$ can be written as

$$\Psi(\mathbf{r}, t) = \psi(\mathbf{r}) f(t).$$

and we obtain

$$\frac{i\hbar}{f} \frac{\partial f}{\partial t} = \left[-\frac{\hbar^2}{2m} \nabla^2 \psi + V(\mathbf{r})\psi \right] \frac{1}{\psi}$$

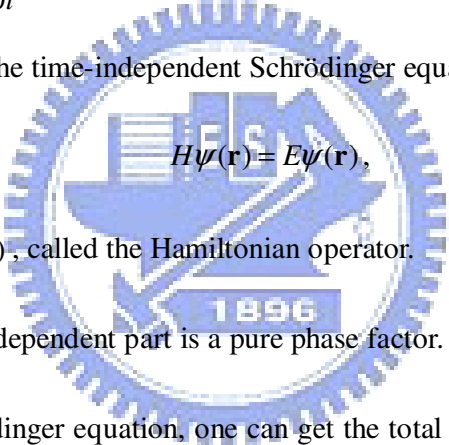
In this equation, the left-hand side has only a variable of t while the right-hand side has only a variable of \mathbf{r} . Let both sides be equal to the same constant, called E (orbital energy). Then,

we can obtain two equations:

$$-\frac{\hbar^2}{2m} \nabla^2 \psi(\mathbf{r}) + V(\mathbf{r})\psi(\mathbf{r}) = E\psi(\mathbf{r}) \quad (1.1)$$

$$i\hbar \frac{\partial f(t)}{\partial t} = Ef(t) \quad (1.2)$$

Equation (1.1) is called the time-independent Schrödinger equation and can be written as:


$$H\psi(\mathbf{r}) = E\psi(\mathbf{r}),$$

where $H = -\frac{\hbar^2}{2m} \nabla^2 + V(\mathbf{r})$, called the Hamiltonian operator.

Evidently the time-dependent part is a pure phase factor. Therefore only dealing with the time-independent Schrödinger equation, one can get the total energy when the particle is in a static potential.

2.2 Density Functional theory and the Kohn-Sham equation

In principle, density functional theory (DFT) is an exact description of the many-body wavefunction, while in practice it spurs on varied types of daring approximations. The starting point of the theory is the observation of Hohenberg and Kohn (1964) that electronic density contains in principle all the information contained in a many-electron wave function. DFT was put on a firm theoretical footing by the two Hohenberg-Kohn theorems (H-K).

The first H-K theorem proved that the ground state properties of a many-electron system are uniquely determined by an electronic density. We can say that the density completely determines the many-body problem.

The second H-K theorem defines energy functional for the system and proves that the correct ground state electronic density minimizes this energy functional. The minimum value of the total energy functional is the ground state energy.

The total energy E of the system as a functional of the charge density $n(\mathbf{r})$ is written:

$$E[n] = T[n] + \int V_{ext}(\mathbf{r})n(\mathbf{r})d\mathbf{r} + \frac{1}{2} \int V_H[\mathbf{r}]n(\mathbf{r})d\mathbf{r} + E_{xc}[n], \quad (2.2)$$

where T is the kinetic energy of the system, V_{ext} is the external potential (e.g. the Coulomb potential due to the nuclear charges), V_H is the electron-electron Coulomb potential, E_{xc} is the exchange-correlation functional.

Only the minimum value of the total energy functional $E[n]$ has an important physical meaning. We can obtain the total ground state energy by minimizing the total energy

functional $E[n]$, and the density that yields this minimum value is the ground state density. In order to minimize the total energy functional (2.2), we use Lagrange multiplier. The method of Lagrange multipliers provides a strategy for finding the maximum/minimum of a function subject to constraints.

$$\frac{\partial}{\partial \psi_i^*(\mathbf{r})} \left(E[n] - \varepsilon_i \left(N - \int n(\mathbf{r}) d\mathbf{r} \right) \right) = 0 \quad (2.3)$$

where ψ_i is the wave function of the state i , ε_i is the Lagrange multiplier and

$\int n(\mathbf{r}) d\mathbf{r} = N$ is the constraint. By equation (2.3), we can obtain the Kohn-Sham equation:

$$\left(-\frac{\hbar^2}{2m} \nabla^2 + V_{ext}(\mathbf{r}) + V_{xc}(\mathbf{r}) + V_H(\mathbf{r}) \right) = \varepsilon_i \psi_i(\mathbf{r}) \quad , \quad (2.4)$$

where ε_i is the Lagrange multiplier and may be interpreted as the orbital energy of the state represented by $\psi_i(\mathbf{r})$, V_{xc} is the exchange-correlation potential, and V_H is electron-electron Coulomb potential. The Kohn-Sham equation (2.4) represents a mapping of the interacting many-electron system onto a system of non-interacting electrons moving in an effective potential due to all the other electrons. The set of wave functions ψ_i are given by the ab initio self-consistent solutions to the Kohn-Sham equation (2.4) so that the occupied electronic states generate a charge density that produces the electronic potential that was used to construct the equations.

2.3 The Local density approximation

The Hohenberg-Kohn theorem provides some motivation for using approximate methods to describe the exchange-correlation energy as a function of the electron density. The local density approximation (LDA) is the simplest method of describing the exchange-correlation energy of an electron system. In LDA, the exchange-correlation energy of an electron system is constructed by assuming that the exchange-correlation energy $\epsilon_{xc}(\mathbf{r})$ per electron at a point \mathbf{r} in the electron gas, it is equal to the exchange-correlation energy per electron in a homogeneous electron gas that has the same density as the electron gas at point \mathbf{r} . Thus,

$$E_{xc}[n(\mathbf{r})] = \int \epsilon_{xc}(\mathbf{r})n(\mathbf{r})d\mathbf{r} \quad (2.5)$$

and by [1], we have:

$$\frac{\delta E_{xc}[n(\mathbf{r})]}{\delta n(\mathbf{r})} = \frac{\partial [n(\mathbf{r})\epsilon_{xc}(\mathbf{r})]}{\partial n(\mathbf{r})} \quad (2.6)$$

The LDA assumes that the exchange-correlation energy functional is purely local. In principle, it ignores correlations to the exchange-correlation energy at a point \mathbf{r} due to nearby inhomogeneities in the electron density. Considering the inexact nature of the approximation, it is remarkable that calculations performed using the LDA have been so successful.

2.4 Pseudopotential

It is well known that most physical properties of solids are dependent on the valence electrons to a much greater extent than on the core electrons. Only the valence electrons play important roles in the calculation of more complex system such as molecules, clusters or band

structures. The orbitals in the inner-core remain mostly unperturbed. It is desirable to replace the whole system by an equivalent one with only the valence orbital and the corresponding pseudopotential. The coulomb interaction of the electrons is the most time-consuming in real calculation and will become trivial in the momentum space. To reduce the number of the basis in the momentum space, it is desirable to reduce the drastic charge near the nucleus for the pseudopotential.

Following Hamann, Schluter and Chiang, the norm-conserving pseudopotential is constructed with the following guiding criteria:

1. Real and pseudo valence eigenvalues agree.
2. Real and pseudo atomic wave functions agree beyond a chosen “core radius” r_c .
3. The integrals from 0 to r of the real and pseudo charge densities agree for $r > r_c$ for each valence state (norm conservation).
4. The logarithmic derivatives of the real and pseudo wave function and their first energy derivatives agree for $r > r_c$.

Properties (3) and (4) are important for the pseudopotential to have optimum transferability among a variety of chemical environments in self-consistent calculations in which the pseudo charge density is treated as a real physical object. Explicitly, the norm-conserving pseudo potential are constructed for each angular momentum l .

3. Process of study

3.1 The expression of the potential of the Kohn-Sham equation

In this note, we use the atomic unit with $\hbar = m = e = 1$. The ground state energy of many-electron system is expressed as a functional $E[n]$ (see(2.2)) of the electronic density $n(\mathbf{r})$:

$$n(\mathbf{r}) = \sum_{i \in \text{occupied}} f_i |\psi_i(\mathbf{r})|^2, \quad (3.1)$$

where f_i are the occupation numbers of the orbital denoted by i . Because we use atomic unit, the Kohn-Sham equation (2.4) becomes:

$$\left(-\frac{1}{2} \nabla^2 + V_{\text{ext}}(\mathbf{r}) + V_{\text{xc}}(\mathbf{r}) + V_H(\mathbf{r}) \right) \psi_i(\mathbf{r}) = \epsilon_i \psi_i(\mathbf{r}) \quad (3.2)$$

Here $V_{\text{ext}}(\mathbf{r})$ is the external potential that is written:

$$V_{\text{ext}}(\mathbf{r}) = -\frac{z}{r}, \quad (3.3)$$

where z is an atomic number (e.g. Mg $z=12$).

The exchange-correlation potential V_{xc} is given by:

$$V_{\text{xc}}(\mathbf{r}) = \frac{\delta E_{\text{xc}}[n]}{\delta n(\mathbf{r})} \quad (3.4)$$

In LDA, by the equation (2.5), we can get further:

$$V_{\text{xc}}(\mathbf{r}) = \frac{\delta E_{\text{xc}}[n]}{\delta n(\mathbf{r})} = \epsilon_{\text{xc}}(\mathbf{r}) + n(\mathbf{r}) \frac{d\epsilon_{\text{xc}}}{dn} \quad (3.5)$$

The electron-electron Coulomb potential V_H is written:

$$V_H(\mathbf{r}) = \int \frac{n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' \quad (3.6)$$

Thus, the effective potential V_{eff} is written:

$$V_{\text{eff}}(\mathbf{r}) = V_{\text{ext}}(\mathbf{r}) + V_H(\mathbf{r}) + V_{\text{xc}}(\mathbf{r}) \quad (3.7)$$

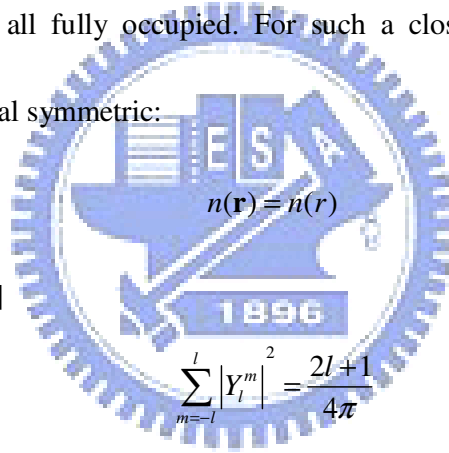
3.2 The simplification of the effective potential

It is convenient to employ the spherical coordinates $\{r, \theta, \phi\}$. The eigenfunctions are written as:

$$\psi_i(\mathbf{r}) = R_{nl}(r)Y_l^m(\theta, \phi) \quad (3.8)$$

The electronic orbitals are represented by the index $i = \{n, l, m\}$, the main quantum number n , the angular momentum quantum number l , and the magnetic quantum number m respectively.

Let us first consider the closed-shell system for example Ne atom with $z=10$. The electronic shells $1s^2, 2s^2, 2p^6$ are all fully occupied. For such a closed-shell system, the electronic density $n(\mathbf{r})$ is spherical symmetric:



due to the identity by [2]

$$\sum_{m=-l}^l |Y_l^m|^2 = \frac{2l+1}{4\pi}$$

Hence the external potential $V_{ext}(n(\mathbf{r}))$ and the exchange-correlation potential $V_{xc}(n(\mathbf{r}))$ are manifestly spherical symmetric. Further more, using the addition theorem of angular momentum:

$$\frac{1}{|\mathbf{r}-\mathbf{r}'|} = \sum_{l=0}^{\infty} \frac{r_{<}^l}{r_{>}^{l+1}} P_l(\cos \theta_{\mathbf{r}, \mathbf{r}'}) = \sum_{l=0}^{\infty} \frac{r_{<}^l}{r_{>}^{l+1}} \frac{4\pi}{2l+1} \sum_{m=-l}^l Y_l^m(\theta, \phi) Y_l^{m*}(\theta', \phi'),$$

where $r_{<} = \min(\mathbf{r}, \mathbf{r}')$, and $r_{>} = \max(\mathbf{r}, \mathbf{r}')$.

We have

$$\begin{aligned}
V_H(\mathbf{r}) &= \int \frac{n(\mathbf{r}')}{|\mathbf{r}-\mathbf{r}'|} d\mathbf{r}' \\
&= \sum_{l=0}^{\infty} \frac{4\pi}{2l+1} \sum_{m=-l}^l Y_l^m(\theta, \phi) \iiint r'^2 n(r') \frac{r_{<}^l}{r_{>}^{l+1}} dr' Y_l^{m*}(\theta', \phi') \sin \theta' d\theta' d\phi' \\
&= 4\pi \int_0^{\infty} r'^2 n(r') \frac{1}{r_{>}} dr' \tag{3.9}
\end{aligned}$$

Thus, the electron-electron Coulomb potential $V_H(n(\mathbf{r}))$ is also spherical symmetric.

The exchange-correlation energy per unit density $\varepsilon_{xc}(n)$ may be decomposed into a sum due to the exchange energy and the correlation energy.

$$\varepsilon_{xc}(n) = \varepsilon_x(n) + \varepsilon_c(n) \tag{3.10}$$

They may be obtained from a system of uniform electron gas with the corresponding density.

They are frequently expressed in terms of the radius r_s of a unit charge defined by

$$r_s = \left(\frac{3}{4\pi n}\right)^{1/3} \tag{3.11}$$

The exchange term can be calculated exactly for a uniform electron gas, and is given by

$$\varepsilon_x(n) = -\frac{3}{4\pi r_s} \left(\frac{9\pi}{4n}\right)^{1/3} = -\frac{0.458}{r_s} \tag{3.12}$$

In general, the magnitude of the correlation energy is much smaller than the exchange energy,

and yet it is much harder to calculate. The correlation energy was calculated by Ceperley and

Alder using Monte-Carlo simulation and was parameterized by Perdew and Zunger as:

$$\varepsilon_c = \begin{cases} -\gamma/(1+\beta_1\sqrt{r_s}+\beta_2r_s) & \text{for } r_s \geq 1, \text{ low density,} \\ A\ln r_s + B + C r_s \ln r_s + D r_s & \text{for } r_s < 1, \text{ high density,} \end{cases} \tag{3.13}$$

where the values of the constants are given by

$$\gamma = 0.1423; \quad \beta_1 = 1.0529; \quad \beta_2 = 0.3334$$

and

$$A = 0.0311; B = -0.0480; C = 0.0020; D = -0.0116$$

By the equation (3.10) to (3.12), the exchange-correlation potential V_{xc} can be simplified as:

$$\begin{aligned}
 V_{xc}(r) &= \frac{\delta E_{xc}[n]}{\delta n(\mathbf{r})} = \varepsilon_{xc}(\mathbf{r}) + n(\mathbf{r}) \frac{d\varepsilon_{xc}}{dn} \\
 &= \varepsilon_{xc}(\mathbf{r}) + n(\mathbf{r}) \frac{d\varepsilon_x}{dn} + n(\mathbf{r}) \frac{d\varepsilon_c}{dn} \\
 &= \left\{ \begin{array}{l} \varepsilon_{xc}(\mathbf{r}) + n(\mathbf{r}) \left[0.458 r_s^{-2} \frac{dr_s}{dn} \right] + \\ n(\mathbf{r}) \left[\gamma(1 + \beta_1 r_s^{0.5} + \beta_2 r_s)^{-2} \cdot (0.5 \beta_1 r_s^{-0.5} \frac{dr_s}{dn} + \beta_2 \frac{dr_s}{dn}) \right] \text{ for } r_s \geq 1, \\ \varepsilon_{xc}(\mathbf{r}) + n(\mathbf{r}) \left[0.458 r_s^{-2} \frac{dr_s}{dn} \right] + \\ n(\mathbf{r}) \left[A r_s^{-1} \frac{dr_s}{dn} + C \frac{dr_s}{dn} \ln r_s + C r_s r_s^{-1} \frac{dr_s}{dn} + D \frac{dr_s}{dn} \right] \text{ for } r_s < 1, \end{array} \right. \quad (3.14)
 \end{aligned}$$

where $\frac{dr_s}{dn}$ is from the equation (3.11) and is written as:

$$\frac{dr_s}{dn} = -4(3)^{-2/3} \pi (4\pi n)^{-4/3}$$

Thus, the terms of the effective potential are obvious to calculate.

3.3 The simplification of the Kohn-Sham equation

Using the expression of the Laplace operator in the spherical coordinates:

$$\nabla^2 \psi(\mathbf{r}) = \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial \psi}{\partial r} \right) + \frac{1}{r^2} \left(\frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial \psi}{\partial \theta} \right) + \frac{1}{\sin^2 \theta} \frac{\partial^2 \psi}{\partial \phi^2} \right) \quad (3.15)$$

And the identity:

$$\left(\frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right) Y_l^m = -l(l+1) Y_l^m \quad (3.16)$$

By the equation (3.15) and (3.16), the Kohn-Sham equation (3.2) now takes the form:

$$-\frac{1}{2} \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial R_{nl}}{\partial r} \right) Y_l^m - \frac{1}{2} \frac{1}{r^2} \left(\frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right) Y_l^m R_{nl} + V_{eff} R_{nl} Y_l^m = \varepsilon_l R_{nl} Y_l^m$$

Using the equation (3.16), we have:

$$-\frac{1}{2} \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial R_{nl}}{\partial r} \right) Y_l^m - \frac{1}{2} \frac{1}{r^2} [-l(l+1)] Y_l^m R_{nl} + V_{eff} R_{nl} Y_l^m = \varepsilon_l R_{nl} Y_l^m$$

Canceling the both sides Y_l^m , we obtain:

$$-\frac{1}{2} \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial R_{nl}}{\partial r} \right) + \frac{l(l+1)}{2r^2} R_{nl} + V_{eff} R_{nl} = \varepsilon_l R_{nl}$$

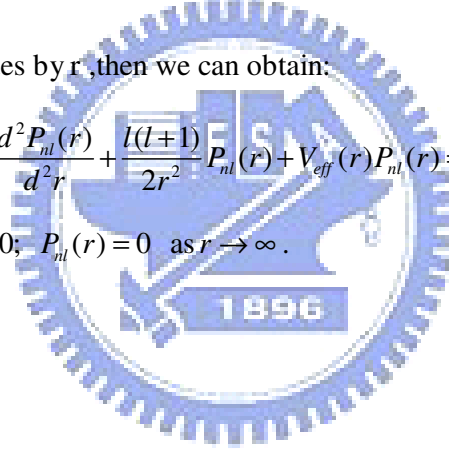
Introducing

$$P_{nl}(r) = r R_{nl}(r)$$

and multiplying both sides by r , then we can obtain:

$$-\frac{1}{2} \frac{d^2 P_{nl}(r)}{dr^2} + \frac{l(l+1)}{2r^2} P_{nl}(r) + V_{eff}(r) P_{nl}(r) = \varepsilon_{nl} P_{nl} \quad (3.17)$$

Here we know $P_{nl}(0) = 0$; $P_{nl}(r) = 0$ as $r \rightarrow \infty$.



3.4 The discretization of the Kohn-Sham equation

By the Taylor series, we can get the discretization of $\frac{d^2 P_{nl}(r)}{dr^2}$ that is written as:

$$\frac{d^2 P_{nl}(x_i)}{dr^2} \approx \frac{P_{nl}(x_{i+1}) - 2P_{nl}(x_i) + P_{nl}(x_{i-1}))}{h^2}, \quad (3.18)$$

where h is the distance of the $x_i - x_{i-1}$. Using the equation (3.18) substitute for (3.17), we

can get:

$$-\frac{1}{2} \frac{P_{nl}(x_{i+1}) - 2P_{nl}(x_i) + P_{nl}(x_{i-1}))}{h^2} + \frac{l(l+1)}{2x_i^2} P_{nl}(x_i) + V_{eff}(x_i) P_{nl}(x_i) = \varepsilon_{nl} P_{nl}(x_i)$$

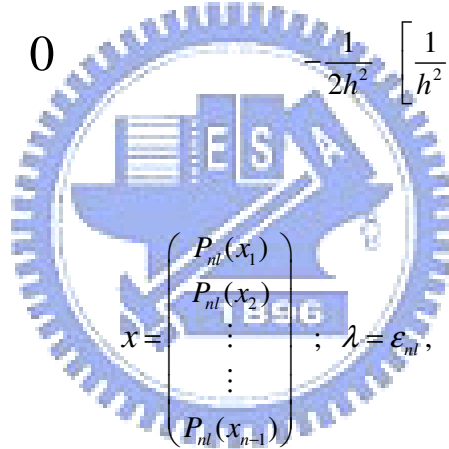
It can be written as a tridiagonal symmetric matrix:

$$\left(\begin{array}{ccc} \left[\frac{1}{h^2} + \frac{l(l+1)}{2x_1^2} + V_{eff}(x_1) \right] & -\frac{1}{2h^2} & 0 \\ -\frac{1}{2h^2} & \ddots & \ddots \\ & \ddots & \ddots & -\frac{1}{2h^2} \\ 0 & & -\frac{1}{2h^2} & \left[\frac{1}{h^2} + \frac{l(l+1)}{2x_{n-1}^2} + V_{eff}(x_{n-1}) \right] \end{array} \right) \begin{pmatrix} P_{nl}(x_1) \\ P_{nl}(x_2) \\ \vdots \\ P_{nl}(x_{n-1}) \end{pmatrix} = \epsilon_{nl} \begin{pmatrix} P_{nl}(x_1) \\ P_{nl}(x_2) \\ \vdots \\ P_{nl}(x_{n-1}) \end{pmatrix}$$

Introducing

$$A = \left(\begin{array}{ccc} \left[\frac{1}{h^2} + \frac{l(l+1)}{2x_1^2} + V_{eff}(x_1) \right] & -\frac{1}{2h^2} & 0 \\ -\frac{1}{2h^2} & \ddots & \ddots \\ & \ddots & \ddots & -\frac{1}{2h^2} \\ 0 & & -\frac{1}{2h^2} & \left[\frac{1}{h^2} + \frac{l(l+1)}{2x_{n-1}^2} + V_{eff}(x_{n-1}) \right] \end{array} \right)$$

and



$$x = \begin{pmatrix} P_{nl}(x_1) \\ P_{nl}(x_2) \\ \vdots \\ P_{nl}(x_{n-1}) \end{pmatrix}; \quad \lambda = \epsilon_{nl},$$

This equation may also be written as:

$$Ax = \lambda x \tag{3.19}$$

In this equation, we can solve the eigenvalues and the eigenvectors to get the orbital energy

and the wave function $P_{nl}(r)$.

3.5 The procedure of Ab Initio self-consistent

Having the equation (3.19), we can start to calculate. The procedure of ab initio self-consistent is shown in the flow diagram in Fig. 1. The procedure requires an initial guess for the electronic charge density. From which the Hartree potential and the exchange-correlation potential can be calculated.

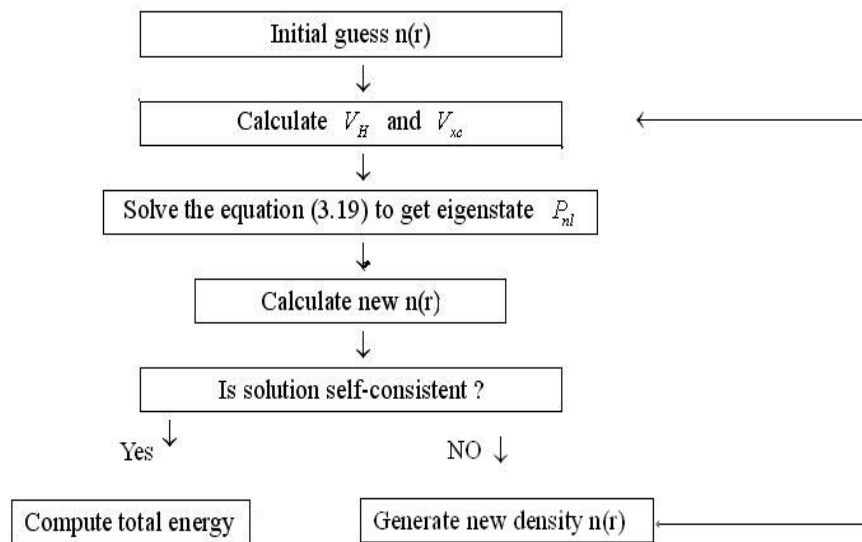


Fig. 1. Flow chart describing the computational procedure for the calculation of the ground state density of atoms.

In the end, we can obtain the ground state density of atoms to calculate total ground state energy.

3.6 The total ground state energy and simplification

From equation (2.2), the ground state energy functional is written as:

$$E[n] = T[n] + \int V_{ext}(\mathbf{r})n(\mathbf{r})d\mathbf{r} + \frac{1}{2} \int V_H[\mathbf{r}]n(\mathbf{r})d\mathbf{r} + E_{xc}[n]$$

The first term on the right hand side is the kinetic energy functional that is expressed in terms of a system of noninteracting electrons.

$$T[n] = \sum_{i \in occupied} f_i \int \psi_i^*(\mathbf{r}) \left(-\frac{\nabla^2}{2} \right) \psi_i(\mathbf{r}) d\mathbf{r}$$

By multiplying both sides of equation (3.15) with $-\frac{1}{2} \psi_i^*(\mathbf{r})$, carrying out the integration, and

then summing over all the occupied states i , the kinetic energy functional may be written as :

$$T[n] = \sum_{i \in occupied} f_i \int_0^\infty \left(\frac{1}{2} P_{nl} \frac{d^2 P_{nl}}{dr^2} + \frac{l(l+1)}{2r^2} P_{nl}^2 \right) dr \quad (3.20)$$

The second term on the right hand side of equation (2.2) is the external energy that can be simplified as:

$$\int V_{ext}(\mathbf{r})n(\mathbf{r})d\mathbf{r} = \iiint -\frac{z}{r} n(r) r^2 dr \sin \theta d\theta d\phi = -4\pi \int_0^\infty z r n(r) dr \quad (3.21)$$

The third term on the right hand side of equation (2.2) is the Hartree energy (electron-electron Coulomb energy) can be simplified as by using the addition theorem of angular momentum:

$$\begin{aligned} \frac{1}{2} \int V_H[\mathbf{r}]n(\mathbf{r})d\mathbf{r} &= \frac{1}{2} \iint \frac{n(\mathbf{r})n(\mathbf{r}')}{|\mathbf{r}-\mathbf{r}'|} d\mathbf{r}d\mathbf{r}' \\ &= \frac{1}{2} \iiint \iiint \frac{n(r)n(r')r^2 r'^2 dr dr' d\Omega d\Omega'}{|\mathbf{r}-\mathbf{r}'|} \\ &= \frac{1}{2} (4\pi)^2 \iint \frac{n(r)n(r')r^2 r'^2}{r_{>}} dr dr' , \end{aligned} \quad (3.22)$$

where $d\Omega = \sin \theta d\theta d\phi$, $d\Omega' = \sin \theta' d\theta' d\phi'$.

In the subsection 3.5, we have gotten the ground state density n . Substituting the ground

state density n for the energy functional equation (2.2), we can obtain the total ground state energy of atoms. In table 1 the resulting values of the ground state energy are displayed, for all atoms from $z=1$ to $z=102$, together with the corresponding values of $E_{tot}^{HF}(Z)$. The ratio $\left| \frac{E[n] - E_{tot}^{HF}}{E_{tot}^{HF}} \right|$ is shown in figure 2. As can be seen, the difference between the HF value and the value computed from equation (2.2) rarely exceeds 7%.

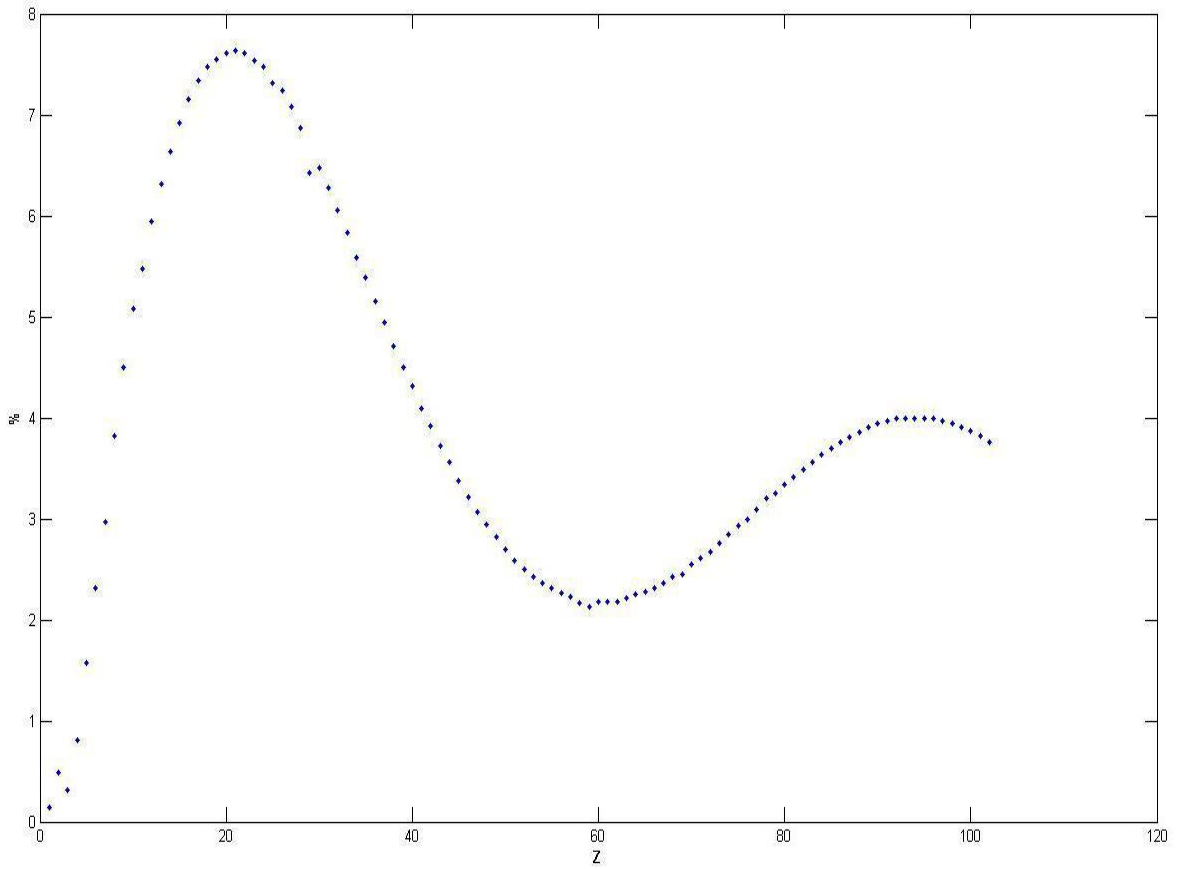


Figure 2. The ratio $\left| \frac{E[n] - E_{tot}^{HF}}{E_{tot}^{HF}} \right|$ plotted against Z . The values $E[n]$ and E_{tot}^{HF} are from table 1.

Table 1. Comparison of $E[n]$, as computed from equation (2.2) of the text, with the Hartree-Fock value $E_{\text{HF}}^{\text{at}}(Z)$ (from Froese-Fischer 1972, and Mann 1973). Energies are in atomic units.

| Z atom | $-E[n]$ (a.u.) | $-E_{\text{HF}}^{\text{at}}$ (a.u.) | error (%) | Z atom | $-E[n]$ (a.u.) | $-E_{\text{HF}}^{\text{at}}$ (a.u.) | error (%) | Z atom | $-E[n]$ (a.u.) | $-E_{\text{HF}}^{\text{at}}$ (a.u.) | error (%) |
|--------|----------------|-------------------------------------|-----------|--------|----------------|-------------------------------------|-----------|--------|----------------|-------------------------------------|-----------|
| 1 H | 0.5007 | 0.5 | 0.14 | 35 Br | 2710.7 | 2572 | 5.39 | 69 Tm | 13257 | 12940 | 2.45 |
| 2 He | 2.8481 | 2.862 | 0.49 | 36 Kr | 2893.9 | 2752 | 5.16 | 70 Yb | 13732 | 13391 | 2.55 |
| 3 Li | 7.4096 | 7.433 | 0.31 | 37 Rb | 3083.4 | 2938 | 4.95 | 71 Lu | 14213 | 13852 | 2.61 |
| 4 Be | 14.688 | 14.57 | 0.81 | 38 Sr | 3279.5 | 3132 | 4.71 | 72 Hf | 14706 | 14323 | 2.67 |
| 5 B | 24.918 | 24.53 | 1.58 | 39 Y | 3482.1 | 3332 | 4.5 | 73 Ta | 15209 | 14800 | 2.76 |
| 6 C | 38.562 | 37.69 | 2.31 | 40 Zr | 3691.5 | 3539 | 4.31 | 74 W | 15722 | 15287 | 2.85 |
| 7 N | 56.016 | 54.4 | 2.97 | 41 Nb | 3907.7 | 3754 | 4.09 | 75 Re | 16246 | 15784 | 2.93 |
| 8 O | 77.665 | 74.81 | 3.82 | 42 Mo | 4131 | 3975 | 3.92 | 76 Os | 16781 | 16293 | 3 |
| 9 F | 103.88 | 99.41 | 4.5 | 43 Tc | 4361.5 | 4205 | 3.72 | 77 Ir | 17326 | 16806 | 3.09 |
| 10 Ne | 135.03 | 128.5 | 5.08 | 44 Ru | 4599.3 | 4441 | 3.56 | 78 Pt | 17888 | 17333 | 3.2 |
| 11 Na | 170.76 | 161.9 | 5.47 | 45 Rh | 4844.4 | 4686 | 3.38 | 79 Au | 18447 | 17866 | 3.25 |
| 12 Mg | 211.48 | 199.6 | 5.95 | 46 Pd | 5097.1 | 4938 | 3.22 | 80 Hg | 19023 | 18409 | 3.34 |
| 13 Al | 257.17 | 241.9 | 6.31 | 47 Ag | 5357.6 | 5198 | 3.07 | 81 Tl | 19609 | 18962 | 3.41 |
| 14 Si | 308.05 | 288.9 | 6.63 | 48 Cd | 5625.7 | 5465 | 2.94 | 82 Pb | 20205 | 19524 | 3.49 |
| 15 P | 364.27 | 340.7 | 6.92 | 49 In | 5901.6 | 5740 | 2.82 | 83 Bi | 20811 | 20096 | 3.56 |
| 16 S | 425.95 | 397.5 | 7.16 | 50 Sn | 6185.4 | 6023 | 2.7 | 84 Po | 21427 | 20676 | 3.63 |
| 17 Cl | 493.21 | 459.5 | 7.34 | 51 Sb | 6477.3 | 6314 | 2.59 | 85 At | 22053 | 21267 | 3.7 |
| 18 Ar | 566.17 | 526.8 | 7.47 | 52 Te | 6777.4 | 6612 | 2.5 | 86 Rn | 22689 | 21867 | 3.76 |
| 19 K | 644.45 | 599.2 | 7.55 | 53 I | 7085.9 | 6918 | 2.43 | 87 Fr | 23333 | 22476 | 3.81 |
| 20 Ca | 728.32 | 676.8 | 7.61 | 54 Xe | 7402.9 | 7232 | 2.36 | 88 Ra | 23986 | 23094 | 3.86 |
| 21 Sc | 817.75 | 759.7 | 7.64 | 55 Cs | 7728.2 | 7554 | 2.31 | 89 Ac | 24649 | 23722 | 3.91 |
| 22 Ti | 912.97 | 848.4 | 7.61 | 56 Ba | 8062.1 | 7884 | 2.26 | 90 Th | 25320 | 24360 | 3.94 |
| 23 V | 1014 | 942.9 | 7.54 | 57 La | 8404.6 | 8221 | 2.23 | 91 Pa | 25999 | 25007 | 3.97 |
| 24 Cr | 1121 | 1043 | 7.48 | 58 Ce | 8753.1 | 8567 | 2.17 | 92 U | 26687 | 25664 | 3.99 |
| 25 Mn | 1234.2 | 1150 | 7.32 | 59 Pr | 9111.3 | 8921 | 2.13 | 93 Np | 27384 | 26331 | 4 |
| 26 Fe | 1353.4 | 1262 | 7.24 | 60 Nd | 9486.1 | 9284 | 2.18 | 94 Pu | 28089 | 27008 | 4 |
| 27 Co | 1478.8 | 1381 | 7.08 | 61 Pm | 9865.6 | 9655 | 2.18 | 95 Am | 28803 | 27696 | 4 |
| 28 Ni | 1610.5 | 1507 | 6.87 | 62 Sm | 10254 | 10035 | 2.18 | 96 Cm | 29525 | 28392 | 3.99 |
| 29 Cu | 1744.4 | 1639 | 6.43 | 63 Eu | 10653 | 10423 | 2.21 | 97 Bk | 30255 | 29100 | 3.97 |
| 30 Zn | 1893.1 | 1778 | 6.47 | 64 Gd | 11063 | 10820 | 2.25 | 98 Cf | 30993 | 29817 | 3.94 |
| 31 Ga | 2043.8 | 1923 | 6.28 | 65 Tb | 11482 | 11226 | 2.28 | 99 Es | 31739 | 30545 | 3.91 |
| 32 Ge | 2200.8 | 2075 | 6.06 | 66 Dy | 11911 | 11641 | 2.32 | 100 Fm | 32493 | 31283 | 3.87 |
| 33 As | 2364.3 | 2234 | 5.83 | 67 Ho | 12350 | 12065 | 2.36 | 101 Md | 33254 | 32031 | 3.82 |
| 34 Se | 2534.2 | 2400 | 5.59 | 68 Er | 12801 | 12498 | 2.42 | 102 No | 34023 | 32790 | 3.76 |

3.7 Construction of Pseudopotential

Explicitly, the norm-conserving pseudopotential are constructed for each angular momentum l in the following steps: We first solve for the radial part $P_{nl}(r)$ of the valence eigenfunctions ψ_i and the corresponding eigenvalues ϵ_{nl} by first carrying out an ab initio self-consistent of the figure 1. The radial parts of the valence states are given by the equation (3.17) or

$$-\frac{1}{2} \frac{d^2 P_{nl}(r)}{d^2 r} + V_l(r) P_{nl}(r) = \epsilon_{nl} P_{nl}(r) \quad , \quad (3.23)$$

where

$$V_l(r) = \frac{l(l+1)}{2r^2} + V_{eff}(r).$$

For each l , we choose a cutoff radius r_c , typically 0.5 to 1.0 times the radius r_m of the outermost peak of $P_{nl}(r)$. In table 2 to table 5, the choices of the cutoff radius r_c are displayed, for all atoms from $z=1$ to $z=102$.

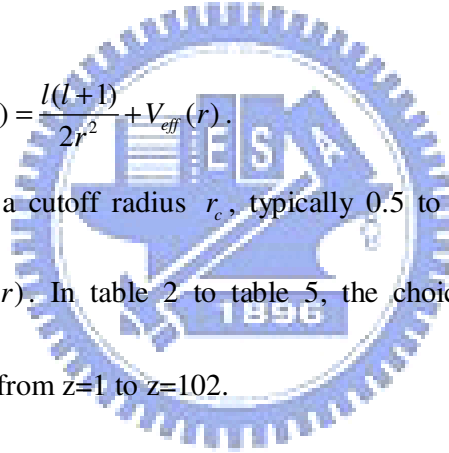


Table 2. The choices of the cutoff radius r_c (Å) for each states of $z=1$ to $z=25$ atoms.

| z | atom | electronic state | 1s | 2s | 3s | 4s | 2p | 3p | 3d |
|----|------|------------------------|-------|-------|-------|-------|-------|-------|-------|
| 1 | H | $1s^1$ | 0.56 | | | | | | |
| 2 | He | $1s^2$ | 0.322 | | | | | | |
| 3 | Li | $1s^2 2s^1$ | 0.21 | 1.75 | | | | | |
| 4 | Be | $1s^2 2s^2$ | 0.154 | 1.148 | | | | | |
| 5 | B | $1s^2 2s^2 2p^1$ | 0.126 | 0.854 | | 0.868 | | | |
| 6 | C | $1s^2 2s^2 2p^2$ | 0.098 | 0.686 | | 0.672 | | | |
| 7 | N | $1s^2 2s^2 2p^3$ | 0.084 | 0.574 | | 0.546 | | | |
| 8 | O | $1s^2 2s^2 2p^4$ | 0.07 | 0.49 | | 0.448 | | | |
| 9 | F | $1s^2 2s^2 2p^5$ | 0.07 | 0.434 | | 0.392 | | | |
| 10 | Ne | $1s^2 2s^2 2p^6$ | 0.056 | 0.392 | | 0.35 | | | |
| 11 | Na | $[\text{Ne}]3s^1$ | 0.056 | 0.35 | 1.82 | | 0.308 | | |
| 12 | Mg | $[\text{Ne}]3s^2$ | 0.056 | 0.308 | 1.428 | | 0.266 | | |
| 13 | Al | $[\text{Ne}]3s^2 3p^1$ | 0.042 | 0.28 | 1.162 | | 0.238 | 1.47 | |
| 14 | Si | $[\text{Ne}]3s^2 3p^2$ | 0.042 | 0.266 | 1.008 | | 0.21 | 1.19 | |
| 15 | P | $[\text{Ne}]3s^2 3p^3$ | 0.042 | 0.238 | 0.896 | | 0.196 | 1.022 | |
| 16 | S | $[\text{Ne}]3s^2 3p^4$ | 0.042 | 0.224 | 0.798 | | 0.182 | 0.896 | |
| 17 | Cl | $[\text{Ne}]3s^2 3p^5$ | 0.042 | 0.21 | 0.728 | | 0.168 | 0.798 | |
| 18 | Ar | $[\text{Ne}]3s^2 3p^6$ | 0.028 | 0.196 | 0.672 | | 0.154 | 0.714 | |
| 19 | K | $[\text{Ar}]4s^1$ | 0.028 | 0.182 | 0.616 | 2.282 | 0.14 | 0.644 | |
| 20 | Ca | $[\text{Ar}]4s^2$ | 0.028 | 0.168 | 0.574 | 1.904 | 0.14 | 0.588 | |
| 21 | Sc | $[\text{Ar}]4s^2 3d^1$ | 0.028 | 0.168 | 0.532 | 1.764 | 0.126 | 0.532 | 0.616 |
| 22 | Ti | $[\text{Ar}]4s^2 3d^2$ | 0.028 | 0.154 | 0.504 | 1.652 | 0.126 | 0.504 | 0.546 |
| 23 | V | $[\text{Ar}]4s^2 3d^3$ | 0.028 | 0.154 | 0.476 | 1.582 | 0.112 | 0.476 | 0.49 |
| 24 | Cr | $[\text{Ar}]4s^1 3d^5$ | 0.028 | 0.14 | 0.448 | 1.582 | 0.112 | 0.448 | 0.462 |
| 25 | Mn | $[\text{Ar}]4s^2 3d^5$ | 0.028 | 0.14 | 0.42 | 1.442 | 0.098 | 0.42 | 0.42 |

Table 3. The choices of the cutoff radius r_c (Å) for each states of $z=26$ to $z=50$ atoms.

| z | atom | electronic state | 1s | 2s | 3s | 4s | 5s | 2p | 3p | 4p | 5p | 3d | 4d |
|----|------|---|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| 26 | Fe | [Ar] 4s ² 3d ⁶ | 0.028 | 0.126 | 0.406 | 1.4 | | 0.098 | 0.392 | | | 0.392 | |
| 27 | Co | [Ar] 4s ² 3d ⁷ | 0.028 | 0.126 | 0.392 | 1.344 | | 0.098 | 0.378 | | | 0.378 | |
| 28 | Ni | [Ar] 4s ² 3d ⁸ | 0.028 | 0.126 | 0.378 | 1.302 | | 0.098 | 0.364 | | | 0.35 | |
| 29 | Cu | [Ar] 4s ¹ 3d ¹⁰ | 0.028 | 0.112 | 0.364 | 1.372 | | 0.084 | 0.35 | | | 0.378 | |
| 30 | Zn | [Ar] 4s ² 3d ¹⁰ | 0.028 | 0.112 | 0.35 | 1.232 | | 0.084 | 0.322 | | | 0.322 | |
| 31 | Ga | [Ar] 4s ² 3d ¹⁰ 4p ¹ | 0.014 | 0.112 | 0.336 | 1.12 | | 0.084 | 0.308 | 1.372 | | 0.294 | |
| 32 | Ge | [Ar] 4s ² 3d ¹⁰ 4p ² | 0.014 | 0.112 | 0.322 | 1.022 | | 0.084 | 0.294 | 1.204 | | 0.28 | |
| 33 | As | [Ar] 4s ² 3d ¹⁰ 4p ³ | 0.014 | 0.098 | 0.308 | 0.952 | | 0.07 | 0.294 | 1.078 | | 0.266 | |
| 34 | Se | [Ar] 4s ² 3d ¹⁰ 4p ⁴ | 0.014 | 0.098 | 0.294 | 0.896 | | 0.07 | 0.28 | 0.98 | | 0.252 | |
| 35 | Br | [Ar] 4s ² 3d ¹⁰ 4p ⁵ | 0.014 | 0.098 | 0.294 | 0.84 | | 0.07 | 0.266 | 0.91 | | 0.238 | |
| 36 | Kr | [Ar] 4s ² 3d ¹⁰ 4p ⁶ | 0.014 | 0.098 | 0.28 | 0.798 | | 0.07 | 0.252 | 0.854 | | 0.224 | |
| 37 | Rb | [Kr] 5s ¹ | 0.014 | 0.098 | 0.266 | 0.756 | 2.52 | 0.07 | 0.252 | 0.784 | | 0.224 | |
| 38 | Sr | [Kr] 5s ² | 0.014 | 0.084 | 0.266 | 0.714 | 2.17 | 0.07 | 0.238 | 0.728 | | 0.21 | |
| 39 | Y | [Kr] 5s ² 4d ¹ | 0.014 | 0.084 | 0.252 | 0.672 | 2.002 | 0.056 | 0.224 | 0.686 | | 0.196 | 1.008 |
| 40 | Zr | [Kr] 5s ² 4d ² | 0.014 | 0.084 | 0.252 | 0.644 | 1.89 | 0.056 | 0.224 | 0.644 | | 0.196 | 0.896 |
| 41 | Nb | [Kr] 5s ¹ 4d ⁴ | 0.014 | 0.084 | 0.238 | 0.616 | 1.876 | 0.056 | 0.21 | 0.616 | | 0.182 | 0.84 |
| 42 | Mo | [Kr] 5s ¹ 4d ⁵ | 0.014 | 0.084 | 0.238 | 0.602 | 1.806 | 0.056 | 0.21 | 0.588 | | 0.182 | 0.77 |
| 43 | Tc | [Kr] 5s ¹ 4d ⁵ | 0.014 | 0.084 | 0.224 | 0.574 | 1.666 | 0.056 | 0.196 | 0.56 | | 0.168 | 0.714 |
| 44 | Ru | [Kr] 5s ² 4d ⁵ | 0.014 | 0.084 | 0.224 | 0.56 | 1.694 | 0.056 | 0.196 | 0.532 | | 0.168 | 0.686 |
| 45 | Rh | [Kr] 5s ¹ 4d ⁷ | 0.014 | 0.07 | 0.21 | 0.532 | 1.652 | 0.056 | 0.182 | 0.518 | | 0.168 | 0.644 |
| 46 | Pd | [Kr] 5s ¹ 4d ⁸ | 0.014 | 0.07 | 0.21 | 0.518 | 1.722 | 0.056 | 0.182 | 0.49 | | 0.154 | 0.616 |
| 47 | Ag | [Kr] 4d ¹⁰ | 0.014 | 0.07 | 0.21 | 0.504 | 1.582 | 0.056 | 0.182 | 0.476 | | 0.154 | 0.588 |
| 48 | Cd | [Kr] 5s ² 4d ¹⁰ | 0.014 | 0.07 | 0.196 | 0.49 | 1.47 | 0.042 | 0.168 | 0.462 | | 0.154 | 0.546 |
| 49 | In | [Kr] 5s ² 4d ¹⁰ 5p ¹ | 0.014 | 0.07 | 0.196 | 0.476 | 1.358 | 0.042 | 0.168 | 0.448 | 1.47 | 0.14 | 0.518 |
| 50 | Sn | [Kr] 5s ² 4d ¹⁰ 5p ² | 0.014 | 0.07 | 0.196 | 0.462 | 1.274 | 0.042 | 0.196 | 0.434 | 1.344 | 0.14 | 0.504 |

Table 4. The choices of the cutoff radius r_c (Å) for each states of $z=51$ to $z=75$ atoms.

| z | atom | electronic state | 1s | 2s | 3s | 4s | 5s | 6s | 2p | 3p | 4p | 5p | 3d | 4d | 5d | 4f |
|----|------|---|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| 51 | Sb | [Kr] 5s ² 4d ¹⁰ 5p ³ | 0.014 | 0.07 | 0.182 | 0.448 | 1.204 | | 0.042 | 0.154 | 0.42 | 1.232 | 0.14 | 0.476 | | |
| 52 | Te | [Kr] 5s ² 4d ¹⁰ 5p ⁴ | 0.014 | 0.07 | 0.182 | 0.434 | 1.134 | | 0.042 | 0.154 | 0.406 | 1.148 | 0.14 | 0.448 | | |
| 53 | I | [Kr] 5s ² 4d ¹⁰ 5p ⁵ | 0.014 | 0.07 | 0.182 | 0.42 | 1.078 | | 0.042 | 0.154 | 0.392 | 1.078 | 0.126 | 0.434 | | |
| 54 | Xe | [Kr] 5s ² 4d ¹⁰ 5p ⁶ | 0.014 | 0.07 | 0.182 | 0.42 | 1.036 | | 0.042 | 0.154 | 0.378 | 1.022 | 0.126 | 0.42 | | |
| 55 | Cs | [Xe] 6s ¹ | 0.014 | 0.07 | 0.168 | 0.406 | 0.98 | 2.912 | 0.042 | 0.14 | 0.364 | 0.952 | 0.126 | 0.406 | | |
| 56 | Ba | [Xe] 6s ² | 0.014 | 0.07 | 0.168 | 0.392 | 0.938 | 2.59 | 0.042 | 0.14 | 0.35 | 0.896 | 0.126 | 0.392 | | |
| 57 | La | [Xe] 5d ¹ 6s ² | 0.014 | 0.056 | 0.168 | 0.378 | 0.896 | 2.422 | 0.042 | 0.14 | 0.336 | 0.854 | 0.112 | 0.378 | 1.246 | |
| 58 | Ce | [Xe] 4f ¹ 5d ¹ 6s ² | 0.014 | 0.056 | 0.168 | 0.378 | 0.882 | 2.408 | 0.042 | 0.14 | 0.336 | 0.826 | 0.112 | 0.364 | 1.246 | 0.406 |
| 59 | Pr | [Xe] 4f ³ 6s ² | 0.014 | 0.056 | 0.168 | 0.364 | 0.854 | 2.282 | 0.042 | 0.126 | 0.322 | 0.798 | 0.112 | 0.364 | | 0.378 |
| 60 | Nd | [Xe] 4f ⁴ 6s ² | 0.014 | 0.056 | 0.154 | 0.364 | 0.854 | 2.436 | 0.042 | 0.126 | 0.322 | 0.798 | 0.112 | 0.35 | | 0.378 |
| 61 | Pm | [Xe] 4f ⁵ 6s ² | 0.014 | 0.056 | 0.154 | 0.35 | 0.84 | 2.422 | 0.042 | 0.126 | 0.308 | 0.77 | 0.112 | 0.336 | | 0.35 |
| 62 | Sm | [Xe] 4f ⁶ 6s ² | 0.014 | 0.056 | 0.154 | 0.35 | 0.826 | 2.394 | 0.028 | 0.126 | 0.294 | 0.756 | 0.112 | 0.336 | | 0.35 |
| 63 | Eu | [Xe] 4f ⁷ 6s ² | 0.014 | 0.056 | 0.154 | 0.336 | 0.812 | 2.38 | 0.028 | 0.126 | 0.294 | 0.742 | 0.098 | 0.322 | | 0.336 |
| 64 | Gd | [Xe] 4f ⁷ 5d ¹ 6s ² | 0.014 | 0.056 | 0.154 | 0.336 | 0.798 | 2.254 | 0.028 | 0.112 | 0.28 | 0.714 | 0.098 | 0.308 | 1.106 | 0.322 |
| 65 | Tb | [Xe] 4f ⁹ 6s ² | 0.014 | 0.056 | 0.14 | 0.336 | 0.784 | 2.324 | 0.028 | 0.112 | 0.28 | 0.7 | 0.098 | 0.308 | | 0.308 |
| 66 | Dy | [Xe] 4f ¹⁰ 6s ² | 0.014 | 0.056 | 0.14 | 0.322 | 0.784 | 2.31 | 0.028 | 0.112 | 0.28 | 0.686 | 0.098 | 0.294 | | 0.308 |
| 67 | Ho | [Xe] 4f ¹¹ 6s ² | 0.014 | 0.056 | 0.14 | 0.322 | 0.784 | 2.352 | 0.028 | 0.112 | 0.266 | 0.294 | 0.098 | 0.294 | | 0.294 |
| 68 | Er | [Xe] 4f ¹² 6s ² | 0.014 | 0.056 | 0.14 | 0.308 | 0.742 | 1.96 | 0.028 | 0.112 | 0.266 | 0.644 | 0.098 | 0.294 | | 0.28 |
| 69 | Tm | [Xe] 4f ¹³ 6s ² | 0.014 | 0.056 | 0.14 | 0.308 | 0.756 | 2.212 | 0.028 | 0.112 | 0.252 | 0.658 | 0.098 | 0.28 | | 0.294 |
| 70 | Yb | [Xe] 4f ¹⁴ 6s ² | 0.014 | 0.056 | 0.14 | 0.308 | 0.714 | 1.862 | 0.028 | 0.098 | 0.252 | 0.616 | 0.084 | 0.28 | | 0.266 |
| 71 | Lu | [Xe] 4f ¹⁴ 6s ² 5d ¹ | 0.014 | 0.056 | 0.14 | 0.294 | 0.728 | 2.142 | 0.028 | 0.098 | 0.252 | 0.616 | 0.084 | 0.266 | 1.036 | 0.266 |
| 72 | Hf | [Xe] 4f ¹⁴ 6s ² 5d ² | 0.014 | 0.056 | 0.14 | 0.294 | 0.7 | 2.044 | 0.028 | 0.098 | 0.238 | 0.602 | 0.084 | 0.266 | 0.952 | 0.252 |
| 73 | Ta | [Xe] 4f ¹⁴ 6s ² 5d ³ | 0.014 | 0.056 | 0.126 | 0.294 | 0.686 | 1.96 | 0.028 | 0.098 | 0.238 | 0.574 | 0.084 | 0.266 | 0.896 | 0.252 |
| 74 | W | [Xe] 4f ¹⁴ 6s ² 5d ⁴ | 0.014 | 0.056 | 0.126 | 0.294 | 0.672 | 1.904 | 0.028 | 0.098 | 0.224 | 0.56 | 0.084 | 0.252 | 0.84 | 0.238 |
| 75 | Re | [Xe] 4f ¹⁴ 6s ² 5d ⁵ | 0.014 | 0.056 | 0.126 | 0.28 | 0.658 | 1.848 | 0.028 | 0.098 | 0.224 | 0.546 | 0.084 | 0.252 | 0.798 | 0.238 |

Table 5. The choices of the cutoff radius r_c (Å) for each states of $Z=76$ to $Z=102$ atoms.

| Z | atom | electronic state | 1s | 2s | 3s | 4s | 5s | 6s | 7s | 2p | 3p | 4p | 5p | 6p | 3d | 4d | 5d | 6d | 4f | 5f |
|-----|------|--|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| 76 | Os | [Xe] 4f ¹⁴ 6s ² 5d ⁶ | 0.014 | 0.056 | 0.126 | 0.28 | 0.644 | 1.806 | | 0.028 | 0.098 | 0.224 | 0.532 | | 0.084 | 0.252 | 0.77 | | 0.224 | |
| 77 | Ir | [Xe] 4f ¹⁴ 6s ² 5d ⁷ | 0.014 | 0.056 | 0.126 | 0.28 | 0.63 | 1.764 | | 0.028 | 0.084 | 0.21 | 0.518 | | 0.084 | 0.238 | 0.728 | | 0.224 | |
| 78 | Pt | [Xe] 4f ¹⁴ 6s ¹ 5d ⁹ | 0.014 | 0.042 | 0.126 | 0.266 | 0.616 | 1.694 | | 0.028 | 0.084 | 0.21 | 0.49 | | 0.084 | 0.238 | 0.7 | | 0.21 | |
| 79 | Au | [Xe] 4f ¹⁴ 6s ¹ 5d ¹⁰ | 0.014 | 0.042 | 0.126 | 0.266 | 0.602 | 1.806 | | 0.028 | 0.084 | 0.21 | 0.49 | | 0.084 | 0.238 | 0.686 | | 0.21 | |
| 80 | Hg | [Xe] 4f ¹⁴ 6s ² 5d ¹⁰ | 0.014 | 0.042 | 0.126 | 0.266 | 0.588 | 1.68 | | 0.028 | 0.084 | 0.21 | 0.476 | | 0.07 | 0.224 | 0.658 | | 0.21 | |
| 81 | Tl | [Xe] 4f ¹⁴ 6s ² 5d ¹⁰ 6p ¹ | 0.014 | 0.042 | 0.126 | 0.266 | 0.574 | 1.582 | | 0.028 | 0.084 | 0.196 | 0.462 | 1.316 | 0.07 | 0.224 | 0.63 | | 0.196 | |
| 82 | Pb | [Xe] 4f ¹⁴ 6s ² 5d ¹⁰ 6p ² | 0.014 | 0.042 | 0.112 | 0.252 | 0.56 | 1.498 | | 0.028 | 0.084 | 0.196 | 0.448 | 1.232 | 0.07 | 0.224 | 0.602 | | 0.196 | |
| 83 | Bi | [Xe] 4f ¹⁴ 6s ² 5d ¹⁰ 6p ³ | 0.014 | 0.042 | 0.112 | 0.252 | 0.546 | 1.442 | | 0.028 | 0.084 | 0.196 | 0.434 | 1.162 | 0.07 | 0.21 | 0.588 | | 0.196 | |
| 84 | Po | [Xe] 4f ¹⁴ 6s ² 5d ¹⁰ 6p ⁴ | 0.014 | 0.042 | 0.112 | 0.252 | 0.546 | 1.372 | | 0.028 | 0.084 | 0.196 | 0.42 | 1.106 | 0.07 | 0.21 | 0.56 | | 0.182 | |
| 85 | At | [Xe] 4f ¹⁴ 6s ² 5d ¹⁰ 6p ⁵ | 0.014 | 0.042 | 0.112 | 0.252 | 0.532 | 1.33 | | 0.028 | 0.084 | 0.182 | 0.42 | 1.05 | 0.07 | 0.21 | 0.546 | | 0.182 | |
| 86 | Rn | [Xe] 4f ¹⁴ 6s ² 5d ¹⁰ 6p ⁶ | 0.014 | 0.042 | 0.112 | 0.238 | 0.518 | 1.288 | | 0.028 | 0.084 | 0.182 | 0.406 | 1.008 | 0.07 | 0.21 | 0.532 | | 0.182 | |
| 87 | Fr | [Rn] 7s ¹ | 0.014 | 0.042 | 0.112 | 0.238 | 0.504 | 1.204 | 3.304 | 0.028 | 0.07 | 0.182 | 0.392 | 0.952 | 0.07 | 0.196 | 0.518 | | 0.182 | |
| 88 | Ra | [Rn] 7s ² | 0.014 | 0.042 | 0.112 | 0.238 | 0.504 | 1.148 | 3.01 | 0.028 | 0.07 | 0.182 | 0.392 | 0.91 | 0.07 | 0.196 | 0.504 | | 0.168 | |
| 89 | Ac | [Rn] 7s ² 6d ¹ | 0.014 | 0.042 | 0.112 | 0.238 | 0.49 | 1.106 | 2.828 | 0.028 | 0.07 | 0.168 | 0.378 | 0.868 | 0.07 | 0.196 | 0.49 | 1.47 | 0.168 | |
| 90 | Th | [Rn] 7s ² 6d ² | 0.014 | 0.042 | 0.112 | 0.238 | 0.476 | 1.064 | 2.688 | 0.014 | 0.07 | 0.168 | 0.364 | 0.84 | 0.07 | 0.196 | 0.476 | 1.358 | 0.168 | |
| 91 | Pa | [Rn] 7s ² 5f ² 6d ¹ | 0.014 | 0.042 | 0.112 | 0.224 | 0.476 | 1.05 | 2.73 | 0.014 | 0.07 | 0.168 | 0.364 | 0.812 | 0.07 | 0.196 | 0.462 | 1.372 | 0.168 | 0.63 |
| 92 | U | [Rn] 7s ² 5f ³ 6d ¹ | 0.014 | 0.042 | 0.112 | 0.224 | 0.462 | 1.036 | 2.702 | 0.014 | 0.07 | 0.168 | 0.35 | 0.798 | 0.07 | 0.182 | 0.448 | 1.344 | 0.168 | 0.602 |
| 93 | Np | [Rn] 7s ² 5f ⁴ 6d ¹ | 0.014 | 0.042 | 0.112 | 0.224 | 0.462 | 1.022 | 2.674 | 0.014 | 0.07 | 0.168 | 0.35 | 0.77 | 0.07 | 0.182 | 0.434 | 1.316 | 0.154 | 0.574 |
| 94 | Pu | [Rn] 7s ² 5f ⁶ | 0.014 | 0.042 | 0.098 | 0.224 | 0.448 | 1.022 | 2.758 | 0.014 | 0.07 | 0.154 | 0.336 | 0.77 | 0.07 | 0.182 | 0.434 | | 0.154 | 0.56 |
| 95 | Am | [Rn] 7s ² 5f ⁷ | 0.014 | 0.042 | 0.098 | 0.224 | 0.448 | 1.008 | 2.744 | 0.014 | 0.07 | 0.154 | 0.336 | 0.742 | 0.056 | 0.182 | 0.42 | | 0.154 | 0.546 |
| 96 | Cm | [Rn] 7s ² 5f ⁷ 6d ¹ | 0.014 | 0.042 | 0.098 | 0.21 | 0.434 | 0.98 | 2.618 | 0.014 | 0.07 | 0.154 | 0.322 | 0.728 | 0.056 | 0.182 | 0.42 | 1.246 | 0.154 | 0.518 |
| 97 | Bk | [Rn] 7s ² 5f ⁹ | 0.014 | 0.042 | 0.098 | 0.21 | 0.434 | 0.98 | 2.716 | 0.014 | 0.07 | 0.154 | 0.322 | 0.714 | 0.056 | 0.168 | 0.406 | | 0.154 | 0.504 |
| 98 | Cf | [Rn] 7s ² 5f ¹⁰ | 0.014 | 0.042 | 0.098 | 0.21 | 0.42 | 0.966 | 2.702 | 0.014 | 0.07 | 0.154 | 0.322 | 0.7 | 0.056 | 0.168 | 0.392 | | 0.14 | 0.49 |
| 99 | Es | [Rn] 7s ² 5f ¹¹ | 0.014 | 0.042 | 0.098 | 0.21 | 0.42 | 0.952 | 2.688 | 0.014 | 0.07 | 0.154 | 0.308 | 0.686 | 0.056 | 0.168 | 0.392 | | 0.14 | 0.476 |
| 100 | Fm | [Rn] 7s ² 5f ¹² | 0.014 | 0.042 | 0.098 | 0.21 | 0.42 | 0.952 | 2.688 | 0.014 | 0.056 | 0.14 | 0.308 | 0.672 | 0.056 | 0.168 | 0.378 | | 0.14 | 0.462 |
| 101 | Md | [Rn] 7s ² 5f ¹³ | 0.014 | 0.042 | 0.098 | 0.21 | 0.406 | 0.938 | 2.674 | 0.014 | 0.056 | 0.14 | 0.294 | 0.658 | 0.056 | 0.168 | 0.378 | | 0.14 | 0.448 |
| 102 | No | [Rn] 7s ² 5f ¹⁴ | 0.014 | 0.042 | 0.098 | 0.21 | 0.406 | 0.924 | 2.66 | 0.014 | 0.056 | 0.14 | 0.294 | 0.644 | 0.056 | 0.168 | 0.364 | | 0.14 | 0.448 |

An intermediate pseudo potential V_1^{ps} :

$$V_1^{ps}(r) = \left[1 - f\left(\frac{r}{r_c}\right) \right] V_1(r) + c_1 f\left(\frac{r}{r_c}\right) \quad (3.24)$$

was first formed with the corresponding the radial Schrödinger equation:

$$-\frac{1}{2} \frac{d^2 w_1(r)}{d^2 r} + V_1^{ps}(r) w_1(r) = \varepsilon_1 w_1(r), \quad (3.25)$$

where introducing a cut-off function $f(x)$ which approaches 0 as $x \rightarrow \infty$, approaches 1 at least as fast as x^3 as $x \rightarrow 0$, and cuts off for $x \approx 1$. A convenient choice was:

$$f(x) = \exp(-x^4)$$

The constant c_1 was adjusted so that the **nodeless** bound solution w_1 of the radial Schrödinger equation (3.25) with $V_1^{ps}(r)$ has energy ε_1 equal to the original eigenvalue of the valence state ε_{nl} . By using the secant method, we can do loop to get c_1 .

Property (1) of the subsection 2.4 is now automatically satisfied. Further, property (2) of subsection 2.4 is satisfied within a multiplicative constant γ_1 for normalized function w_1 :

$$\gamma_1 w_1(r) \xrightarrow{r > r_c} p_{nl}(r) \quad (3.26)$$

Since both satisfy the same differential equation for $r > r_c$ and with the same homogeneous boundary condition. Here we choose

$$\gamma_1 = \frac{P_{nl}(a)}{w_1(a)},$$

where a is about the middle point ($\approx 5 \text{ \AA}$) of the domain (0~10 \AA).

To satisfy properties (2)-(4) of the subsection 2.4, the intermediate pseudo wave function w_1 is modified to

$$w_2(r) = \gamma_1 \left[w_1(r) + \delta_l g_l \left(\frac{r}{r_c} \right) \right], \quad (3.27)$$

where another cut off function $g_l(x)$ was introduced, which cuts off to zero for $x > 1$, and behaves as x^{l+1} at small x . The convenient choice is written as:

$$g_l(x) = x^{l+1} \exp(-x^4).$$

The asymptotic behavior of $f(x)$ and $g_l(x)$ guarantees the potential to be finite at the origin. We choose δ_l that is the smaller solution of the quadratic equation of the

normalization condition:

$$\int_0^\infty \gamma_1^2 \left[w_1(r) + \delta_l g_l \left(\frac{r}{r_c} \right) \right]^2 dr = 1 \quad (3.28)$$

Expanding the equation (3.28), we have:

$$\left[\gamma_1^2 \int_0^\infty g_l^2 \left(\frac{r}{r_c} \right) dr \right] \delta_l^2 + \left[2\gamma_1^2 \int_0^\infty w_1(r) g_l \left(\frac{r}{r_c} \right) dr \right] \delta_l + \left[\gamma_1^2 \int_0^\infty w_1^2(r) dr - 1 \right] = 0 \quad (3.29)$$

Solving the quadratic equation (3.29), we can get smaller solution δ_l .

Notice that this condition (3.28) guarantees that the norm conservation condition because for w_2 with P_{nl} since both function coincide for $r > r_c$. The pseudopotential $V_{ps}(r)$ producing the **nodeless** eigenfunction $w_2(r)$ at eigenvalue ϵ_{nl} is now found by inverting the radial

Schrödinger equation:

$$V_{ps}(r) = \left[\epsilon_{nl} w_2(r) + \frac{1}{2} \frac{d^2 w_2(r)}{d^2 r} \right] \frac{1}{w_2(r)} \quad (3.30)$$

To form the final bare-ion pseudopotential, the valence pseudo charge density is found by using the wave functions w_2 in the same configuration as the original atom calculation. The Hartree and exchange-correlation potentials due to this density are then calculated and subtracted from each V_{ps} . Analytical expressions containing a few parameters can subsequently be fitted to the numerical potential functions.



A Appendix

A.1 The main program of No atom

```
module toole
implicit none
real(kind=8)::z1=102.0 !atomic number of NO
real::a0=0.529 !Bohr radius
real,parameter::pi=3.14159265
integer,parameter::g=400
real,parameter::h=10.0/real(g) !切割間距
real,parameter::h1=10.0 !範圍 0-h1
real(kind=8)
a(g,g),v_l(g),v1l_ps(g),v2l_ps(g),e_nl,r_c,c_l,r1,delta_l,w10(0:g),w20(0:g),pnl(0:g)

real(kind=8)
v(g),r_s(0:g),v_xc(0:g),v_ee(0:g),v_eff(g),e_xc(0:g),e_x(0:g),e_c(0:g),electron,electron1
,ev0(7),ev1(6),ev2(4),ev3(2)

real(kind=8)
k(0:g),p_10(0:g),p_20(0:g),p_30(0:g),p_40(0:g),p_21(0:g),p_31(0:g),p_32(0:g),p_41(0:g)
,p_50(0:g),p_42(0:g),p_51(0:g),p_60(0:g),p_52(0:g)

real(kind=8) p_43(0:g),p_61(0:g),p_70(0:g),p_62(0:g),p_53(0:g)

real(kind=8)
f(0:g,0:g),f1(0:g),diff_r_s(0:g),f2(g,0:g),f2_1(0:g),f3(0:g),f4(0:g),n(0:g),n_out(0:g),dex4(0:g)
,dex5(0:g)

real(kind=8)
dex_10(g),dex_20(g),dex_30(g),dex_40(g),dex_21(g),dex_31(g),dex_32(g),dex_41(g)
,dex_50(g),dex_42(g),dex_51(g),dex_60(g),dex_52(g)

real(kind=8) dex_43(g),dex_61(g),dex_70(g),dex_62(g),dex_53(g)
integer key,ierr
double precision w(g-1),e(g-1),z(g-1,g-1)
real::l
```

```
real(kind=8) kinetic_energy,E_xc_functional,external_energy,coulomb_energy,total_energy
end module
```

```
program main
use toole
implicit none
real(kind=8) norm,norm1
integer i,count,error
real*8,external::normal_P
logical alive
```

```
call gp()
```

```
inquire(file="DENSITY MD.txt",exist=alive)
if (.not. alive) then
write(*,*) trim("DENSITY MD.txt"),"doesn't exist."
end if
open(10,file="DENSITY MD.txt")
do while(.true.)
read(10,*,iostat=error)n
if (error/=0) exit
end do
close(10)
```



```
do while (.true.)
norm=0.0
norm1=0.0
call sub11()
call sub12()
if (abs(electron-z1)<1e-3) then
n=n
else
n=(z1/electron)*n
end if
```

```

call radius() !the radius r_s of a unit charge
call ee()      !the exchange energy e_x
call ce()      !the correlation energy e_c
call ece()     !the exchange-correlation energy per unit density e_xc
call sub0()    !diff. of r_s
call ecp()     !The exchange-correlation potential v_xc
call sub10()   !計算 electron-electron coulomb potential 中的項
call sub10_1()
call eecp()    !the electron-electron coulomb potential v_ee
call ep()      !external potential v
call effp()    !effective potential v_eff

```

```
l=0.0
```

```
call computing_ev_evt() !計算 eigenvalue & eigenvector
```

```

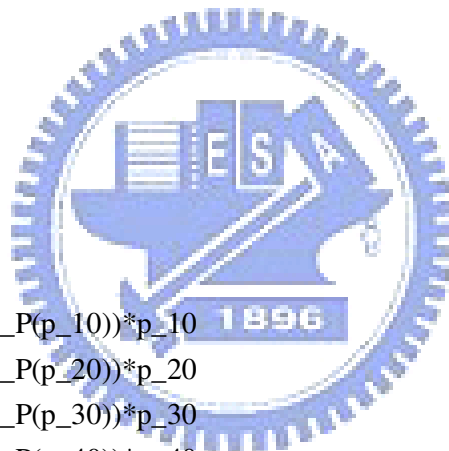
p_10(1:g-1)=z(:,1)
p_20(1:g-1)=z(:,2)
p_30(1:g-1)=z(:,3)
p_40(1:g-1)=z(:,4)
p_50(1:g-1)=z(:,5)
p_60(1:g-1)=z(:,6)
p_70(1:g-1)=z(:,7)
p_10=sqrt(1.0/normal_P(p_10))*p_10
p_20=sqrt(1.0/normal_P(p_20))*p_20
p_30=sqrt(1.0/normal_P(p_30))*p_30
p_40=sqrt(1.0/normal_P(p_40))*p_40
p_50=sqrt(1.0/normal_P(p_50))*p_50
p_60=sqrt(1.0/normal_P(p_60))*p_60
p_70=sqrt(1.0/normal_P(p_70))*p_70
ev0(1)=w(1)
ev0(2)=w(2)
ev0(3)=w(3)
ev0(4)=w(4)
ev0(5)=w(5)
ev0(6)=w(6)
ev0(7)=w(7)

```

```
l=1.0
```

```
call computing_ev_evt() !計算 eigenvalue & eigenvector
```

```
p_21(1:g-1)=z(:,1)
```




```

p_31(1:g-1)=z(:,2)
p_41(1:g-1)=z(:,3)
p_51(1:g-1)=z(:,4)
p_61(1:g-1)=z(:,5)
p_21=sqrt(1.0/normal_P(p_21))*p_21
p_31=sqrt(1.0/normal_P(p_31))*p_31
p_41=sqrt(1.0/normal_P(p_41))*p_41
p_51=sqrt(1.0/normal_P(p_51))*p_51
p_61=sqrt(1.0/normal_P(p_61))*p_61
ev1(1)=w(1)
ev1(2)=w(2)
ev1(3)=w(3)
ev1(4)=w(4)
ev1(5)=w(5)

```

l=2.0

call computing_ev_evt() !計算 eigenvalue & eigenvector

```

p_32(1:g-1)=z(:,1)
p_42(1:g-1)=z(:,2)
p_52(1:g-1)=z(:,3)
p_62(1:g-1)=z(:,4)
p_32=sqrt(1.0/normal_P(p_32))*p_32
p_42=sqrt(1.0/normal_P(p_42))*p_42
p_52=sqrt(1.0/normal_P(p_52))*p_52
p_62=sqrt(1.0/normal_P(p_62))*p_62
ev2(1)=w(1)
ev2(2)=w(2)
ev2(3)=w(3)
ev2(4)=w(4)

```

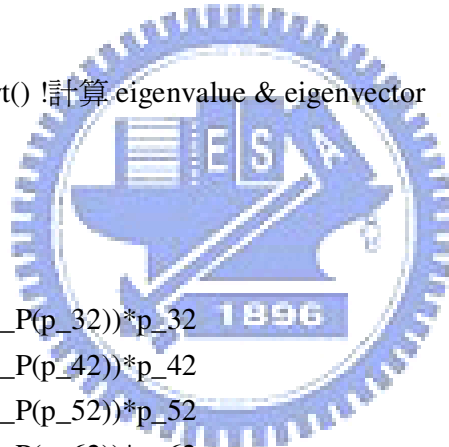
l=3.0

call computing_ev_evt() !計算 eigenvalue & eigenvector

```

p_43(1:g-1)=z(:,1)
p_53(1:g-1)=z(:,2)
p_43=sqrt(1.0/normal_P(p_43))*p_43
p_53=sqrt(1.0/normal_P(p_53))*p_53
ev3(1)=w(1)
ev3(2)=w(2)

```



```

call density_out() !density n_out

do i=0,g
    norm=norm+(n(i)-n_out(i))**2.0
    norm1=norm1+n(i)**2.0
end do
if(sqrt(norm/norm1)<1e-4) then

    exit
else
    do i=0,g
        n(i)=0.5*n_out(i)+0.5*n(i)
    end do
end if

end do

open(unit=10,file='energy.txt')
call E_G() !Total energy E_G[n]

write(10,*) "動能=",kinetic_energy
write(10,*) "coulomb_energy=",coulomb_energy
write(10,*) "exchange-correlation fuctional E_xc=",E_xc_functional
write(10,*) "external_energy=",external_energy
write(10,*) "總能=",total_energy
write(10,*) "誤差=",(abs(-total_energy-32790)/32790)*100.0,"%

write(*,*) "動能=",kinetic_energy
write(*,*) "coulomb_energy=",coulomb_energy
write(*,*) "exchange-correlation fuctional E_xc=",E_xc_functional
write(*,*) "external_energy=",external_energy
write(*,*) "總能=",total_energy

open(unit=10,file='DENSITY NO.txt')
write(10,*) (n_out(i),i=0,g)

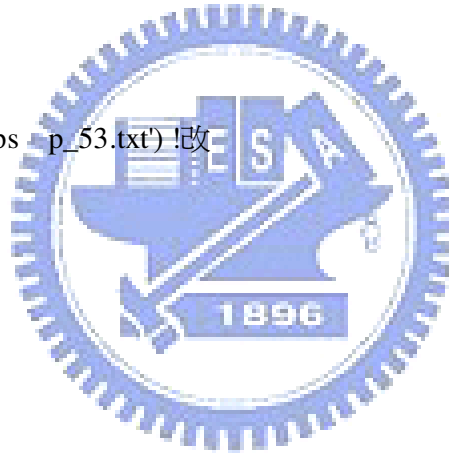
```

```

l=3.0 !改
pnl=p_53 !改
e_nl=ev3(2) !改
call find_c_l01()
call sub15()
call sub16()
call computing_ev_evt_ps()
w10(1:g-1)=z(:,1)
w10=sqrt(1.0/normal_P(w10))*w10
r1=p_53(200)/w10(200) !改
call find_delta_l()
call sub_w2_0()
call pseudo()
open(unit=10,file='w20 p_53.txt') !改
  do i=0,g
    write(10,*)w20(i)
  end do
open(unit=10,file='v21_ps p_53.txt') !改
  do i=1,g-1
    write(10,*)v21_ps(i)
  end do

end

```



```

subroutine gp() !grid point
use toole
implicit none
integer i
  do i=0,g
    k(i)=(real(i)/real(g))*h1
  end do
end

```

```

subroutine radius() !the radius r_s of a unit charge
use toole
implicit none
integer i
do i=0,g-1
  r_s(i)=(3.0/(4.0*pi*n(i)))**(1.0/3.0)
end do
end

```

```

subroutine ee() !the exchange energy
use toole
implicit none
integer i
do i=0,g-1
  e_x(i)=-0.458/r_s(i)
end do
end

```



```

subroutine ce() !the correlation energy
use toole
implicit none
integer i
do i=0,g-1
  if (r_s(i)>=1.0) then
    e_c(i)=-0.1423/(1.0+1.0529*sqrt(r_s(i))+0.3334*r_s(i))
  else if (r_s(i)<1.0) then
    e_c(i)=0.0311*log(r_s(i))+(-0.0480)+0.002*r_s(i)*log(r_s(i))+(-0.0116)*r_s(i)
  end if
end do
end

```

```

subroutine ece() !the exchange-correlation energy per unit density
use toole
implicit none
integer i
do i=0,g-1
  e_xc(i)=e_x(i)+e_c(i)
end do
end

```

```

end do
end

```

```

subroutine sub0() !diff. of r_s
use toole
implicit none
integer i
do i=0,g-1
diff_r_s(i)=-4.0*pi*3.0**(-2.0/3.0)*(4.0*pi*n(i))**(-4.0/3.0)
end do
end

```

```

subroutine ecp() !The exchange-correlation potential v_xc
use toole
implicit none
integer i
do i=0,g-1

```

```

if (r_s(i)>=1.0) then
v_xc(i)=e_xc(i)+n(i)*(0.458*r_s(i)**(-2.0)*diff_r_s(i)+0.1423*(1.0+1.0529*r_s(i)**(0.5)+0.3334*r_s(i))**(-2.0)*(0.5*1.0529*r_s(i)**(-0.5)*diff_r_s(i)+0.3334*diff_r_s(i)))

```

```

else if (r_s(i)<1.0) then
v_xc(i)=e_xc(i)+n(i)*(0.458*r_s(i)**(-2.0)*diff_r_s(i)+0.0311*r_s(i)**(-1.0)*diff_r_s(i)+0.002*diff_r_s(i)*log(r_s(i))+0.002*diff_r_s(i)-0.0116*diff_r_s(i))

```

```

end if
end do
end

```

```

subroutine ep() !external potential v
use toole
implicit none
integer i
do i=1,g
v(i)=-z1/k(i)

```

```

end do
end

```

```

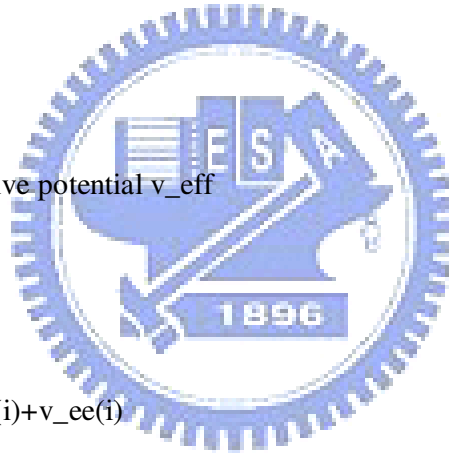
subroutine eecp() !The electron-electron coulomb potential V_ee
use toole          !v_ee(r)= ∫ (0.0,10.0)n(r1)*(r1**2.0)*1.0/max(r,r1)
implicit none      !f2 from sub10()
integer i
real,external::trape_integral4,trape_integral5
do i=0,g
  if(i==0) then
    v_ee(i)=4.0*pi*trape_integral4(f2_1,h,0)
  else
    v_ee(i)=4.0*pi*trape_integral5(f2,h,i)
  end if
end do
end

```

```

subroutine effp() !effective potential v_eff
use toole
implicit none
integer i
do i=1,g-1
  v_eff(i)=v(i)+v_xc(i)+v_ee(i)
end do
end

```



```

real function trape_integral3(datas2,width,j) !simpson 積分
use toole
implicit none
real(kind=8) datas2(0:g,0:g)
real width
real sum
integer i,j
sum=(datas2(0,j)+datas2(g,j))
do i=1,g-1
  if (mod(i,2)==0) then
    sum=sum+4.0*datas2(i,j)
  else

```

```

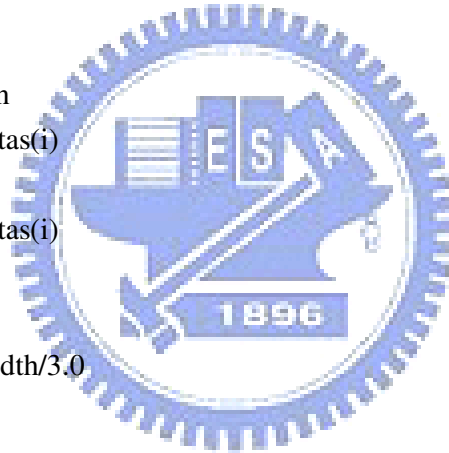
        sum=sum+2.0*datas2(i,j)
    end if
end do
trape_integral3=sum*width/3.0
return
end

```

```

real function trape_integral4(datas,width,j) !simpson 積分
use toole
implicit none
real(kind=8) datas(0:g)
real width
real sum
integer i,j
sum=(datas(j)+datas(g))
do i=j+1,g-1
    if (mod(i,2)==0) then
        sum=sum+4.0*datas(i)
    else
        sum=sum+2.0*datas(i)
    end if
end do
trape_integral4=sum*width/3.0
return
end

```



```

real function trape_integral41(datas,width,j) !simpson 積分
use toole
implicit none
real(kind=8) datas(g)
real width
real sum
integer i,j
sum=(datas(j)+datas(g-1))
do i=j+1,g-2
    if (mod(i,2)==0) then
        sum=sum+4.0*datas(i)
    else
        sum=sum+2.0*datas(i)
    end if
end do

```

```

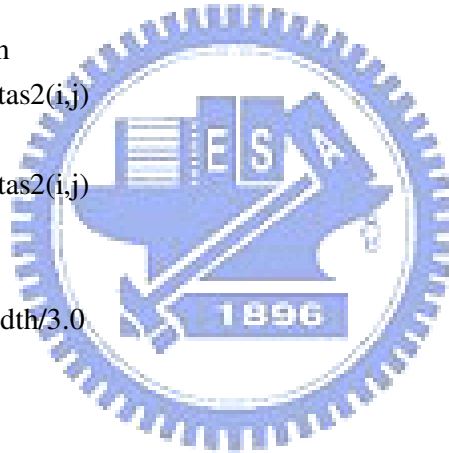
    end if
  end do
  trape_integral41=sum*width/3.0
  return
end

```

```

real function trape_integral5(datas2,width,i) !simpson 積分
use toole
implicit none
real(kind=8) datas2(g,0:g)
real width
real sum
integer i,j
sum=(datas2(i,0)+datas2(i,g))
do j=1,g-1
  if (mod(j,2)==0) then
    sum=sum+4.0*datas2(i,j)
  else
    sum=sum+2.0*datas2(i,j)
  end if
end do
trape_integral5=sum*width/3.0
return
end

```



```

subroutine computing_ev_evt()
use toole
implicit none
integer:: i,j
do j=1,g-1 !離散化之後產生的矩陣
  do i=1,g-1
    if (i==j) then
      a(i,j)=(1.0/(h**2.0)+1*(l+1.0)/(2.0*k(i)**2.0)+v_eff(i))
    else if (abs(i-j)==1) then
      a(i,j)=(-1.0/(2.0*h**2.0))
    else
      a(i,j)=0.0
    end if
  end do
end do

```

!計算 eigenvalue and eigenvector


```

    end do
end do

do i=1,g-1
    w(i)=a(i,i)
end do

do i=2,g-1
    e(i)=(-1.0/(2.0*h**2.0))
end do
call rst(g-1,g-1,w,e,1,z,ierr)

end subroutine

```

```

subroutine sub3() !計算動能的項
use toole
implicit none
integer i
l=0.0
do i=1,g-1
    dex_10(i)=(-1.0/2.0)*p_10(i)*(p_10(i+1)-2.0*p_10(i)+p_10(i-1))/(h**2.0)&
        &+(1*(l+1.0)/(2.0*k(i)**2.0))*p_10(i)**2.0
end do

do i=1,g-1
    dex_20(i)=(-1.0/2.0)*p_20(i)*(p_20(i+1)-2.0*p_20(i)+p_20(i-1))/(h**2.0)&
        &+(1*(l+1.0)/(2.0*k(i)**2.0))*p_20(i)**2.0

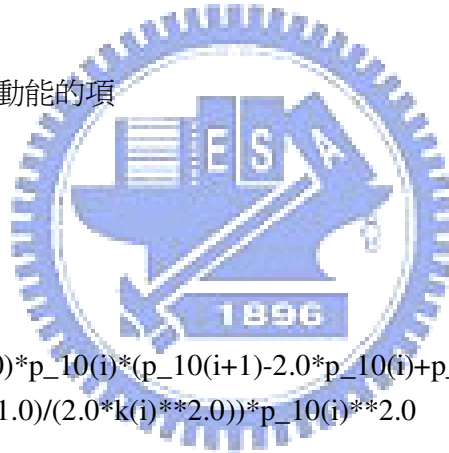
end do

do i=1,g-1
    dex_30(i)=(-1.0/2.0)*p_30(i)*(p_30(i+1)-2.0*p_30(i)+p_30(i-1))/(h**2.0)&
        &+(1*(l+1.0)/(2.0*k(i)**2.0))*p_30(i)**2.0

end do

do i=1,g-1

```



```
dex_40(i)=(-1.0/2.0)*p_40(i)*(p_40(i+1)-2.0*p_40(i)+p_40(i-1))/(h**2.0)&
&+(1*(l+1.0)/(2.0*k(i)**2.0))*p_40(i)**2.0
```

```
end do
```

```
do i=1,g-1
```

```
dex_50(i)=(-1.0/2.0)*p_50(i)*(p_50(i+1)-2.0*p_50(i)+p_50(i-1))/(h**2.0)&
&+(1*(l+1.0)/(2.0*k(i)**2.0))*p_50(i)**2.0
```

```
end do
```

```
do i=1,g-1
```

```
dex_60(i)=(-1.0/2.0)*p_60(i)*(p_60(i+1)-2.0*p_60(i)+p_60(i-1))/(h**2.0)&
&+(1*(l+1.0)/(2.0*k(i)**2.0))*p_60(i)**2.0
```

```
end do
```

```
do i=1,g-1
```

```
dex_70(i)=(-1.0/2.0)*p_70(i)*(p_70(i+1)-2.0*p_70(i)+p_70(i-1))/(h**2.0)&
&+(1*(l+1.0)/(2.0*k(i)**2.0))*p_70(i)**2.0
```

```
end do
```

```
l=1.0
```

```
do i=1,g-1
```

```
dex_21(i)=(-1.0/2.0)*p_21(i)*(p_21(i+1)-2.0*p_21(i)+p_21(i-1))/(h**2.0)&
&+(1*(l+1.0)/(2.0*k(i)**2.0))*p_21(i)**2.0
```

```
end do
```

```
do i=1,g-1
```

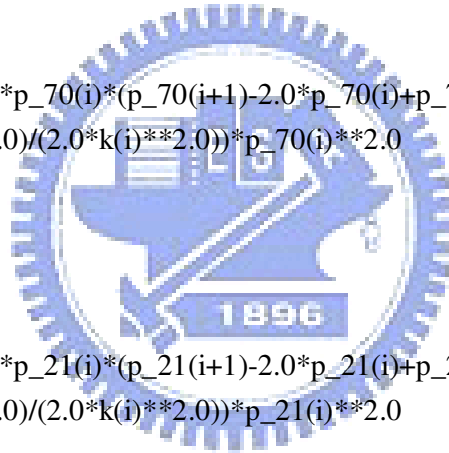
```
dex_31(i)=(-1.0/2.0)*p_31(i)*(p_31(i+1)-2.0*p_31(i)+p_31(i-1))/(h**2.0)&
&+(1*(l+1.0)/(2.0*k(i)**2.0))*p_31(i)**2.0
```

```
end do
```

```
do i=1,g-1
```

```
dex_41(i)=(-1.0/2.0)*p_41(i)*(p_41(i+1)-2.0*p_41(i)+p_41(i-1))/(h**2.0)&
&+(1*(l+1.0)/(2.0*k(i)**2.0))*p_41(i)**2.0
```

```
end do
```



```
do i=1,g-1
  dex_51(i)=(-1.0/2.0)*p_51(i)*(p_51(i+1)-2.0*p_51(i)+p_51(i-1))/(h**2.0)&
    &+(1*(1+1.0)/(2.0*k(i)**2.0))*p_51(i)**2.0
```

```
end do
```

```
do i=1,g-1
  dex_61(i)=(-1.0/2.0)*p_61(i)*(p_61(i+1)-2.0*p_61(i)+p_61(i-1))/(h**2.0)&
    &+(1*(1+1.0)/(2.0*k(i)**2.0))*p_61(i)**2.0
```

```
end do
```

```
l=2.0
```

```
do i=1,g-1
  dex_32(i)=(-1.0/2.0)*p_32(i)*(p_32(i+1)-2.0*p_32(i)+p_32(i-1))/(h**2.0)&
    &+(1*(1+1.0)/(2.0*k(i)**2.0))*p_32(i)**2.0
```

```
end do
```

```
do i=1,g-1
  dex_42(i)=(-1.0/2.0)*p_42(i)*(p_42(i+1)-2.0*p_42(i)+p_42(i-1))/(h**2.0)&
    &+(1*(1+1.0)/(2.0*k(i)**2.0))*p_42(i)**2.0
```

```
end do
```

```
do i=1,g-1
  dex_52(i)=(-1.0/2.0)*p_52(i)*(p_52(i+1)-2.0*p_52(i)+p_52(i-1))/(h**2.0)&
    &+(1*(1+1.0)/(2.0*k(i)**2.0))*p_52(i)**2.0
```

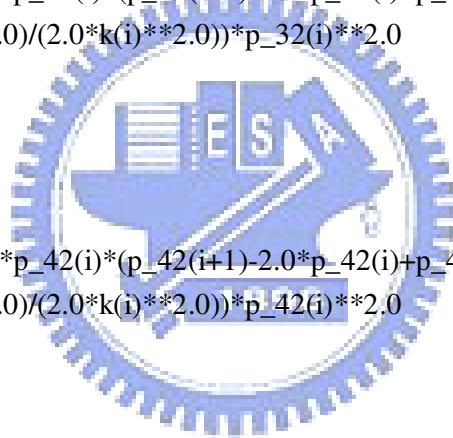
```
end do
```

```
do i=1,g-1
  dex_62(i)=(-1.0/2.0)*p_62(i)*(p_62(i+1)-2.0*p_62(i)+p_62(i-1))/(h**2.0)&
    &+(1*(1+1.0)/(2.0*k(i)**2.0))*p_62(i)**2.0
```

```
end do
```

```
l=3.0
```

```
do i=1,g-1
```



```

dex_43(i)=(-1.0/2.0)*p_43(i)*(p_43(i+1)-2.0*p_43(i)+p_43(i-1))/(h**2.0)&
&+(1*(1+1.0)/(2.0*k(i)**2.0))*p_43(i)**2.0

end do

do i=1,g-1
dex_53(i)=(-1.0/2.0)*p_53(i)*(p_53(i+1)-2.0*p_53(i)+p_53(i-1))/(h**2.0)&
&+(1*(1+1.0)/(2.0*k(i)**2.0))*p_53(i)**2.0

end do
end

subroutine KE() !動能 T
use toole
implicit none
real,external::trape_integral41
kinetic_energy=2.0*trape_integral41(dex_10,h,1)+2.0*trape_integral41(dex_20,h,1)+6.0*trap
e_integral41(dex_21,h,1)+2.0*trape_integral41(dex_30,h,1)&
&+6.0*trape_integral41(dex_31,h,1)+2.0*trape_integral41(dex_40,h,1)+10.0
*trape_integral41(dex_32,h,1)+6.0*trape_integral41(dex_41,h,1)&
&+2.0*trape_integral41(dex_50,h,1)+10.0*trape_integral41(dex_42,h,1)+6.0
*trape_integral41(dex_51,h,1)+2.0*trape_integral41(dex_60,h,1)&
&+10.0*trape_integral41(dex_52,h,1)+14.0*trape_integral41(dex_43,h,1)+6.
0*trape_integral41(dex_61,h,1)+2.0*trape_integral41(dex_70,h,1)&
&+0.0*trape_integral41(dex_62,h,1)+14.0*trape_integral41(dex_53,h,1)
! |
end

subroutine sub4() !計算 exchange-correlation fuctional E_xc 中使用
use toole
implicit none
integer i
do i=0,g
if(i==g) then
dex4(i)=0.0
else
dex4(i)=e_xc(i)*n_out(i)*(k(i)**2.0)!
end if
end do
end do

```

```
end
```

```
subroutine Excfuntional() !exchange-correlation fuctional E_xc
use toole
implicit none
real,external::trape_integral4
E_xc_functional=4.0*pi*trape_integral4(dex4,h,0) !
End
```

```
subroutine sub5() !計算 external energy  $\int v(r)n(r)dr$  使用
use toole
implicit none
integer i
do i=0,g
    dex5(i)=z1*k(i)*n_out(i) !
end do
end
```

```
subroutine sub6() !計算總能中的  $\int v(r)n(r)dr$ 
use toole
implicit none
real,external::trape_integral4
external_energy=-4.0*pi*trape_integral4(dex5,h,0)
end
```



```
subroutine sub7() !計算 coulomb energy 使用
use toole
implicit none
integer i,j
do j=0,g
do i=0,g
    if(i==0 .and. j==0) then
        f(i,j)=0.0
    else
        f(i,j)=(k(i)**2.0)*(k(j)**2.0)*(1.0/max(k(i),k(j)))*n_out(i)*n_out(j)
    end if
end do
end do
```

```
end
```

```
subroutine sub8()      !計算 coulomb energy 使用
```

```
use toole
```

```
implicit none
```

```
integer j
```

```
real,external::trape_integral3
```

```
do j=0,g
```

```
    f1(j)=trape_integral3(f,h,j)
```

```
end do
```

```
end
```

```
subroutine sub9() !coulomb energy  $(1/2) * \int \int (Vee*n)dr$ 
```

```
use toole
```

```
implicit none
```

```
real,external::trape_integral4
```

```
coulomb_energy=0.5*(4.0*pi)**2.0*trape_integral4(f1,h,0)
```

```
end
```

```
subroutine sub10()
```

```
use toole
```

```
implicit none
```

```
integer i,j
```

```
do j=1,g
```

```
do i=1,g
```

```
    f2(i,j)=k(j)**2.0*n(j)*(1.0/max(k(i),k(j)))
```

```
end do
```

```
end do
```

```
end
```

```
subroutine sub10_1()
```

```
use toole
```

```
implicit none
```

```
integer i
```

```
do i=0,g
```

```
    f2_1(i)=k(i)*n(i)
```

```
end do
```



```
end
```

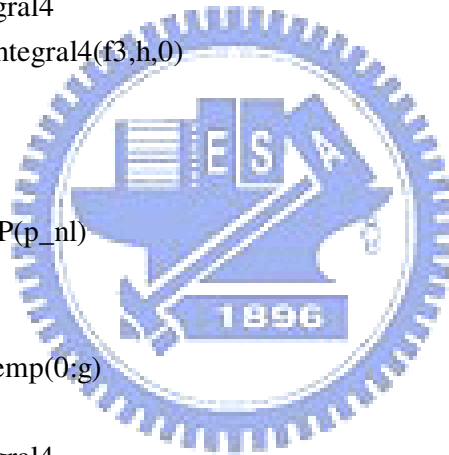
```
subroutine sub11()
use toole
implicit none
integer i
  do i=0,g
    f3(i)=k(i)**2.0*n(i)
  end do
end
```

```
subroutine sub12()
use toole
implicit none
real,external::trape_integral4
electron=4.0*pi*trape_integral4(f3,h,0)
end
```

```
real*8 function normal_P(p_nl)
use toole
implicit none
real(kind=8) p_nl(0:g),temp(0:g)
integer i
real,external::trape_integral4
  do i=0,g
    temp(i)=P_nl(i)**2.0
  end do
normal_P=trape_integral4(temp,h,0)
end
```

```
subroutine density_out() !density n_out
use toole
implicit none
integer i
```

```
p_10(0)=0.0
```



p_20(0)=0.0
 p_30(0)=0.0
 p_21(0)=0.0
 p_31(0)=0.0
 p_40(0)=0.0
 p_32(0)=0.0
 p_41(0)=0.0
 p_50(0)=0.0
 p_42(0)=0.0
 p_51(0)=0.0
 p_60(0)=0.0
 p_52(0)=0.0
 p_43(0)=0.0
 p_61(0)=0.0
 p_70(0)=0.0
 p_53(0)=0.0

p_10(g)=0.0
 p_20(g)=0.0
 p_30(g)=0.0
 p_21(g)=0.0
 p_31(g)=0.0
 p_40(g)=0.0
 p_32(g)=0.0
 p_41(g)=0.0
 p_50(g)=0.0
 p_42(g)=0.0
 p_51(g)=0.0
 p_60(g)=0.0
 p_52(g)=0.0
 p_43(g)=0.0
 p_61(g)=0.0
 p_70(g)=0.0
 p_53(g)=0.0



do i=0,g

 if (i==0) then

$$n_out(i)=(1.0/(4.0*\pi))*(2.0*(2.0*(p_10(1)/k(1))-(p_10(2)/k(2)))**2.0+2.0*(2.0*(p_20(1)/k(1))-(p_20(2)/k(2)))**2.0+6.0*(2.0*(p_21(1)/k(1))-(p_21(2)/k(2)))**2.0&$$


```

&+2.0*(2.0*(p_30(1)/k(1))-(p_30(2)/k(2)))**2.0+6.0*(2.0*(p_31(1)/k(1))-(p_31(2)/k(2)))**2
.0+2.0*(2.0*(p_40(1)/k(1))-(p_40(2)/k(2)))**2.0&
&+10.0*(2.0*(p_32(1)/k(1))-(p_32(2)/k(2)))**2.0+6.0*(2.0*(p_41(1)/k(1))-(p_41(2)/k(2)))**
2.0+2.0*(2.0*(p_50(1)/k(1))-(p_50(2)/k(2)))**2.0&
&+10.0*(2.0*(p_42(1)/k(1))-(p_42(2)/k(2)))**2.0+6.0*(2.0*(p_51(1)/k(1))-(p_51(2)/k(2)))**
2.0+2.0*(2.0*(p_60(1)/k(1))-(p_60(2)/k(2)))**2.0&
&+10.0*(2.0*(p_52(1)/k(1))-(p_52(2)/k(2)))**2.0+14.0*(2.0*(p_43(1)/k(1))-(p_43(2)/k(2)))*
*2.0+6.0*(2.0*(p_61(1)/k(1))-(p_61(2)/k(2)))**2.0&
&+2.0*(2.0*(p_70(1)/k(1))-(p_70(2)/k(2)))**2.0+0.0*(2.0*(p_62(1)/k(1))-(p_62(2)/k(2)))**2
.0+14.0*(2.0*(p_53(1)/k(1))-(p_53(2)/k(2)))**2.0)

```

else

```

n_out(i)=(1.0/(4.0*pi))*(2.0*(p_10(i)/k(i))**2.0+2.0*(p_20(i)/k(i))**2.0+6.0*(p_21(i)/k(i))*
*2.0+2.0*(p_30(i)/k(i))**2.0+6.0*(p_31(i)/k(i))**2.0&
&+2.0*(p_40(i)/k(i))**2.0+10.0*(p_32(i)/k(i))**2.0+6.0*(p_41(i)/k(i))**2.0+2.0*(p_50(i)/k(
i))**2.0+10.0*(p_42(i)/k(i))**2.0&
&+6.0*(p_51(i)/k(i))**2.0+2.0*(p_60(i)/k(i))**2.0+10.0*(p_52(i)/k(i))**2.0+14.0*(p_43(i)/
k(i))**2.0+6.0*(p_61(i)/k(i))**2.0&
&+2.0*(p_70(i)/k(i))**2.0+0.0*(p_62(i)/k(i))**2.0+14.0*(p_53(i)/k(i))**2.0)
end if
end do
end

```

subroutine computing_ev_evt_ps()

use toole

!計算 eigenvalue and eigenvector of ps.

implicit none

integer:: i,j

do j=1,g-1 !離散化之後產生的矩陣

do i=1,g-1

if (i==j) then

a(i,j)=(1.0/(h**2.0)+v11_ps(i))

else if (abs(i-j)==1) then

a(i,j)=(-1.0/(2.0*h**2.0))

else

a(i,j)=0.0

end if

end do

end do

```

do i=1,g-1
    w(i)=a(i,i)
end do

do i=2,g-1
    e(i)=(-1.0/(2.0*h**2.0))
end do
call rst(g-1,g-1,w,e,1,z,ierr)

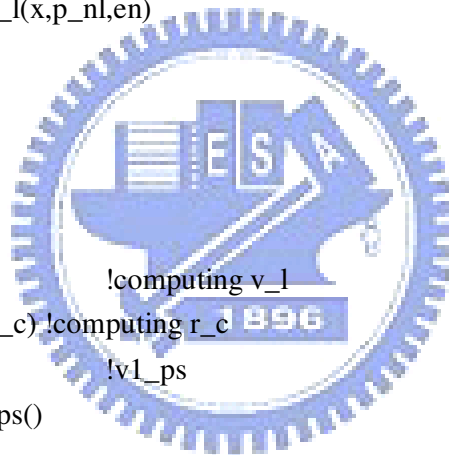
end subroutine

```

```

real*8 function adjust_c_l(x,p_nl,en)
use toole
implicit none
real(kind=8) x,en
real(kind=8) p_nl(0:g)
c_l=x
call sub15()
call cutoff_radius(p_nl,r_c)
call sub16()
call computing_ev_evt_ps()
adjust_c_l=w(1)-en
return
end

```



```

subroutine find_c_l01() !find c_l
use toole
implicit none
real*8,external::adjust_c_l

real(kind=8)::a1,b1,c1
real(kind=8)::fa,fb,fc
a1=-2.0
b1=0.0
fa=adjust_c_l(a1,pnl,e_nl)

```

```

fb=adjust_c_l(b1,pnl,e_nl)
c1=a1-fa*(b1-a1)/(fb-fa)
fc=adjust_c_l(c1,pnl,e_nl)
do while(abs(fc)>1e-4)
a1=b1
b1=c1
fa=adjust_c_l(a1,pnl,e_nl)
fb=adjust_c_l(b1,pnl,e_nl)
c1=a1-fa*(b1-a1)/(fb-fa)
fc=adjust_c_l(c1,pnl,e_nl)
end do
c_l=c1
end

```

```

subroutine sub15()
use toole
implicit none
integer i
do i=1,g-1
v_l(i)=(1*(1+1.0)/(2.0*k(i)**2.0))+v_eff(i)
end do
end

```



```

subroutine sub16()
use toole
implicit none
integer i
real*8,external::func
do i=1,g-1

```

$$v_{l_ps}(i)=(1.0-func(k(i)/r_c))*v_l(i)+c_l*func(k(i)/r_c)$$

```
end do
```

```
end
```

```
real*8 function func(x)
```

```

use toole
implicit none
real(kind=8) x

```

```

    func=exp(-x**4.0)

```

```

return
end

```

```

real*8 function  gl(x)
use toole
implicit none
real(kind=8) x

```

```

    gl=x**(l+1)*exp(-x**4.0)

```

```

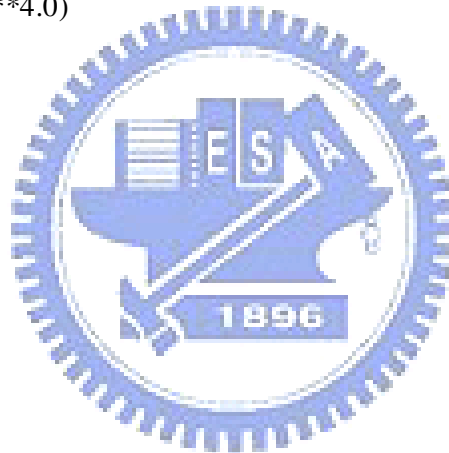
return
end

```

```

subroutine sub_w2_0()
use toole
implicit none
integer i
real*8,external::gl
    do i=1,g-1
        w20(i)=r1*(w10(i)+delta_l*gl(k(i)/r_c))
    end do
end

```



```

subroutine find_delta_l()
use imsl
use toole
implicit none
real(kind=8) p(3),temp1(0:g),temp2(0:g),temp3(0:g)
complex(kind=8) r(2)
integer i
real,external::trape_integral4,gl
    do i=0,g
        temp1(i)=(w10(i))**2.0

```

```

end do
p(1)=r1**2.0*trape_integral4(temp1,h,0)-1.0

```

```

do i=0,g
temp2(i)=w10(i)*gl(k(i)/r_c)
end do
p(2)=2.0*r1**2.0*trape_integral4(temp2,h,0)

```

```

do i=0,g
temp3(i)=(gl(k(i)/r_c))**2.0
end do
p(3)=r1**2.0*trape_integral4(temp3,h,0)

```

```

call dzplrc(2,p,r)
delta_l=r(1)
end

```



```

subroutine cutoff_radius(p_nl,rc)
use toole
implicit none
integer,external::sequential_search
real(kind=8) P_nl(0:g),rc,temp,b(0:g)
integer i,j
b=p_nl
do j=1,g-1
if (abs(b(j))>abs(b(j+1))) then
temp=b(j)
b(j)=b(j+1)
b(j+1)=temp
end if
end do
r_c=0.56*k(sequential_search(p_nl,b(g)))
end

```

```
integer function sequential_search(in,value)!循序搜尋法
```

```
use toole
```

```
implicit none
```

```
real(kind=8) value
```

```
real(kind=8) in(0:g)
```

```
integer i
```

```
do i=0,g
```

```
    if (value==in(i)) then
```

```
        sequential_search=i
```

```
    return
```

```
    end if
```

```
end do
```

```
sequential_search=0
```

```
return
```

```
end function
```



```
subroutine pseudo()
```

```
use toole
```

```
implicit none
```

```
integer i
```

```
do i=1,g-1
```

```
    v2l_ps(i)=(e_n1*w20(i)+0.5*((w20(i+1)-2.0*w20(i)+w20(i-1))/h**2.0))*(1.0/w20(i))
```

```
end do
```

```
end
```

```
subroutine E_G() !total energy 計算
```

```
use toole
```

```
implicit none
```

```
call sub3()
```

```
call KE() !動能 T
```

```
n=n_out
```

```
call radius() !the radius r_s of a unit charge
```

```

call ee()      !the exchange energy e_x
call ce()      !the correlation energy e_c
call ece()     !the exchange-correlation energy per unit density e_xc
call sub4()
call Excfuctional() !exchange-correlation fuctional E_xc
call sub5()
call sub6()     計算總能中的  $\int v(r)n(r)dr$ 
call sub7()
call sub8()
call sub9()     !!coulomb energy  $(1/2)*\int\int(Vee*n)dr$ 
total_energy=kinetic_energy+E_xc_functional+external_energy+coulomb_energy
end

```

```

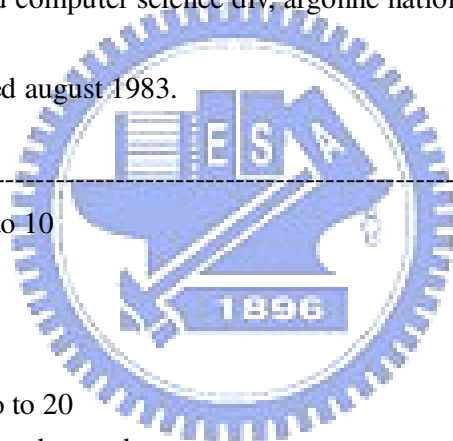
!find the eigenvalues and eigenvectors (if desired)
!of a real symmetric tridiagonal matrix.
subroutine rst(nm,n,w,e,matz,z,ierr)
integer i,j,n,nm,ierr,matz
double precision w(n),e(n),z(nm,n)
!c
!c   this subroutine calls the recommended sequence of
!c   subroutines from the eigensystem subroutine package (eispack)
!c   to find the eigenvalues and eigenvectors (if desired)
!c   of a real symmetric tridiagonal matrix.
!c
!c   on input
!c
!c       nm   must be set to the row dimension of the two-dimensional
!c       array parameters as declared in the calling program
!c       dimension statement.
!c
!c       n    is the order of the matrix.
!c
!c       w    contains the diagonal elements of the real
!c       symmetric tridiagonal matrix.
!c
!c       e    contains the subdiagonal elements of the matrix in
!c       its last n-1 positions.  e(1) is arbitrary.
!c

```

```

!c      matz  is an integer variable set equal to zero if
!c      only eigenvalues are desired.  otherwise it is set to
!c      any non-zero integer for both eigenvalues and eigenvectors.
!c
!c      on output
!c
!c
!c      z  contains the eigenvectors if matz is not zero.
!c
!c      ierr  is an integer output variable set equal to an error
!c      completion code described in the documentation for imtql1
!c      and imtql2.  the normal completion code is zero.
!c
!c      questions and comments should be directed to burton s. garbow,
!c      mathematics and computer science div, argonne national laboratory
!c
!c      this version dated august 1983.
!c
!c      -----
!c      if (n .le. nm) go to 10
!c      ierr = 10 * n
!c      go to 50
!c
!c      10 if (matz .ne. 0) go to 20
!c      ..... find eigenvalues only .....
!c      call  imtql1(n,w,e,ierr)
!c      go to 50
!c      ..... find both eigenvalues and eigenvectors .....
!c      20 do 40 i = 1, n
!c
!c          do 30 j = 1, n
!c              z(j,i) = 0.0d0
!c      30  continue
!c
!c          z(i,i) = 1.0d0
!c      40 continue
!c
!c      call  imtql2(nm,n,w,e,z,ierr)
!c      50 return

```



end

```
subroutine imtql1(n,d,e,ierr)
```

```
integer i,j,l,m,n,ii,mml,ierr
```

```
double precision d(n),e(n)
```

```
double precision b,c,f,g,p,r,s,tst1,tst2,pythag
```

```
!c
```

```
!c   this subroutine is a translation of the algol procedure imtql1,
```

```
!c   num. math. 12, 377-383(1968) by martin and wilkinson,
```

```
!c   as modified in num. math. 15, 450(1970) by dubrulle.
```

```
!c   handbook for auto. comp., vol.ii-linear algebra, 241-248(1971).
```

```
!c
```

```
!c   this subroutine finds the eigenvalues of a symmetric
```

```
!c   tridiagonal matrix by the implicit ql method.
```

```
!c
```

```
!c   on input
```

```
!c
```

```
!c       n is the order of the matrix.
```

```
!c
```

```
!c       d contains the diagonal elements of the input matrix.
```

```
!c
```

```
!c       e contains the subdiagonal elements of the input matrix
```

```
!c       in its last n-1 positions.  e(1) is arbitrary.
```

```
!c
```

```
!c   on output
```

```
!c
```

```
!c       d contains the eigenvalues in ascending order.  if an
```

```
!c       error exit is made, the eigenvalues are correct and
```

```
!c       ordered for indices 1,2,...ierr-1, but may not be
```

```
!c       the smallest eigenvalues.
```

```
!c
```

```
!c       e has been destroyed.
```

```
!c
```

```
!c       ierr is set to
```

```
!c         zero      for normal return,
```

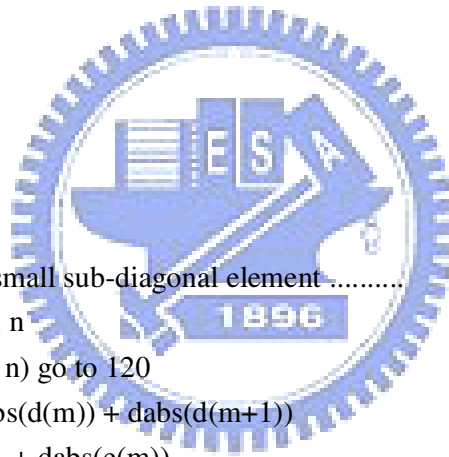
```
!c         j        if the j-th eigenvalue has not been
```

```
!c         determined after 30 iterations.
```

```

!c
!c      calls pythag for  dsqrt(a*a + b*b) .
!c
!c      questions and comments should be directed to burton s. garbow,
!c      mathematics and computer science div, argonne national laboratory
!c
!c      this version dated august 1983.
!c
!c      -----
!c
      ierr = 0
      if (n .eq. 1) go to 1001
!c
      do 100 i = 2, n
100 e(i-1) = e(i)
!c
      e(n) = 0.0d0
!c
      do 290 l = 1, n
          j = 0
!c      ..... look for small sub-diagonal element .....
105      do 110 m = l, n
          if (m .eq. n) go to 120
          tst1 = dabs(d(m)) + dabs(d(m+1))
          tst2 = tst1 + dabs(e(m))
          if (tst2 .eq. tst1) go to 120
110      continue
!c
120      p = d(l)
          if (m .eq. l) go to 215
          if (j .eq. 30) go to 1000
          j = j + 1
!c      ..... form shift .....
          g = (d(l+1) - p) / (2.0d0 * e(l))
          r = pythag(g,1.0d0)
          g = d(m) - p + e(l) / (g + dsign(r,g))
          s = 1.0d0
          c = 1.0d0
          p = 0.0d0

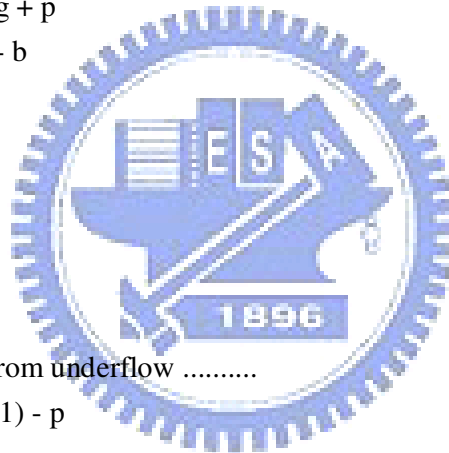
```



```

        mml = m - 1
!c      ..... for i=m-1 step -1 until 1 do -- .....
        do 200 ii = 1, mml
            i = m - ii
            f = s * e(i)
            b = c * e(i)
            r = pythag(f,g)
            e(i+1) = r
            if (r .eq. 0.0d0) go to 210
            s = f / r
            c = g / r
            g = d(i+1) - p
            r = (d(i) - g) * s + 2.0d0 * c * b
            p = s * r
            d(i+1) = g + p
            g = c * r - b
200    continue
!c
        d(l) = d(l) - p
        e(l) = g
        e(m) = 0.0d0
        go to 105
!c      ..... recover from underflow .....
210    d(i+1) = d(i+1) - p
        e(m) = 0.0d0
        go to 105
!c      ..... order eigenvalues .....
215    if (l .eq. 1) go to 250
!c      ..... for i=1 step -1 until 2 do -- .....
        do 230 ii = 2, l
            i = l + 2 - ii
            if (p .ge. d(i-1)) go to 270
            d(i) = d(i-1)
230    continue
!c
250    i = 1
270    d(i) = p
290    continue
!c

```



```

        go to 1001
!c      ..... set error -- no convergence to an
!c      eigenvalue after 30 iterations .....
    1000 ierr = 1
    1001 return
end

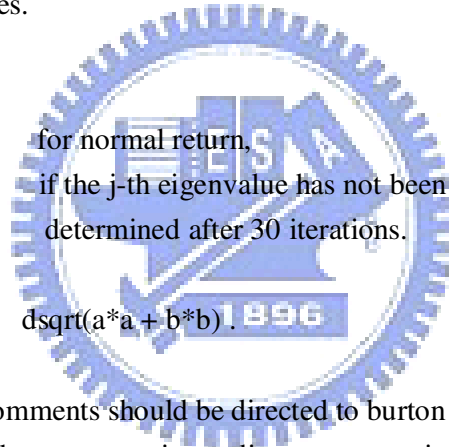
subroutine imtql2(nm,n,d,e,z,ierr)
integer i,j,k,l,m,n,ii,nm,mml,ierr
double precision d(n),e(n),z(nm,n)
double precision b,c,f,g,p,r,s,tst1,tst2,pythag
!c
!c      this subroutine is a translation of the algol procedure imtql2,
!c      num. math. 12, 377-383(1968) by martin and wilkinson,
!c      as modified in num. math. 15, 450(1970) by dubrulle.
!c      handbook for auto. comp., vol.ii-linear algebra, 241-248(1971).
!c
!c      this subroutine finds the eigenvalues and eigenvectors
!c      of a symmetric tridiagonal matrix by the implicit ql method.
!c      the eigenvectors of a full symmetric matrix can also
!c      be found if  tred2 has been used to reduce this
!c      full matrix to tridiagonal form.
!c
!c      on input
!c
!c      nm must be set to the row dimension of two-dimensional
!c      array parameters as declared in the calling program
!c      dimension statement.
!c
!c      n is the order of the matrix.
!c
!c      d contains the diagonal elements of the input matrix.
!c
!c      e contains the subdiagonal elements of the input matrix
!c      in its last n-1 positions.  e(1) is arbitrary.
!c
!c      z contains the transformation matrix produced in the
!c      reduction by  tred2, if performed.  if the eigenvectors

```

```

!c      of the tridiagonal matrix are desired, z must contain
!c      the identity matrix.
!c
!c      on output
!c
!c      d contains the eigenvalues in ascending order.  if an
!c      error exit is made, the eigenvalues are correct but
!c      unordered for indices 1,2,...,ierr-1.
!c
!c      e has been destroyed.
!c
!c      z contains orthonormal eigenvectors of the symmetric
!c      tridiagonal (or full) matrix.  if an error exit is made,
!c      z contains the eigenvectors associated with the stored
!c      eigenvalues.
!c
!c      ierr is set to
!c      zero      for normal return,
!c      j        if the j-th eigenvalue has not been
!c              determined after 30 iterations.
!c
!c      calls pythag for  dsqrt(a*a + b*b) .
!c
!c      questions and comments should be directed to burton s. garbow,
!c      mathematics and computer science div, argonne national laboratory
!c
!c      this version dated august 1983.
!c
!c      -----
!c
!c      ierr = 0
!c      if (n .eq. 1) go to 1001
!c
!c      do 100 i = 2, n
100 e(i-1) = e(i)
!c
!c      e(n) = 0.0d0
!c
!c      do 240 l = 1, n

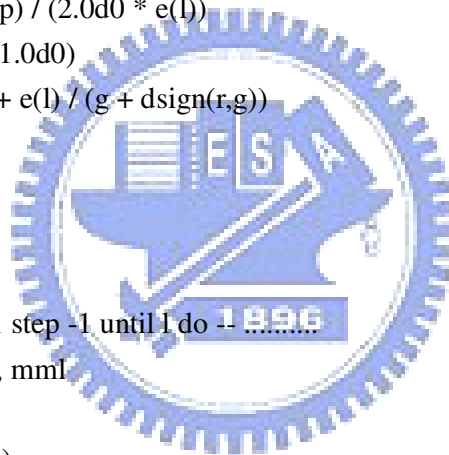
```



```

        j = 0
!c ..... look for small sub-diagonal element .....
105   do 110 m = 1, n
        if (m .eq. n) go to 120
        tst1 = dabs(d(m)) + dabs(d(m+1))
        tst2 = tst1 + dabs(e(m))
        if (tst2 .eq. tst1) go to 120
110   continue
!c
120   p = d(l)
        if (m .eq. l) go to 240
        if (j .eq. 30) go to 1000
        j = j + 1
!c ..... form shift .....
        g = (d(l+1) - p) / (2.0d0 * e(l))
        r = pythag(g, 1.0d0)
        g = d(m) - p + e(l) / (g + dsign(r, g))
        s = 1.0d0
        c = 1.0d0
        p = 0.0d0
        mml = m - 1
!c ..... for i=m-1 step -1 until l do - .....
        do 200 ii = 1, mml
            i = m - ii
            f = s * e(i)
            b = c * e(i)
            r = pythag(f, g)
            e(i+1) = r
            if (r .eq. 0.0d0) go to 210
            s = f / r
            c = g / r
            g = d(i+1) - p
            r = (d(i) - g) * s + 2.0d0 * c * b
            p = s * r
            d(i+1) = g + p
            g = c * r - b
!c ..... form vector .....
        do 180 k = 1, n
            f = z(k, i+1)

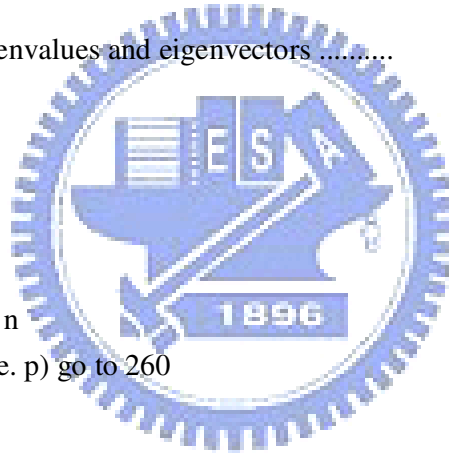
```



```

                z(k,i+1) = s * z(k,i) + c * f
                z(k,i) = c * z(k,i) - s * f
180          continue
!c
200          continue
!c
                d(l) = d(l) - p
                e(l) = g
                e(m) = 0.0d0
                go to 105
!c          ..... recover from underflow .....
210          d(i+1) = d(i+1) - p
                e(m) = 0.0d0
                go to 105
240          continue
!c          ..... order eigenvalues and eigenvectors .....
do 300 ii = 2, n
    i = ii - 1
    k = i
    p = d(i)
!c
do 260 j = ii, n
    if (d(j) .ge. p) go to 260
    k = j
    p = d(j)
260          continue
!c
                if (k .eq. i) go to 300
                d(k) = d(i)
                d(i) = p
!c
do 280 j = 1, n
    p = z(j,i)
    z(j,i) = z(j,k)
    z(j,k) = p
280          continue
!!c
300          continue
!c

```



```

        go to 1001
!c      ..... set error -- no convergence to an
!c      eigenvalue after 30 iterations .....
    1000 ierr = 1
    1001 return
end

```

```

double precision function pythag(a,b)

```

```

double precision a,b

```

```

!c

```

```

!c  finds dsqrt(a**2+b**2) without overflow or destructive underflow

```

```

!c

```

```

    double precision p,r,s,t,u

```

```

    p = dmax1(dabs(a),dabs(b))

```

```

    if (p .eq. 0.0d0) go to 20

```

```

    r = (dmin1(dabs(a),dabs(b))/p)**2

```

```

10 continue

```

```

    t = 4.0d0 + r

```

```

    if (t .eq. 4.0d0) go to 20

```

```

    s = r/t

```

```

    u = 1.0d0 + 2.0d0*s

```

```

    p = u*p

```

```

    r = (s/u)**2 * r

```

```

    go to 10

```

```

20 pythag = p

```

```

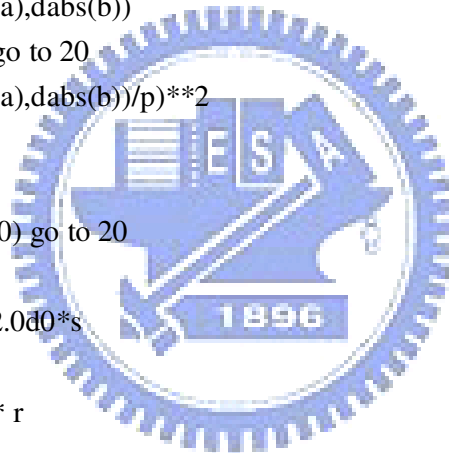
    return

```

```

end

```



A.2 The initial electron density of No atom(1*401)

| | | |
|------------------------|------------------------|------------------------|
| 107084.007828308 | 34372.7202003888 | 5300.30345368854 |
| 2745.91161917885 | 1624.66343877385 | 892.197653574079 |
| 517.051763519848 | 354.800421382601 | 280.003233664372 |
| 231.320654298285 | 188.341687507305 | 148.569806588318 |
| 113.740401626887 | 85.3953312399476 | 63.6774371773841 |
| 47.8322918811604 | 36.6624632645909 | 28.9606046635084 |
| 23.6807816232239 | 20.0194356502229 | 17.3967305601694 |
| 15.4231953497467 | 13.8477810162582 | 12.5191465439582 |
| 11.3492259742482 | 10.2911567867160 | 9.32069403295397 |
| 8.42701541242964 | 7.60486812302368 | 6.85187952990541 |
| 6.16574673671114 | 5.54406555171051 | 4.98350001575553 |
| 4.48028895432457 | 4.03005286824163 | 3.62830563879969 |
| 3.27040769438993 | 2.95191799334773 | 2.66854854916613 |
| 2.41636727839534 | 2.19173325135725 | 1.99139802790508 |
| 1.81243131970336 | 1.65226375522610 | 1.50861503142306 |
| 1.37950656016855 | 1.26319882912490 | 1.15819069653529 |
| 1.06316840210656 | 0.976999821221959 | 0.898695023259516 |
| 0.827399583719429 | 0.762364868721077 | 0.702942201908803 |
| 0.648560898054416 | 0.598723661220932 | 0.552990586063213 |
| 0.510975795157975 | 0.472335824607880 | 0.436767214016821 |
| 0.403998090272881 | 0.373786537892698 | 0.345914497227430 |
| 0.320186606536367 | 0.296425802473405 | 0.274472443596859 |
| 0.254181144073415 | 0.235420159665273 | 0.218068965152489 |
| 0.202017856902356 | 0.187166163485957 | 0.173421923097319 |
| 0.160700544202954 | 0.148924491324130 | 0.138022306315634 |
| 0.127928344025784 | 0.118582005664314 | 0.109927520435997 |
| 0.101913332498783 | 9.449192099516436E-002 | 8.761932895483111E-002 |
| 8.125498364289073E-002 | 7.536132896241667E-002 | 6.990367048803577E-002 |
| 6.484987503584268E-002 | 6.017024046721889E-002 | 5.583724402556516E-002 |
| 5.182542753444918E-002 | 4.811119642185761E-002 | 4.467271542712876E-002 |
| 4.148974157580070E-002 | 3.854353246913057E-002 | 3.581671468731099E-002 |
| 3.329319502591192E-002 | 3.095805076599340E-002 | 2.879745787173793E-002 |
| 2.679859747752530E-002 | 2.494959032376140E-002 | 2.323941942108340E-002 |
| 2.165787433951473E-002 | 2.019548439481859E-002 | 1.884347232932811E-002 |
| 1.759369667040717E-002 | 1.643860878903685E-002 | 1.537120840670903E-002 |
| 1.438500497240101E-002 | 1.347397809970459E-002 | 1.263254693531655E-002 |

| | | |
|------------------------|------------------------|------------------------|
| 1.185553407455260E-002 | 1.113813960804344E-002 | 1.047591147786275E-002 |
| 9.864722160225423E-003 | 9.300743863008295E-003 | 8.780427516106309E-003 |
| 8.300481899761514E-003 | 7.857856086626652E-003 | 7.449721203108031E-003 |
| 7.073454903081042E-003 | 6.726625618606037E-003 | 6.406980137083121E-003 |
| 6.112429498216537E-003 | 5.841038107483337E-003 | 5.591011940225279E-003 |
| 5.360689047079300E-003 | 5.148529188640743E-003 | 4.953105720501421E-003 |
| 4.773097140783451E-003 | 4.607279767040080E-003 | 4.454520103278698E-003 |
| 4.313768928726015E-003 | 4.184054900614255E-003 | 4.064479111616121E-003 |
| 3.954209759697231E-003 | 3.852477476473913E-003 | 3.758570222046024E-003 |
| 3.671829862411742E-003 | 3.591648038451772E-003 | 3.517462724052065E-003 |
| 3.448754308609661E-003 | 3.385043110725487E-003 | 3.325886409134098E-003 |
| 3.270875381509599E-003 | 3.219633054924623E-003 | 3.171811762731331E-003 |
| 3.127091196957642E-003 | 3.085176423897589E-003 | 3.045795870980201E-003 |
| 3.008699695388421E-003 | 2.973658235972852E-003 | 2.940460565578269E-003 |
| 2.908913158824198E-003 | 2.878838408990432E-003 | 2.850073623823818E-003 |
| 2.822470043278500E-003 | 2.795891543766766E-003 | 2.770213858375975E-003 |
| 2.745323703078859E-003 | 2.721117957650124E-003 | 2.697502897497303E-003 |
| 2.674393472946103E-003 | 2.651712573239848E-003 | 2.629390629829836E-003 |
| 2.607364923534576E-003 | 2.585579058293296E-003 | 2.563982437714587E-003 |
| 2.542529953206961E-003 | 2.521181418041916E-003 | 2.499901211479095E-003 |
| 2.478658027306466E-003 | 2.457424453632664E-003 | 2.436176558278047E-003 |
| 2.414893696350247E-003 | 2.393558203744460E-003 | 2.372155294234838E-003 |
| 2.350672575447646E-003 | 2.329100071057797E-003 | 2.307429861672135E-003 |
| 2.285656019340413E-003 | 2.263774214852109E-003 | 2.241781791908709E-003 |
| 2.219677495588686E-003 | 2.197461331112933E-003 | 2.175134536064552E-003 |
| 2.152699472602142E-003 | 2.130159293665480E-003 | 2.107518102411886E-003 |
| 2.084780655473636E-003 | 2.061952400499435E-003 | 2.039039288663764E-003 |
| 2.016047870133048E-003 | 1.992985135609288E-003 | 1.969858403951470E-003 |
| 1.946675281901410E-003 | 1.923443632740960E-003 | 1.900171567406406E-003 |
| 1.876867266208779E-003 | 1.853539117905898E-003 | 1.830195594001334E-003 |
| 1.806845158029764E-003 | 1.783496274244414E-003 | 1.760157467748888E-003 |
| 1.736837183780744E-003 | 1.713543843074242E-003 | 1.690285828883776E-003 |
| 1.667071255161842E-003 | 1.643908202517696E-003 | 1.620804612982832E-003 |
| 1.597768324217861E-003 | 1.574806982149893E-003 | 1.551928041618833E-003 |
| 1.529138801069894E-003 | 1.506446337823151E-003 | 1.483857526295779E-003 |
| 1.461379109343330E-003 | 1.439017530768938E-003 | 1.416779063764879E-003 |
| 1.394669765906052E-003 | 1.372695530528835E-003 | 1.350861920173128E-003 |
| 1.329174369132426E-003 | 1.307638102807593E-003 | 1.286258039648673E-003 |

| | | |
|------------------------|------------------------|------------------------|
| 1.265038930184872E-003 | 1.243985324631714E-003 | 1.223101534021123E-003 |
| 1.202391640081171E-003 | 1.181859535426908E-003 | 1.161508907636419E-003 |
| 1.141343220490764E-003 | 1.121365722369299E-003 | 1.101579499595279E-003 |
| 1.081987414842048E-003 | 1.062592179433795E-003 | 1.043396265656977E-003 |
| 1.024401978048282E-003 | 1.005611430924838E-003 | 9.870265691125674E-004 |
| 9.686491649160720E-004 | 9.504808334294193E-004 | 9.325230324215833E-004 |
| 9.147770038020224E-004 | 8.972439105427312E-004 | 8.799247079612382E-004 |
| 8.628202102971768E-004 | 8.459311125501529E-004 | 8.292579726991195E-004 |
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| | | |
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| 1.058210769637239E-008 | 0.000000000000000E+000 | |

A.3 The diagrams of output resulting

In these diagrams, pseudo wave functions and radial functions P_{nl} are described by red lines and green lines, respectively.

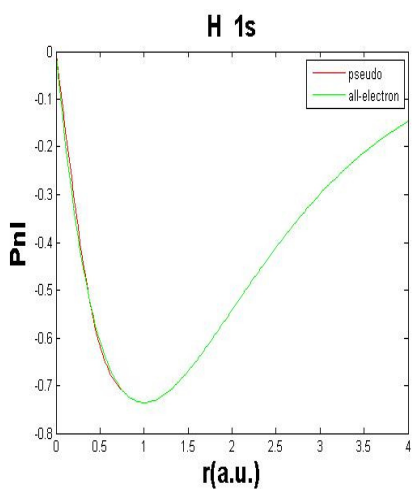


Figure 3.

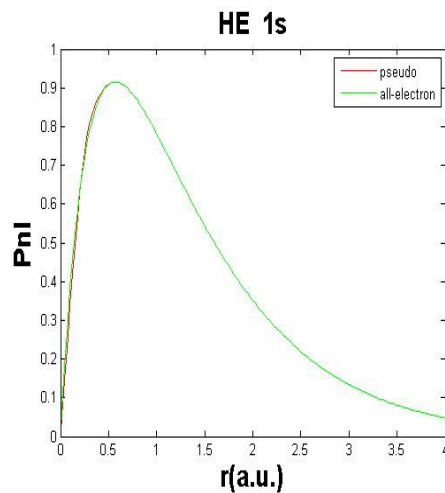


Figure 4.

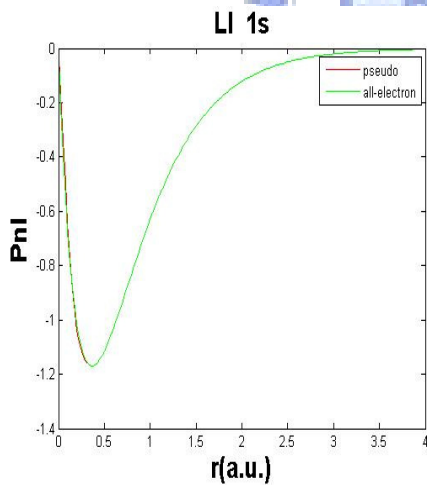


Figure 5.

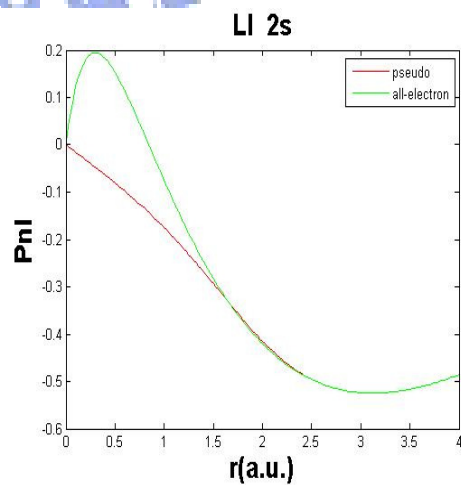


Figure 6.

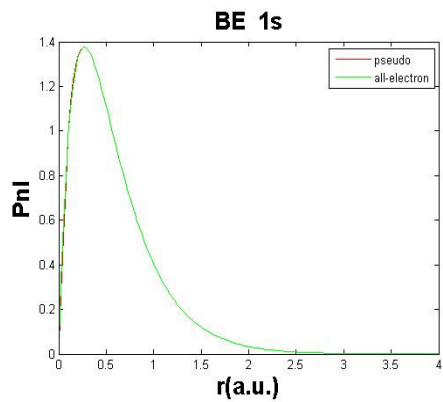


Figure 7.

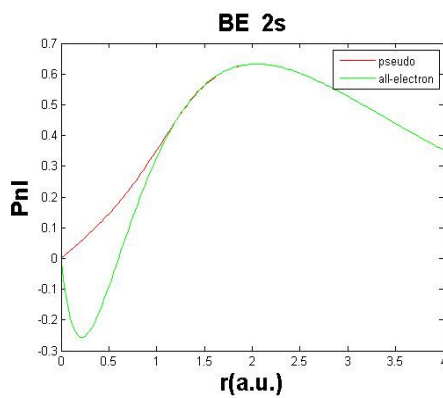


Figure 8.

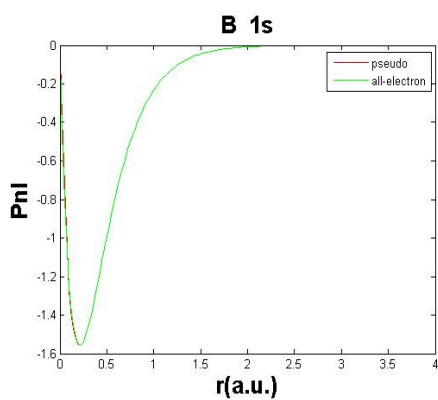


Figure 9.

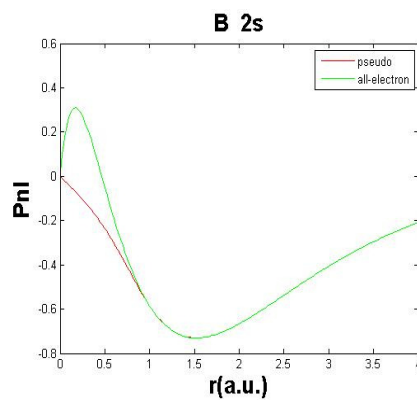


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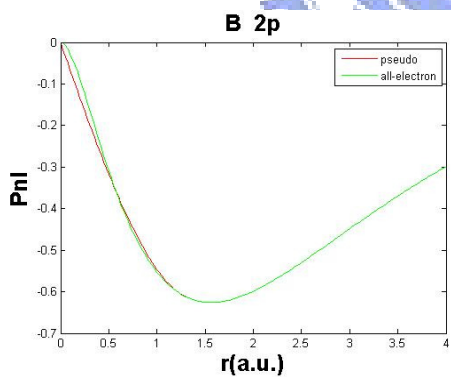


Figure 11.

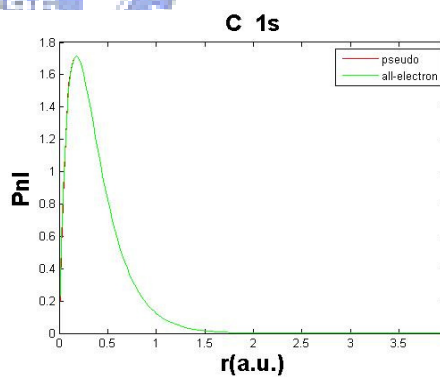


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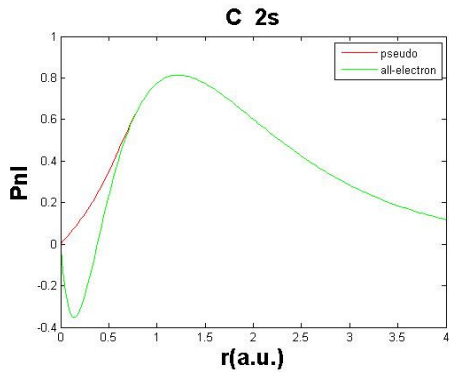


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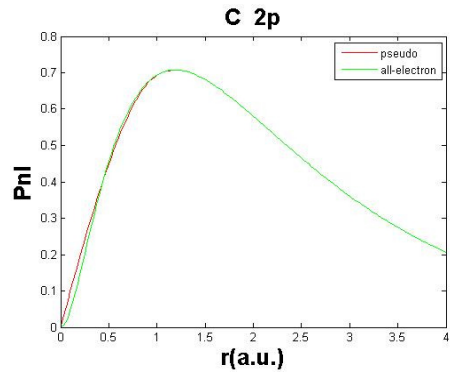


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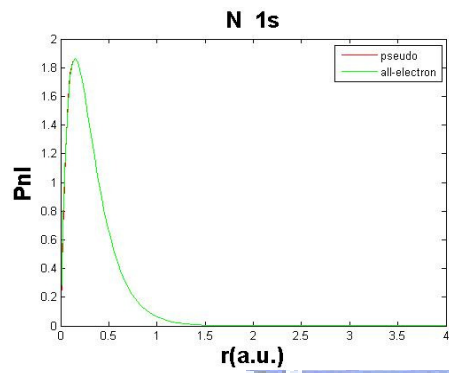


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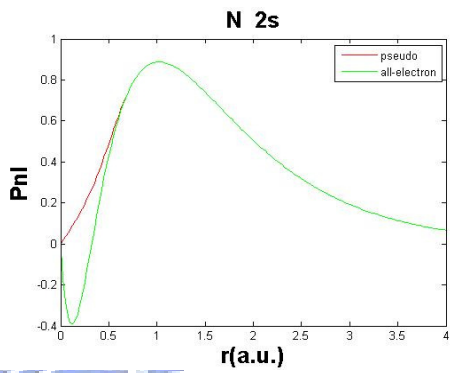


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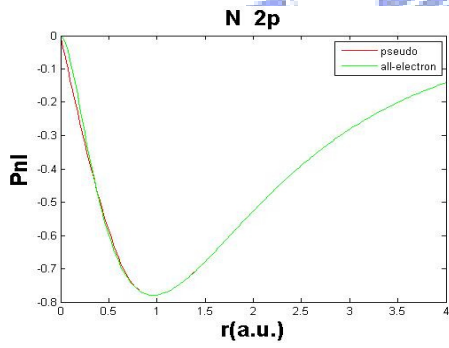


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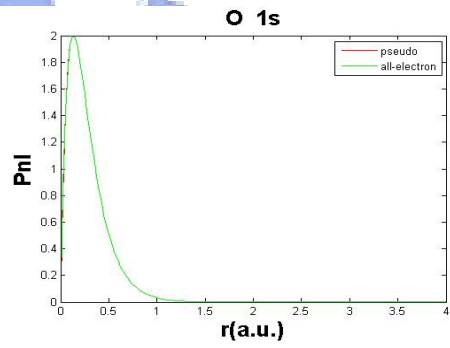


Figure 18.

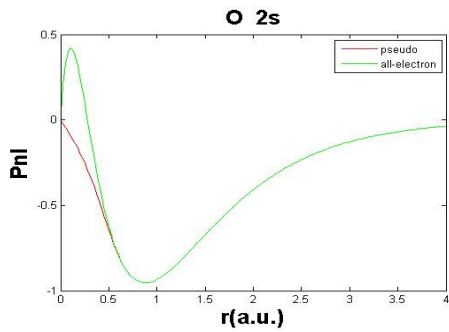


Figure 19.

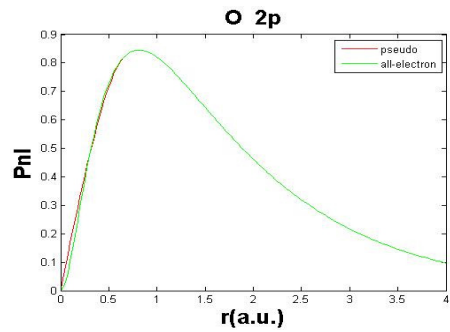


Figure 20.

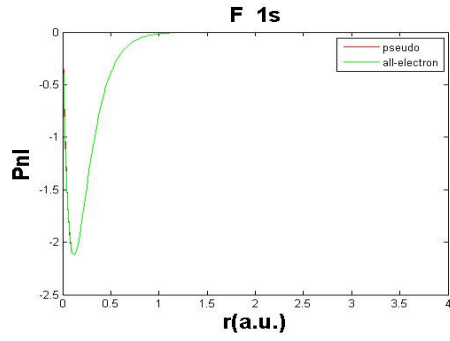


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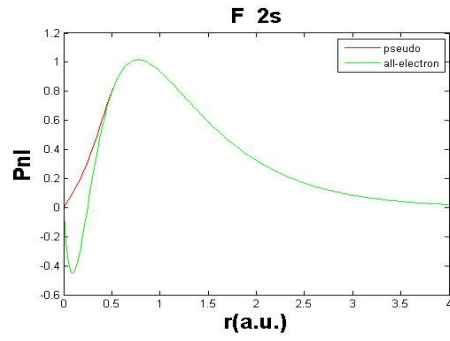


Figure 22.

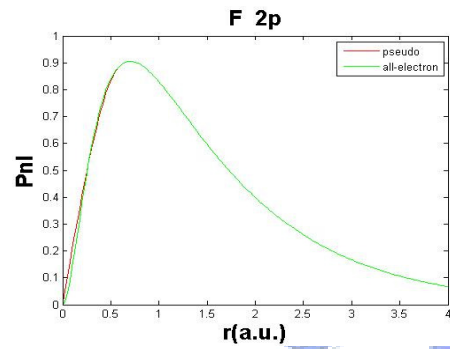


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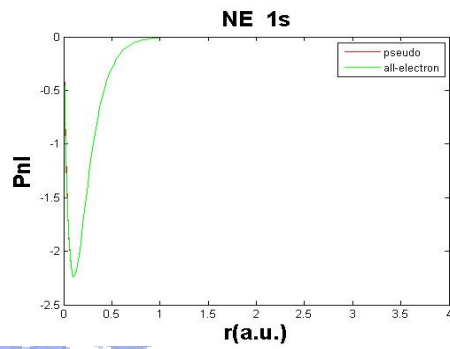


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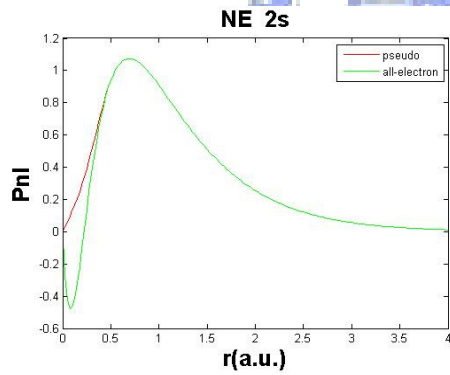


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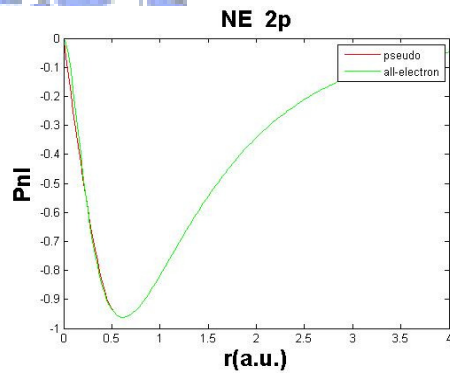


Figure 26.

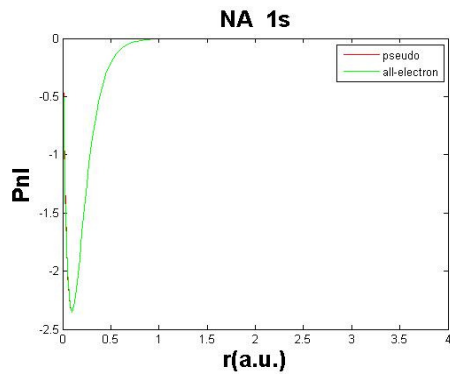


Figure 27.

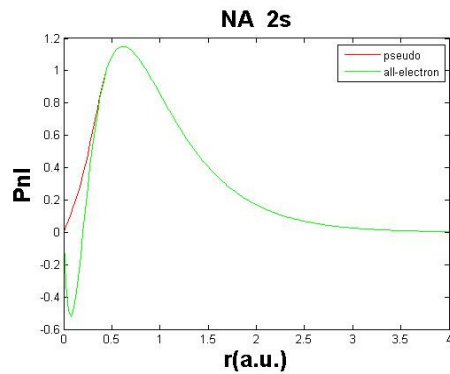


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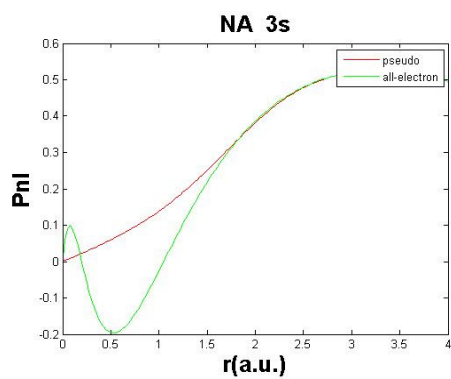


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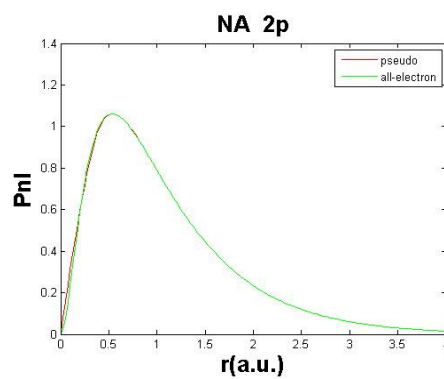


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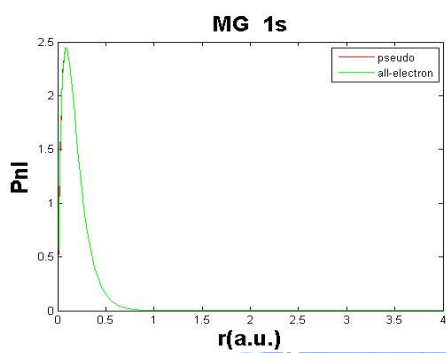


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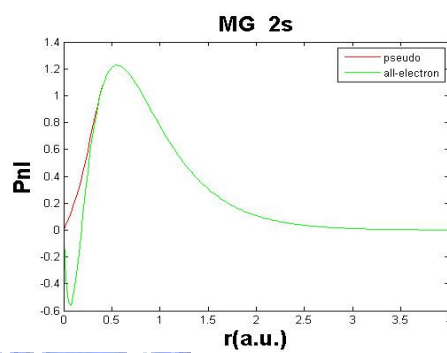


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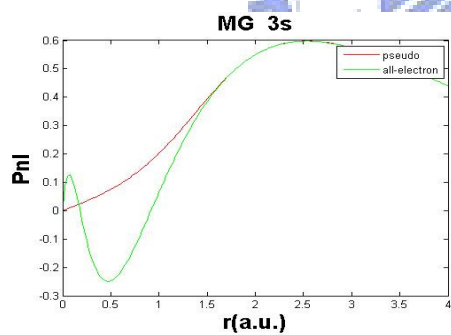


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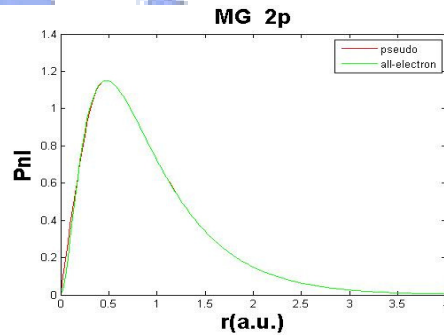


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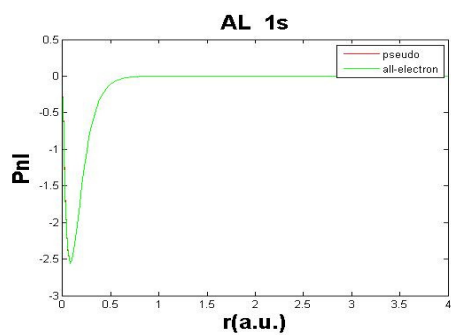


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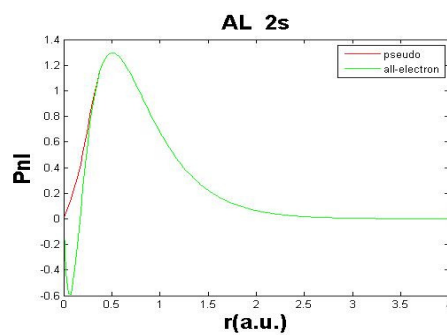


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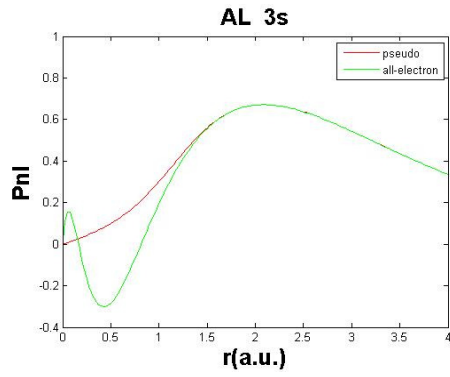


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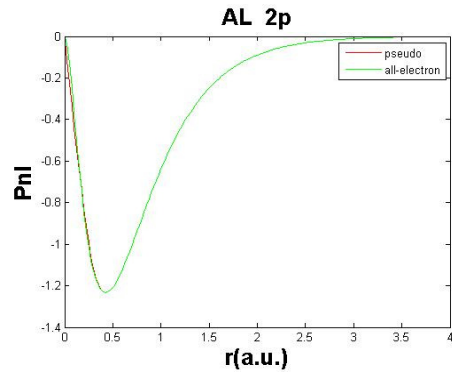


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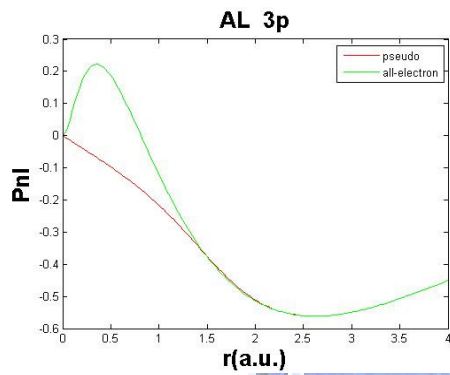


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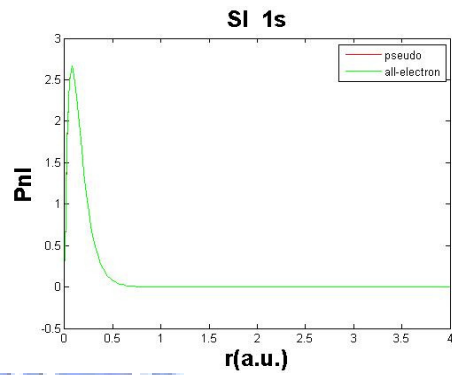


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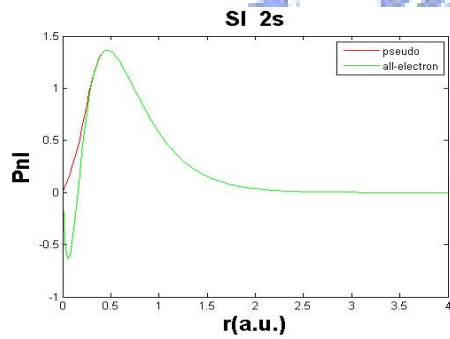


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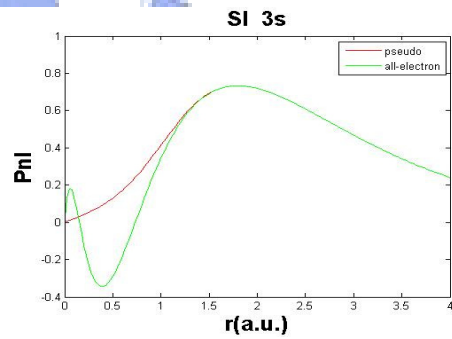


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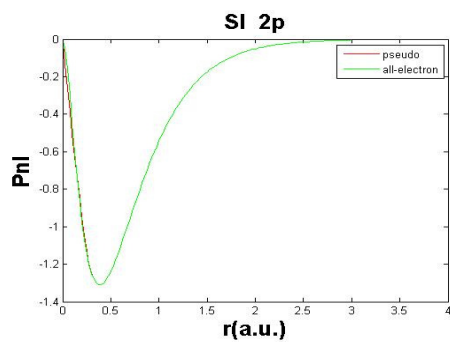


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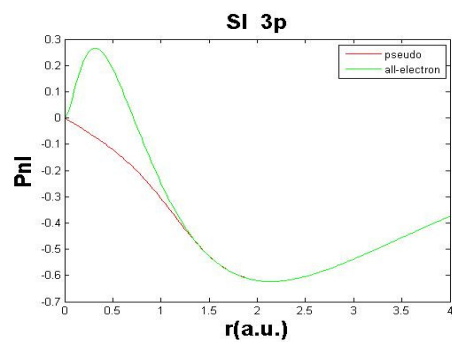


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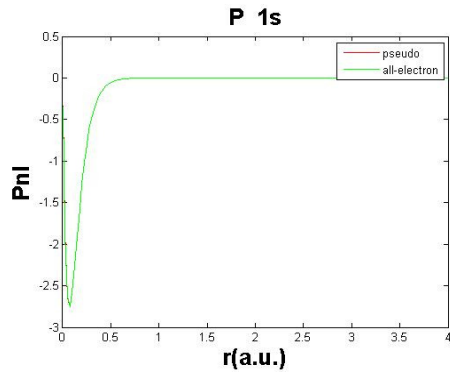


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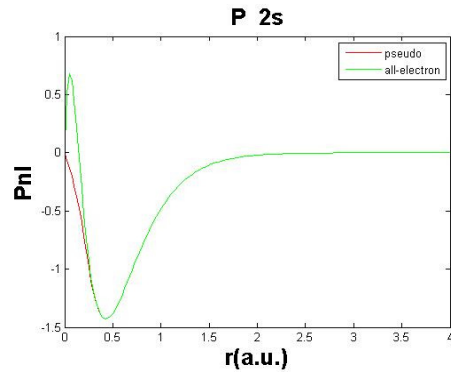


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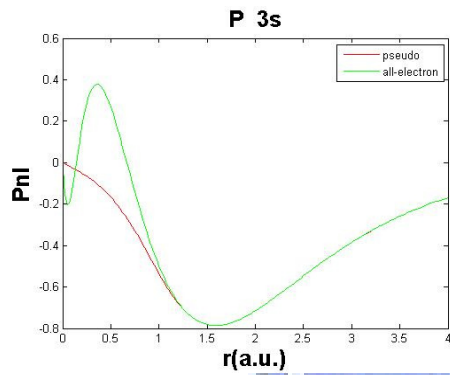


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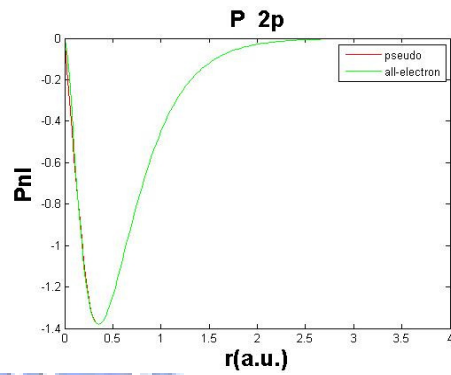


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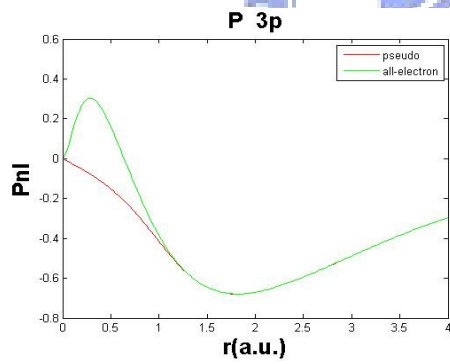


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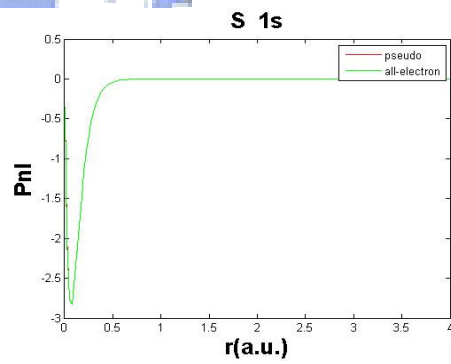


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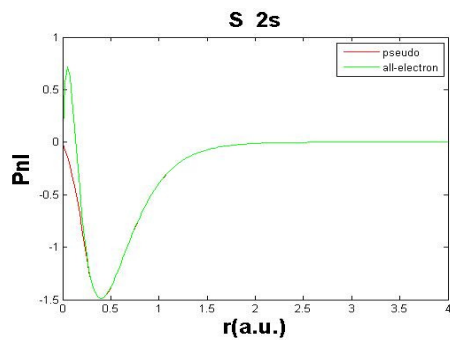


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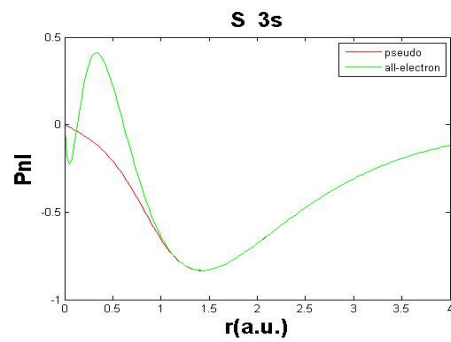


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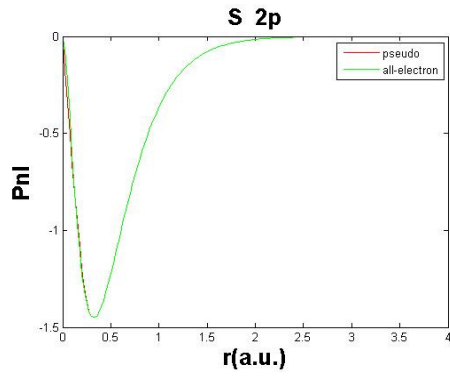


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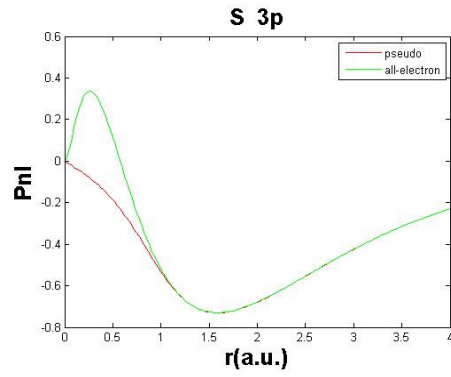


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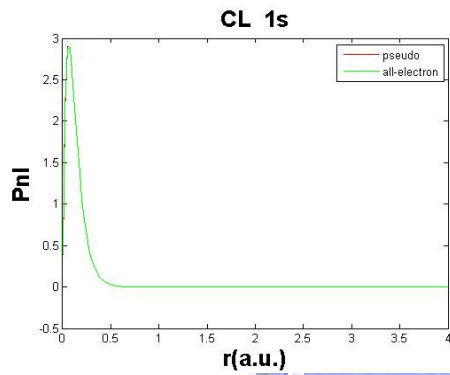


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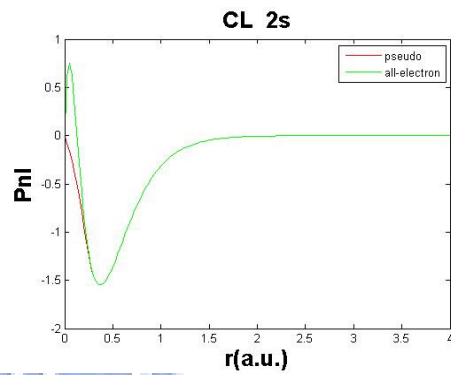


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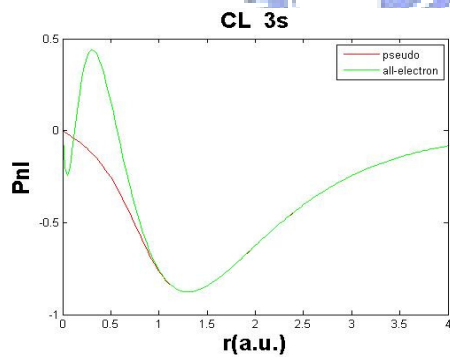


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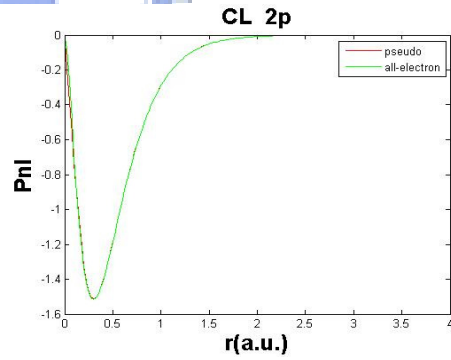


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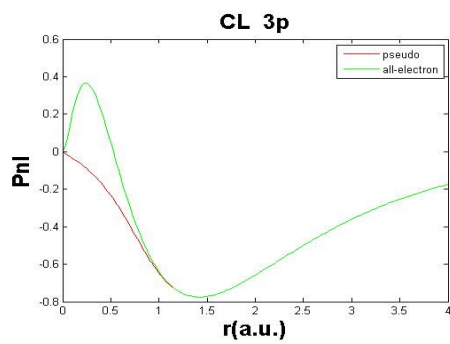


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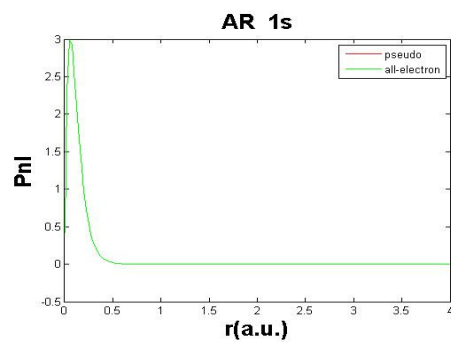


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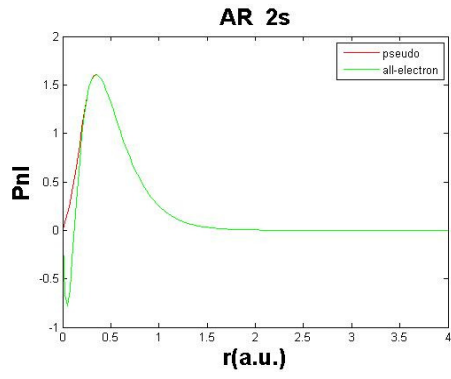


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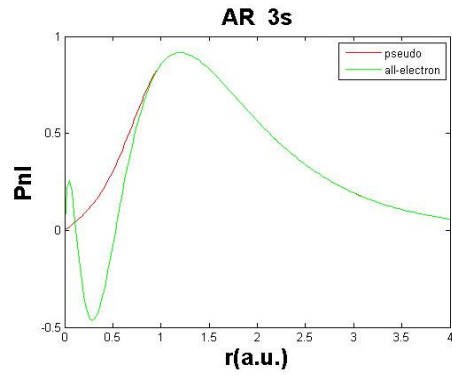


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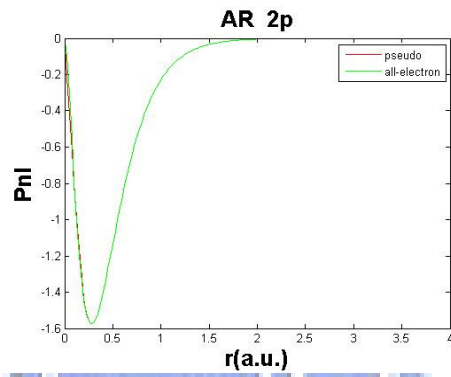
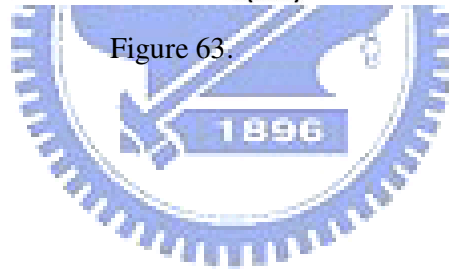


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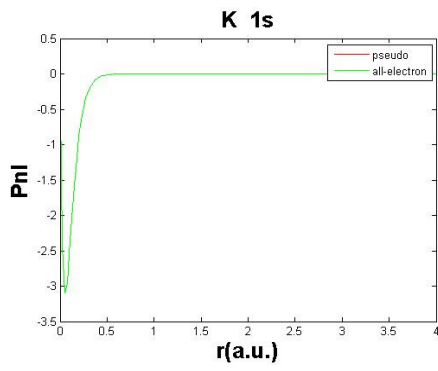


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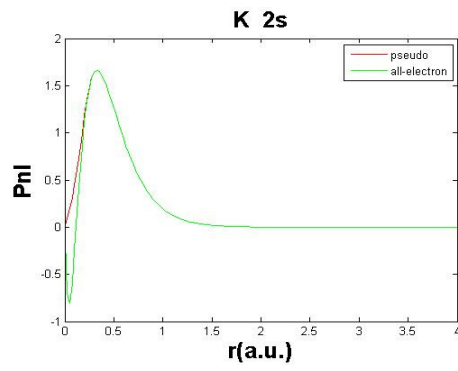


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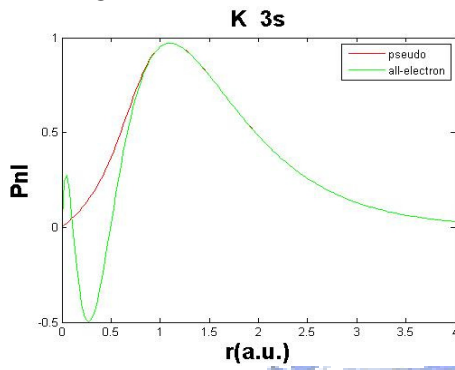


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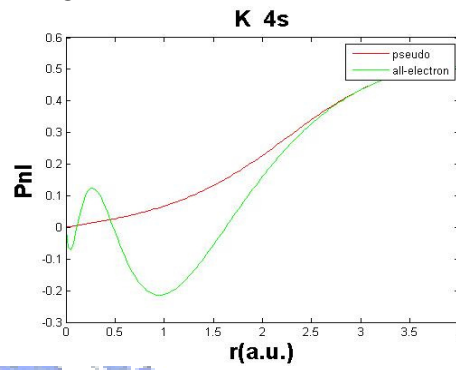


Figure 67.

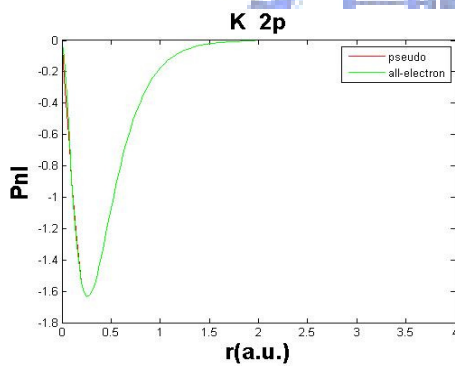


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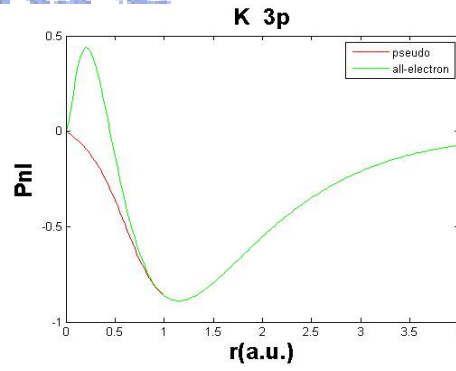


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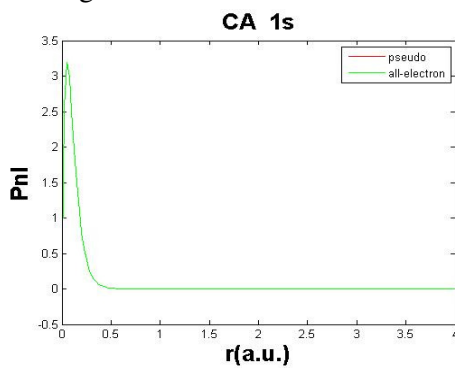


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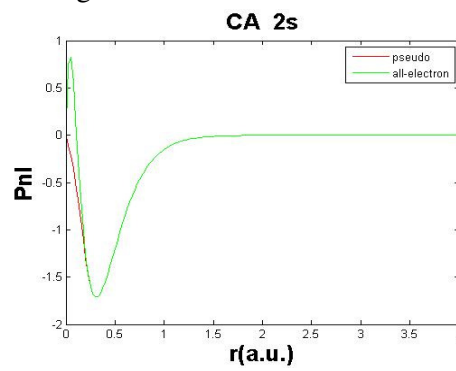


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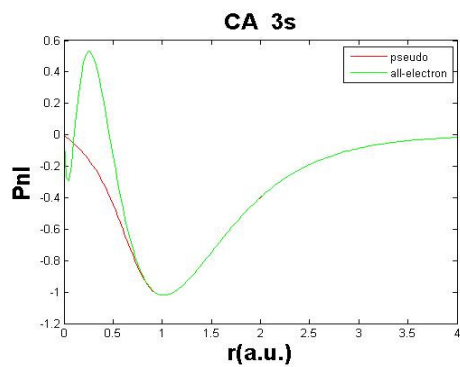


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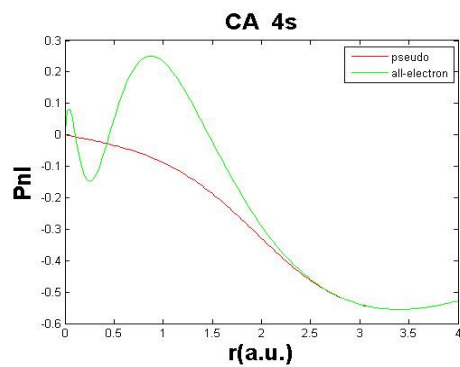


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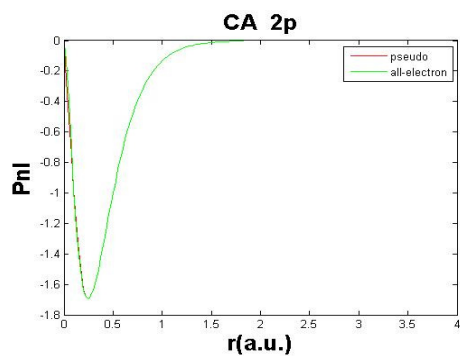


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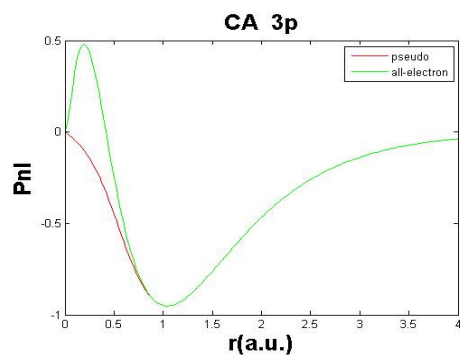


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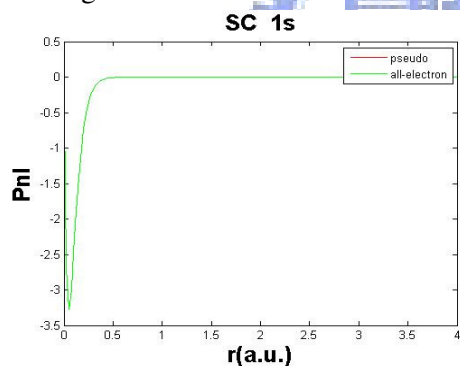


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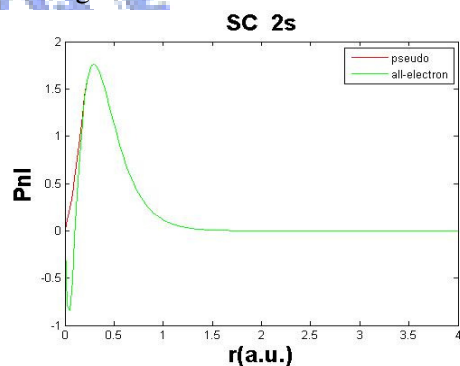


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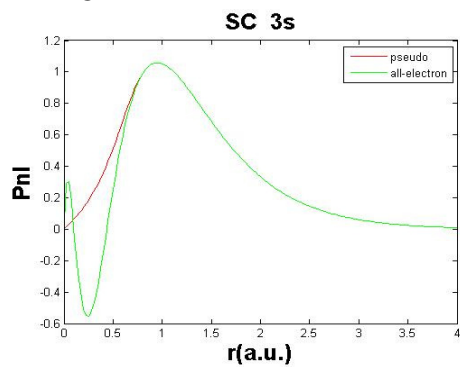


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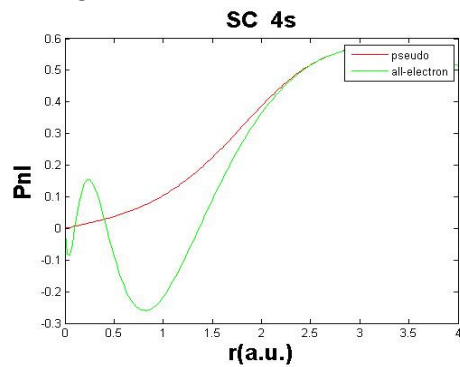


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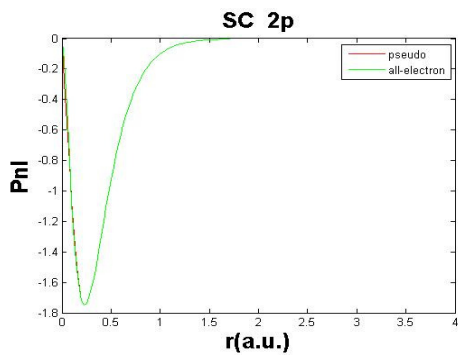


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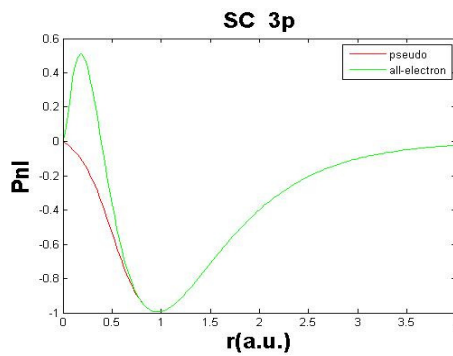


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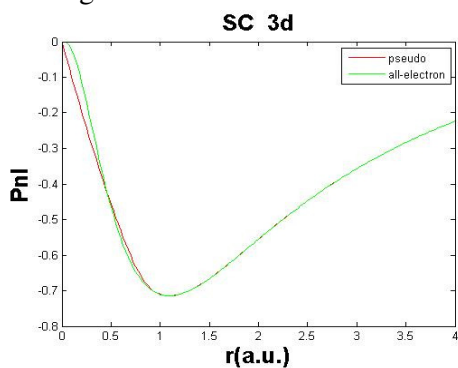


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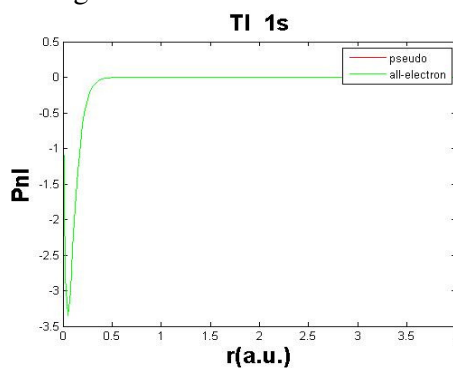


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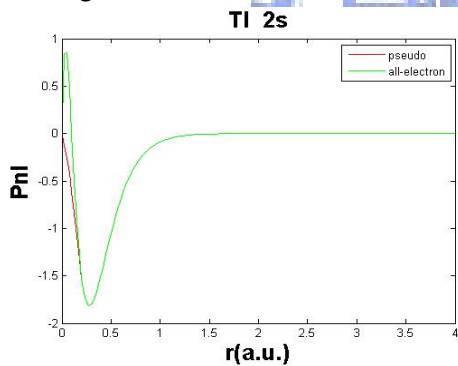


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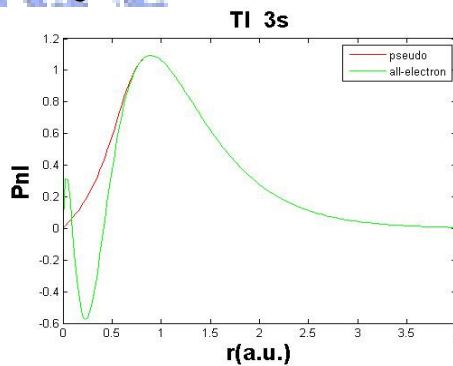


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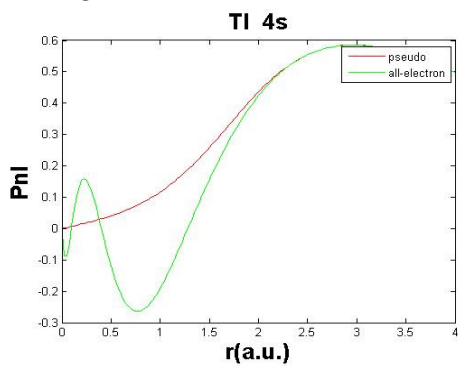


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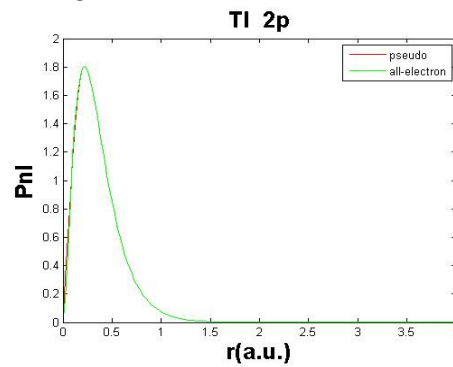


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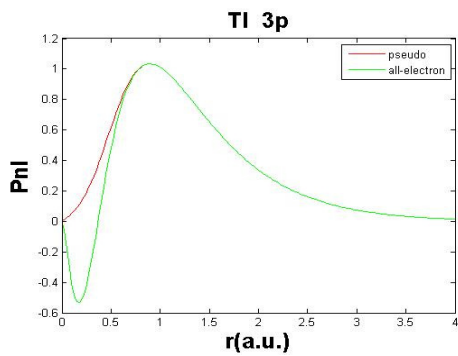


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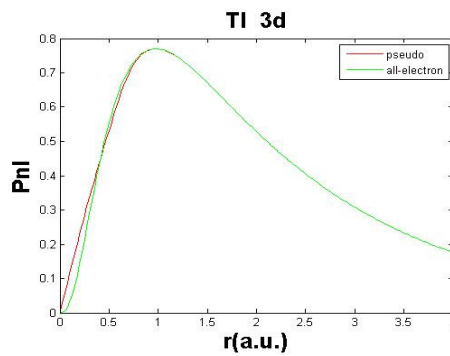


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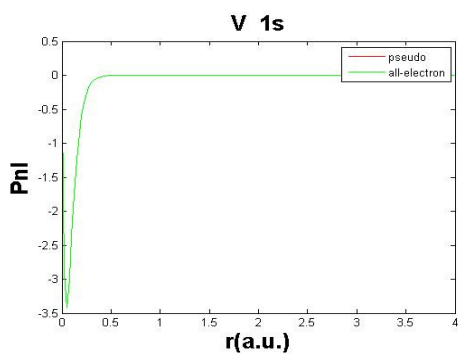


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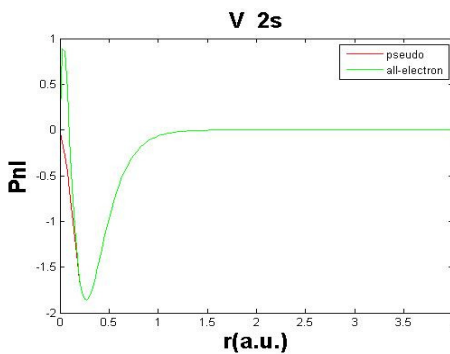


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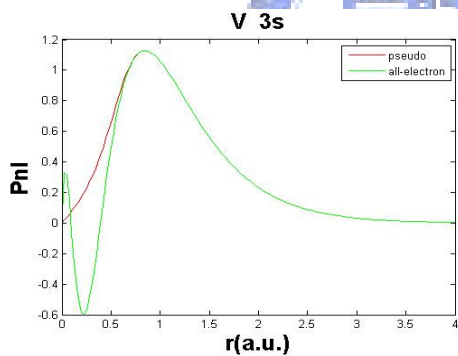


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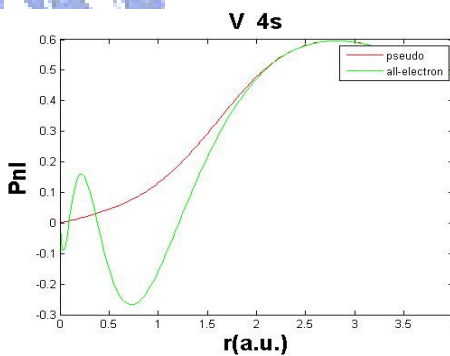


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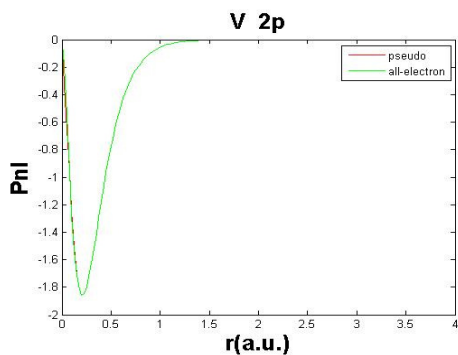


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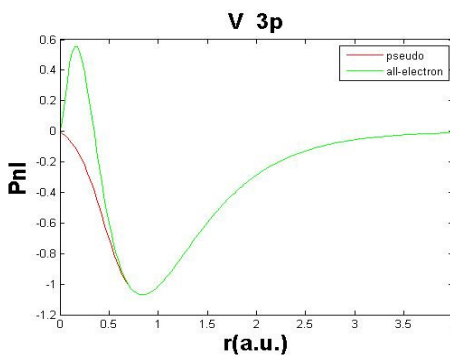


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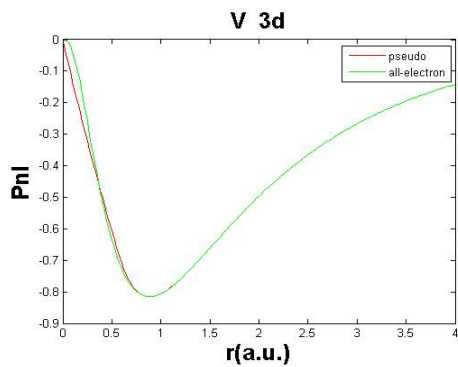


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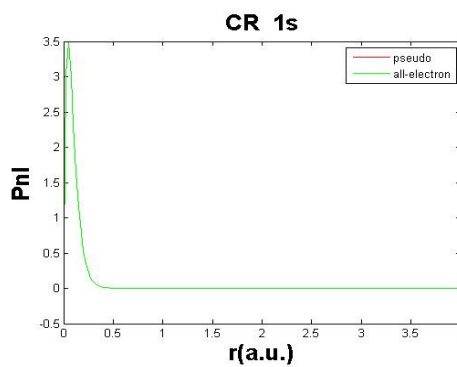


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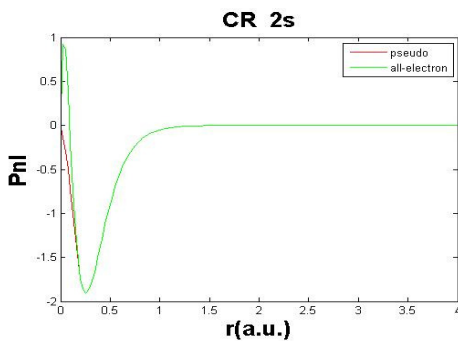


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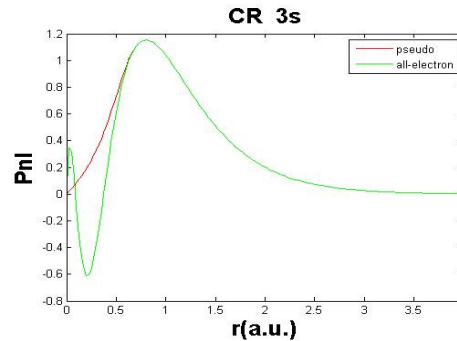


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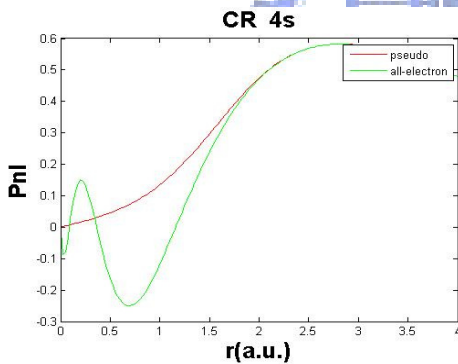


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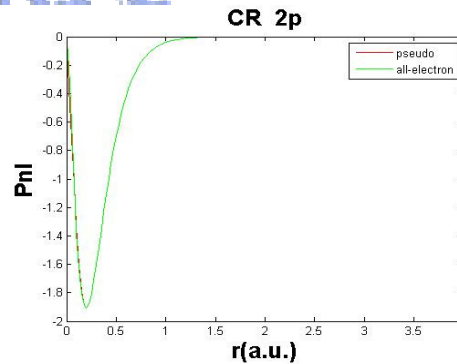


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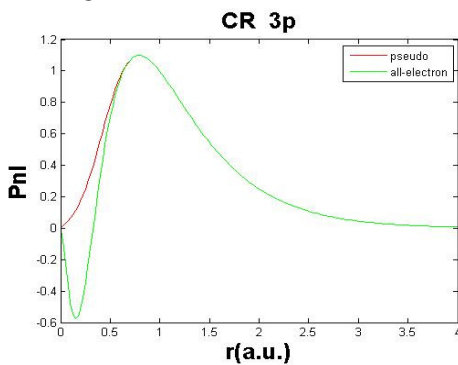


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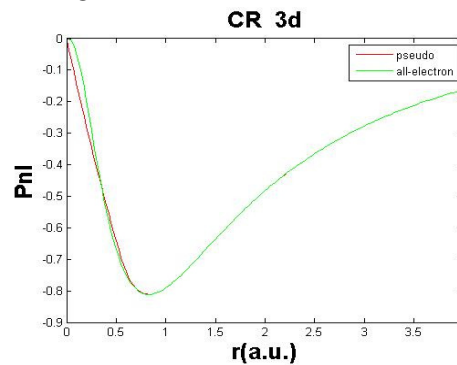


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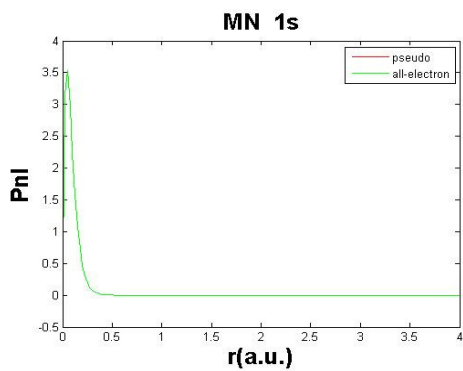


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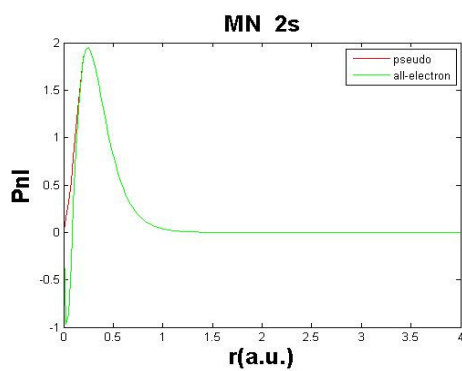


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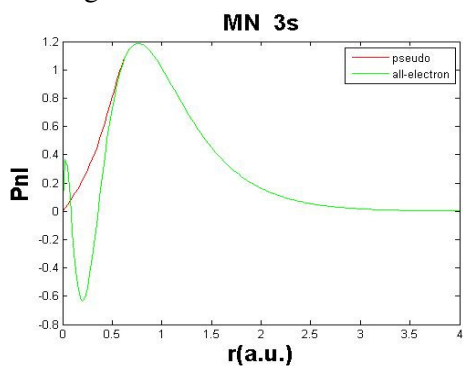


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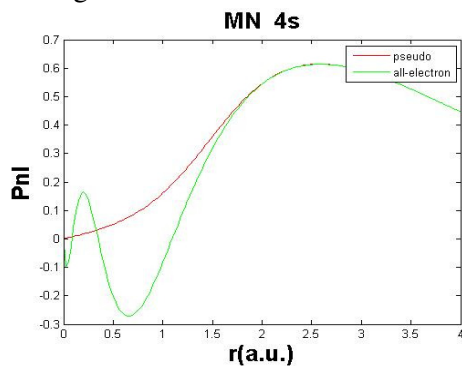


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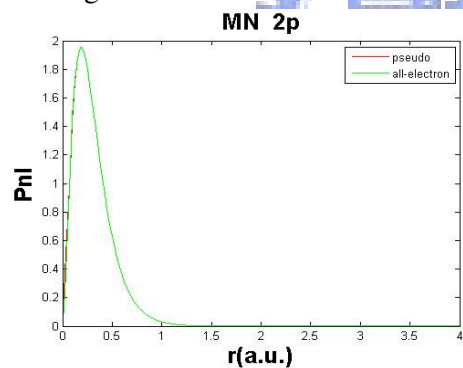


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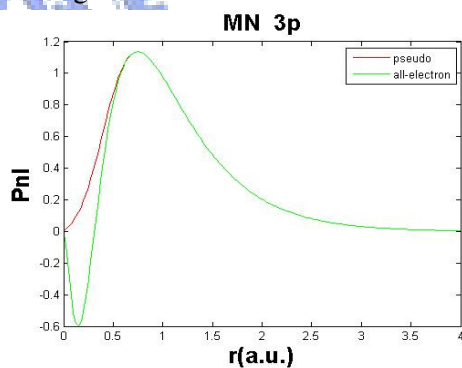


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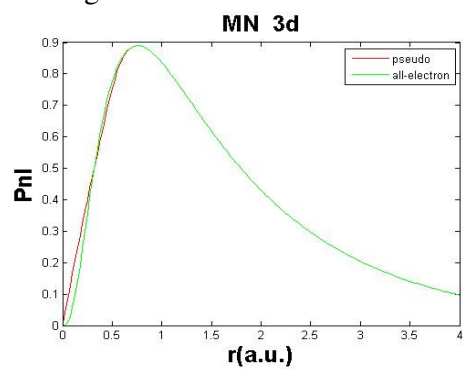


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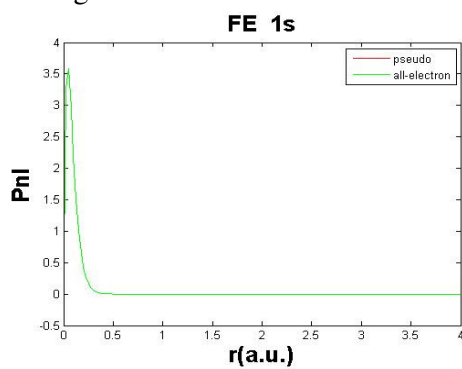


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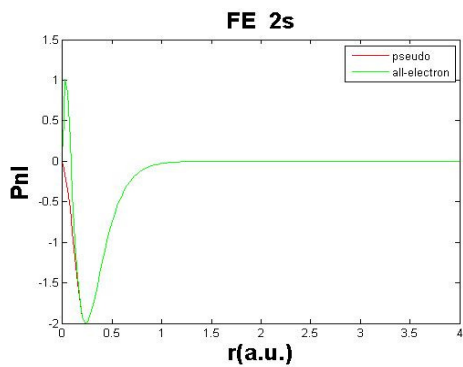


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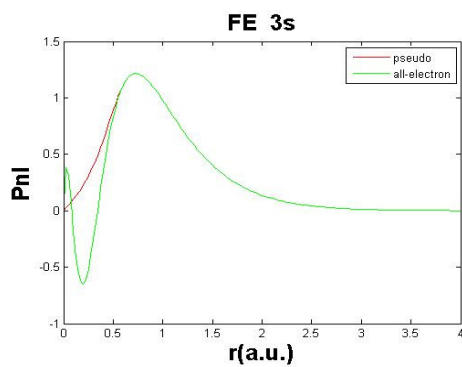


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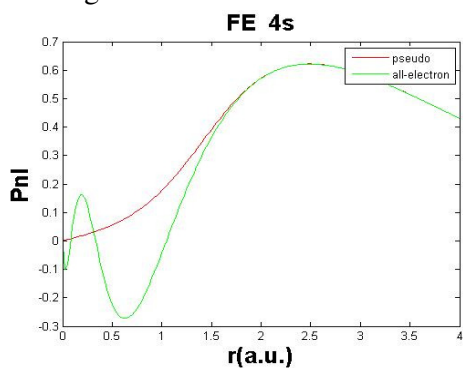


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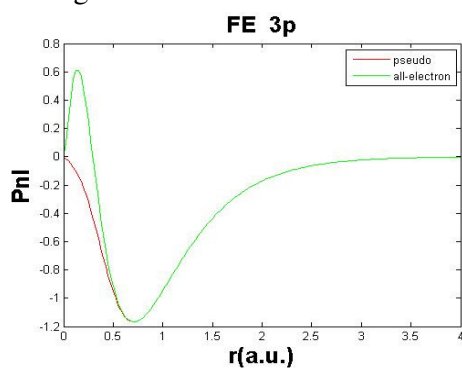


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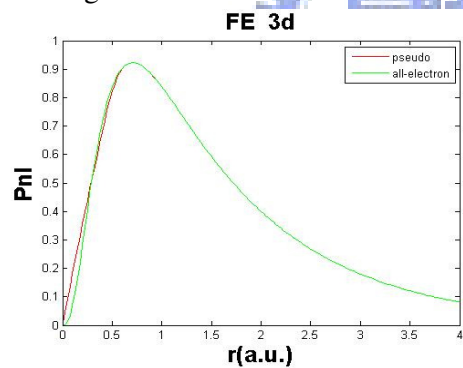


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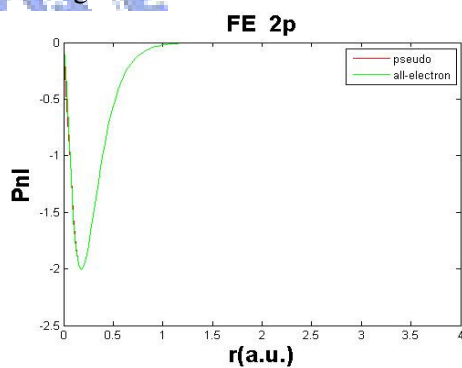


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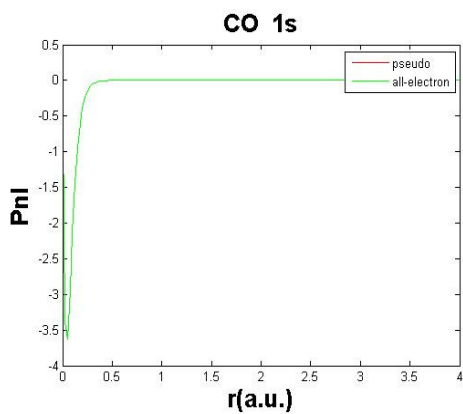


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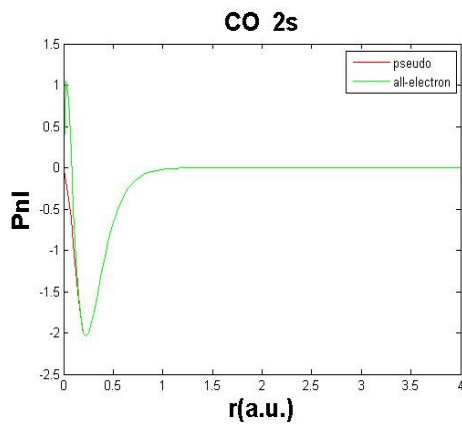


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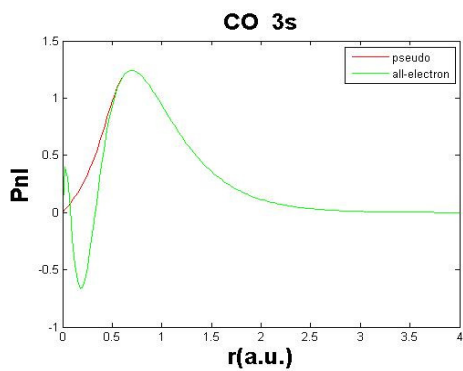


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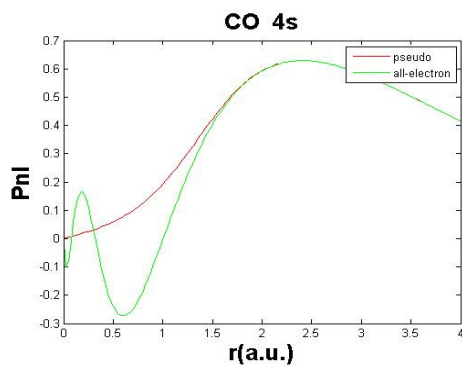


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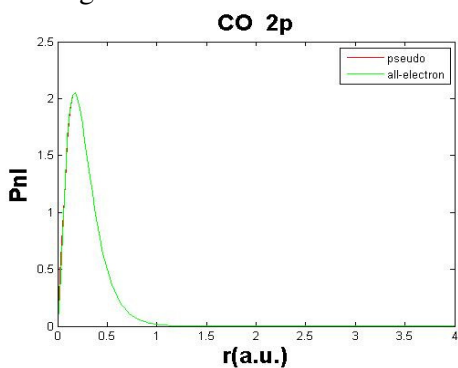


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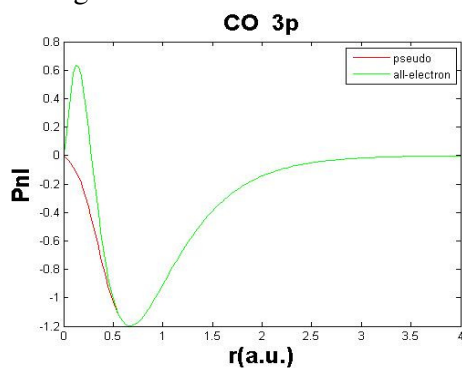


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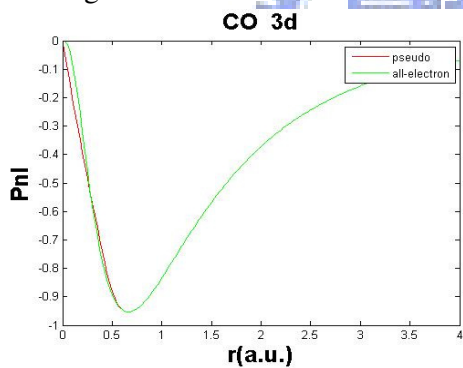


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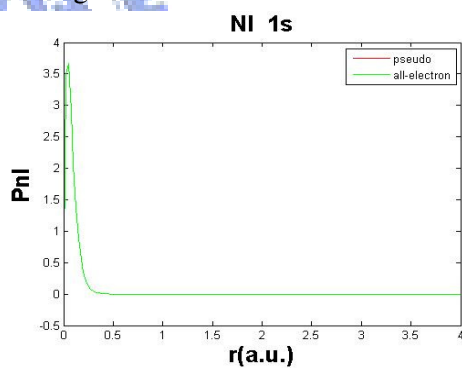


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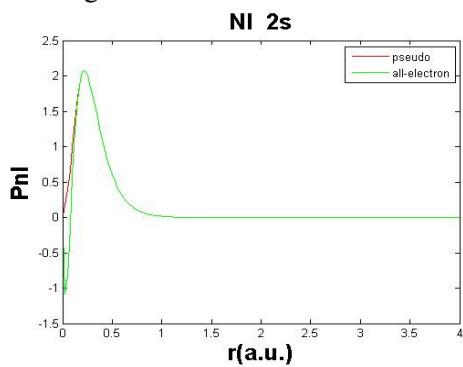


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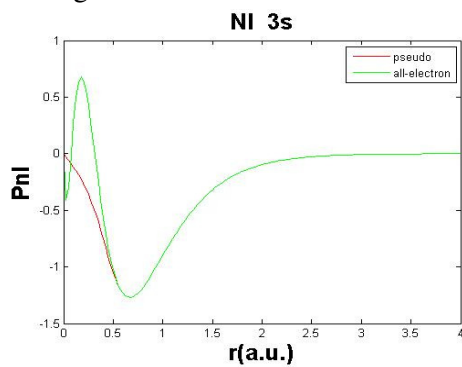


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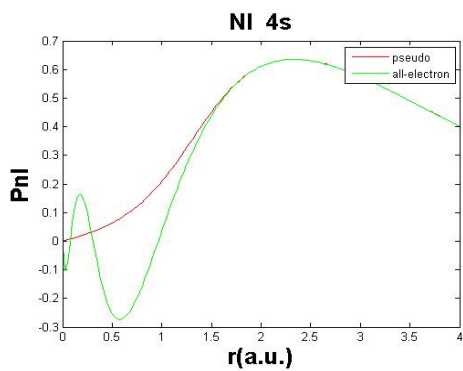


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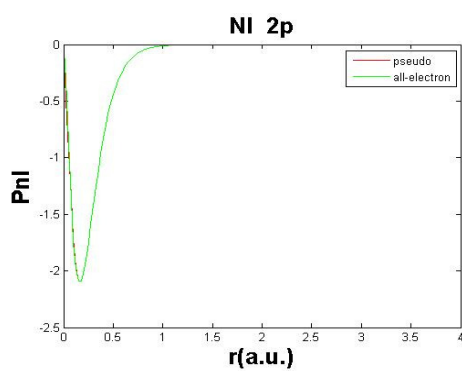


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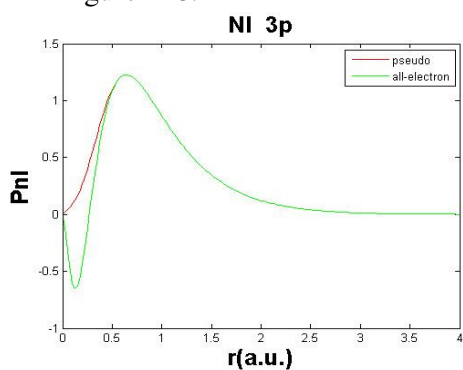


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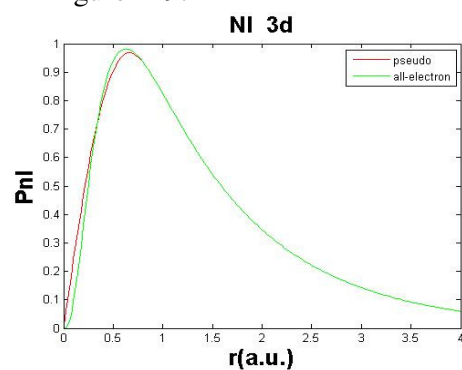


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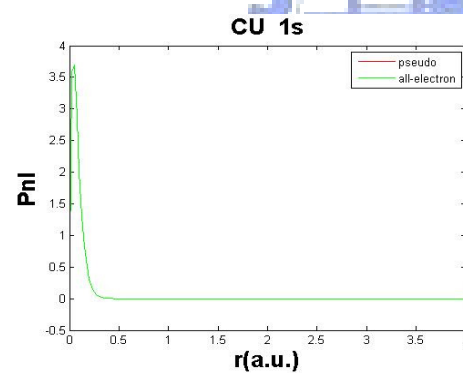


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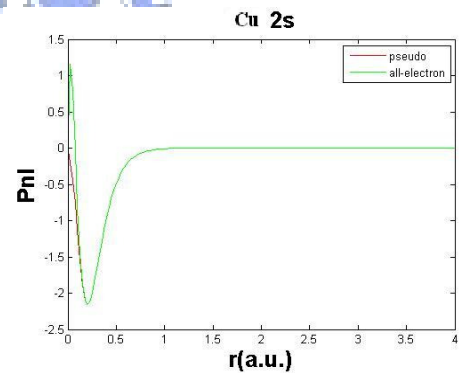


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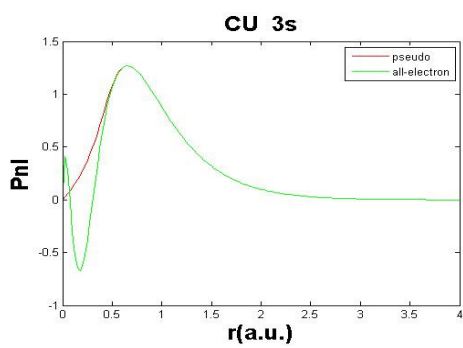


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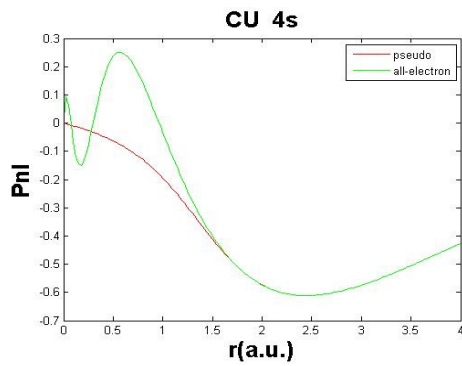


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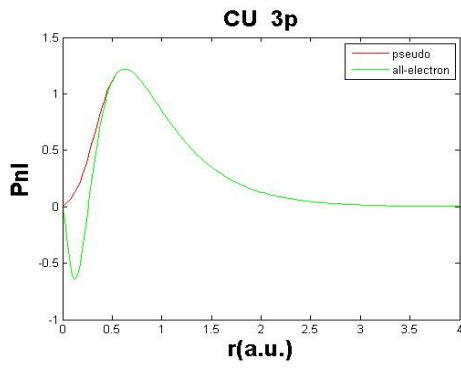


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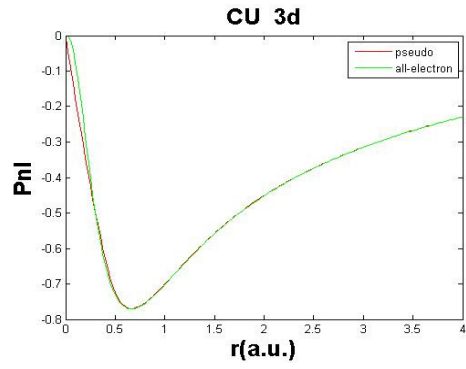


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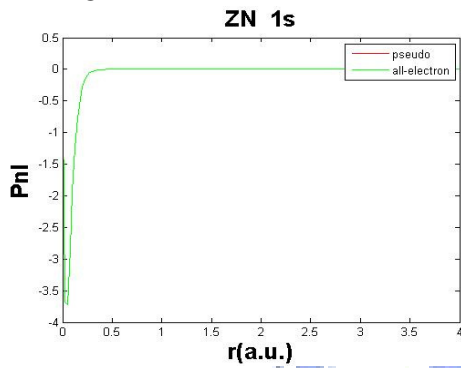


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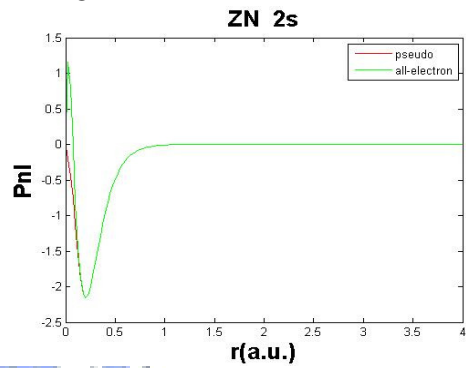


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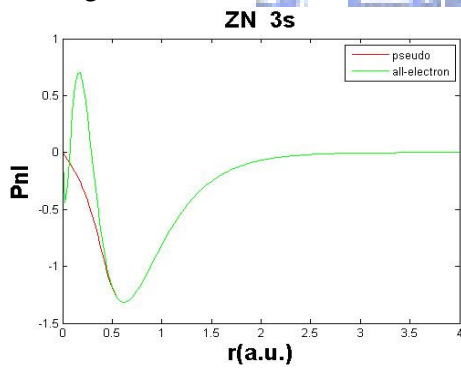


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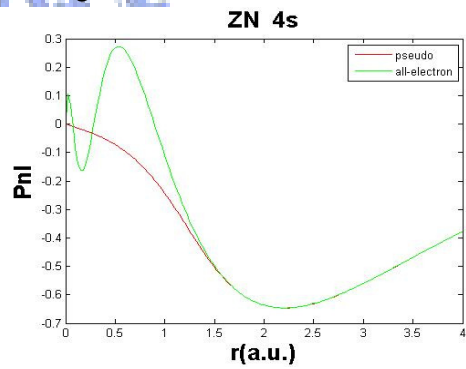


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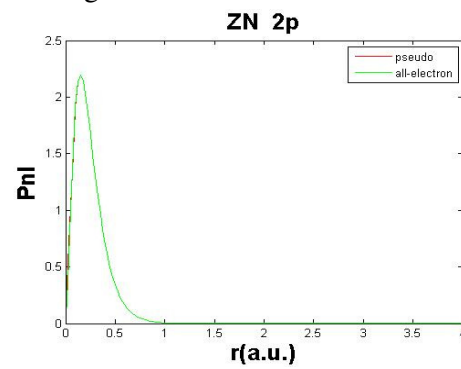


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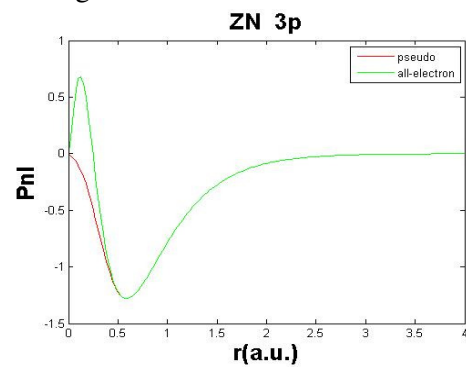


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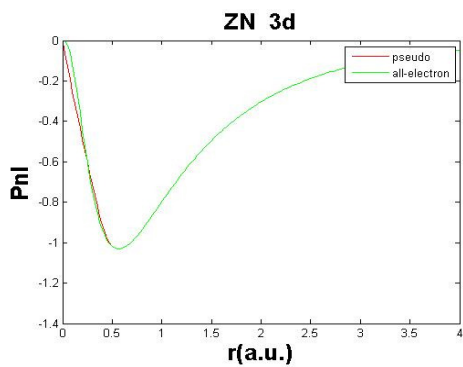


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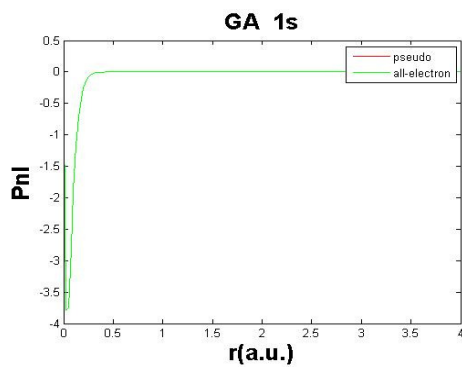


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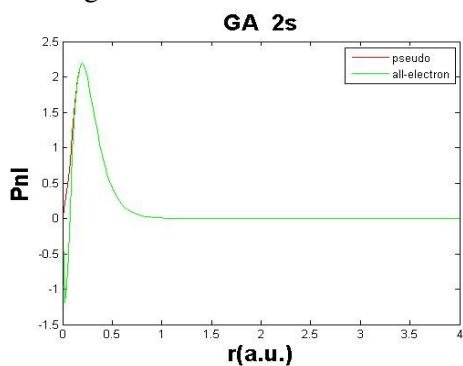


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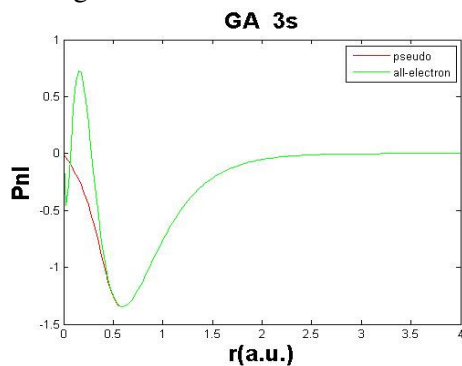


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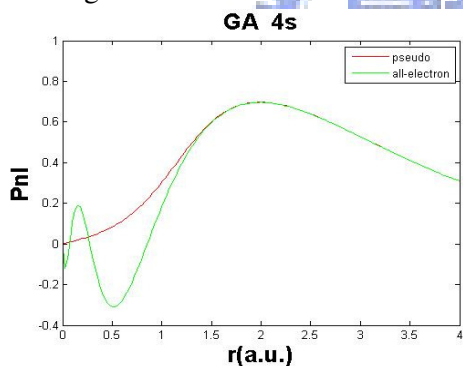


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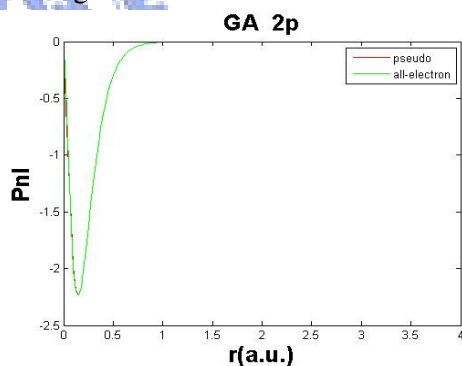


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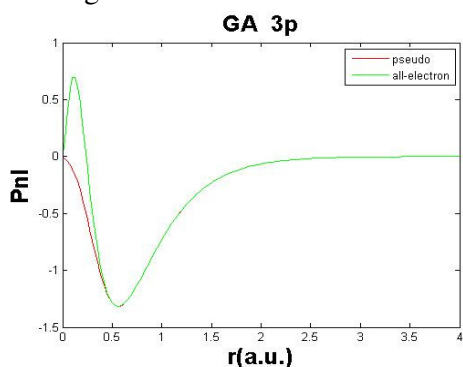


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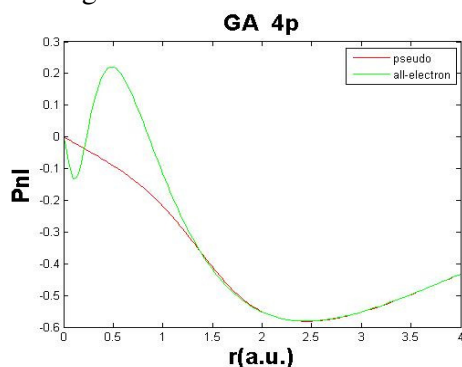


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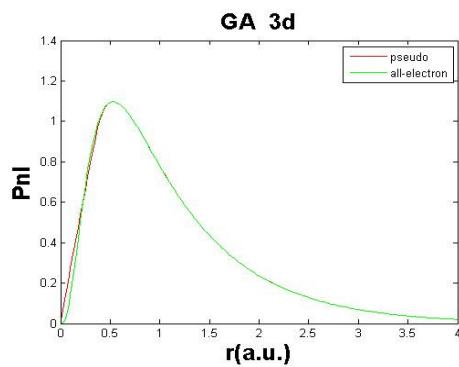


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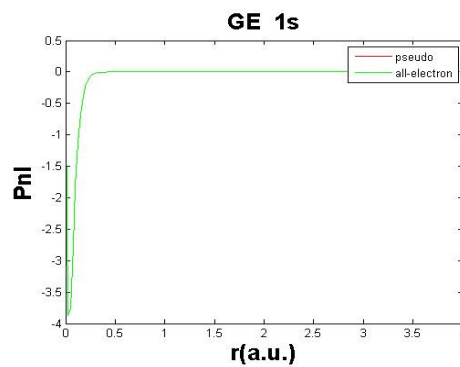


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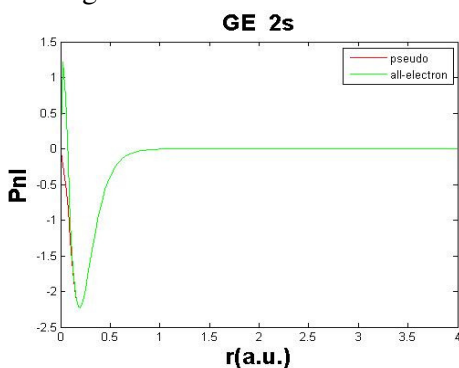


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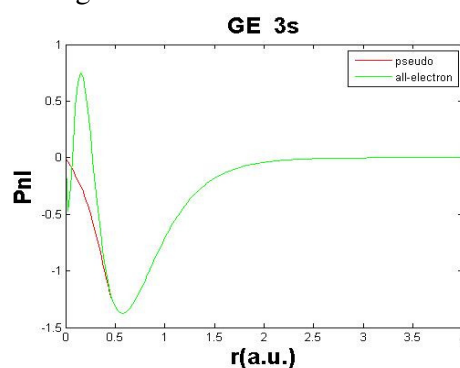


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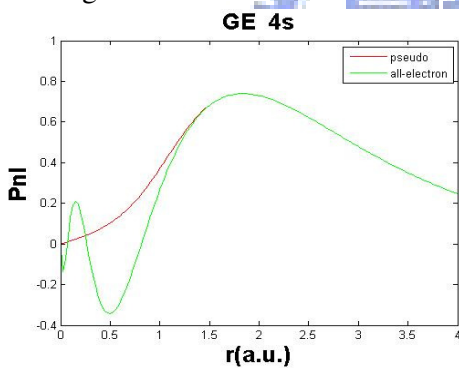


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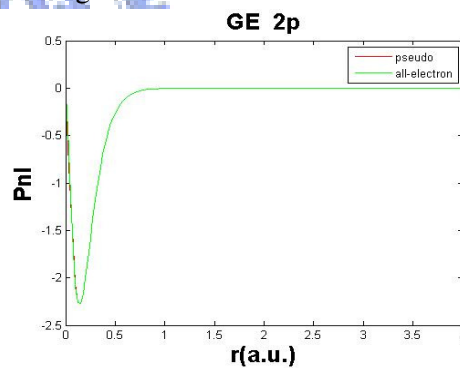


Figure 157.

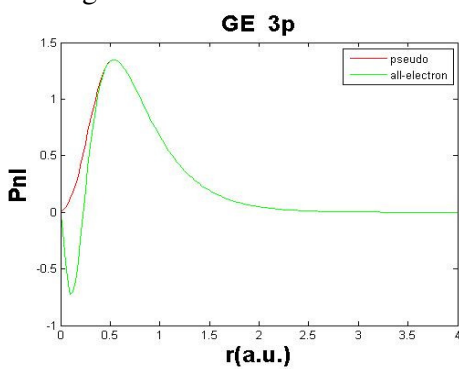


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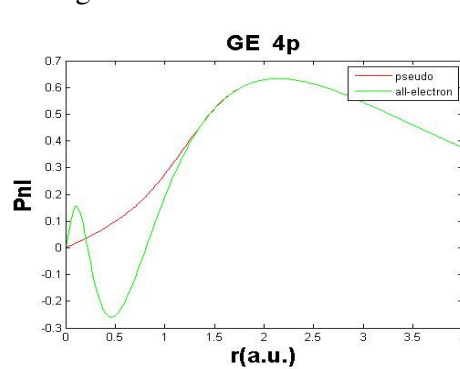


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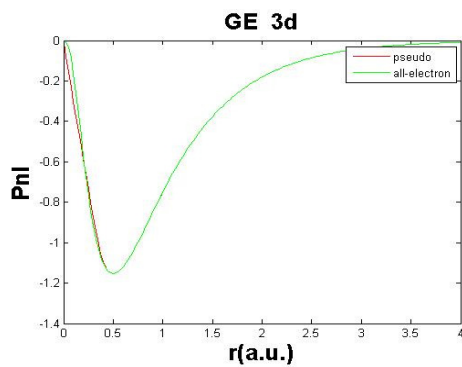


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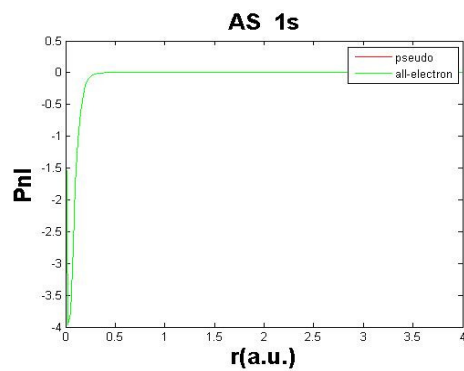


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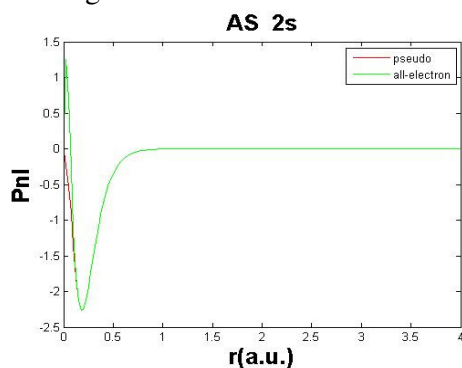


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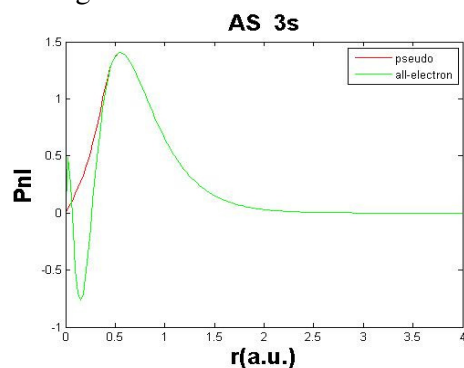


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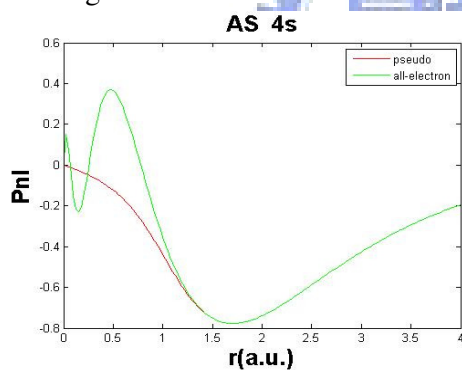


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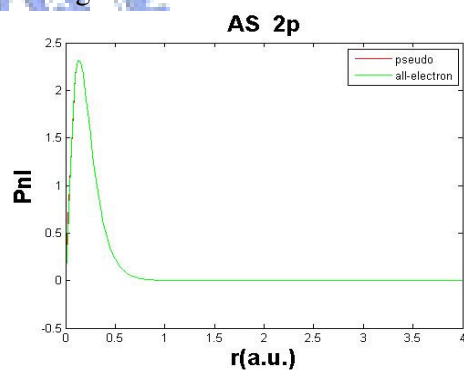


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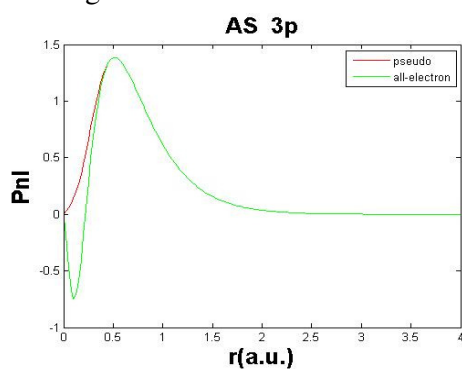


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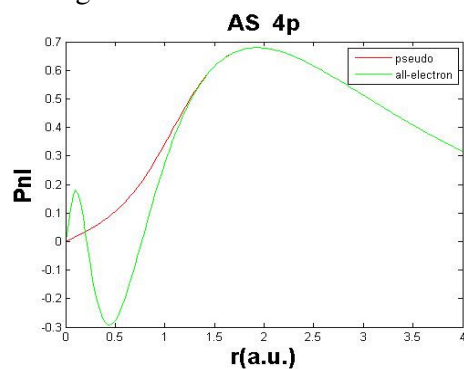


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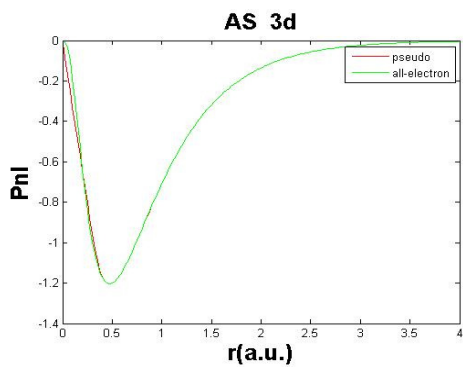


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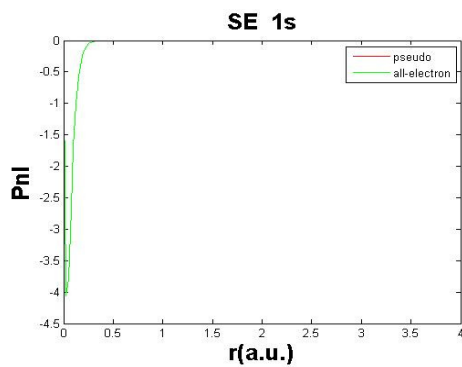


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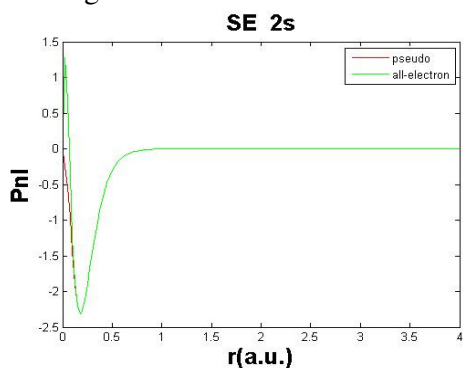


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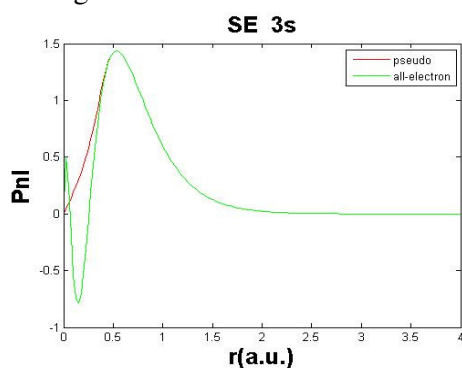


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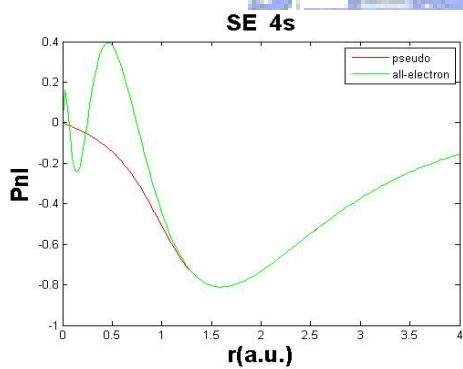


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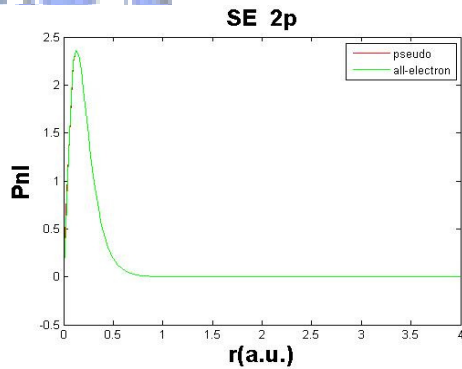


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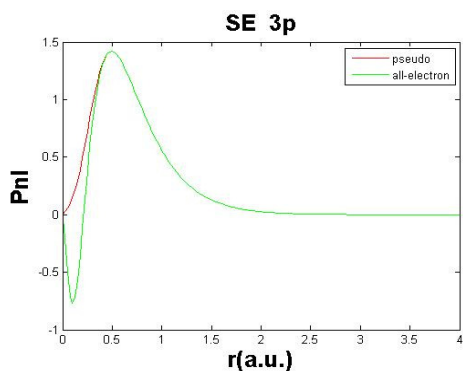


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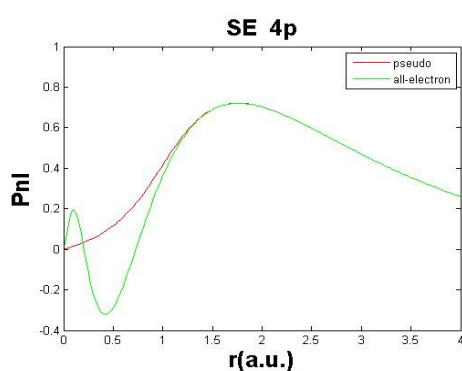


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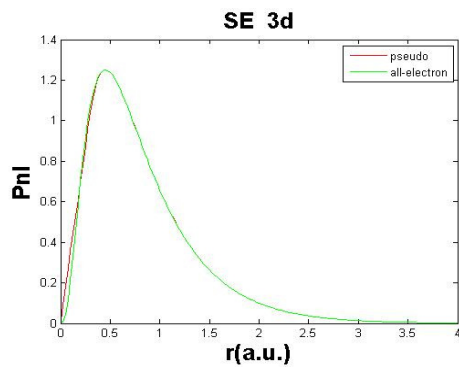


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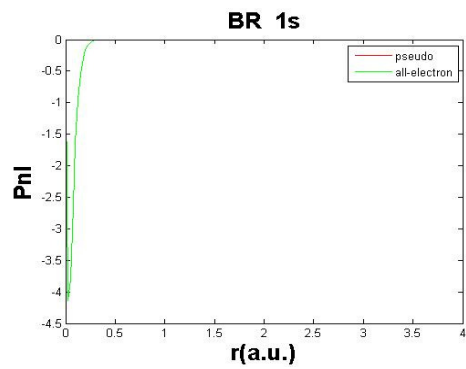


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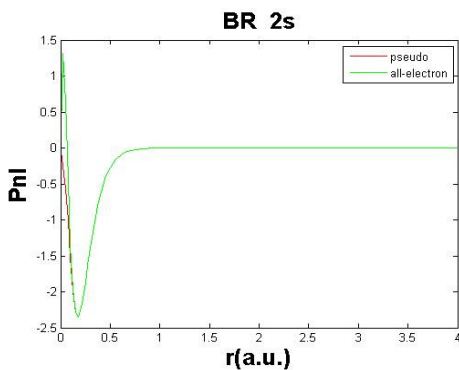


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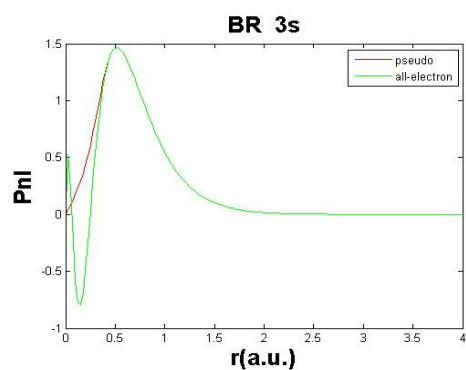


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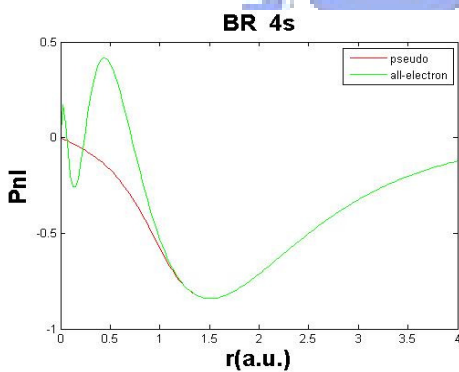


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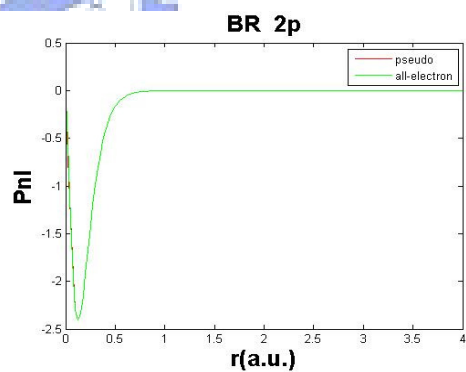


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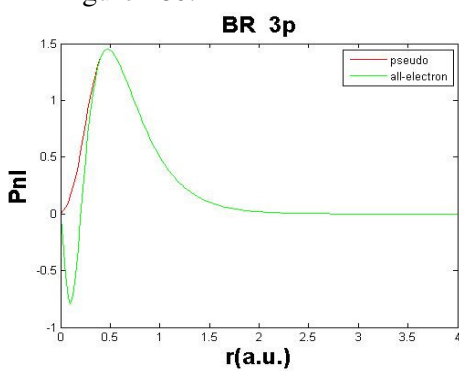


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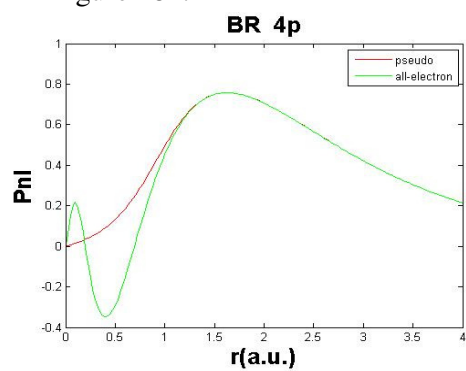


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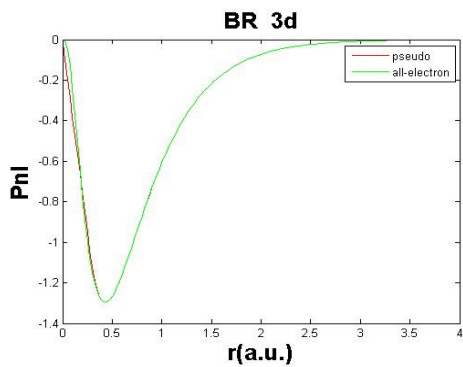


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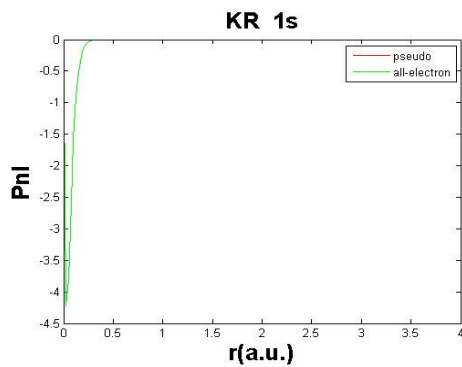


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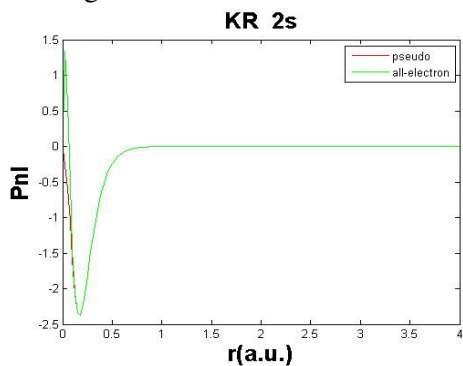


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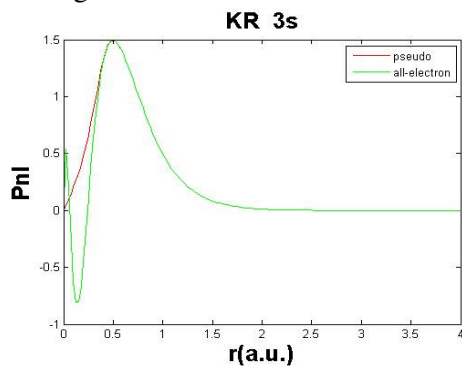


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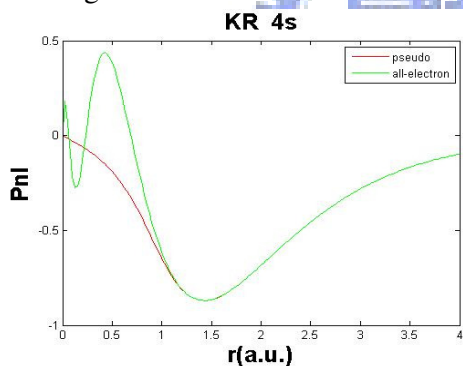


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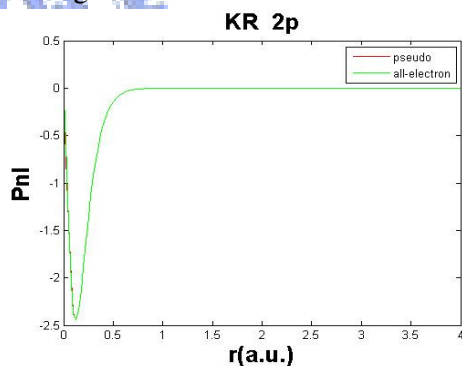


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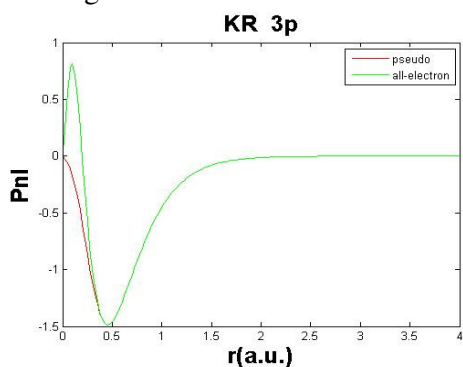


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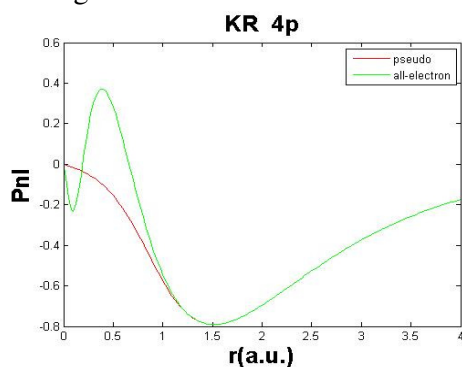


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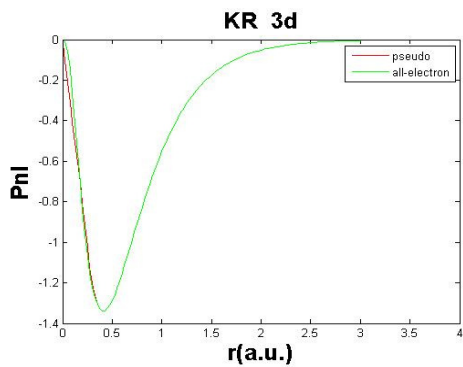


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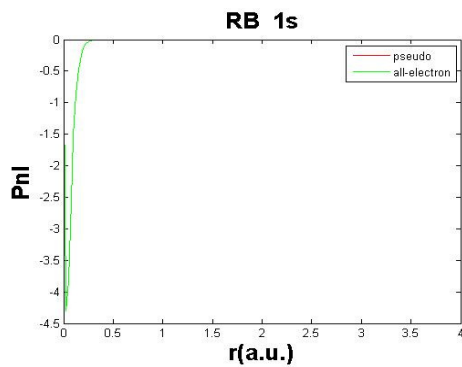


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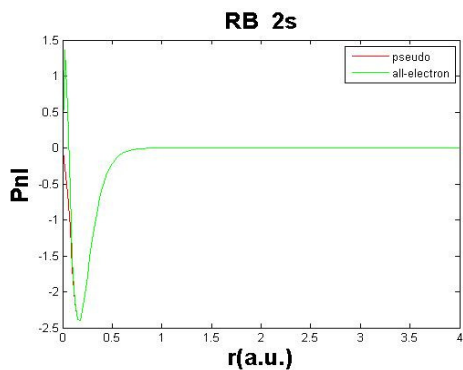


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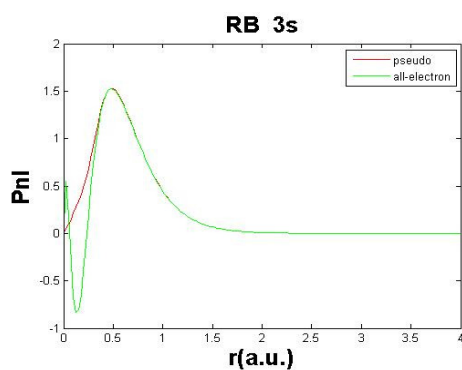


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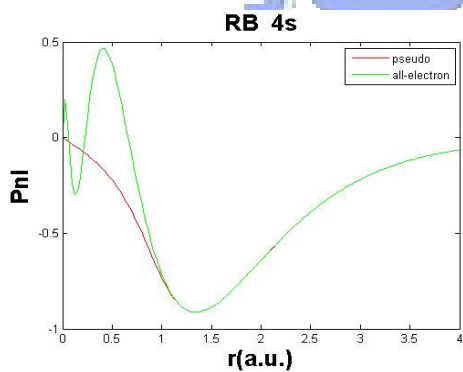


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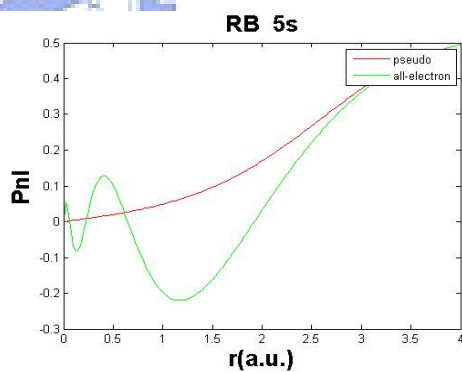


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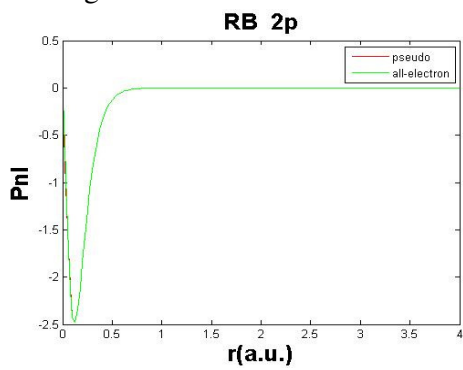


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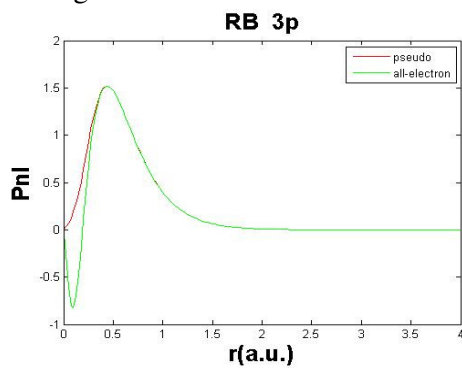


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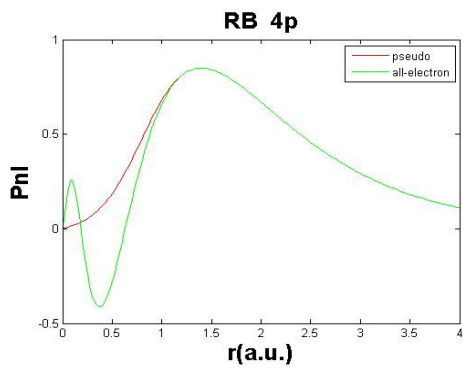


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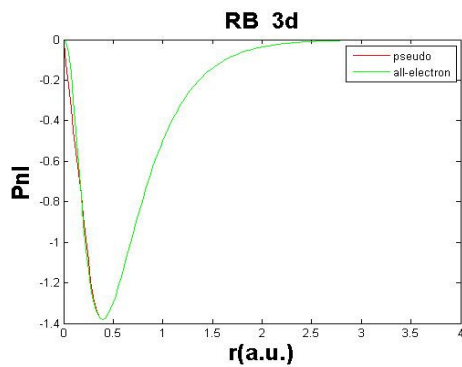


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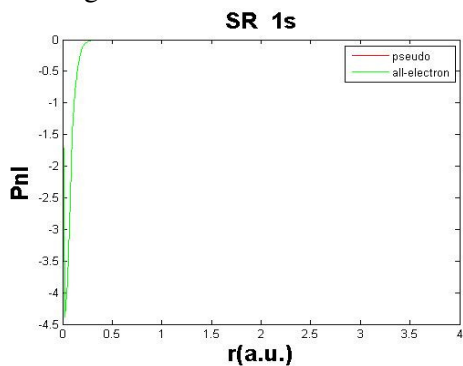


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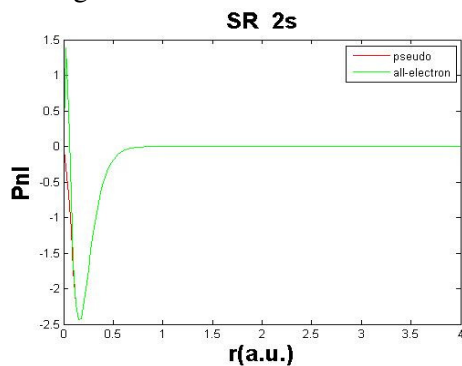


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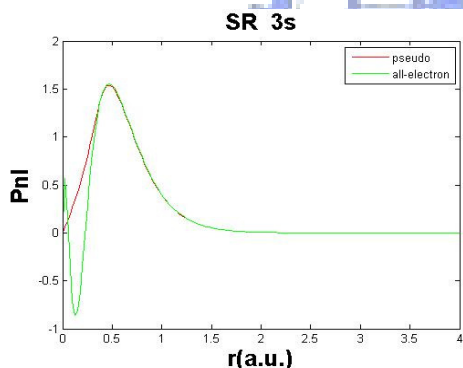


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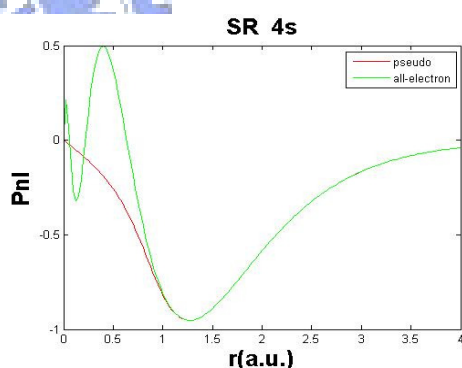


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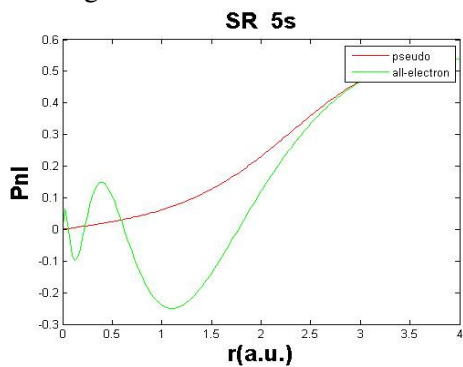


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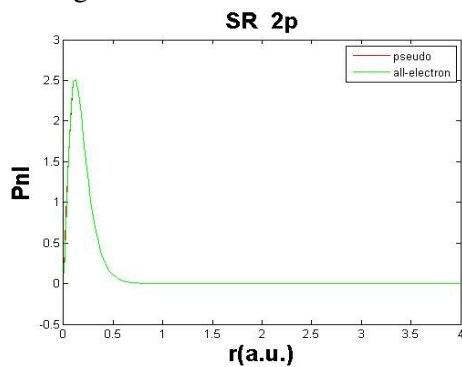


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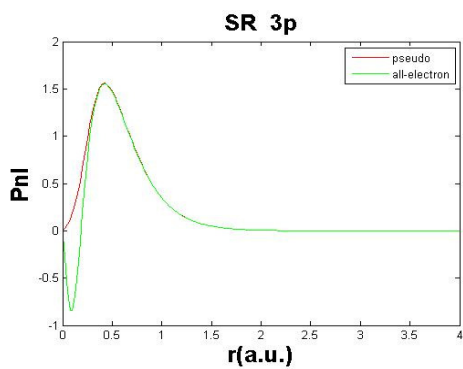


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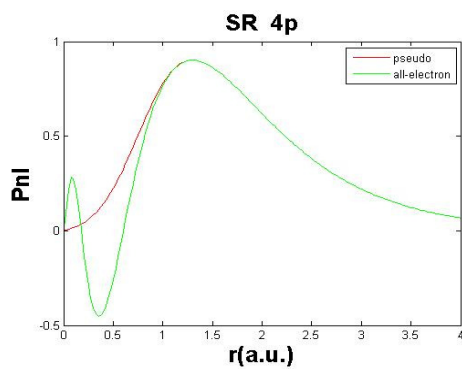


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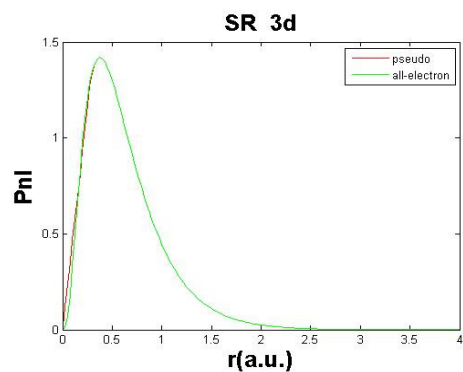


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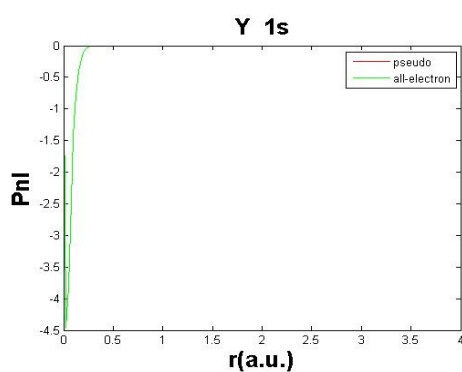


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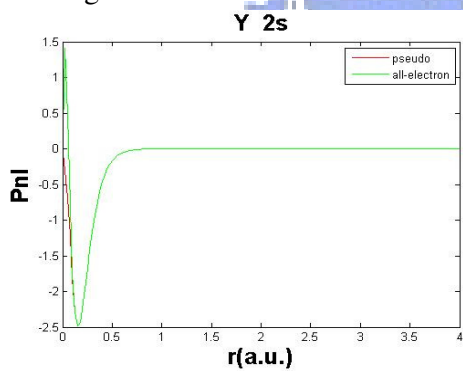


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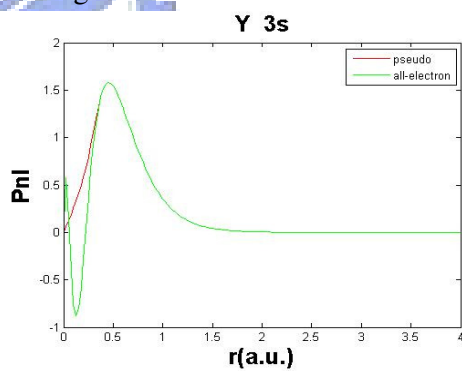


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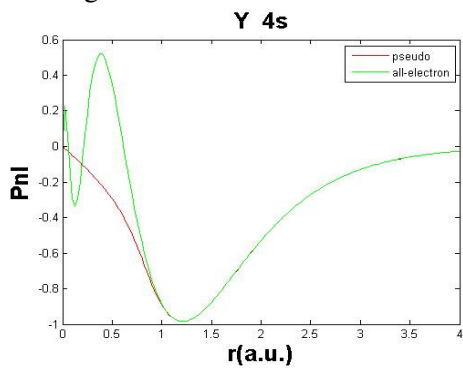


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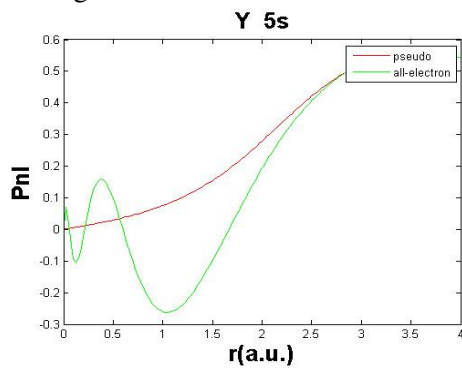


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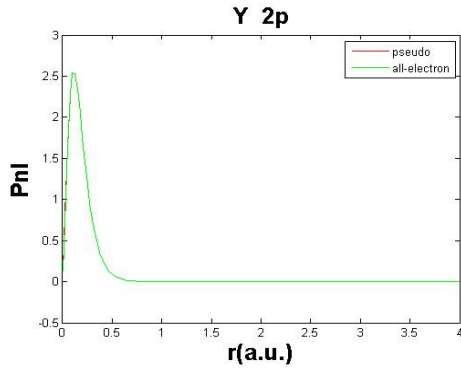


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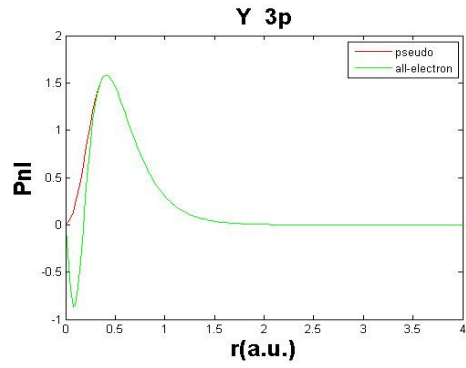


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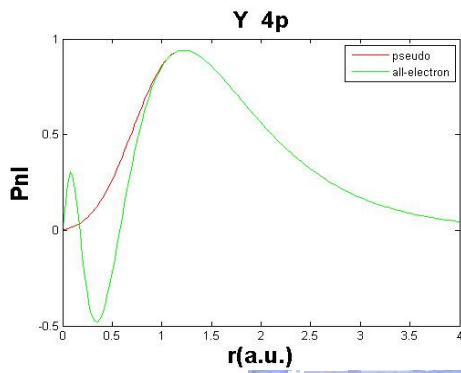


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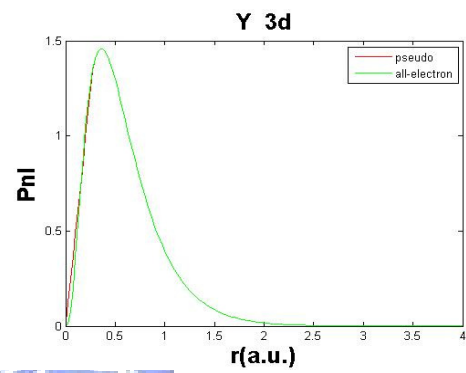


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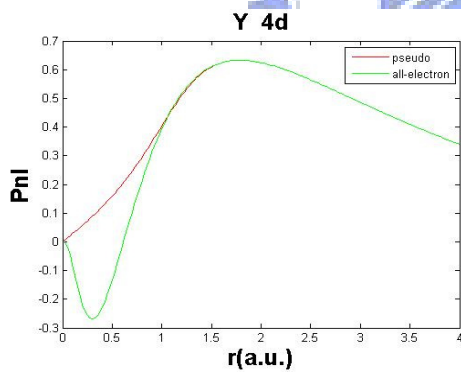


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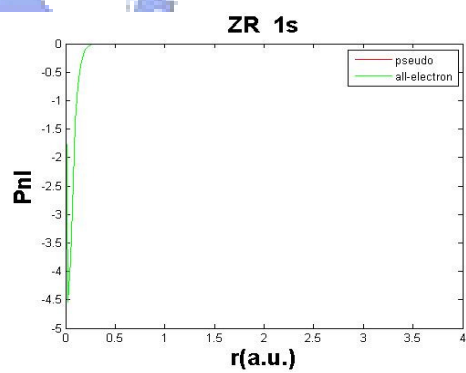


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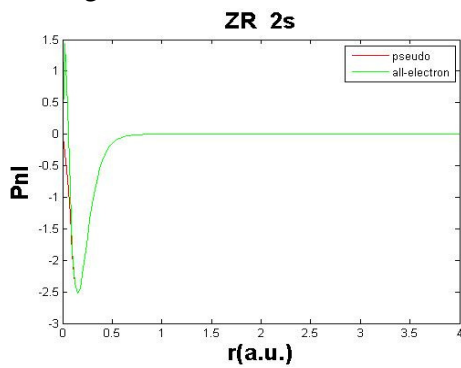


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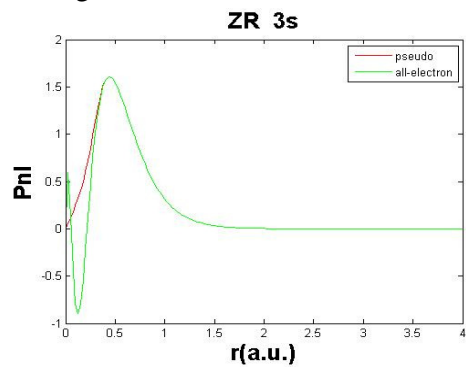


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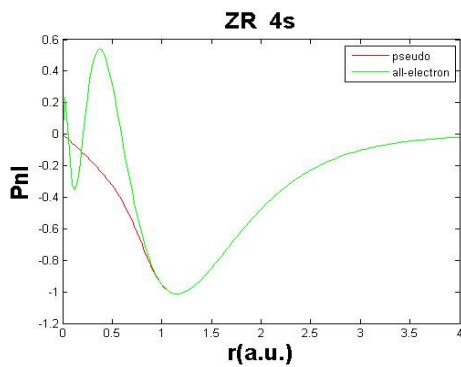


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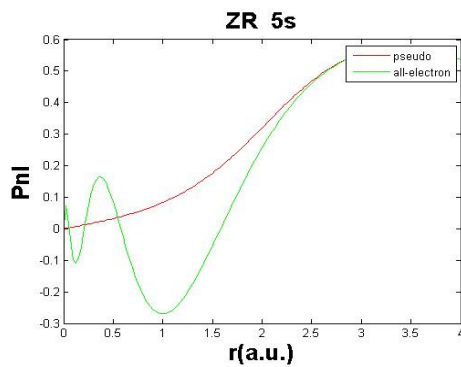


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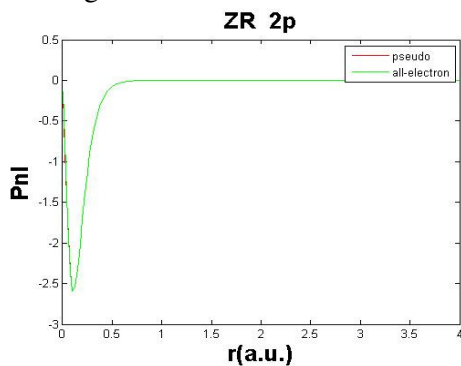


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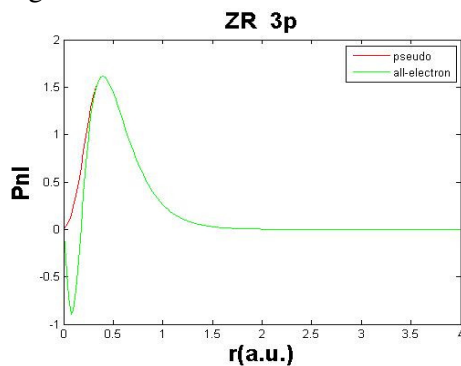


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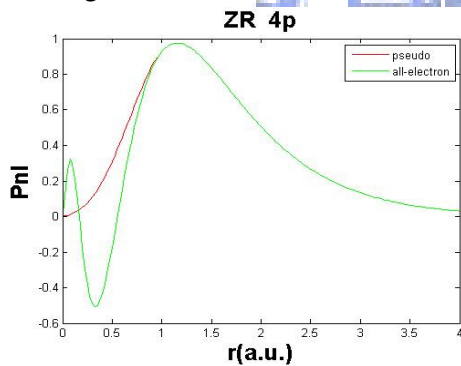


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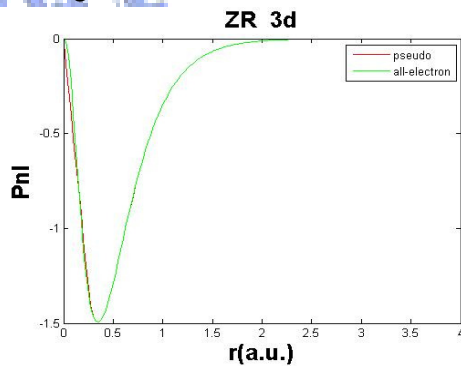


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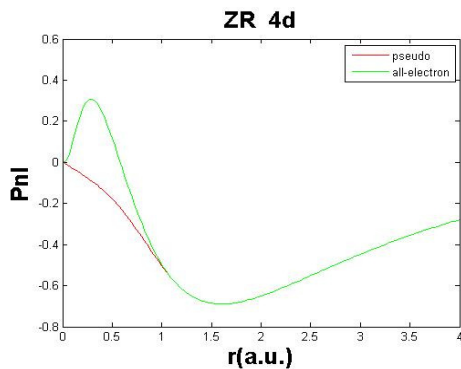


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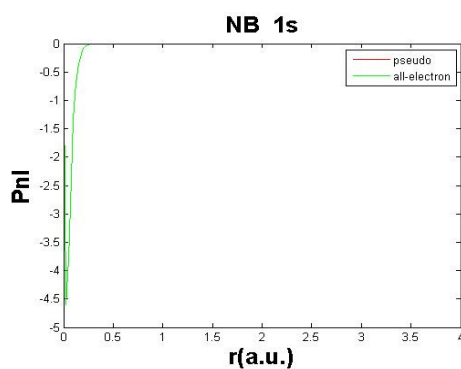


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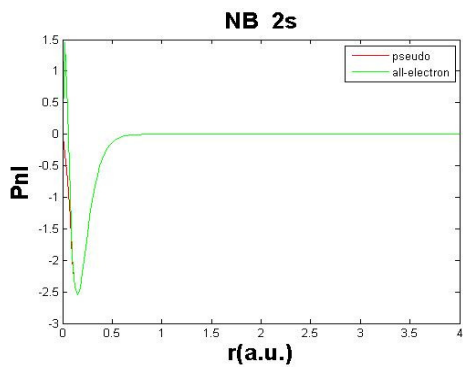


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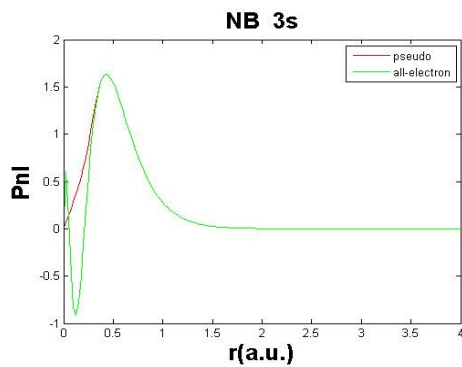


Figure 233.

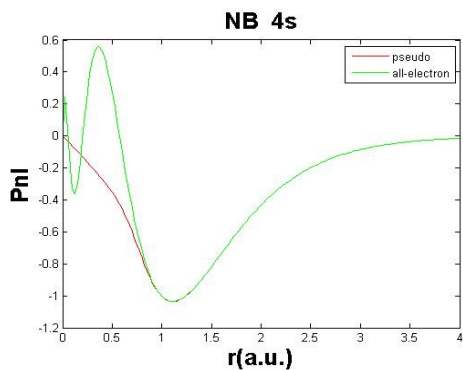


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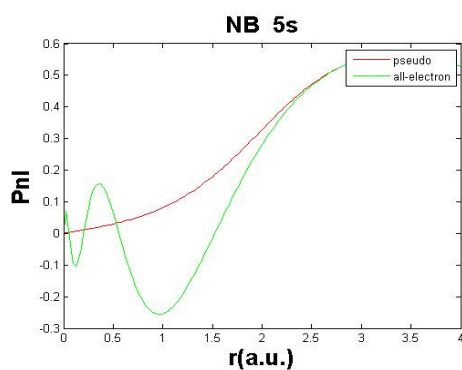


Figure 235.

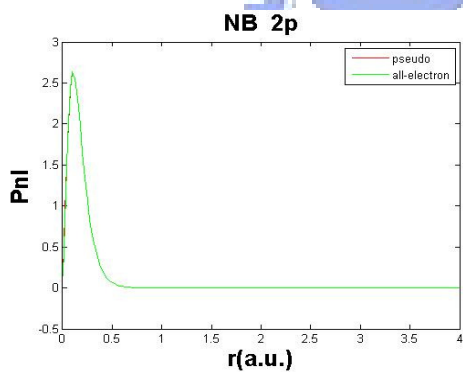


Figure 236.

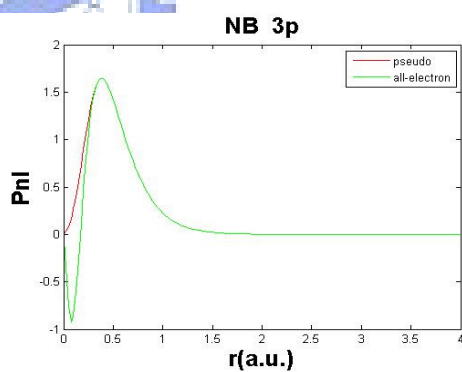


Figure 237.

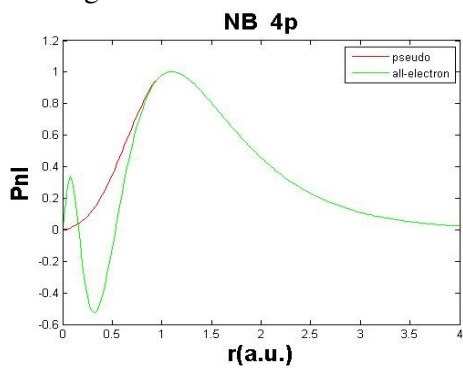


Figure 238.

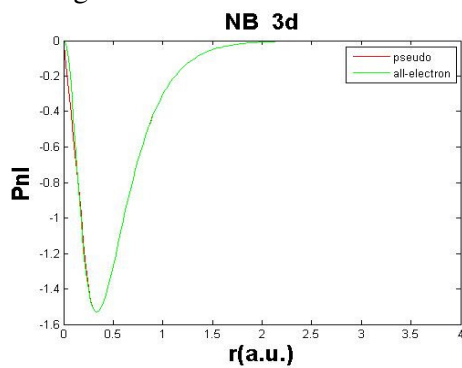


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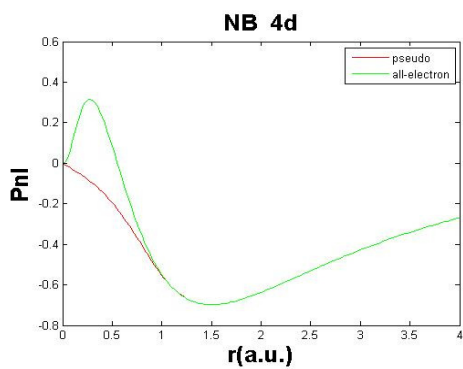


Figure 240.

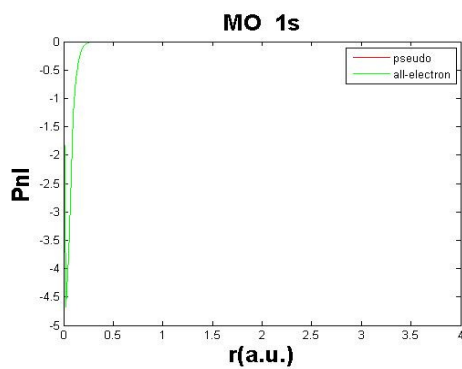


Figure 241.

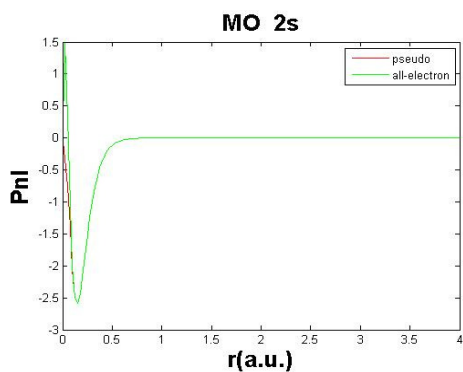


Figure 242.

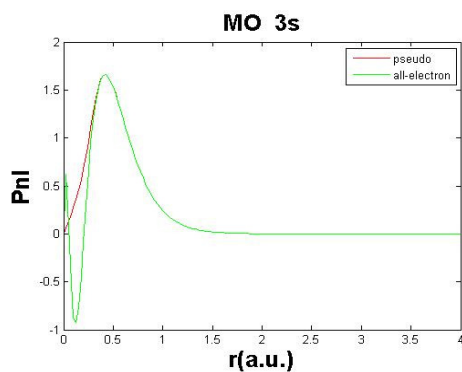


Figure 243.

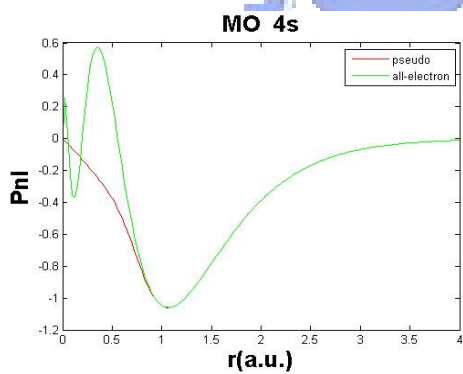


Figure 244.

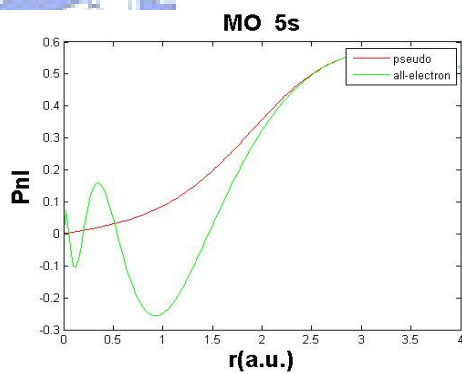


Figure 245.

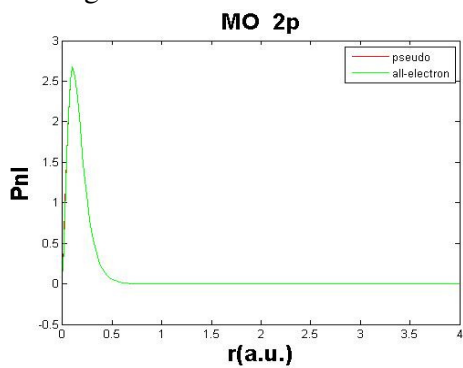


Figure 246.

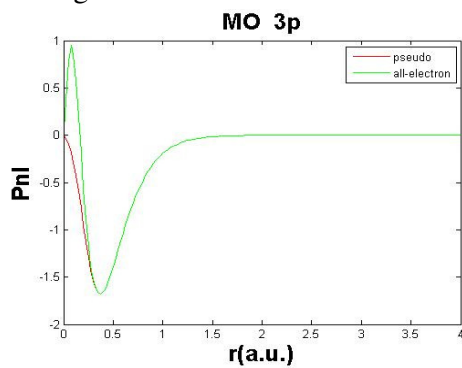


Figure 247.

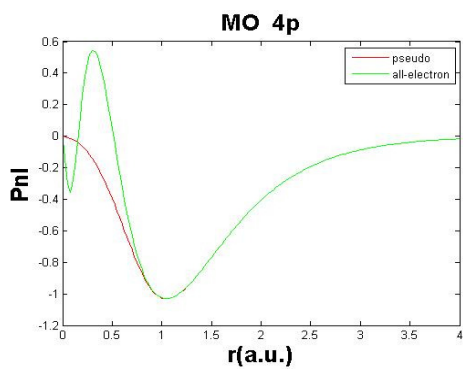


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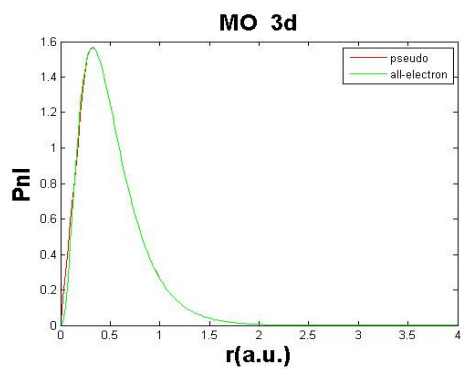


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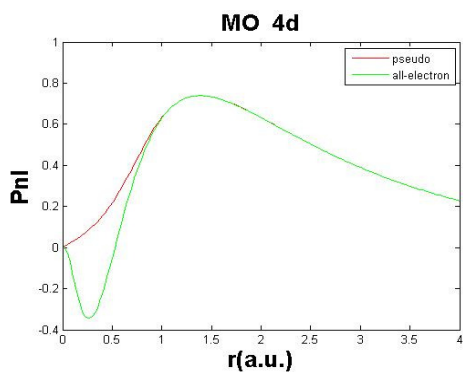


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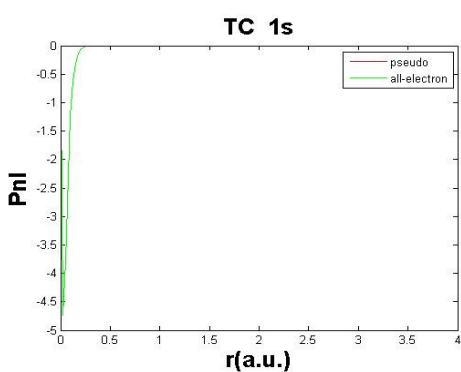


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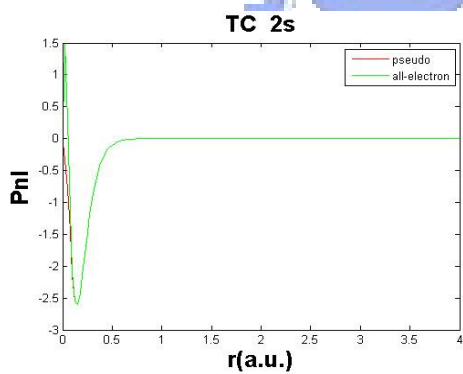


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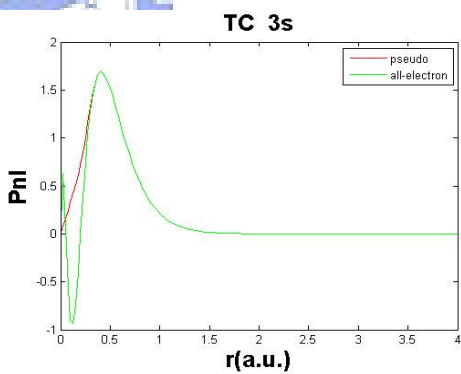


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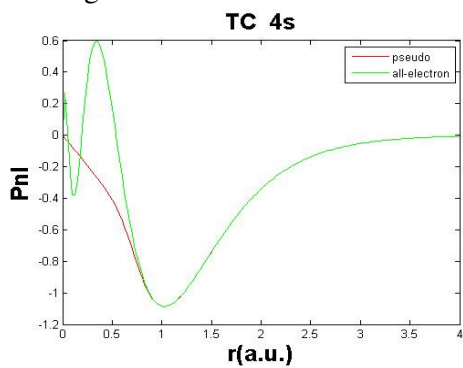


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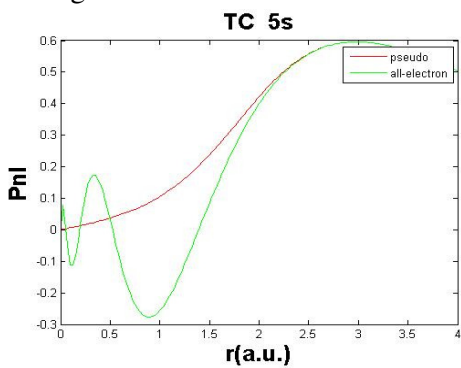


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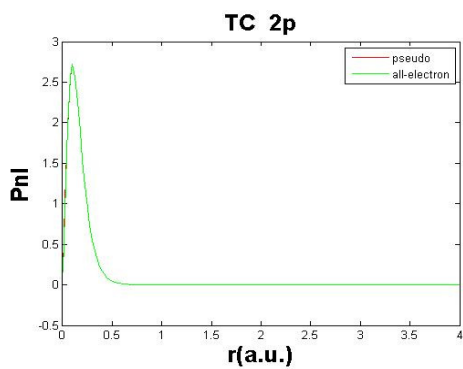


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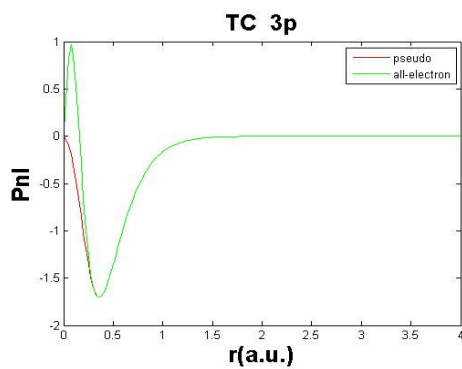


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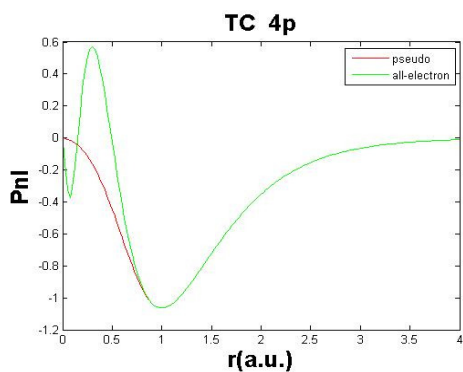


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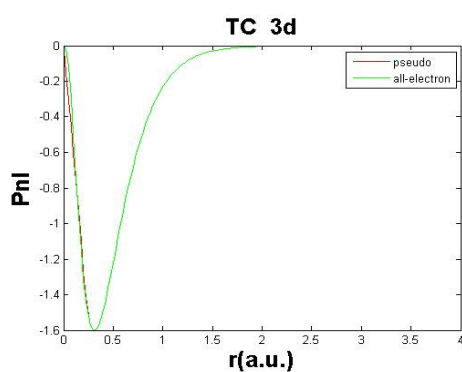


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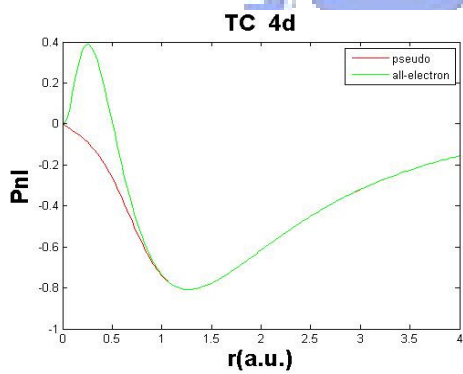


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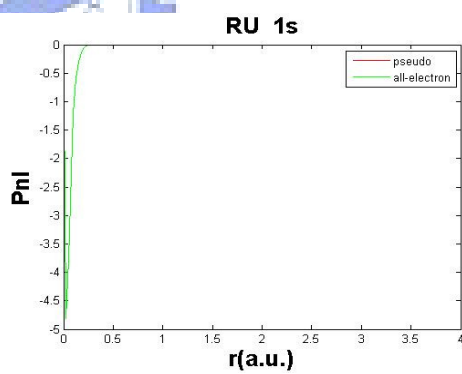


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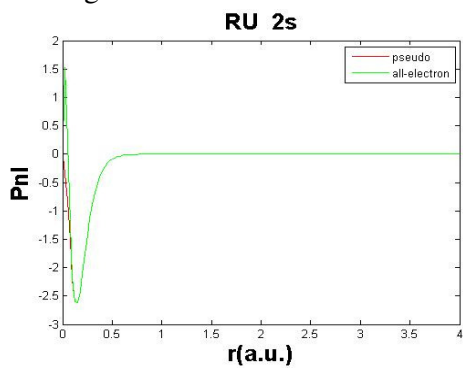


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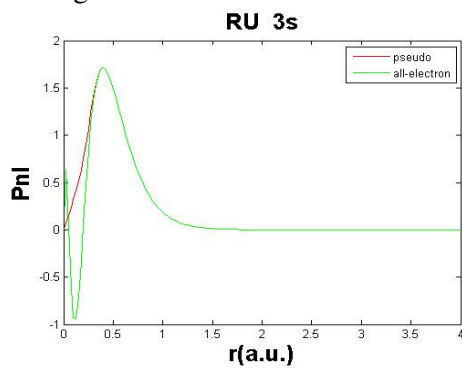


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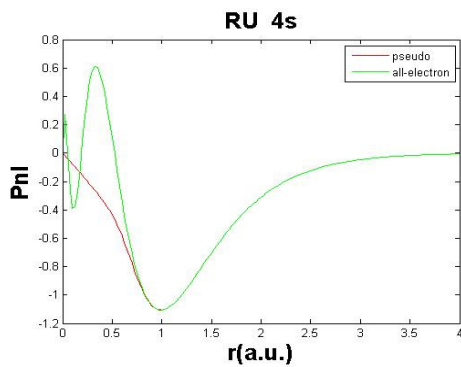


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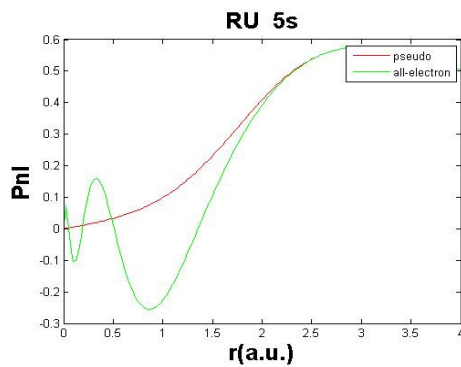


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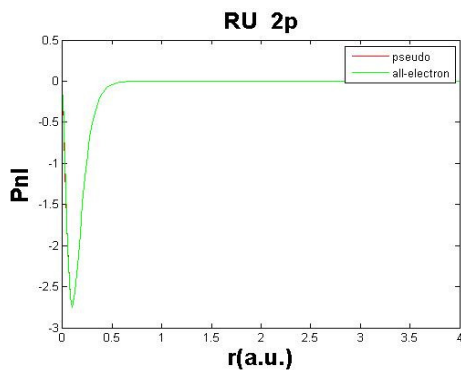


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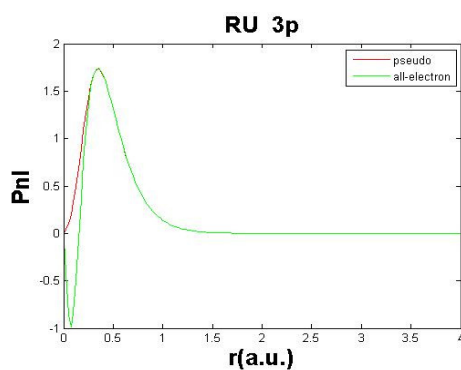


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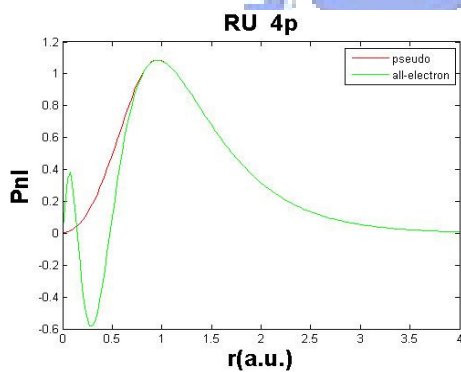


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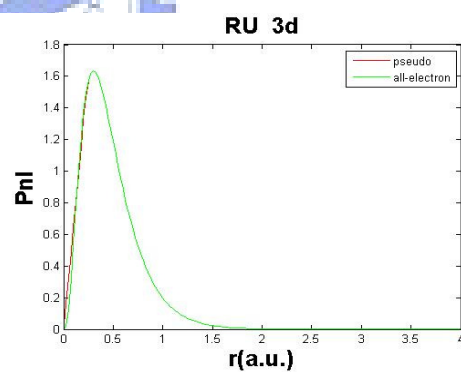


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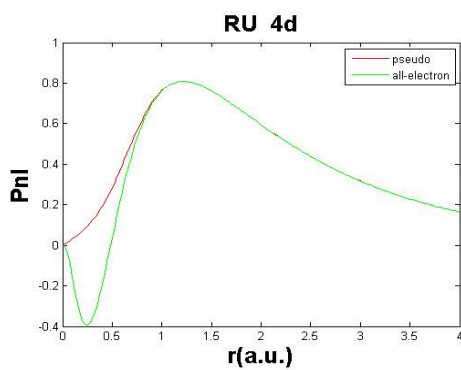


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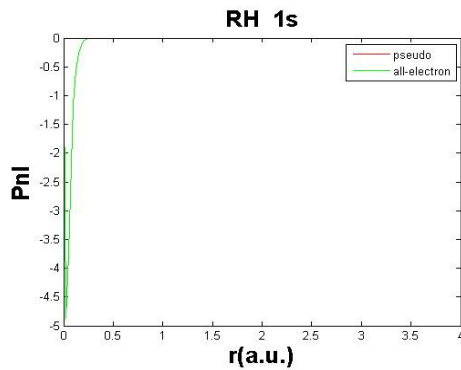


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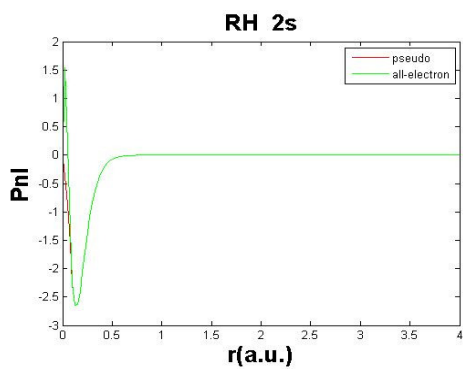


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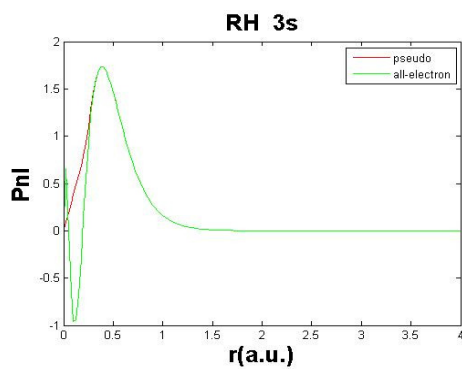


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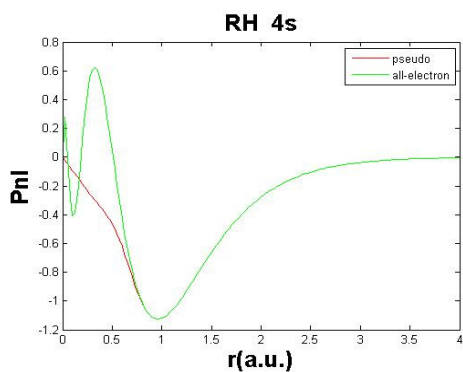


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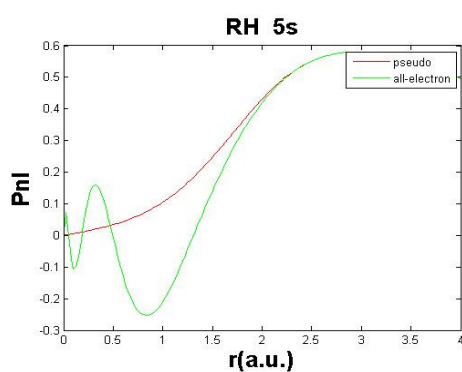


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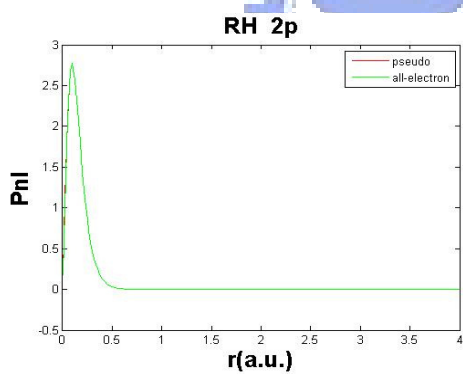


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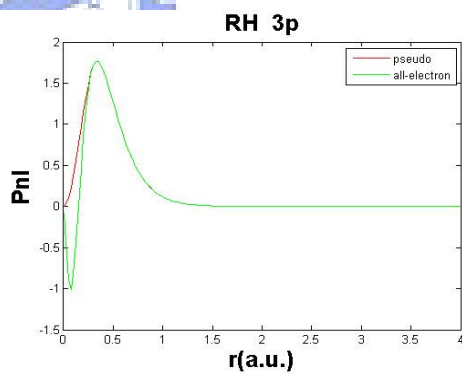


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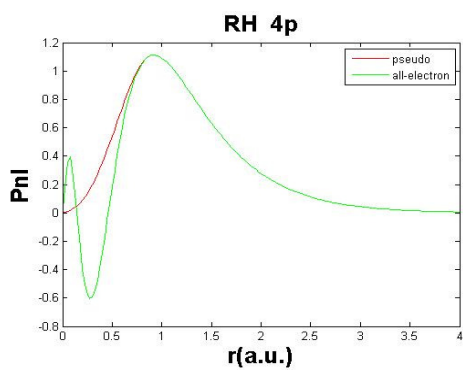


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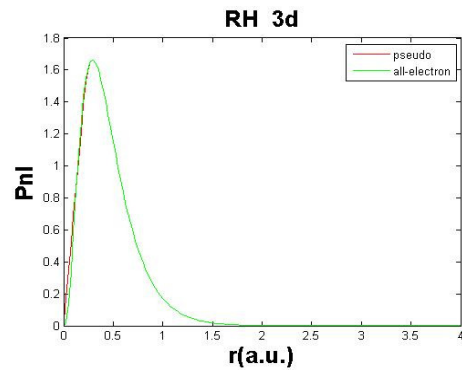


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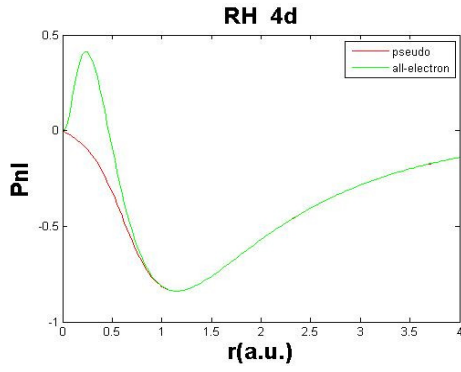


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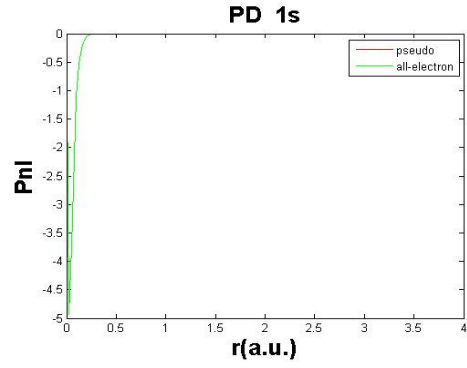


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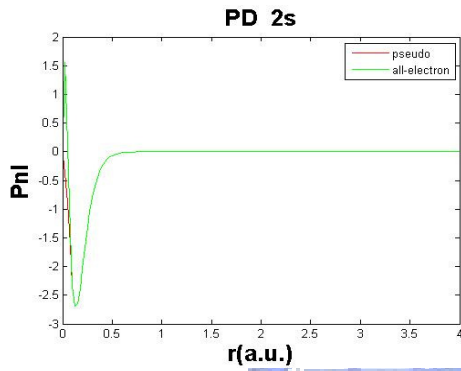


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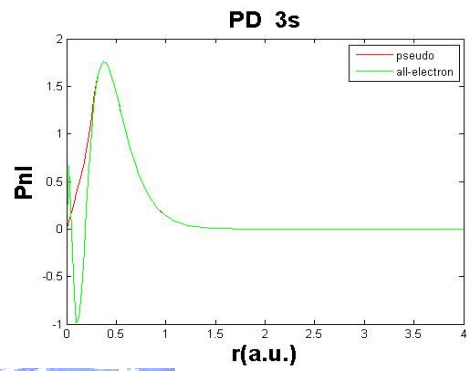


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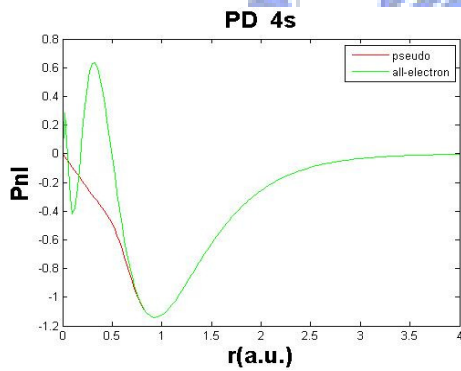


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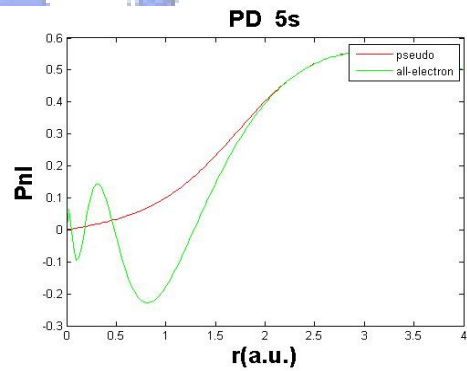


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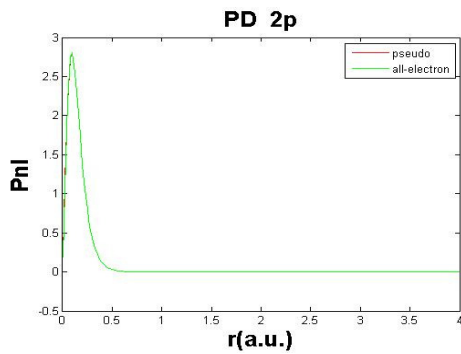


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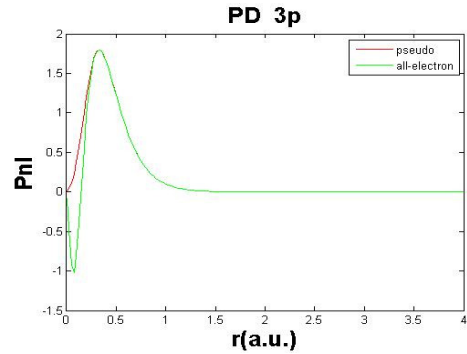


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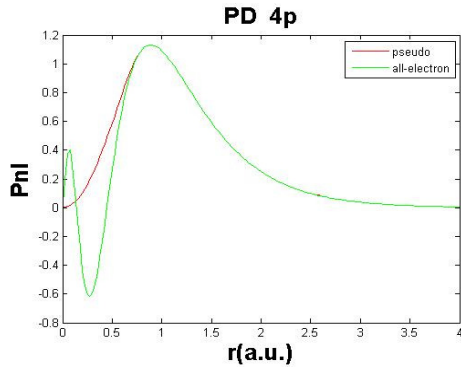


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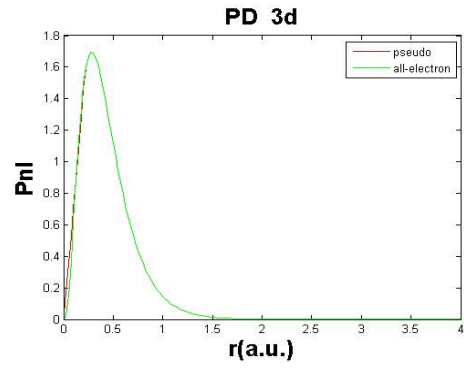


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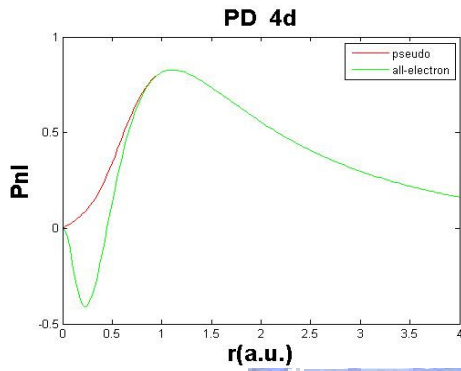


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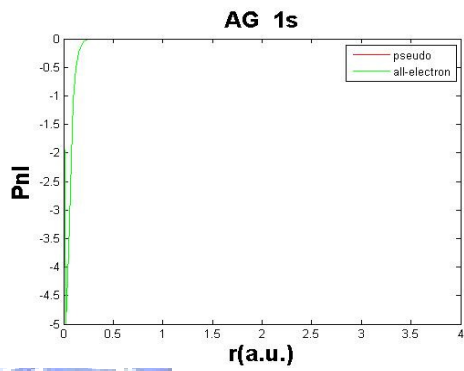


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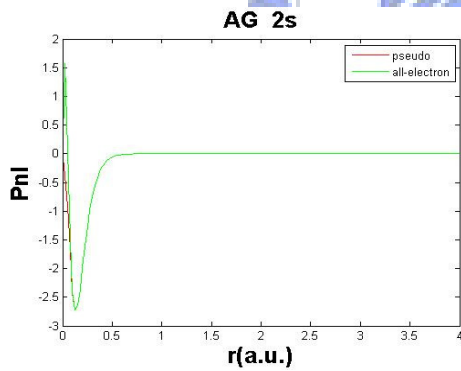


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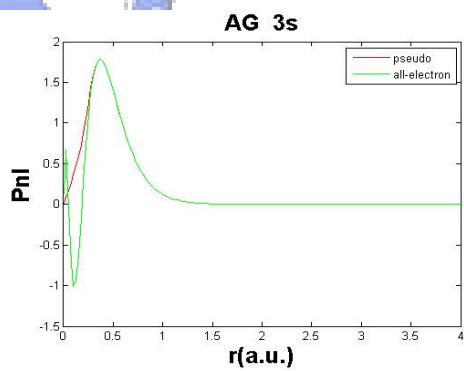


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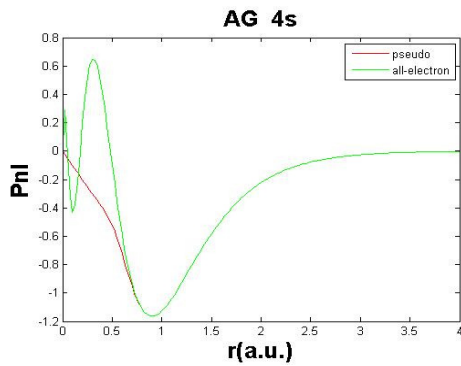


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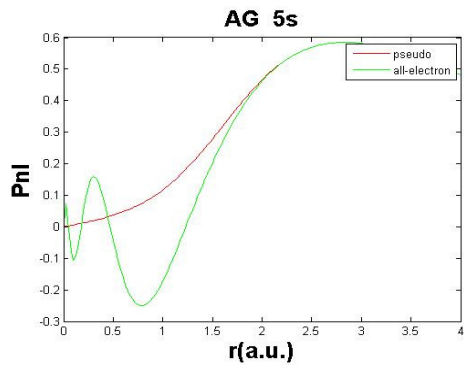


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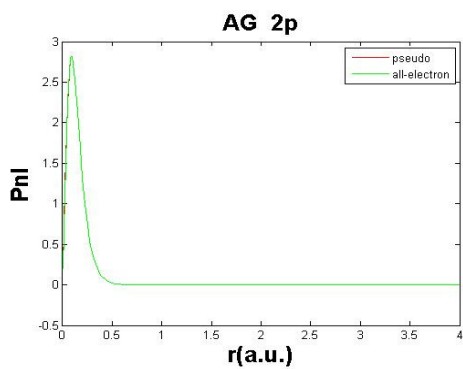


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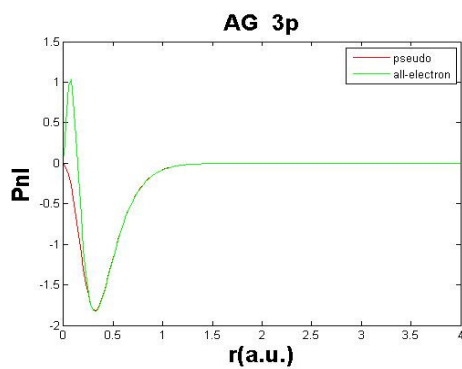


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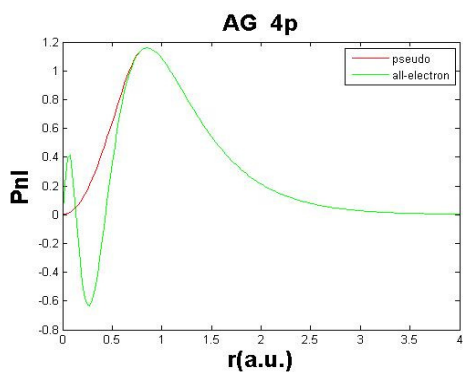


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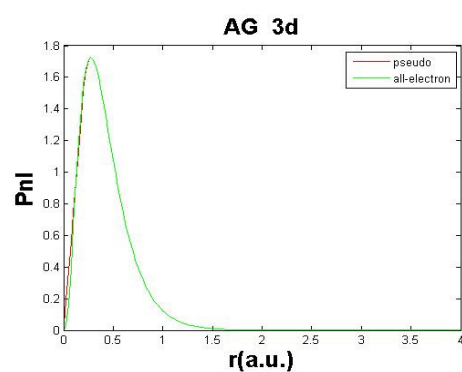


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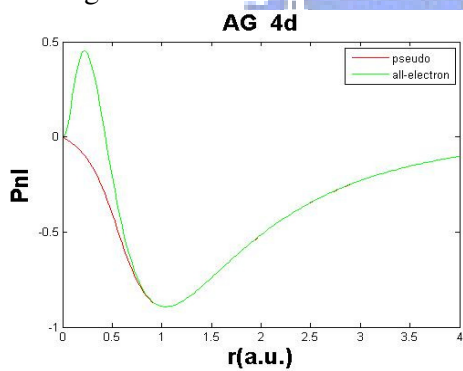


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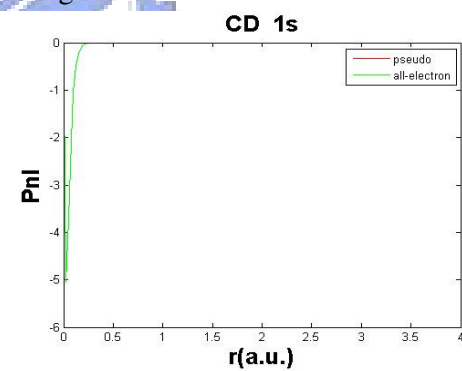


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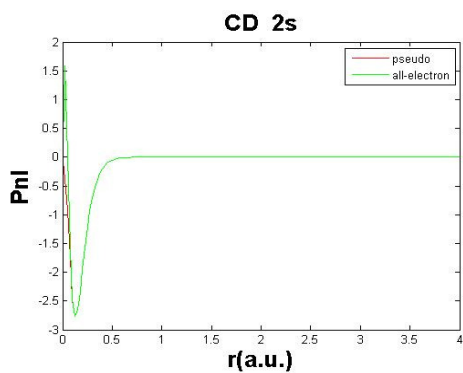


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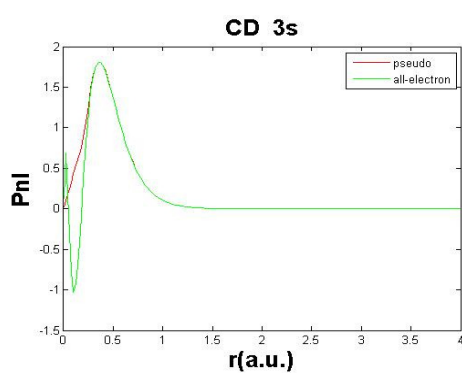


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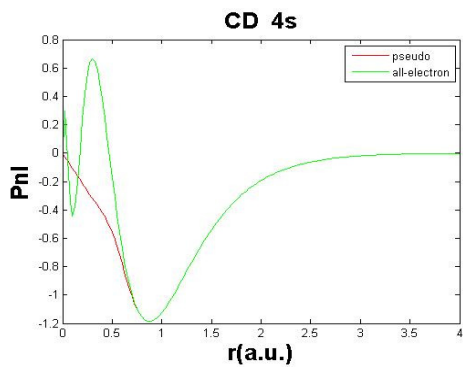


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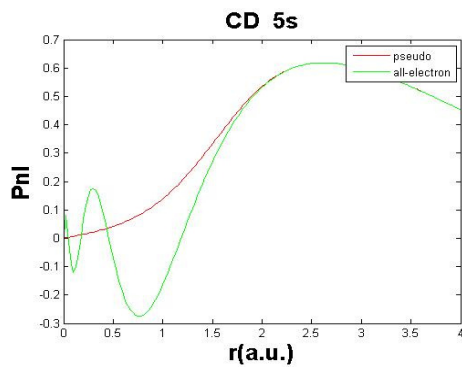


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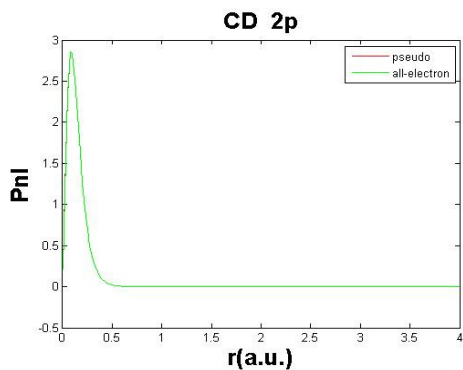


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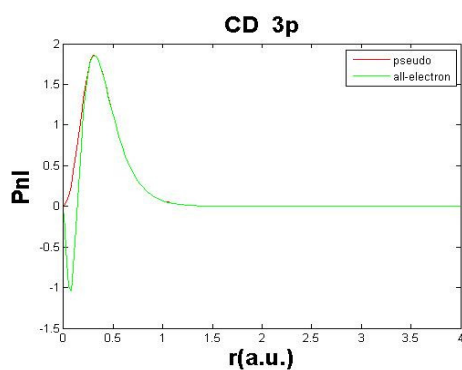


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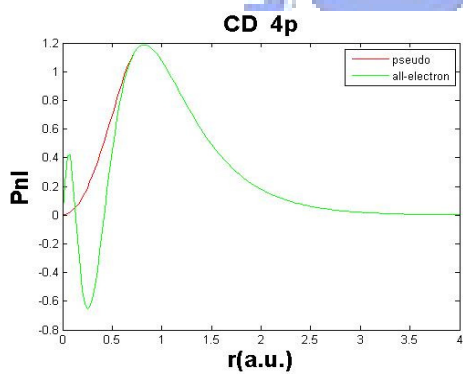


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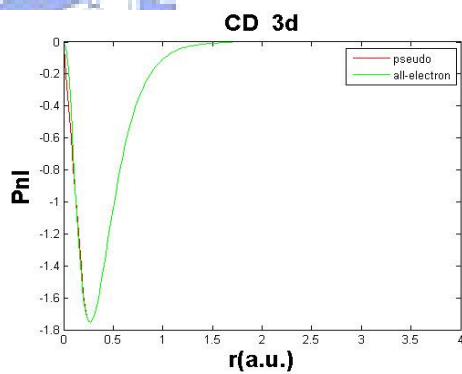


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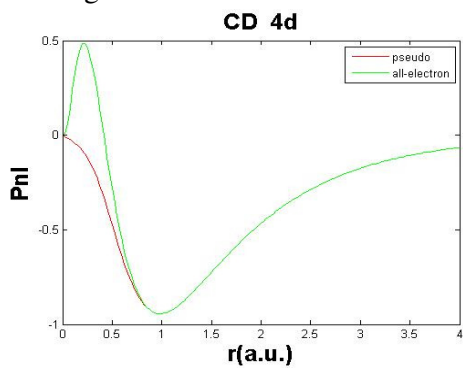


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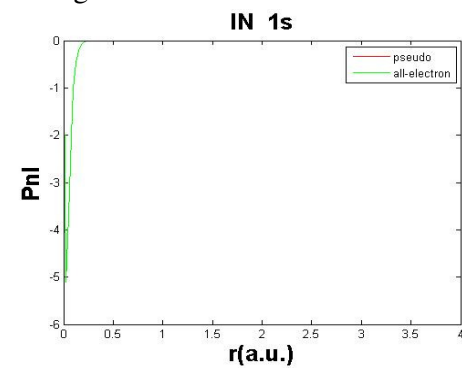


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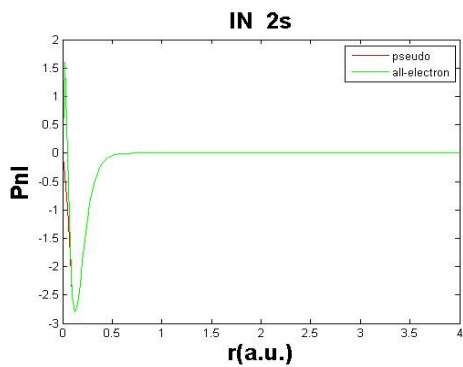


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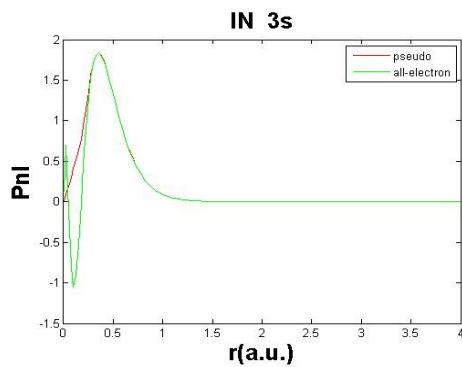


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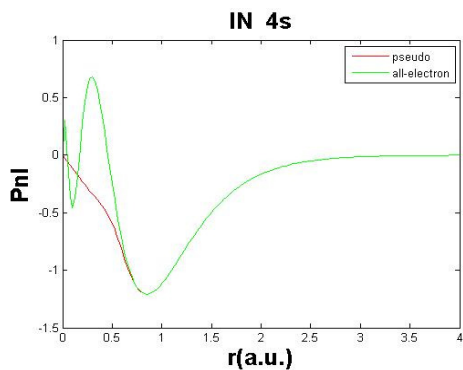


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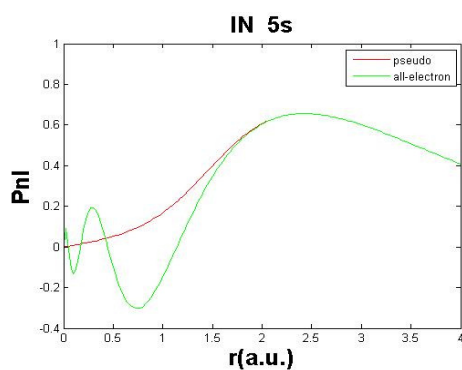


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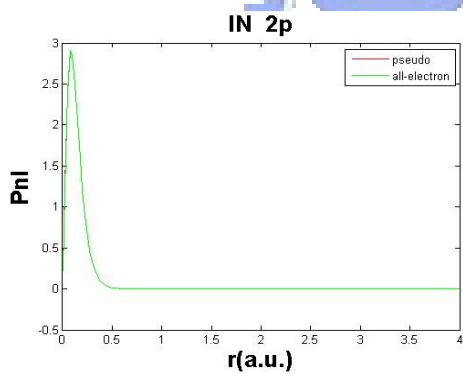


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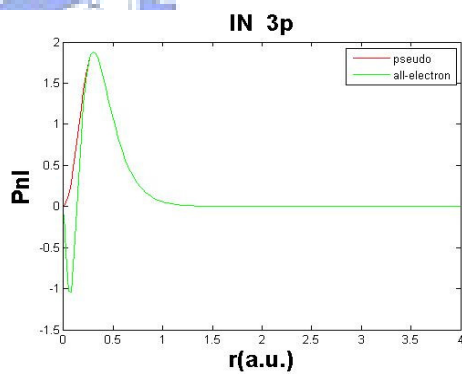


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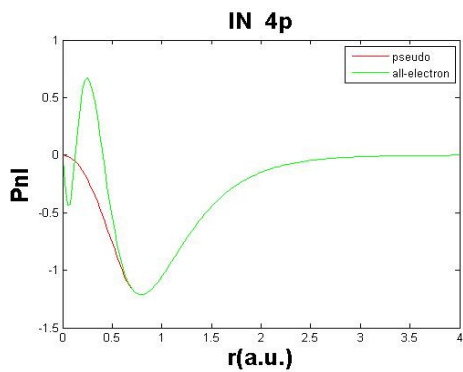


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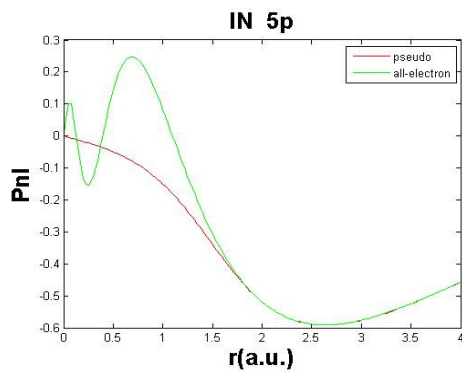


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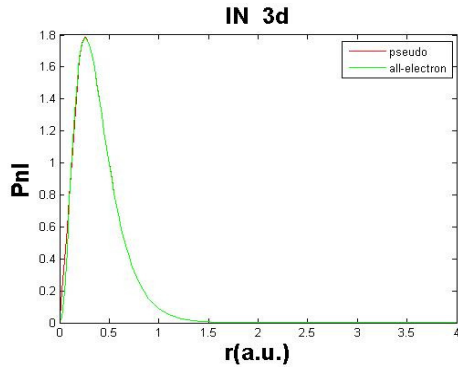


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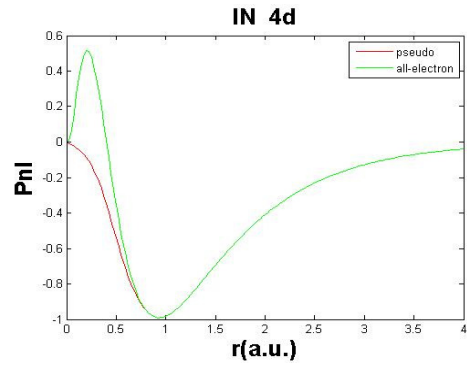


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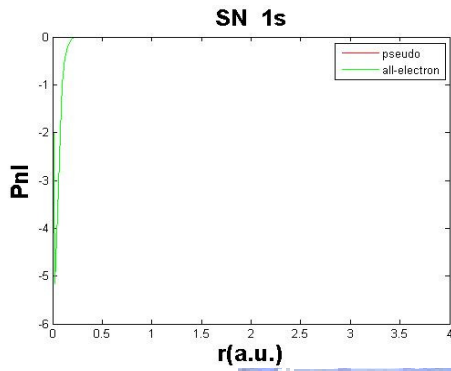


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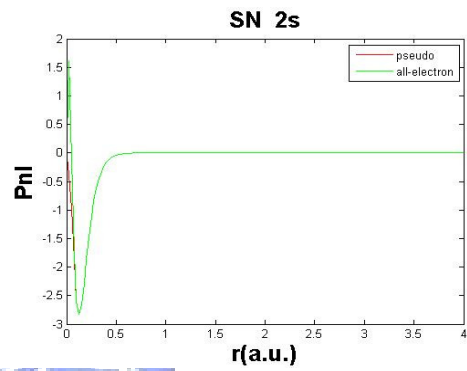


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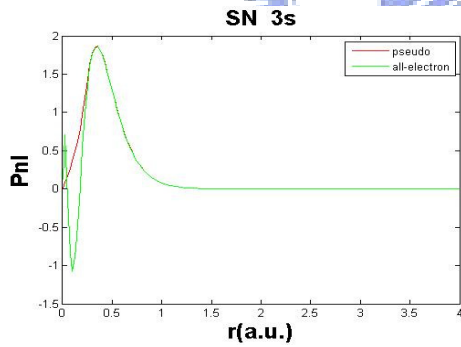


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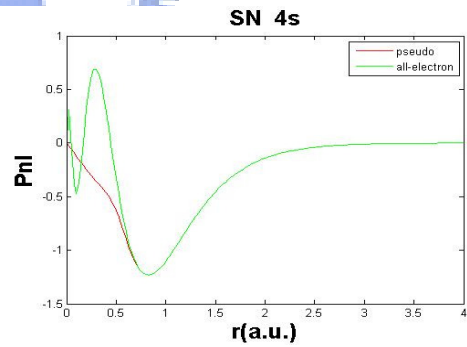


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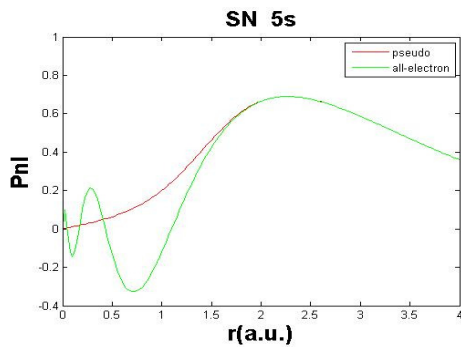


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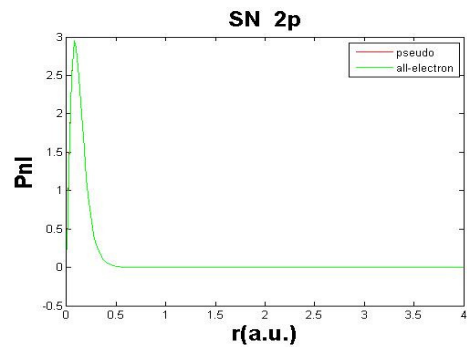


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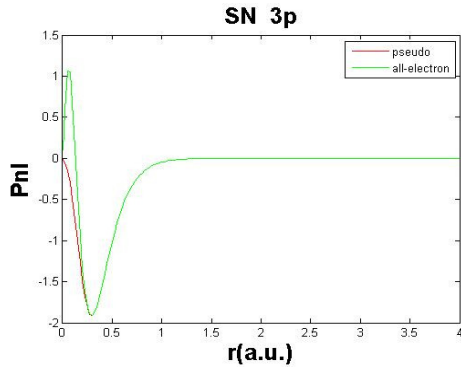


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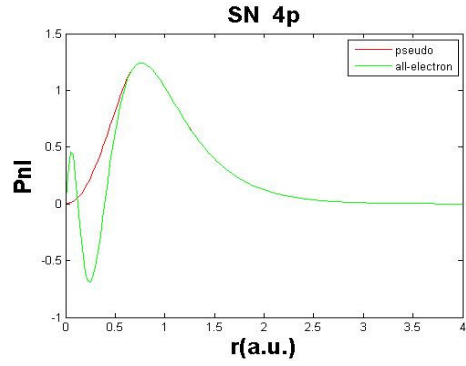


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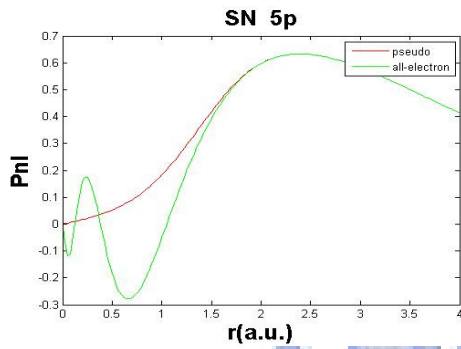


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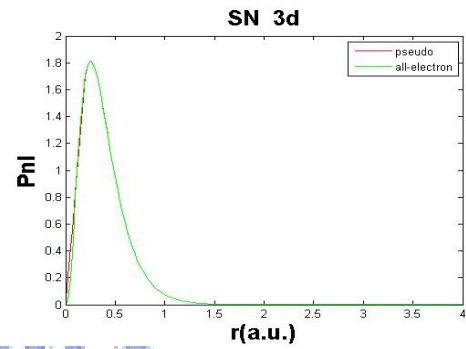


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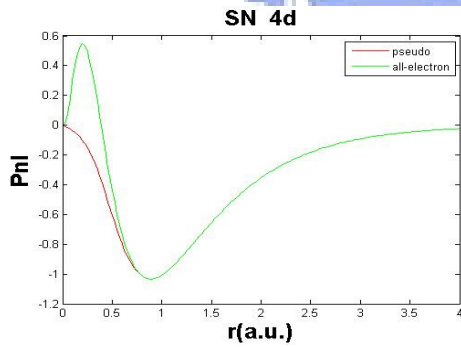


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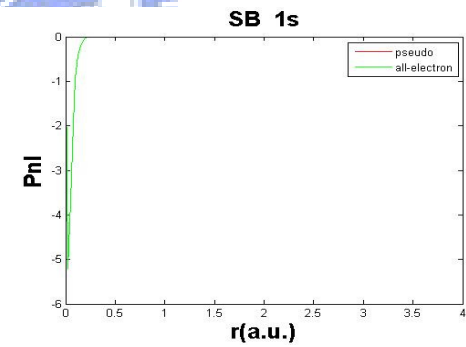


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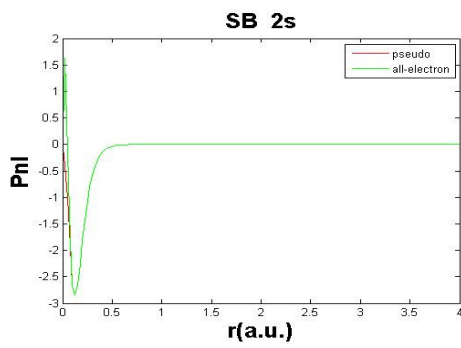


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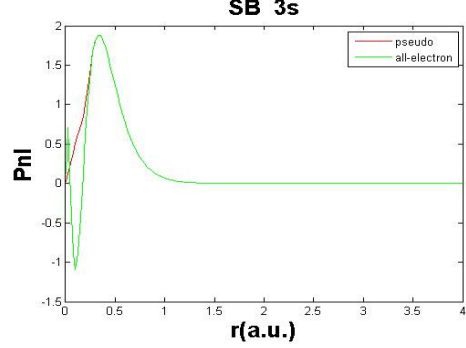


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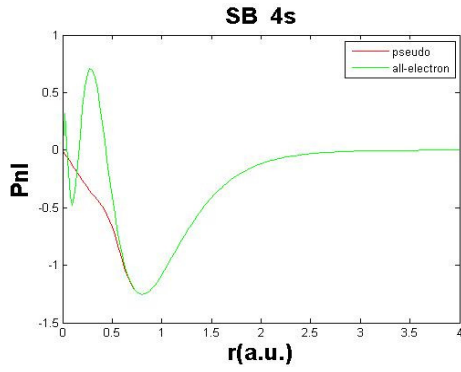


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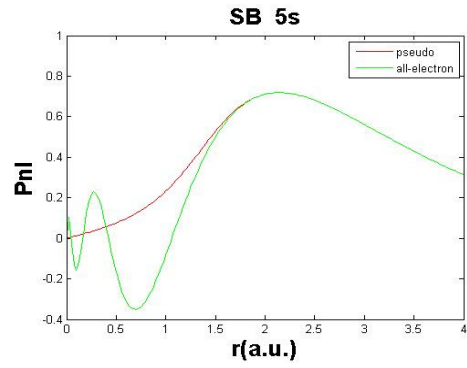


Figure 337.

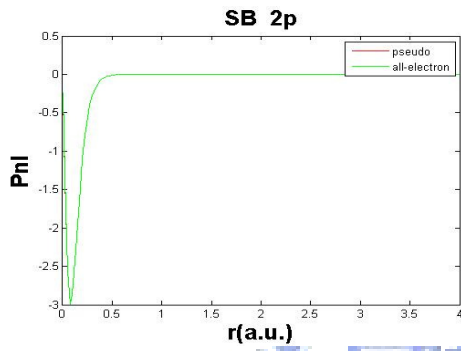


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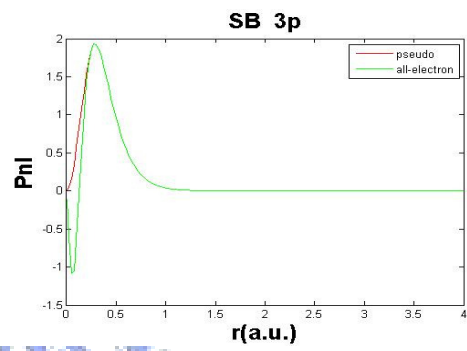


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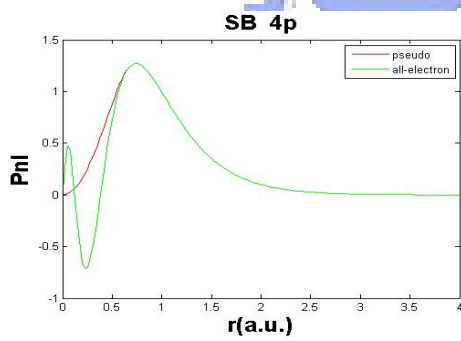


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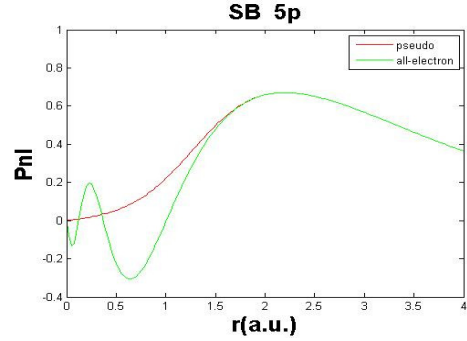


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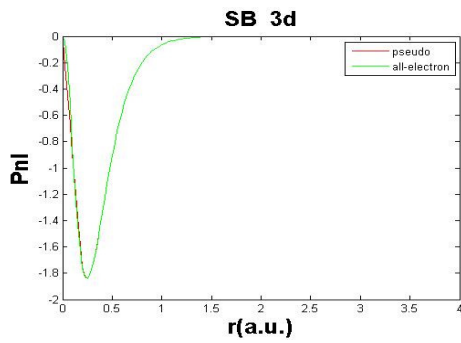


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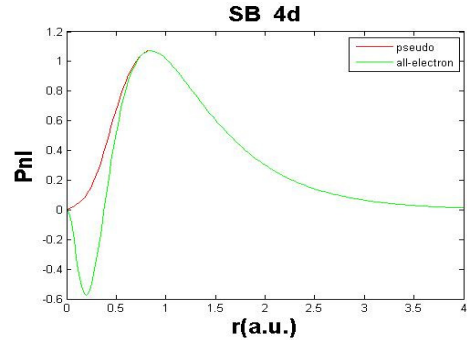


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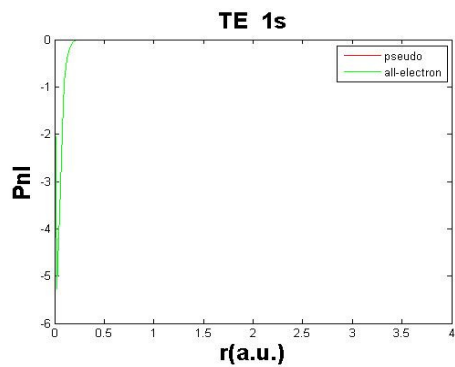


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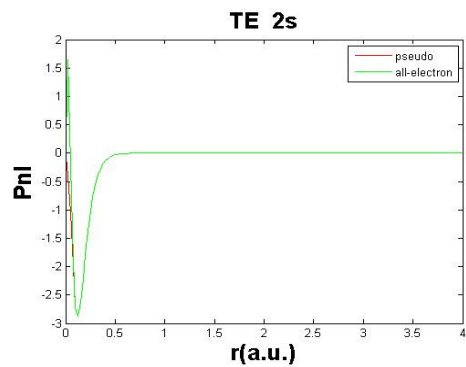


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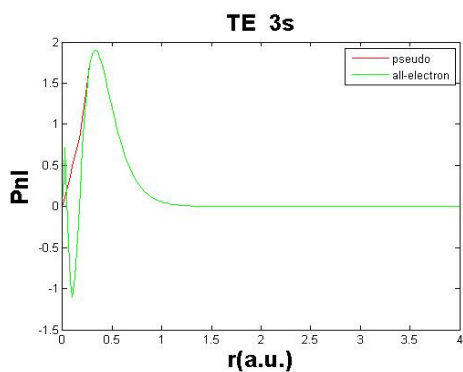


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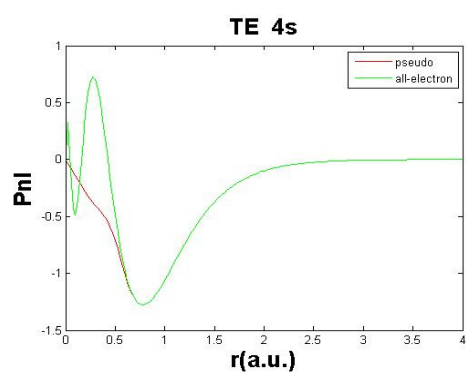


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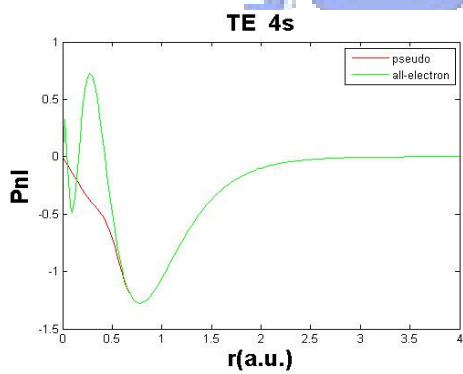


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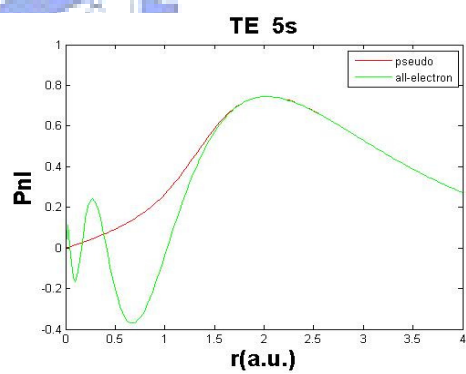


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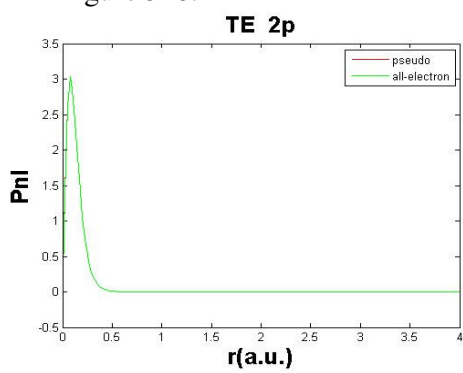


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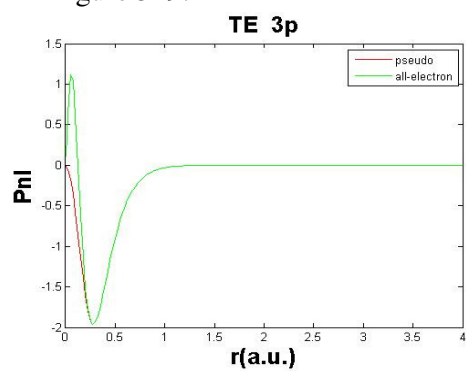


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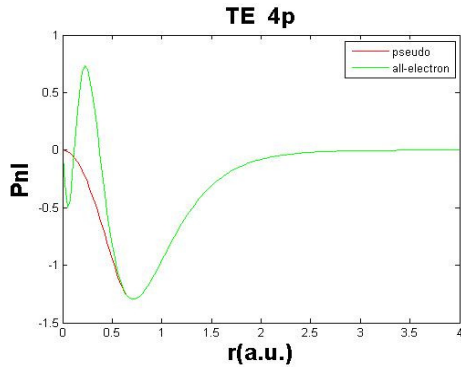


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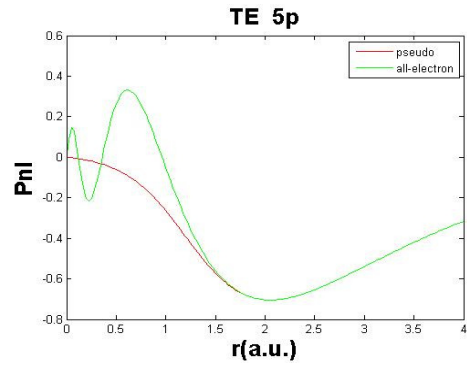


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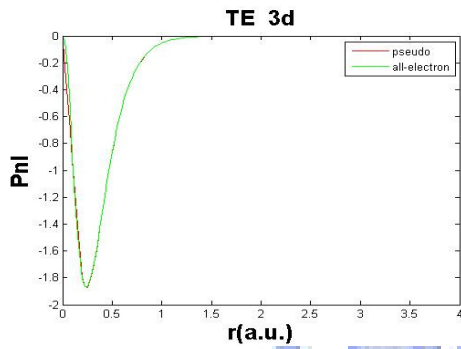


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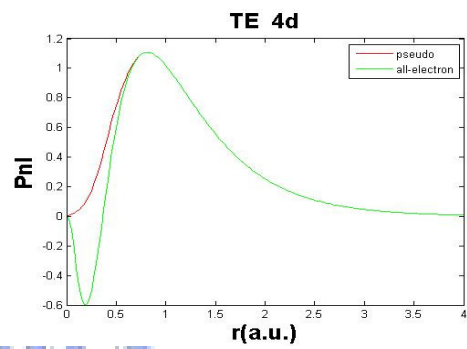


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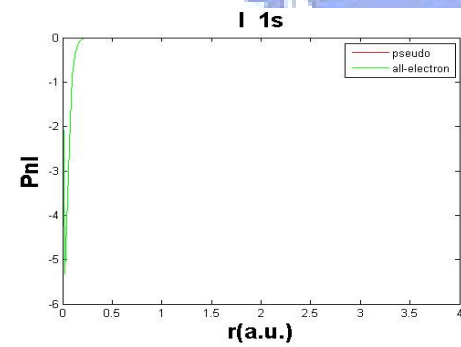


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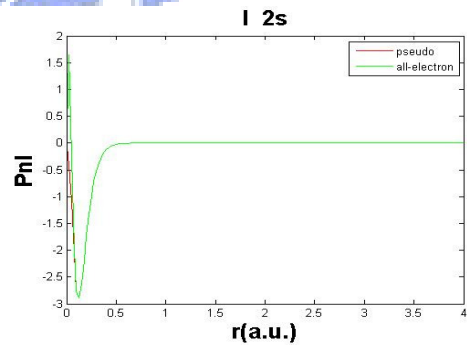


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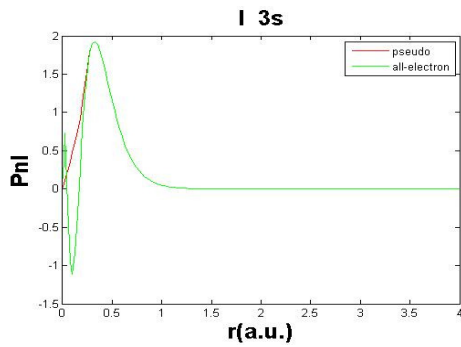


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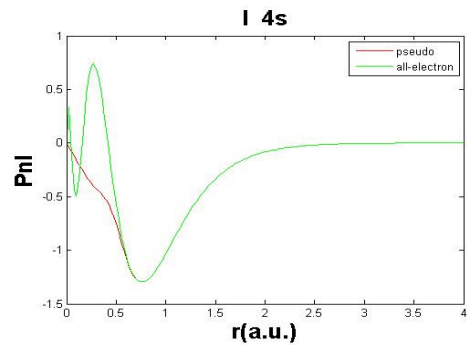


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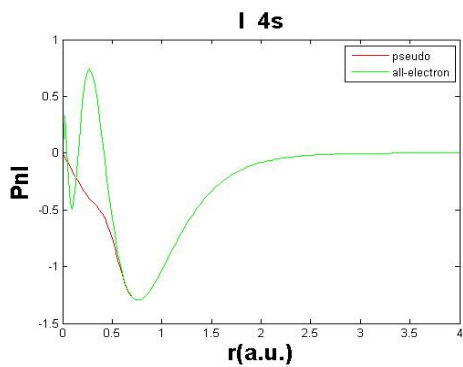


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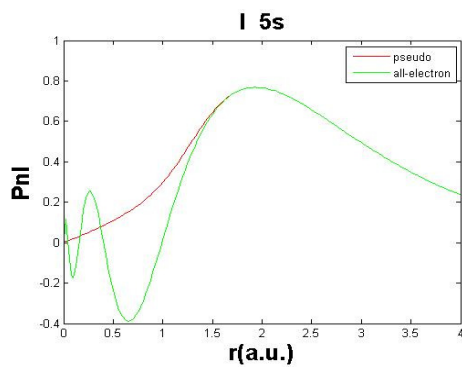


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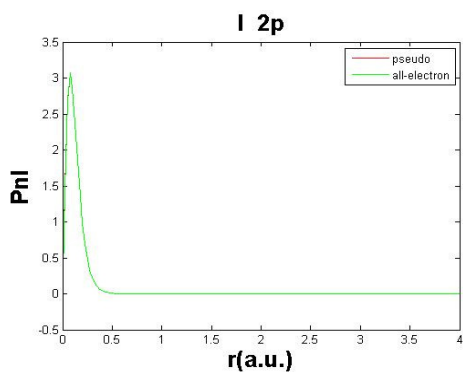


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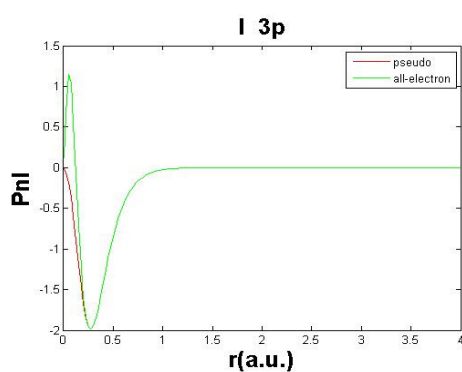


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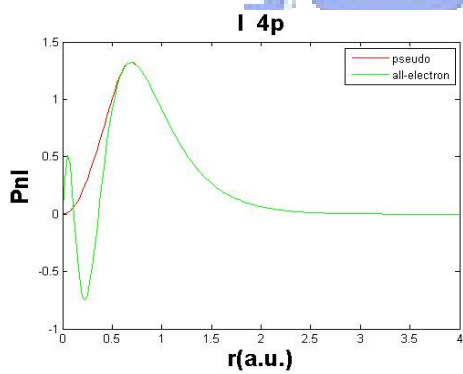


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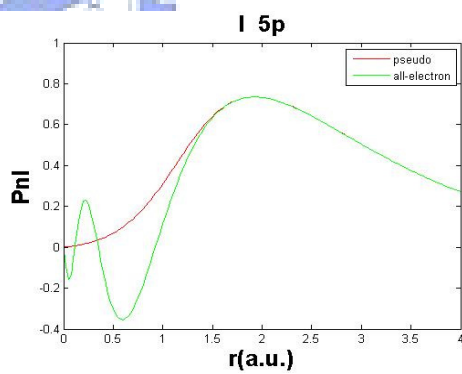


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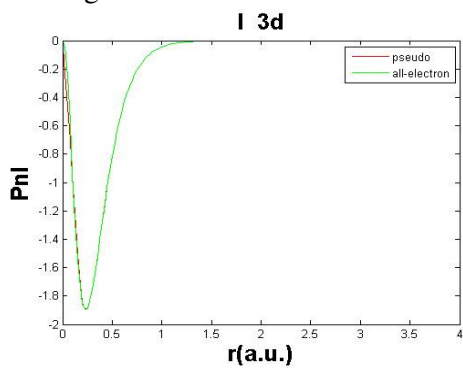


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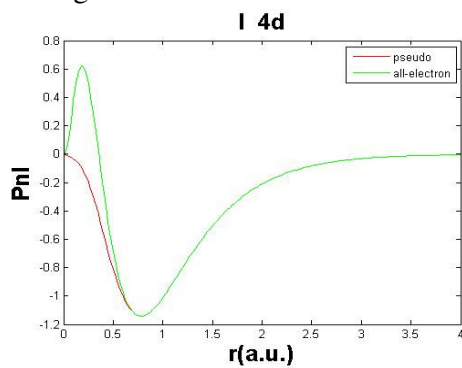


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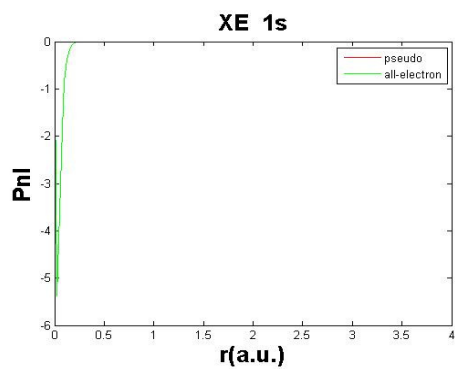


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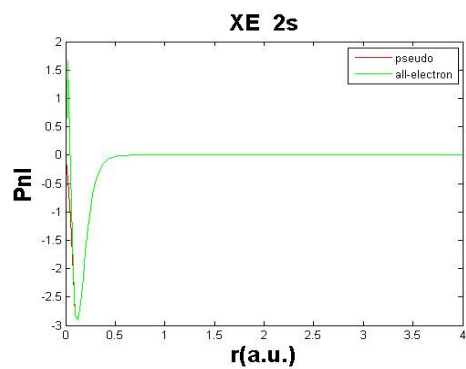


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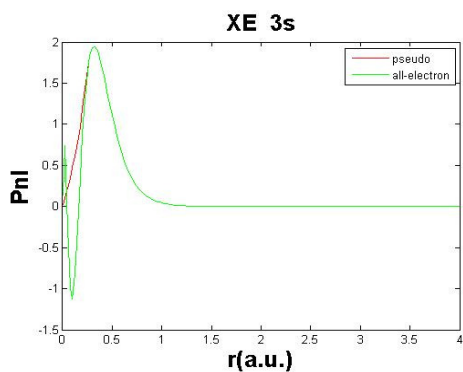


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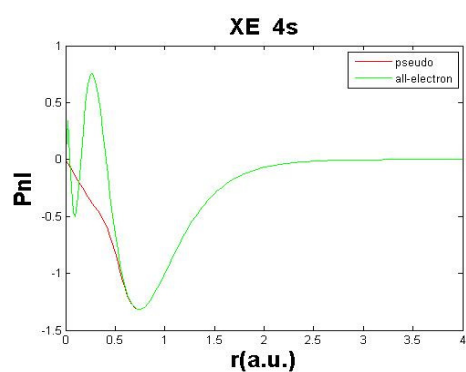


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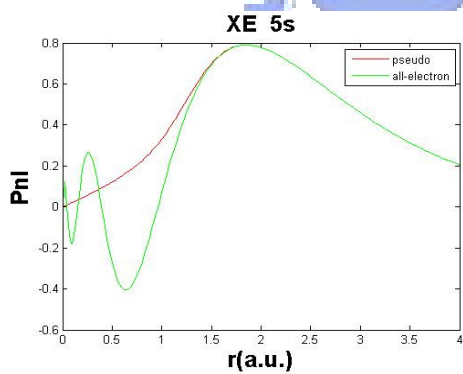


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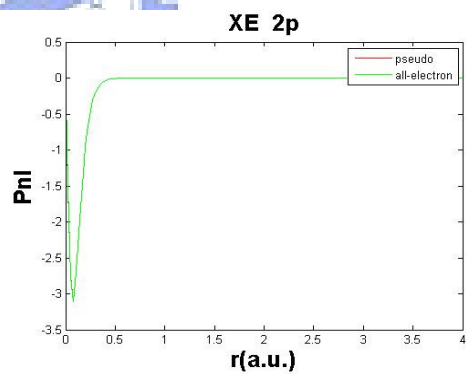


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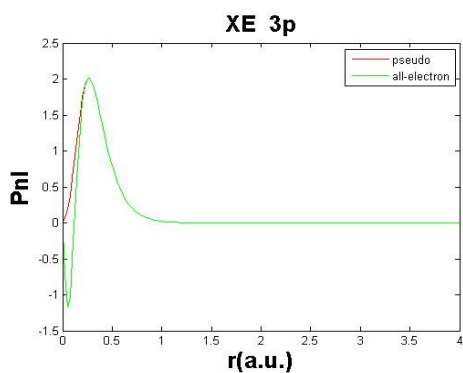


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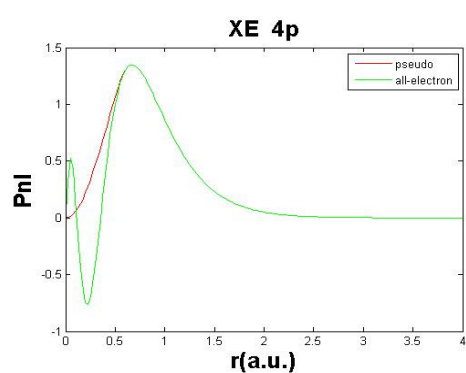


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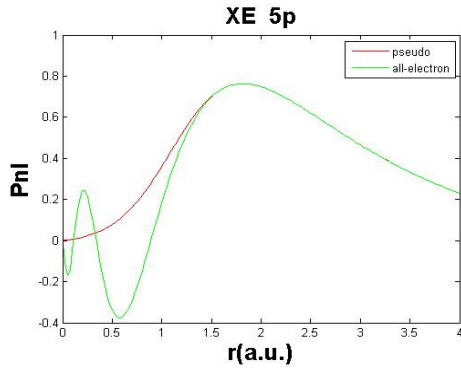


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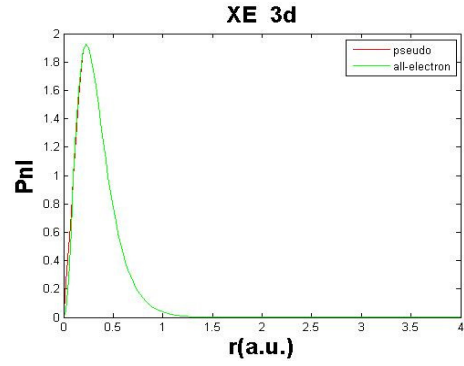


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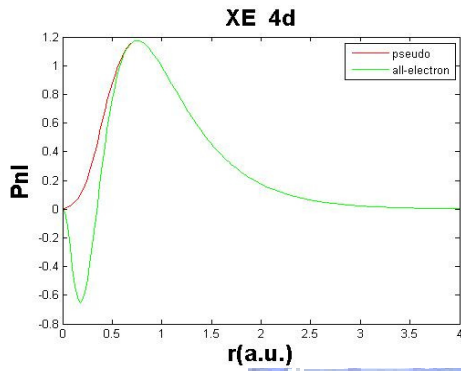


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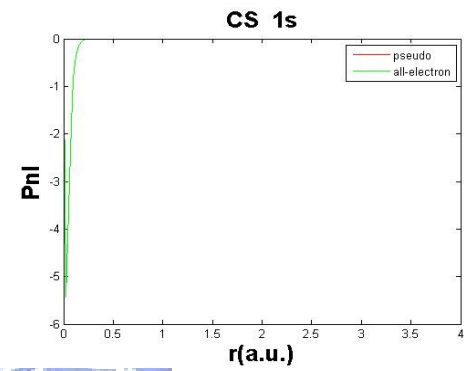


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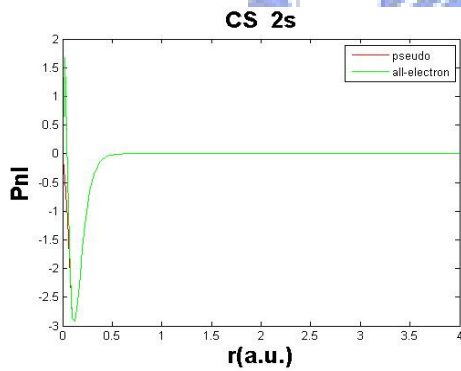


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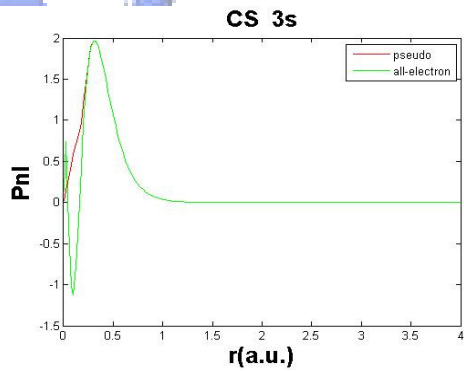


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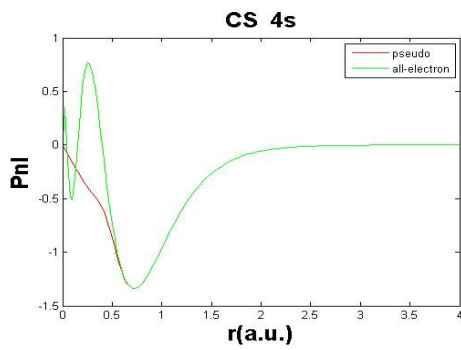


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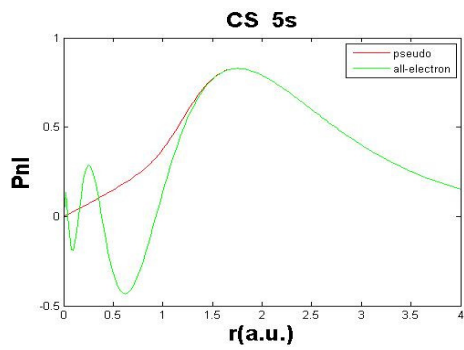


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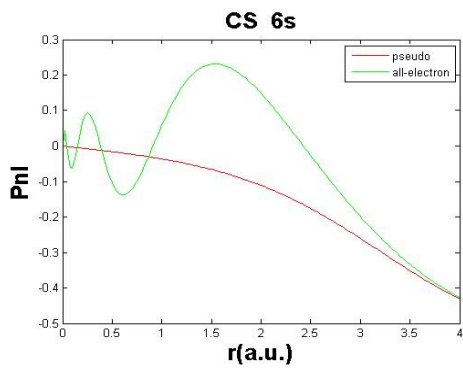


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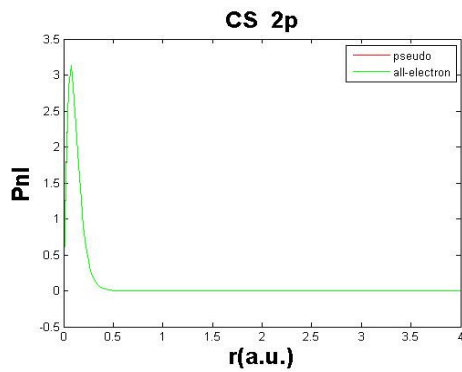


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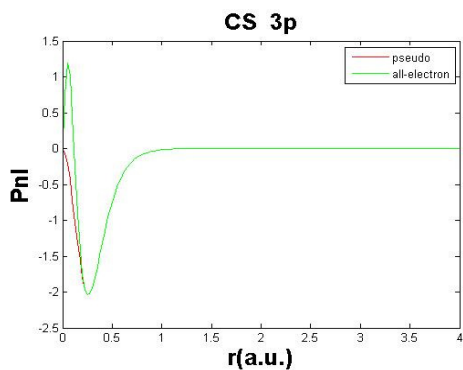


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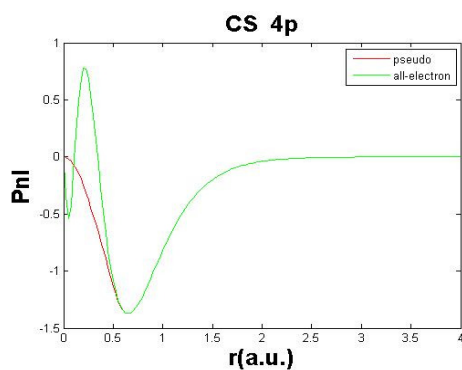


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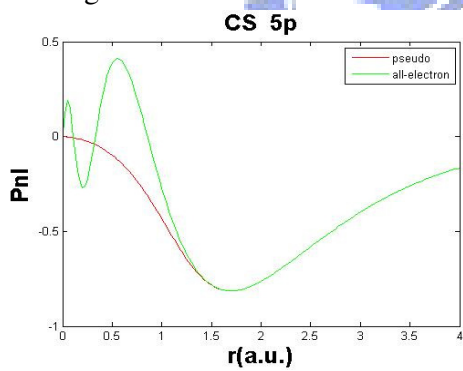


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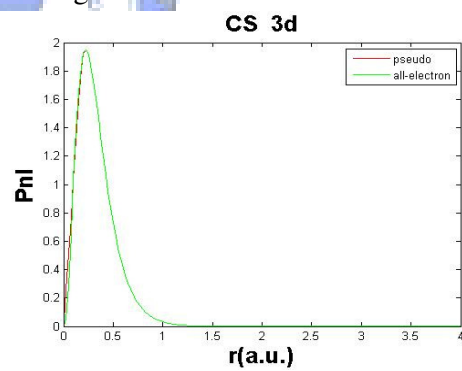


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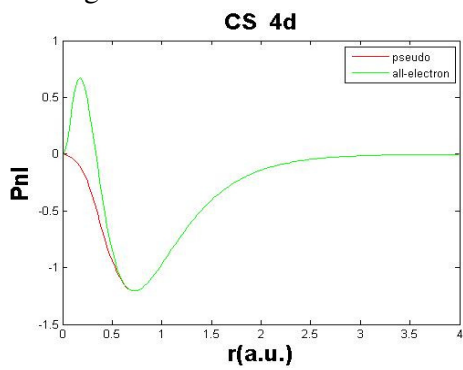


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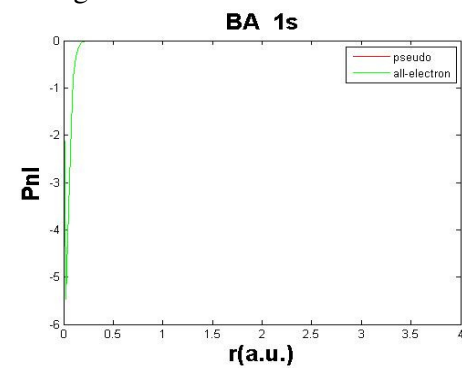


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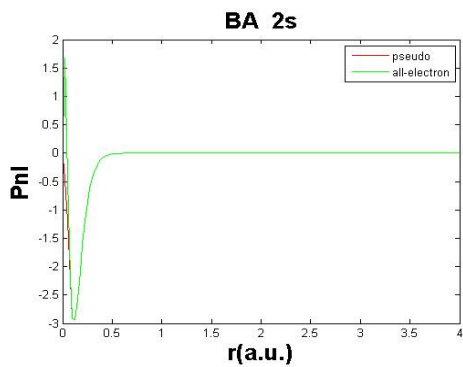


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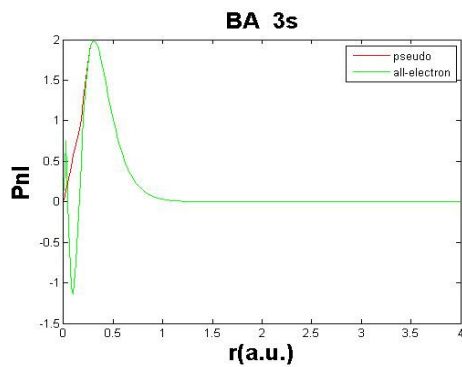


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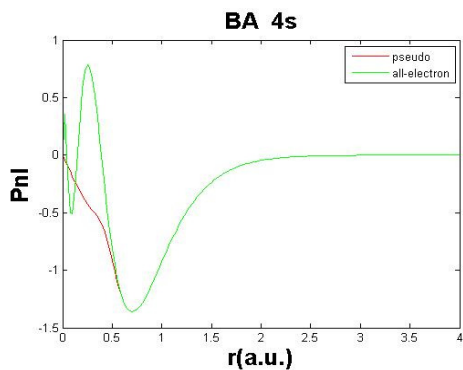


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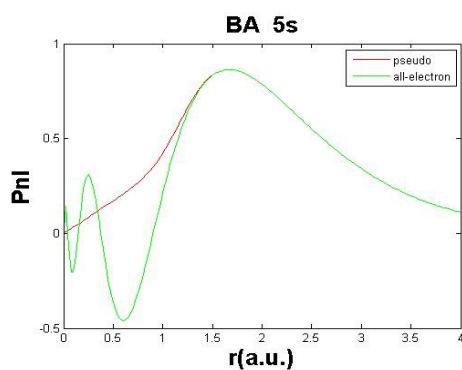


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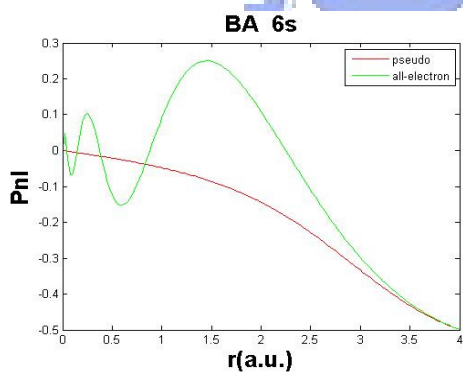


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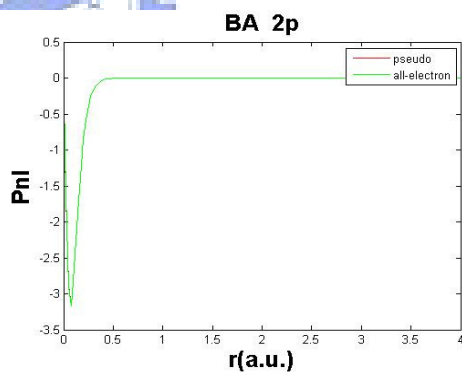


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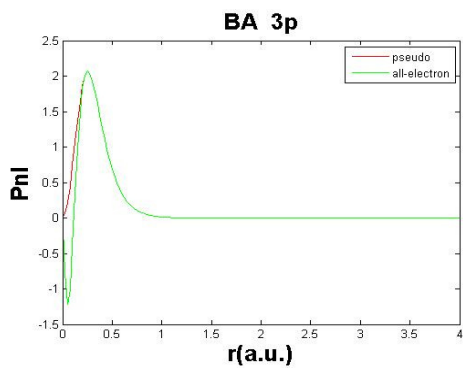


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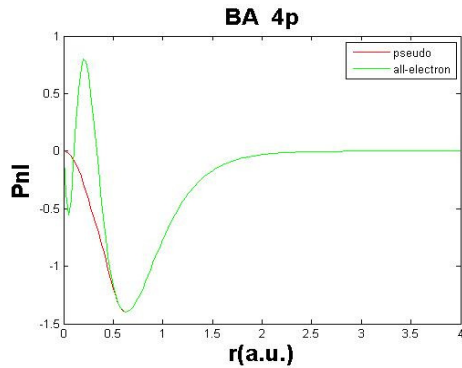


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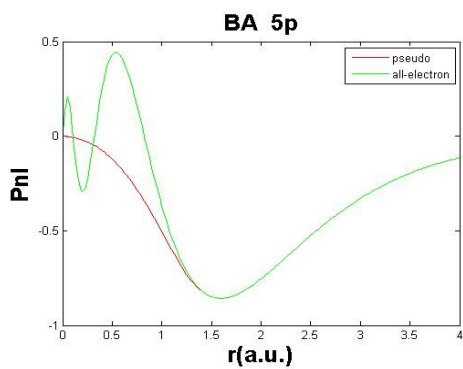


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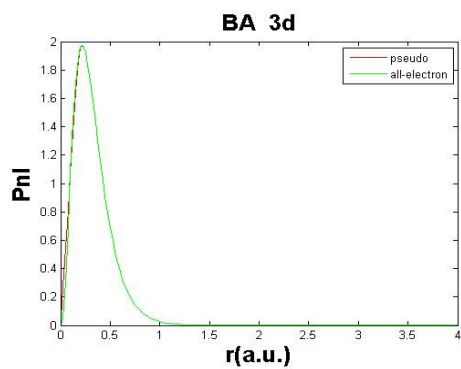


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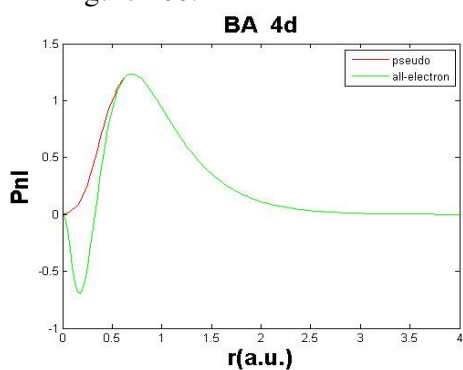


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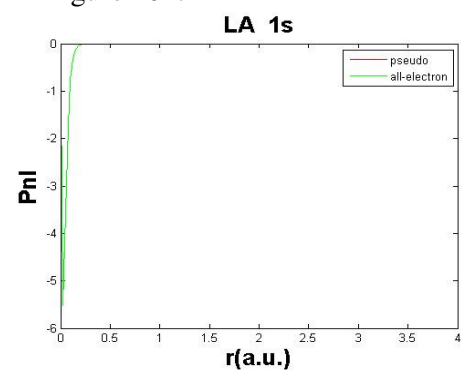


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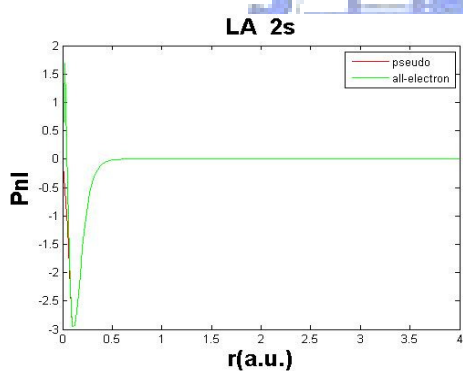


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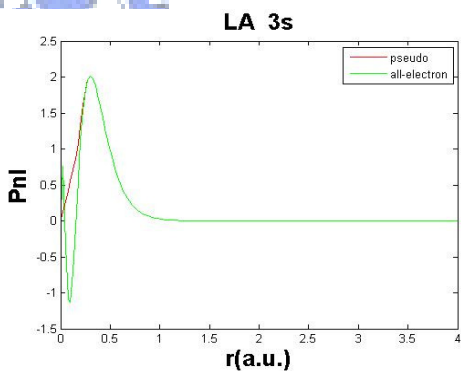


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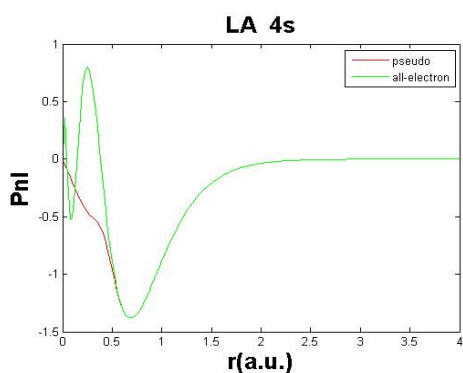


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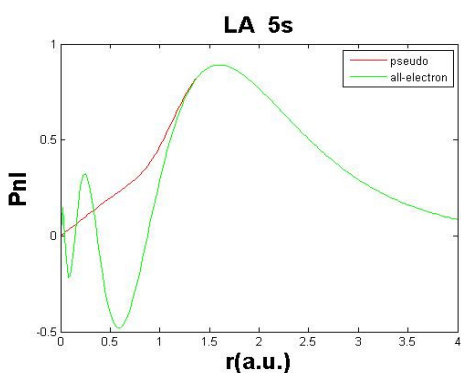


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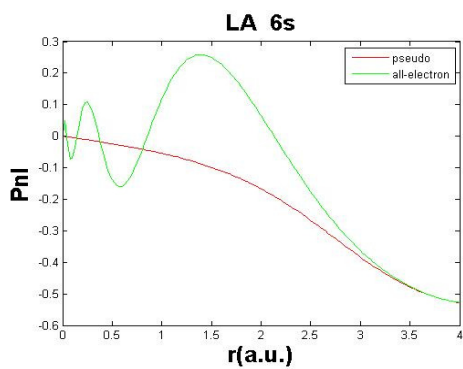


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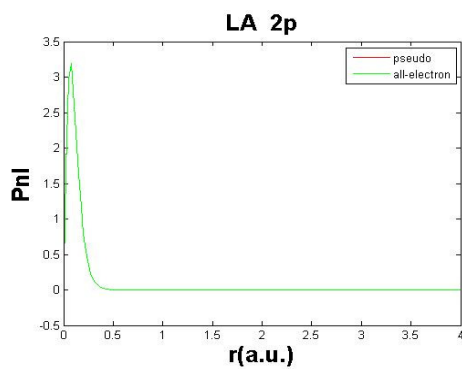


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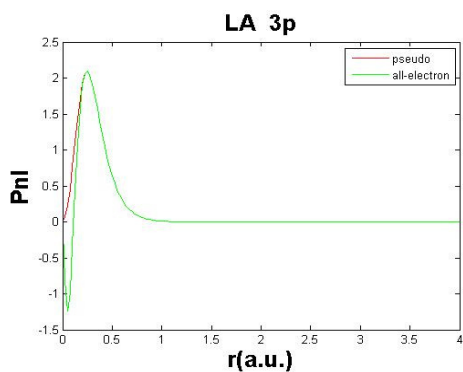


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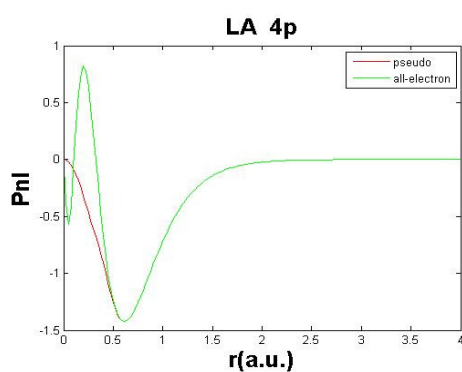


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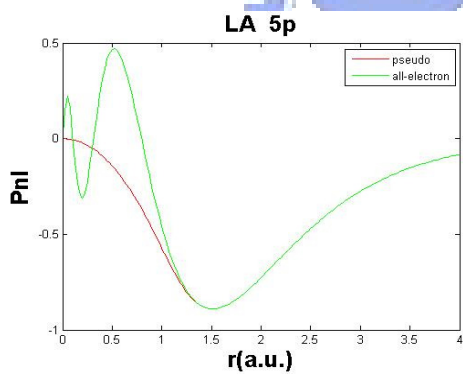


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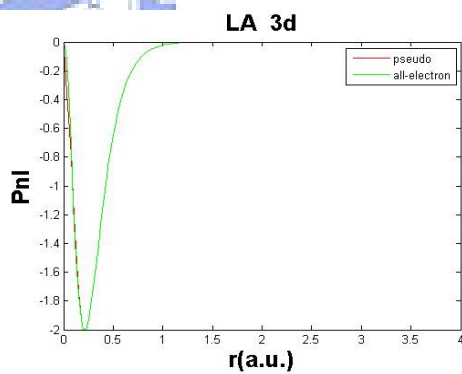


Figure 413.

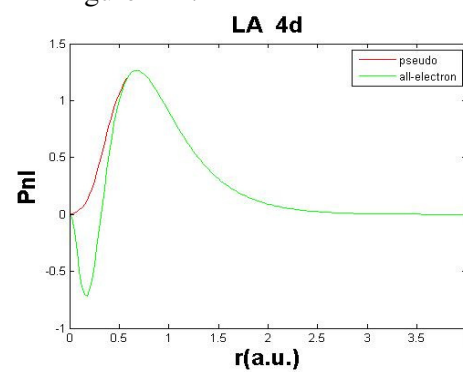


Figure 414.

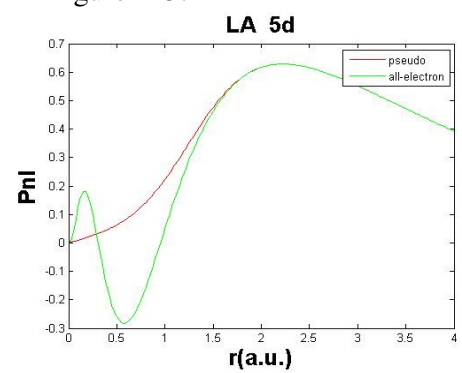


Figure 415.

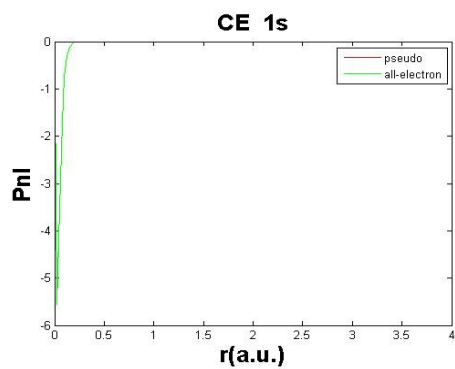


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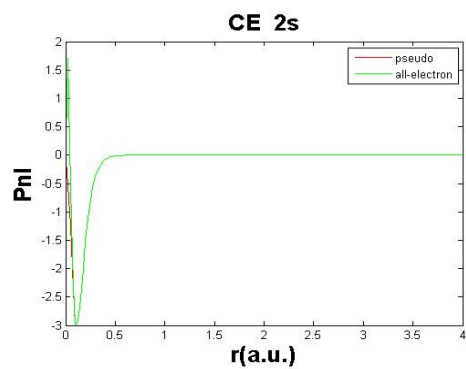


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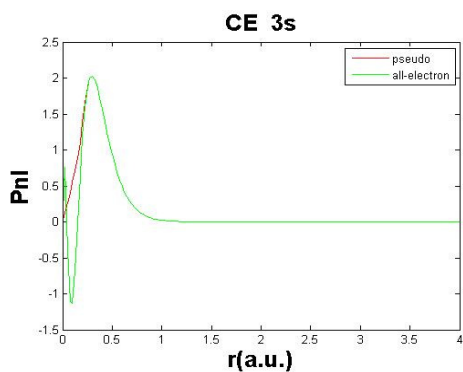


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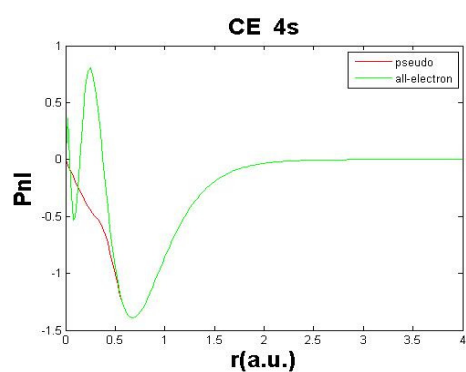


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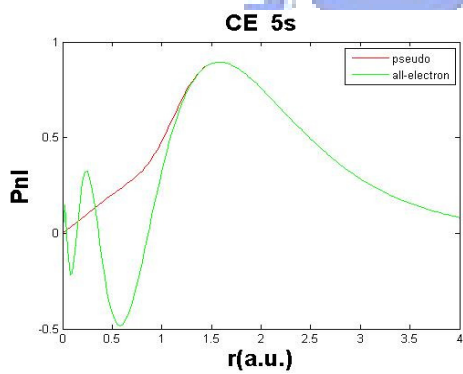


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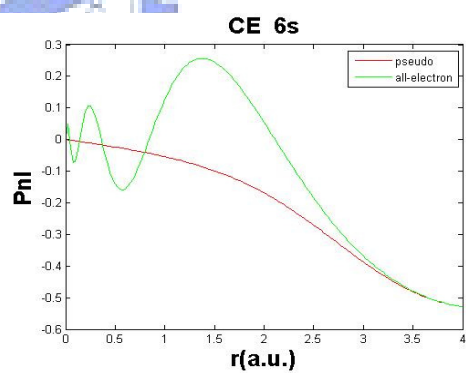


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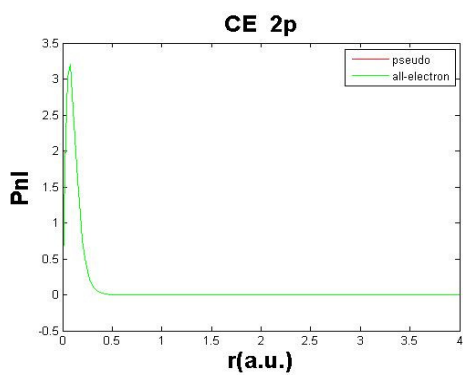


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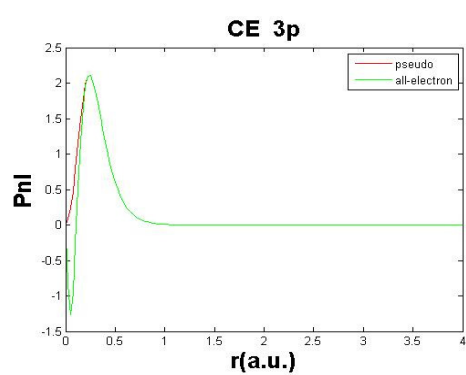


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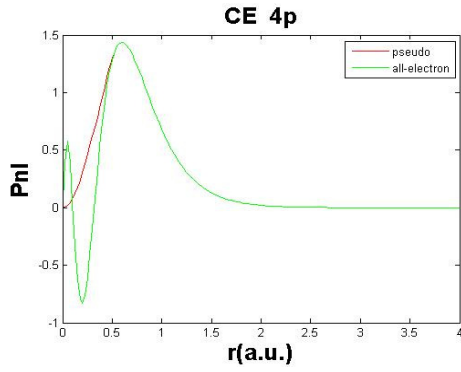


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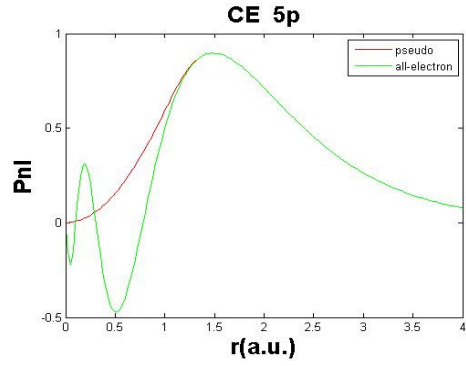


Figure 425.

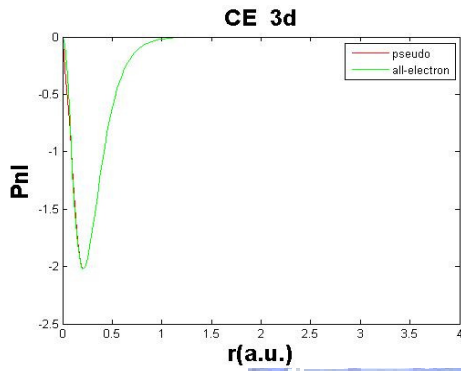


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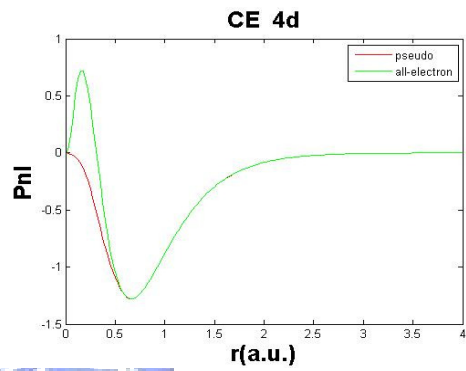


Figure 427.

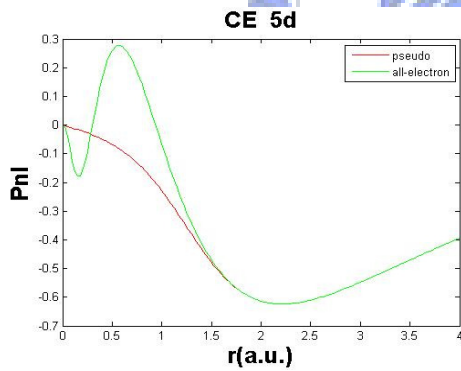


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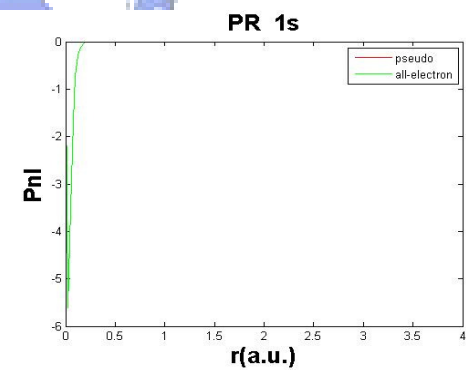


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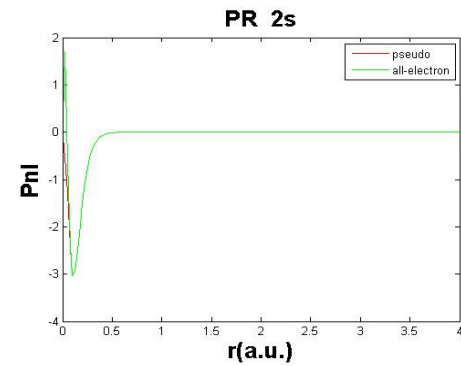


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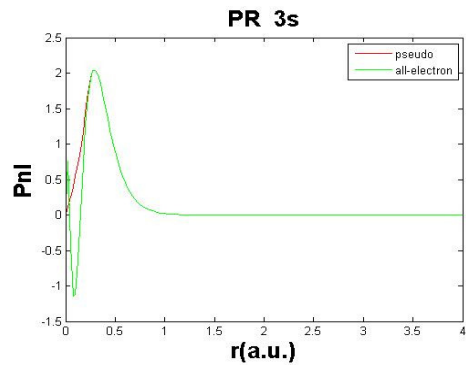


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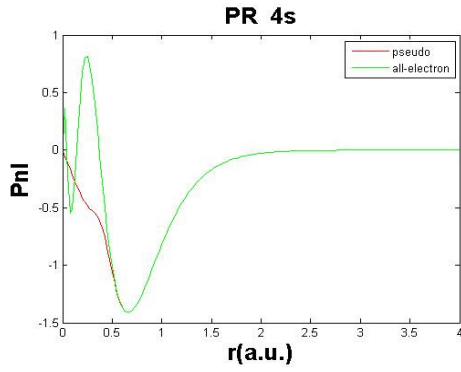


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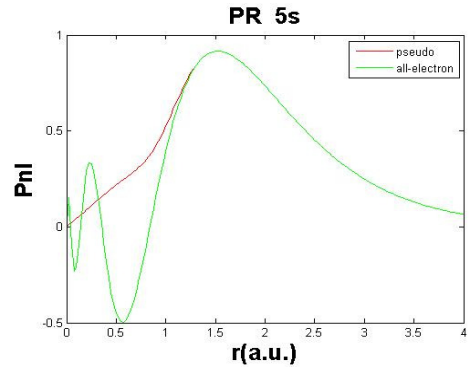


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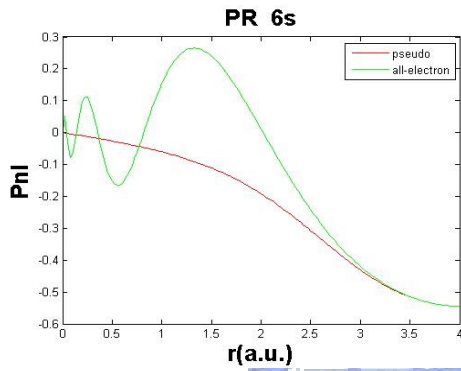


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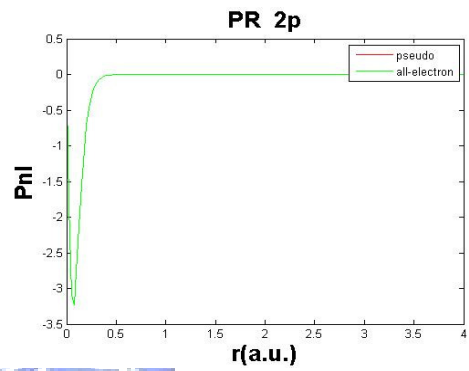


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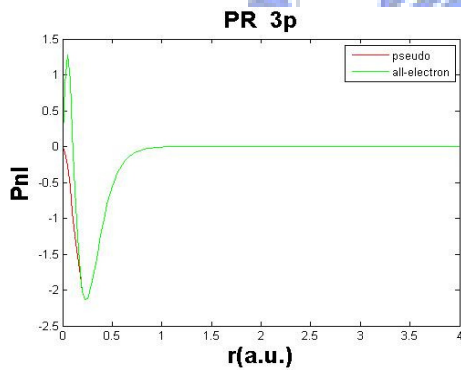


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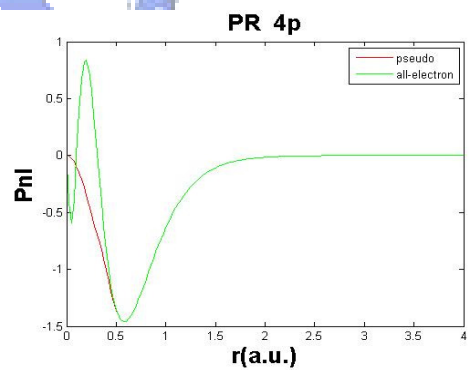


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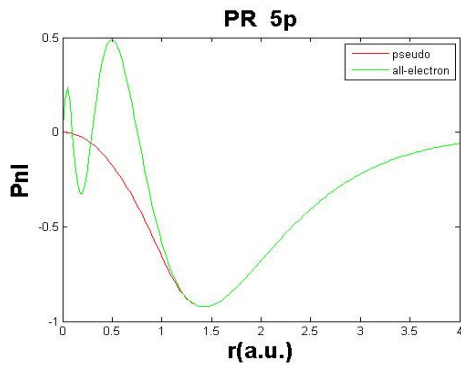


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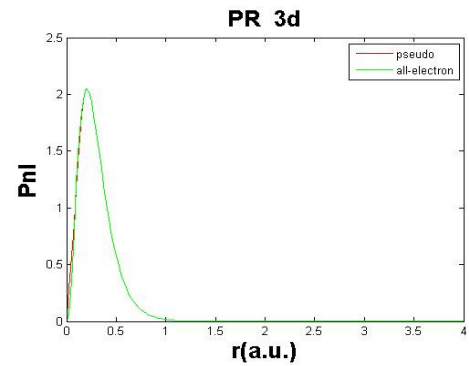


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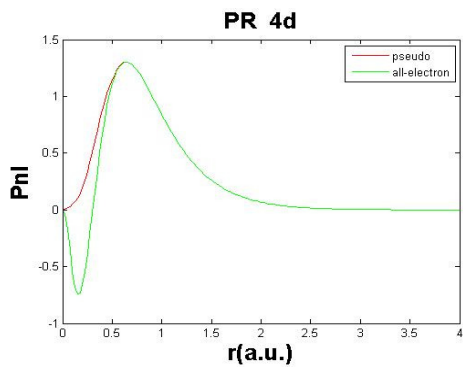


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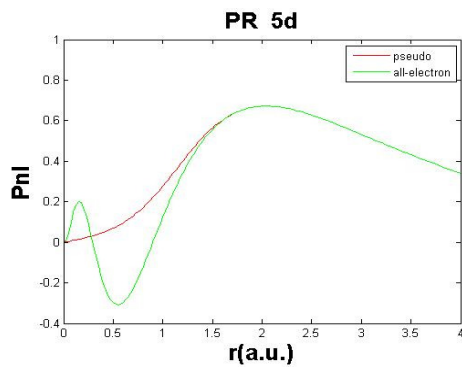


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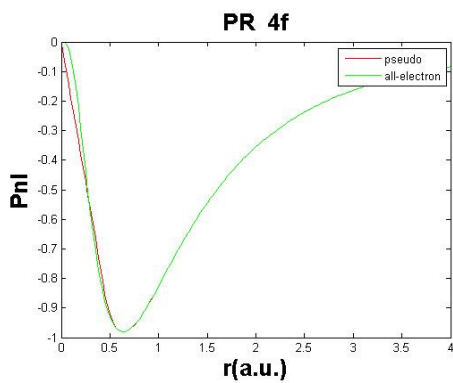


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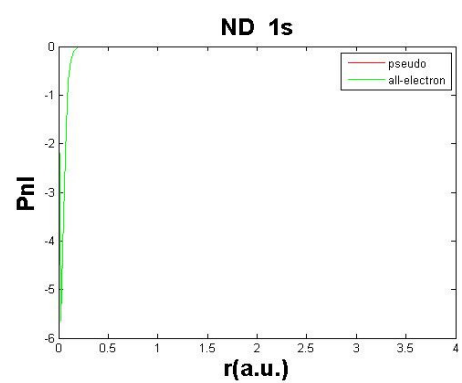


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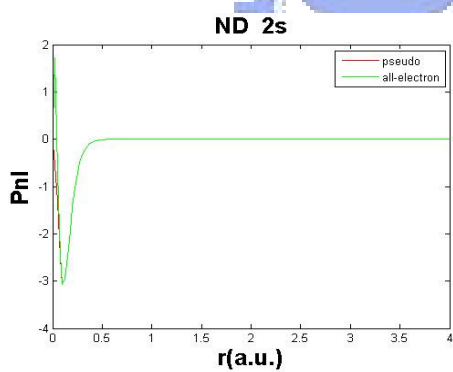


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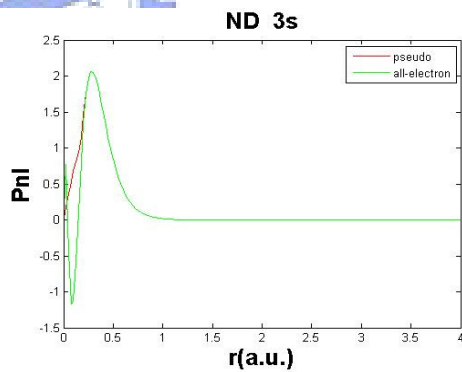


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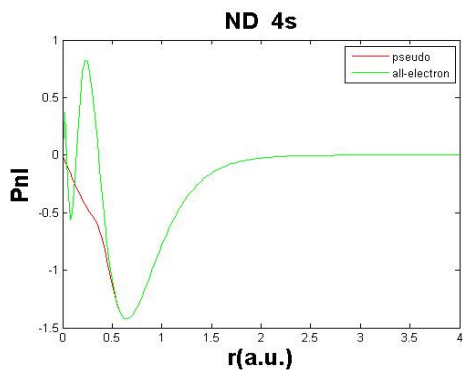


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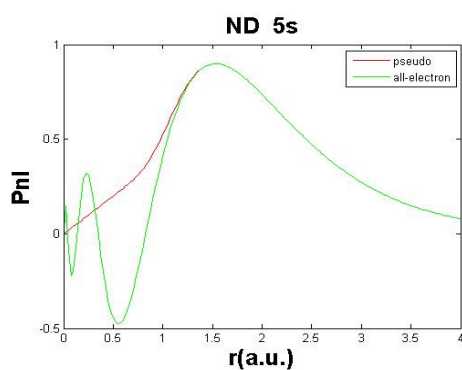


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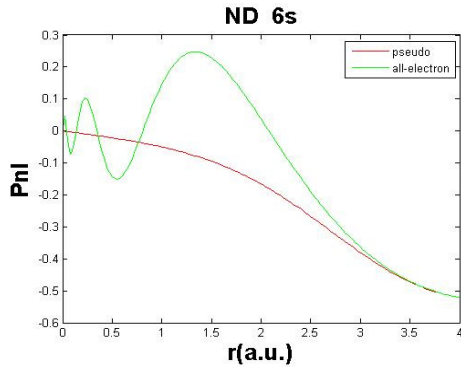


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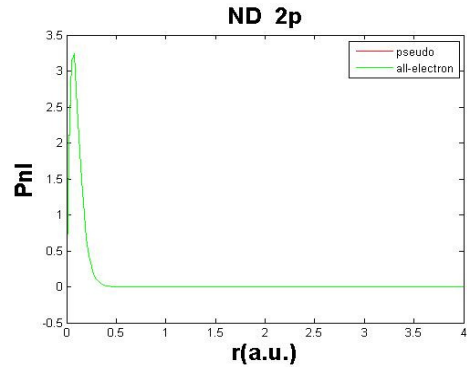


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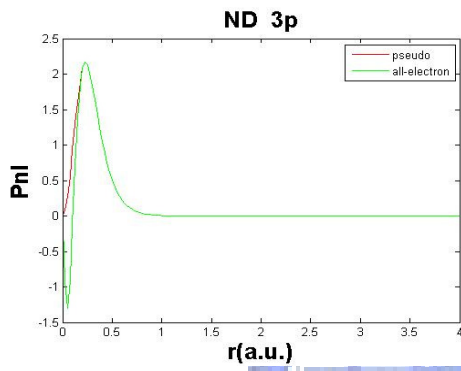


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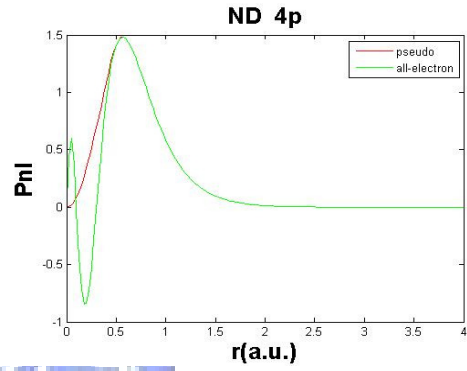


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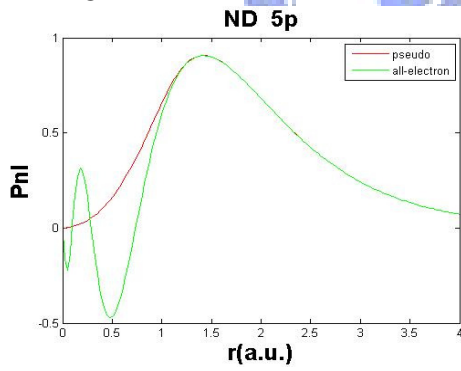


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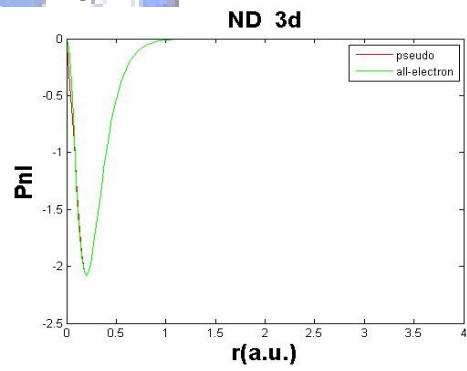


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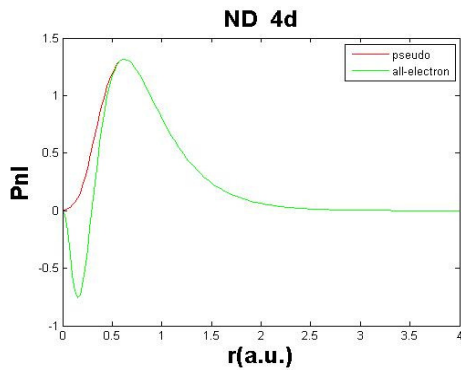


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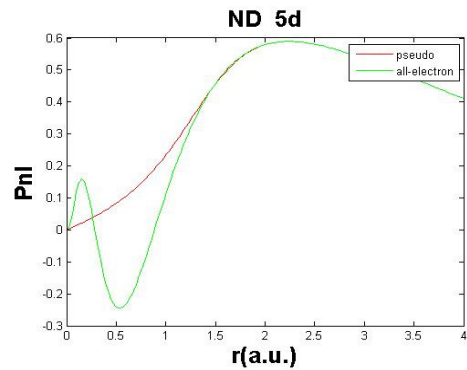


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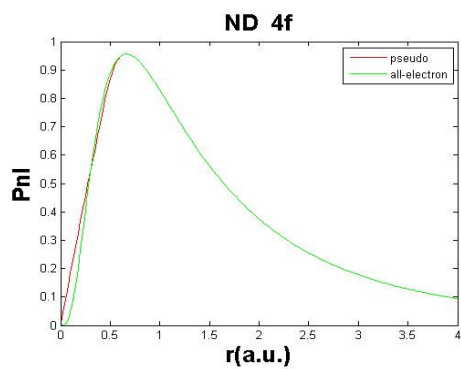


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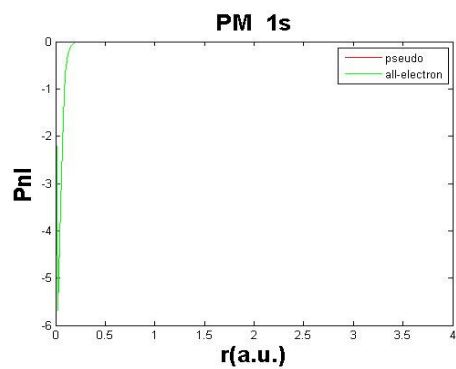


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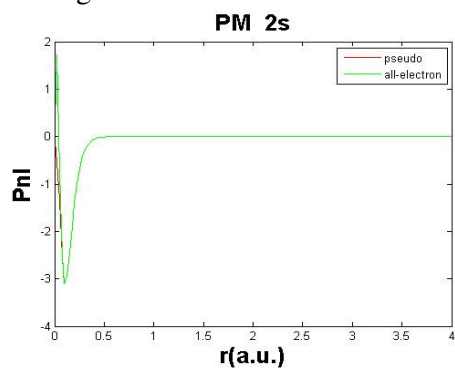


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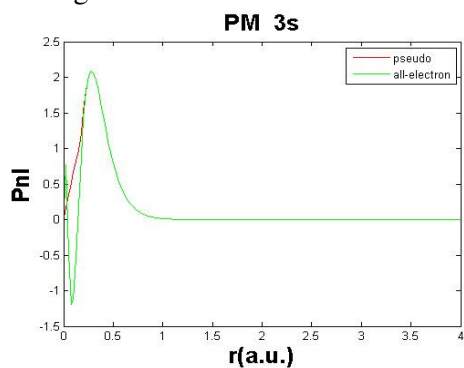


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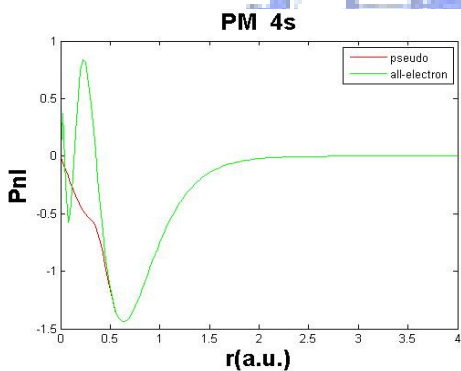


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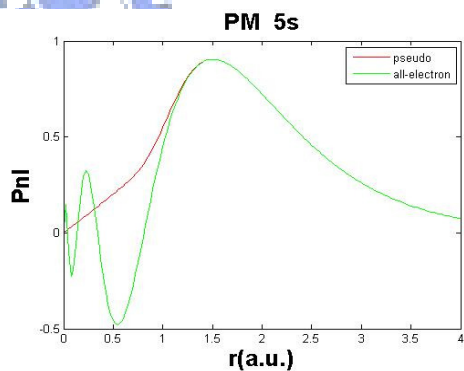


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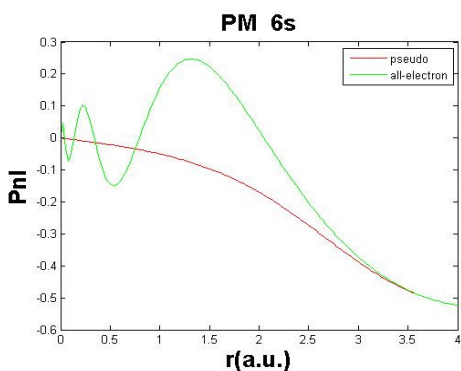


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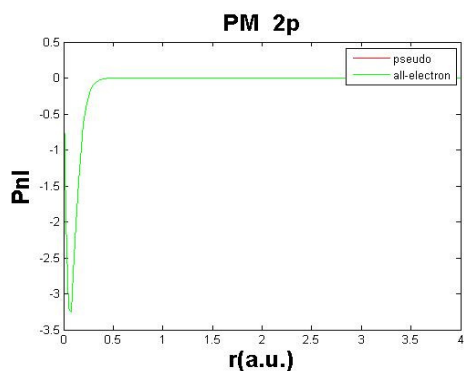


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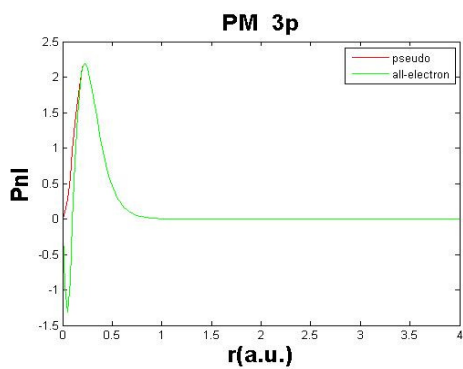


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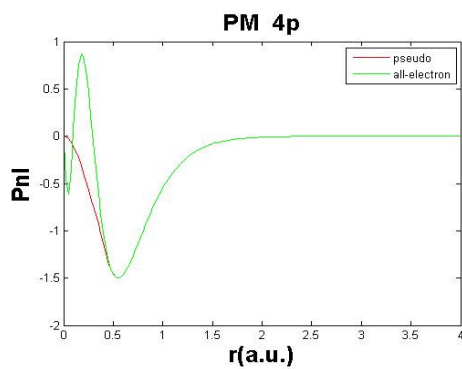


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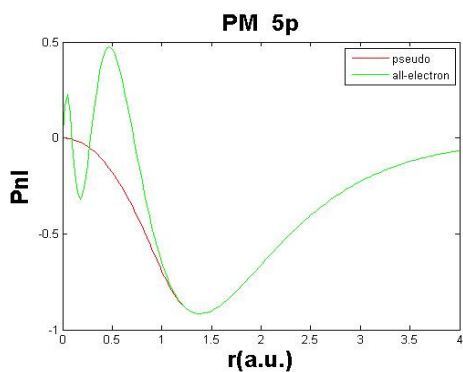


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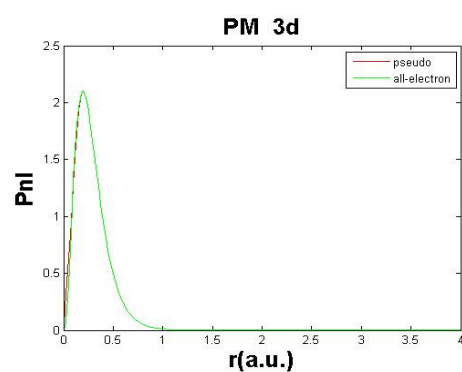


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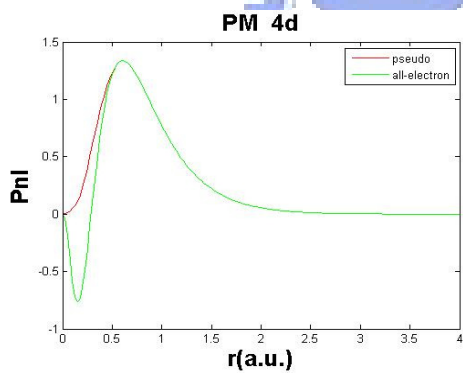


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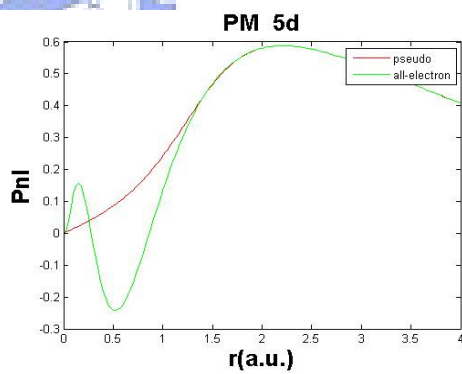


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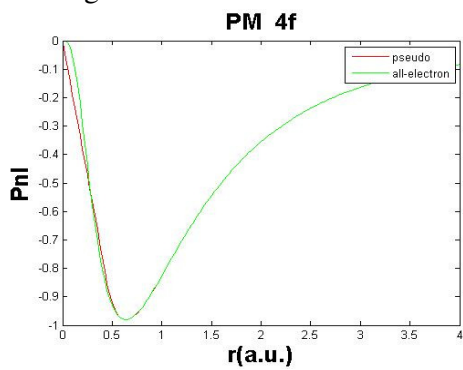


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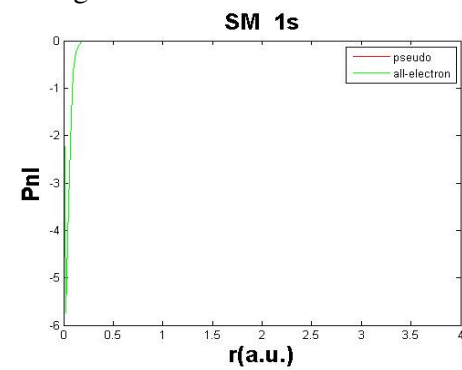


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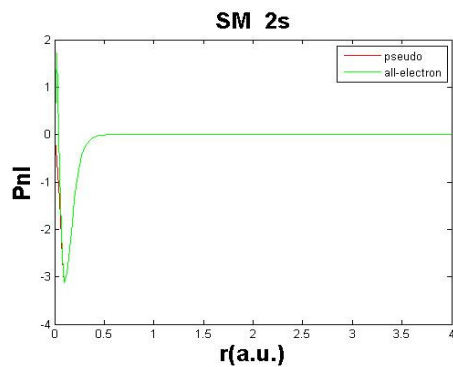


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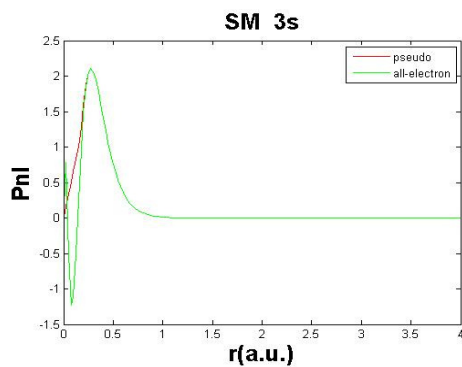


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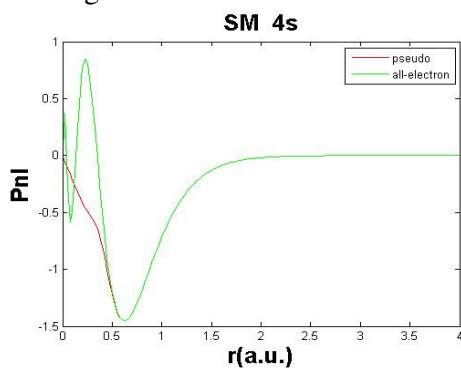


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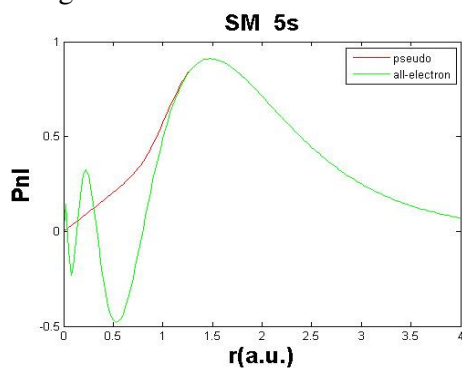


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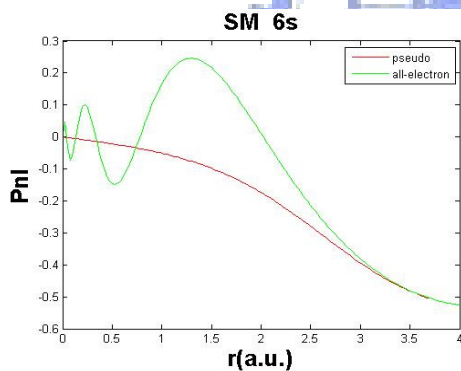


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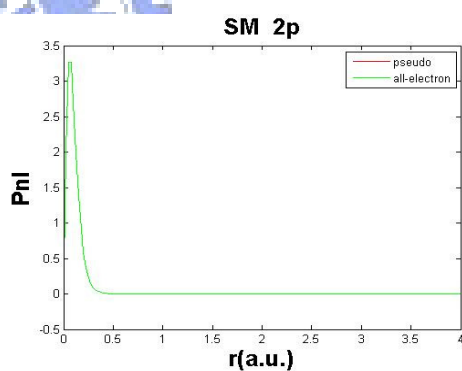


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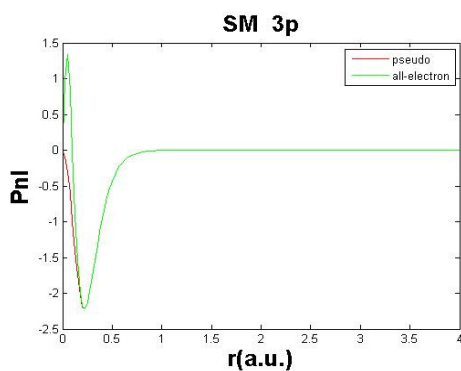


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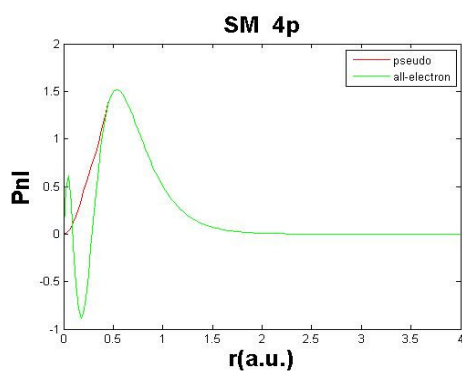


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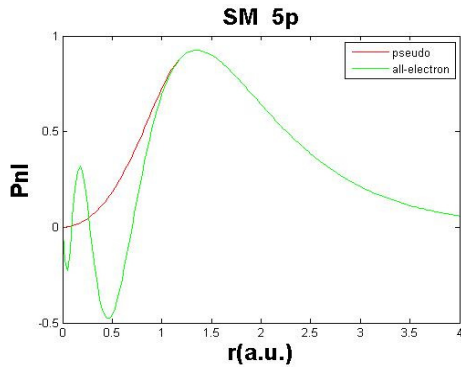


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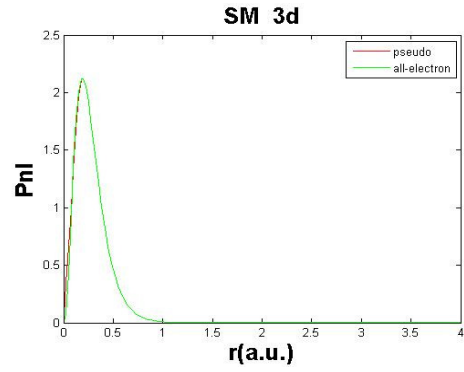


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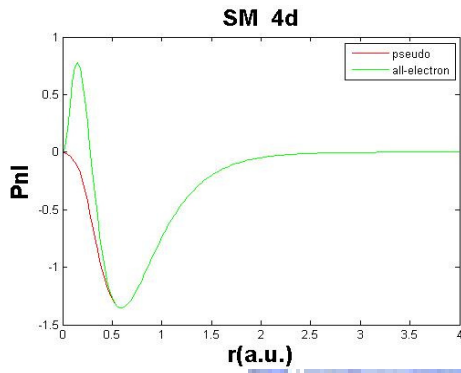


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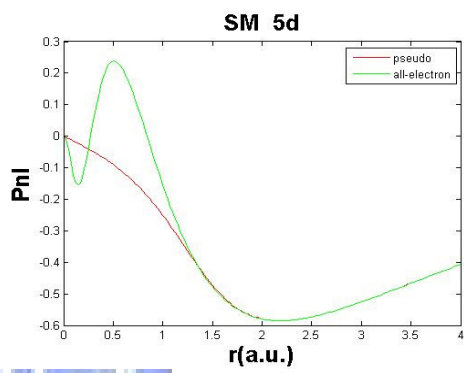


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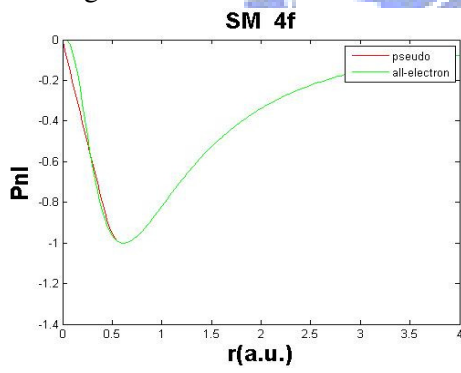


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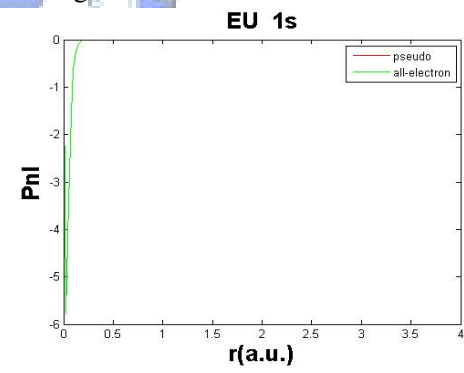


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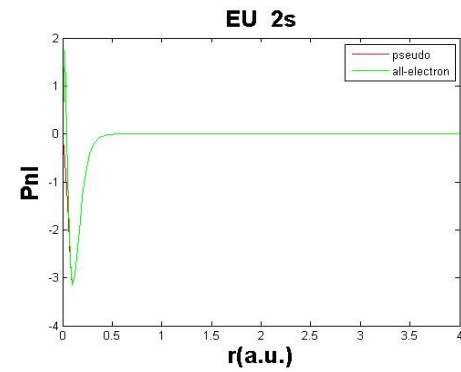


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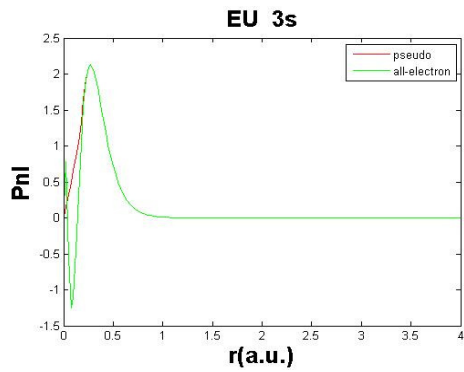


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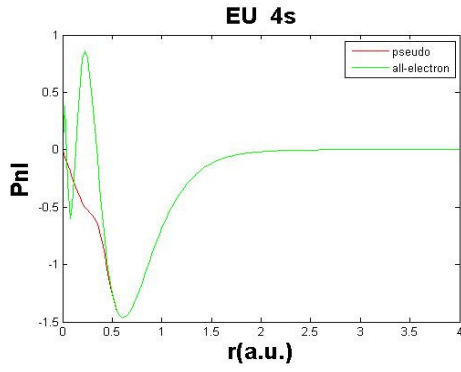


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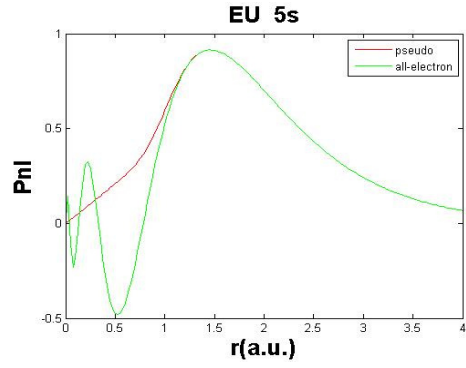


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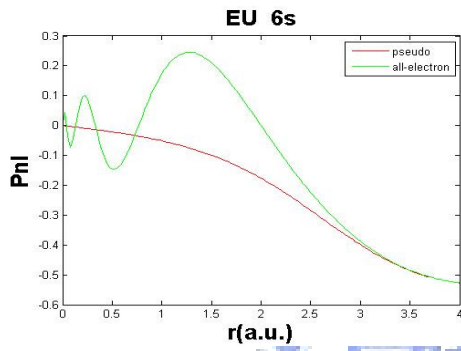


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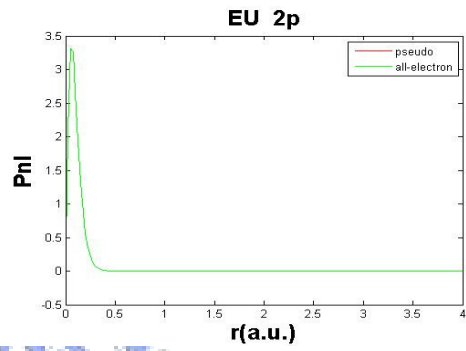


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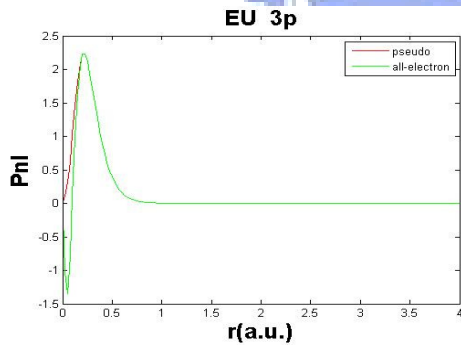


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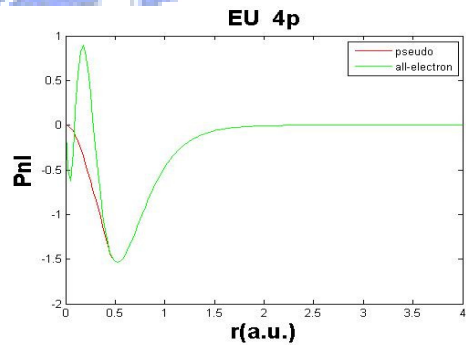


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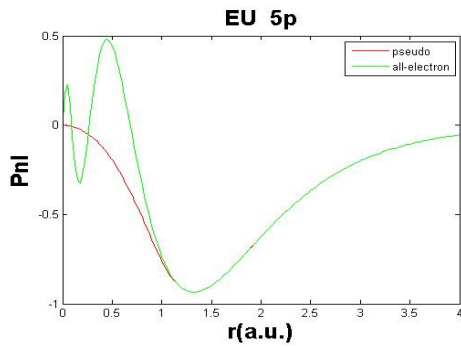


Figure 494.

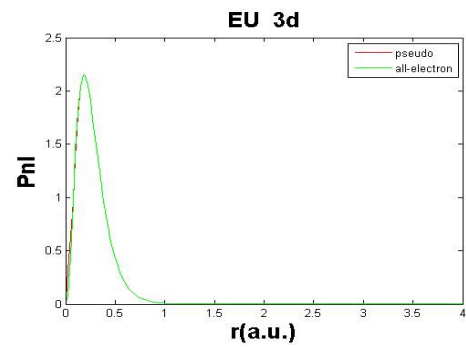


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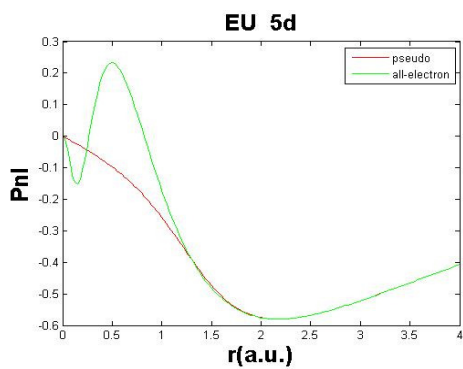


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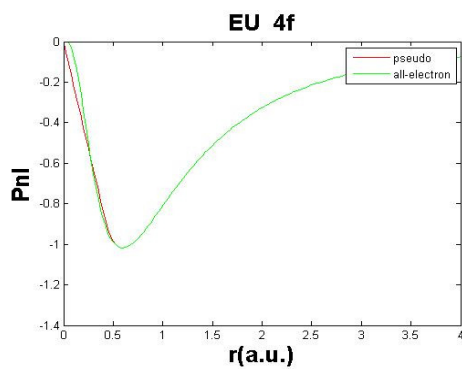


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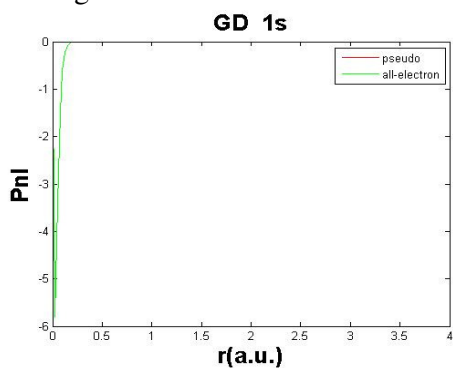


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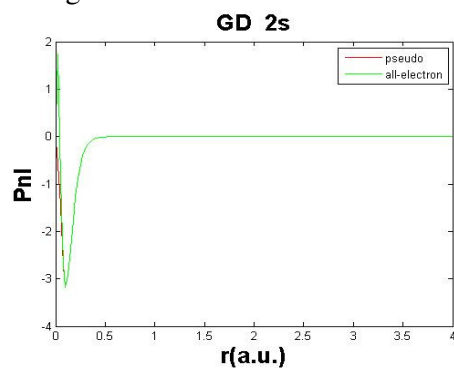


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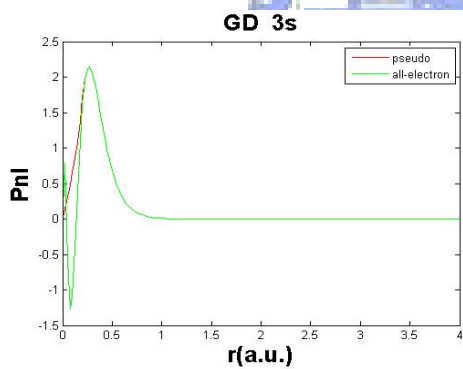


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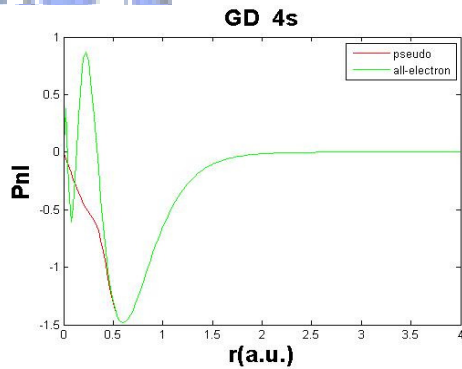


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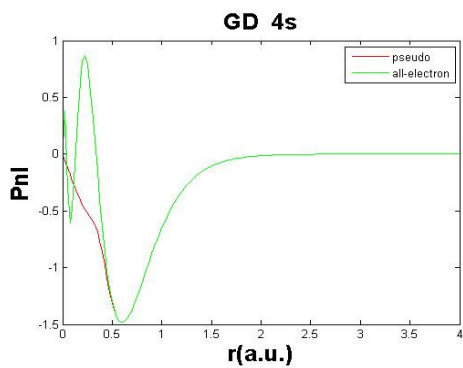


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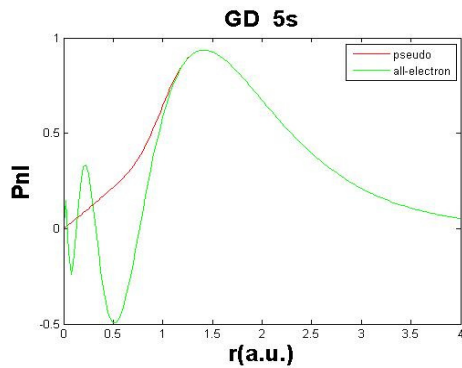


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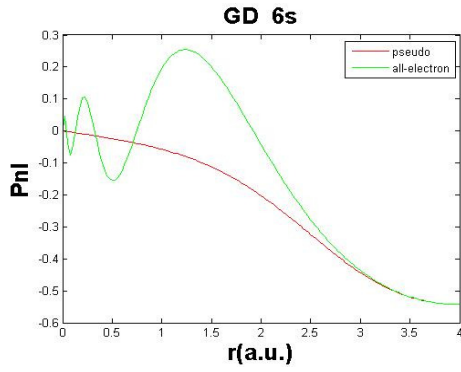


Figure 504.

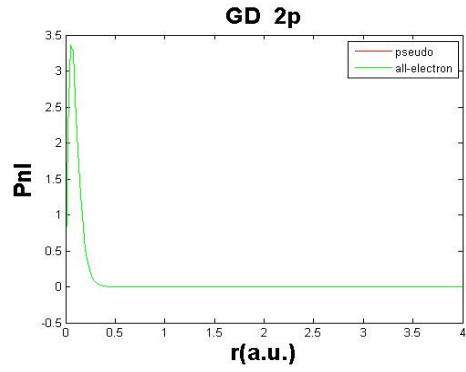


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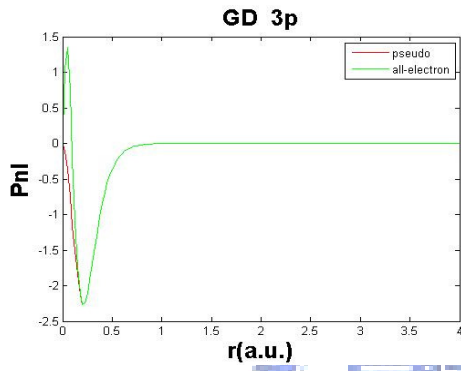


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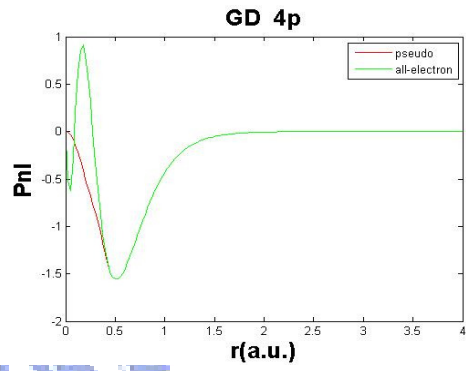


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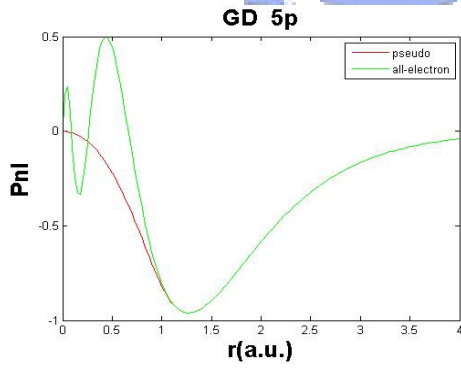


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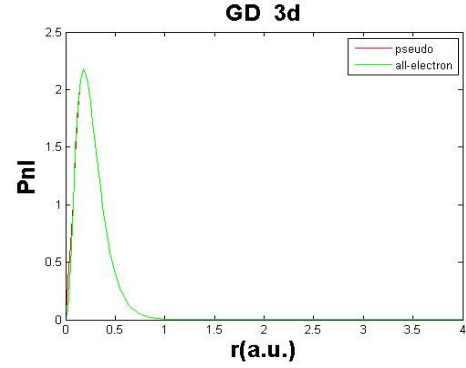


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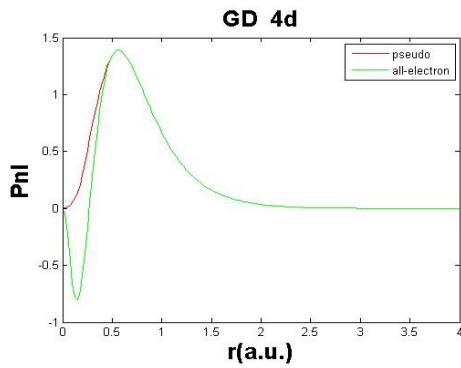


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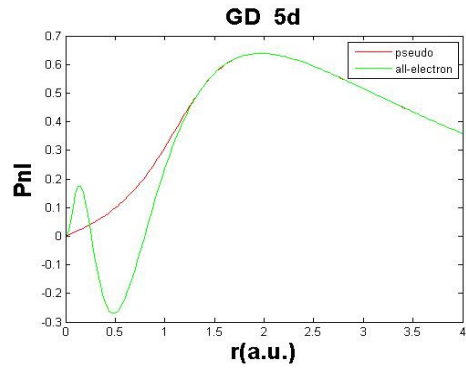


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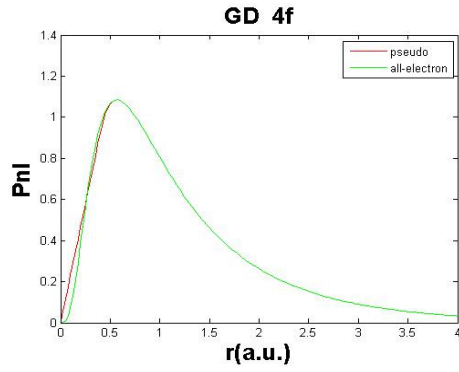


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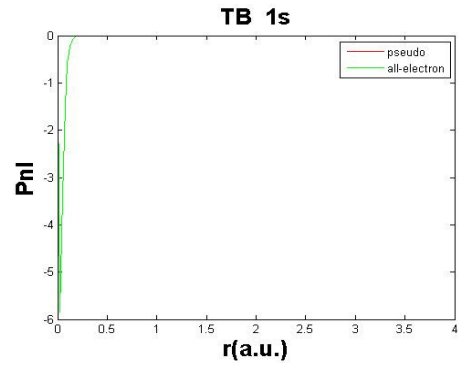


Figure 513.

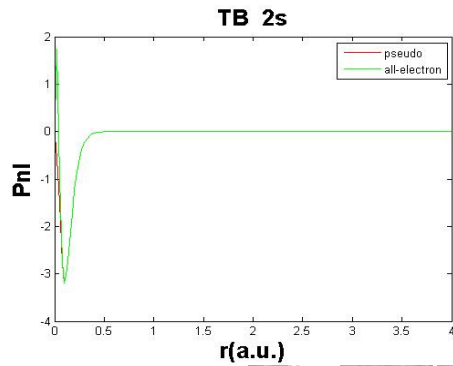


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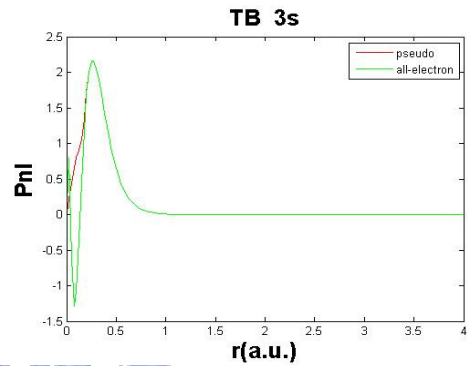


Figure 515.

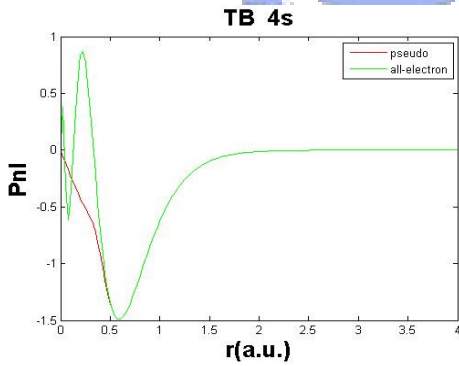


Figure 516.

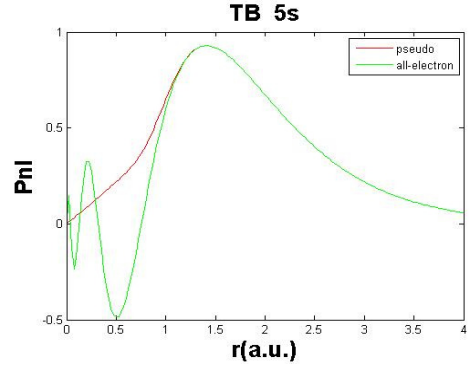


Figure 517.

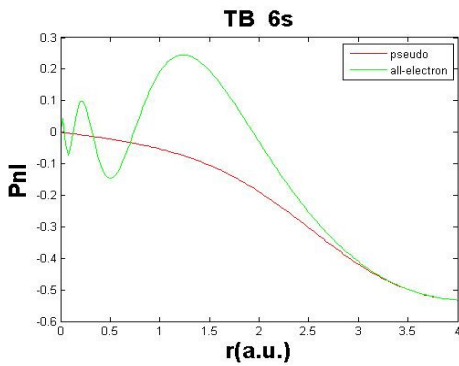


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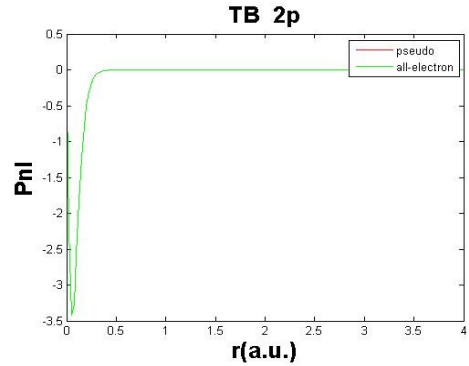


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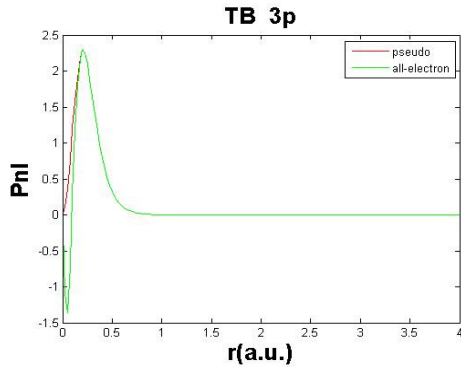


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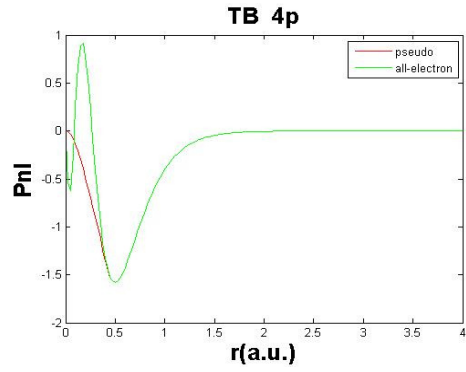


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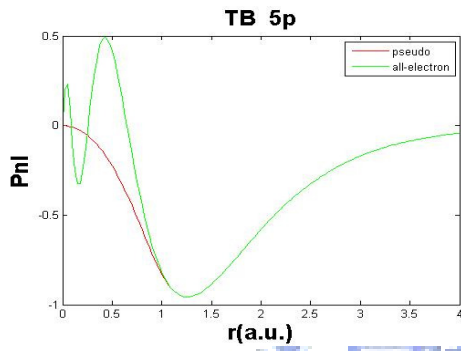


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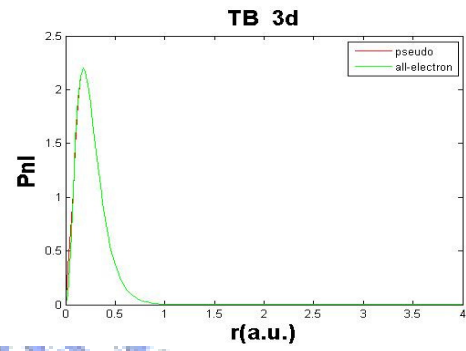


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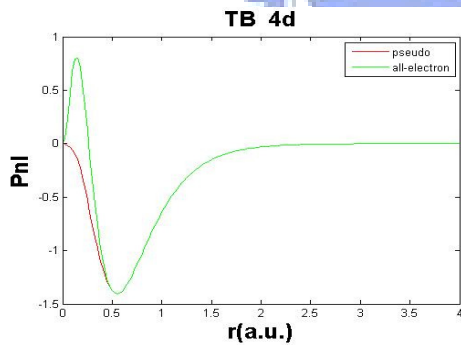


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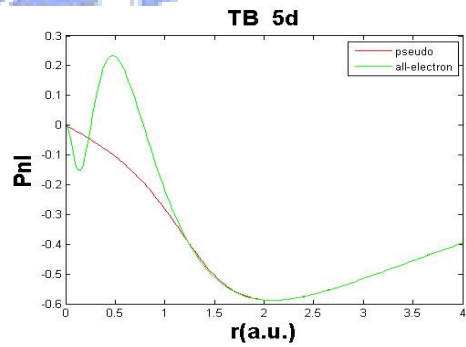


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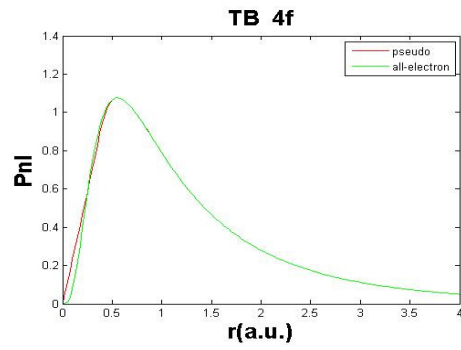


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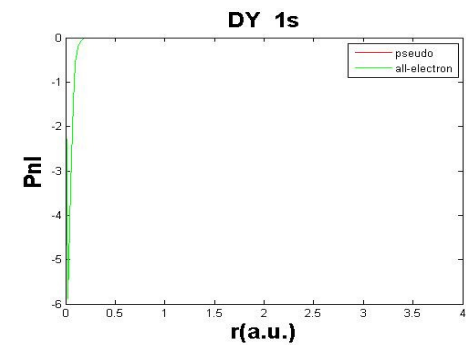


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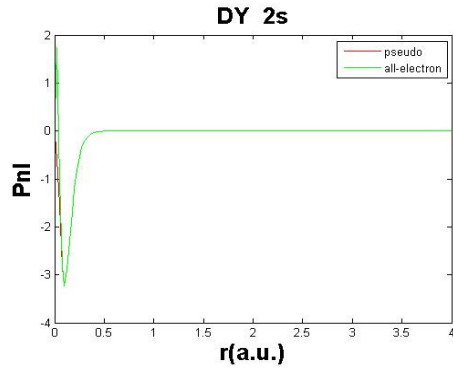


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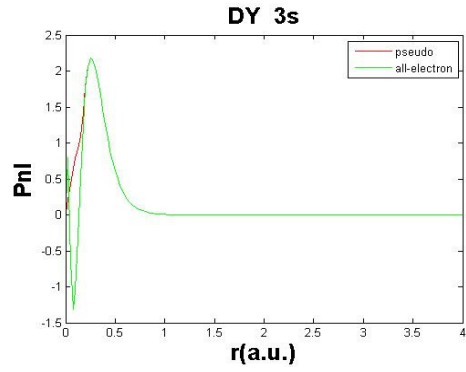


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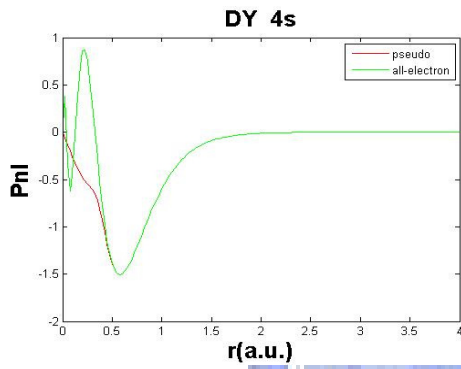


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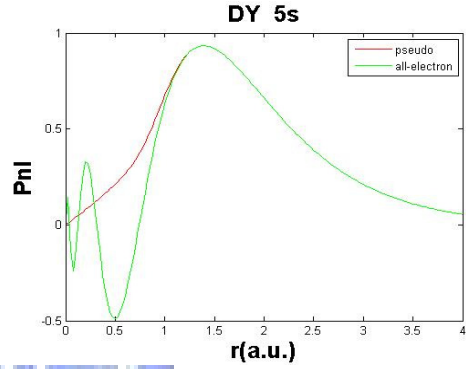


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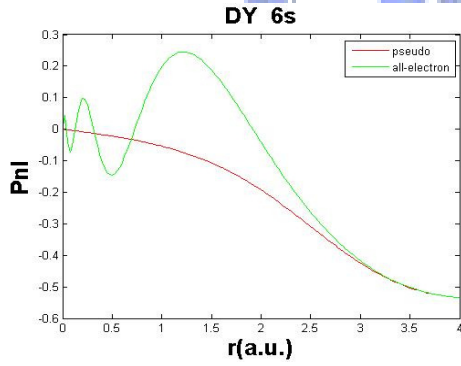


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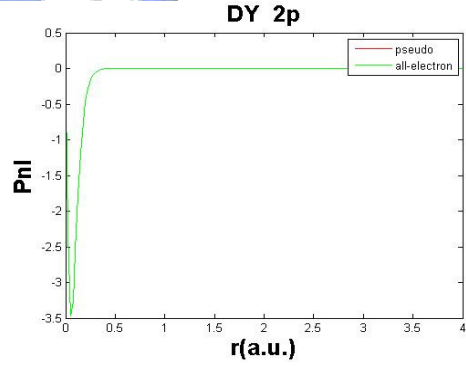


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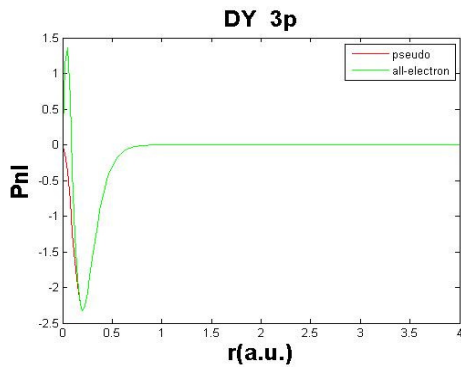


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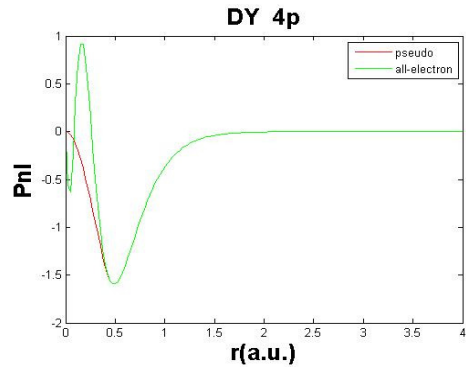


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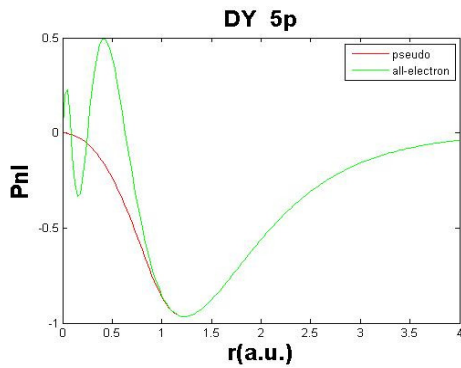


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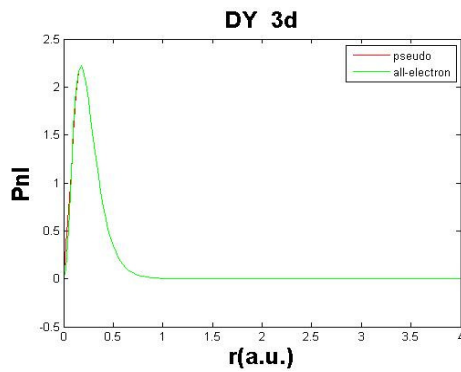


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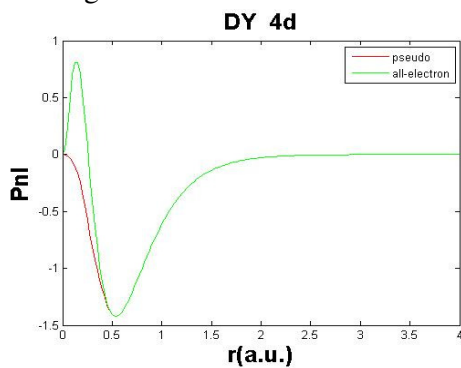


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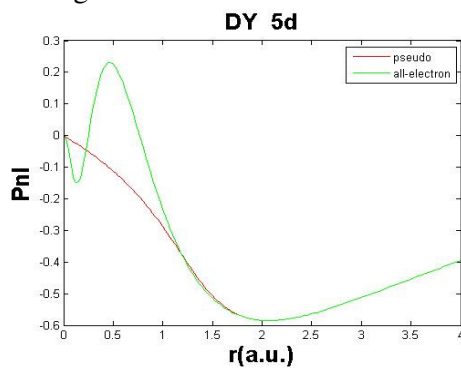


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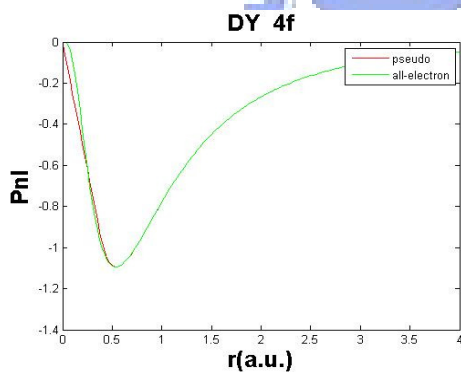


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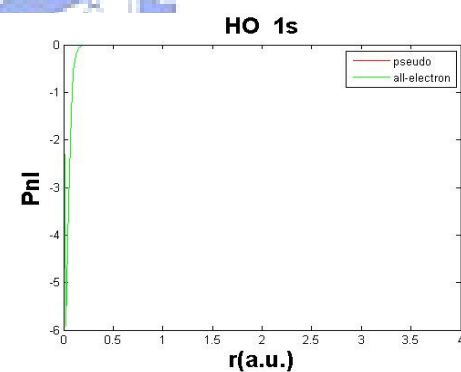


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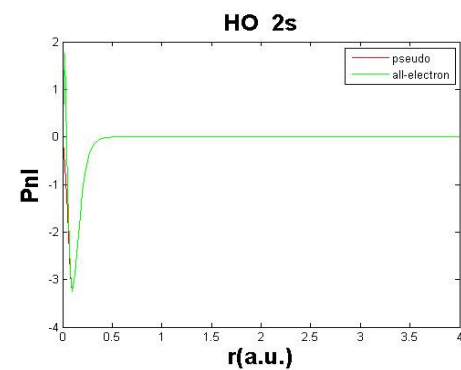


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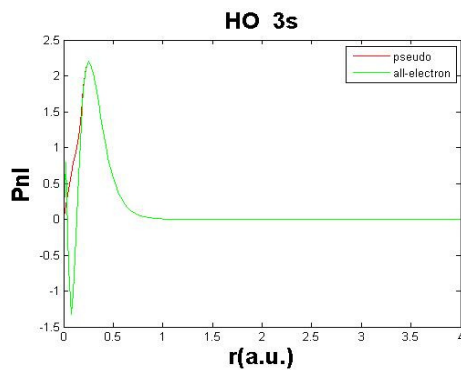


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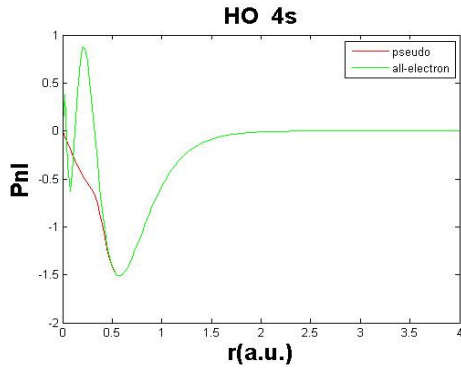


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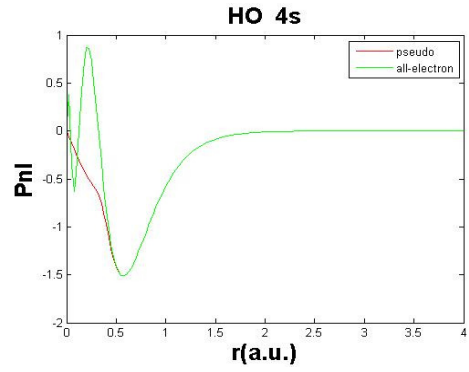


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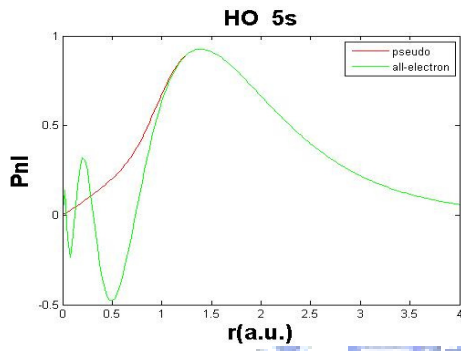


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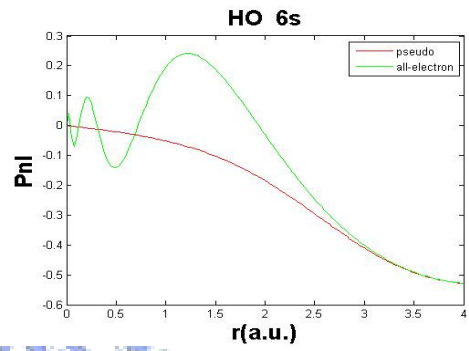


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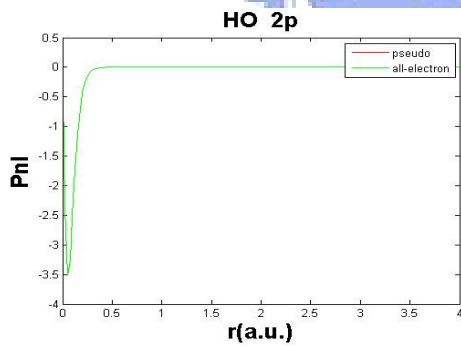


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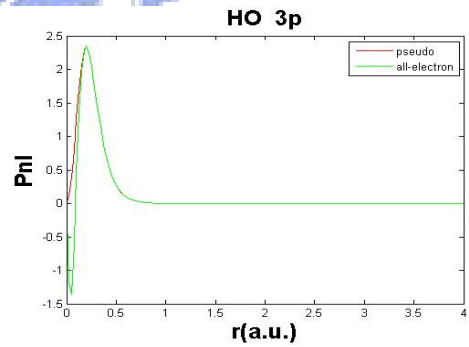


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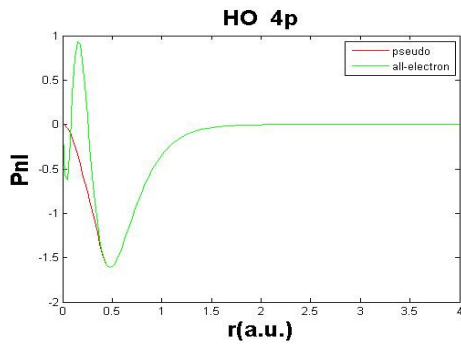


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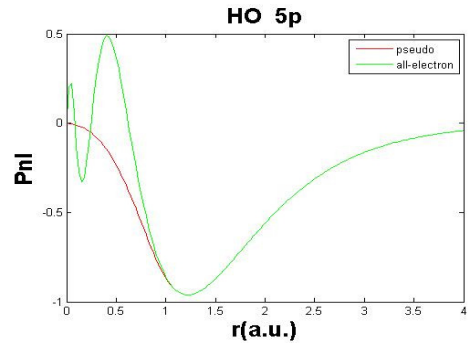


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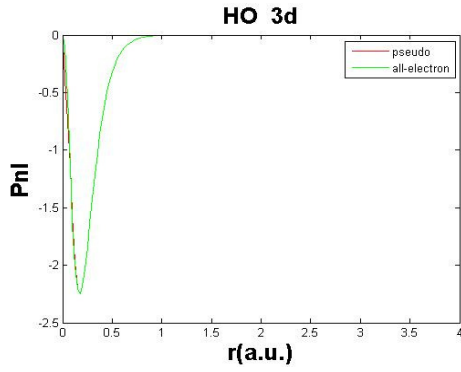


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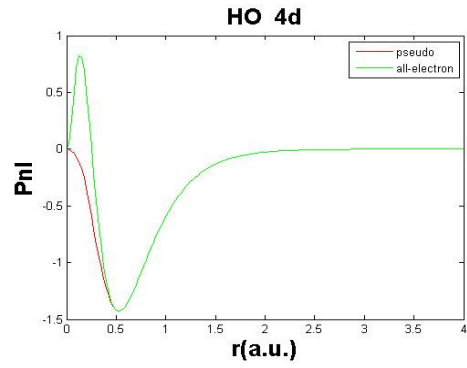


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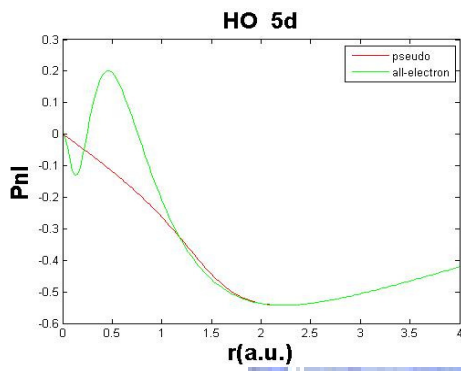


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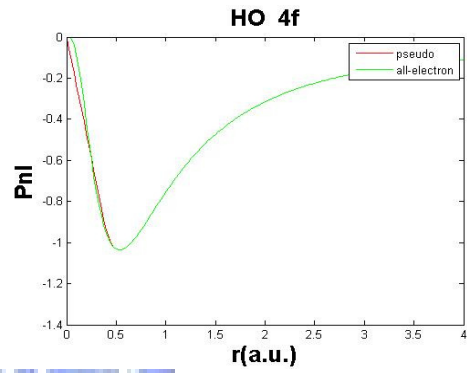


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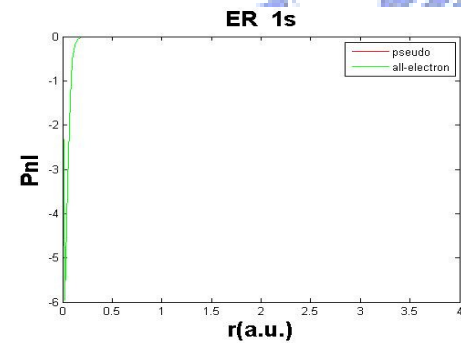


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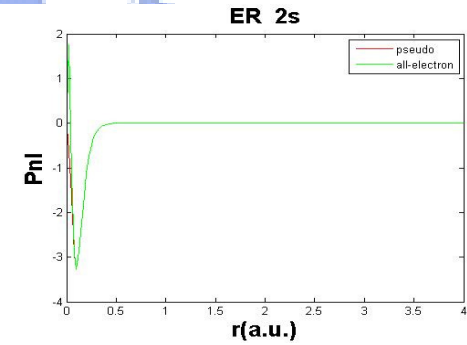


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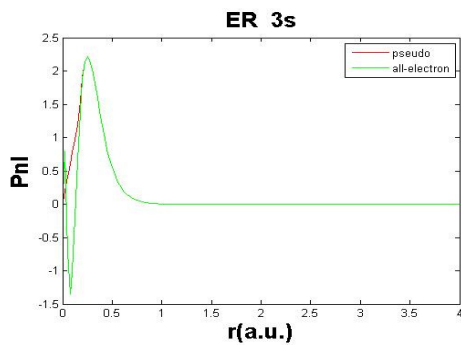


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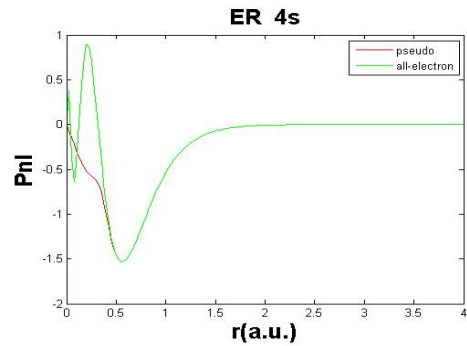


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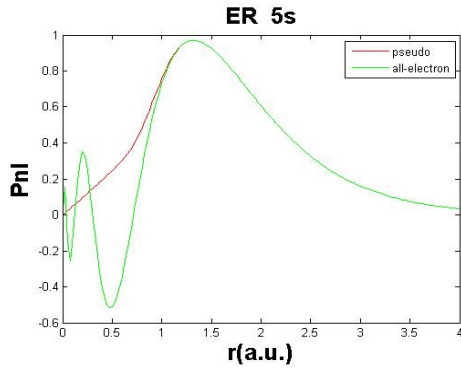


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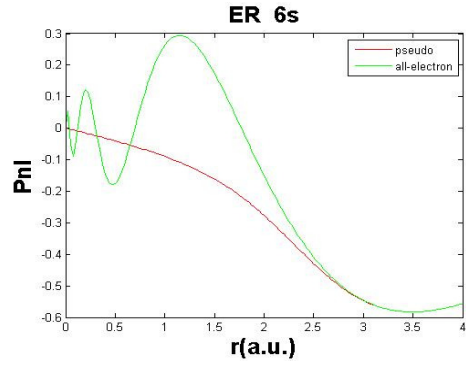


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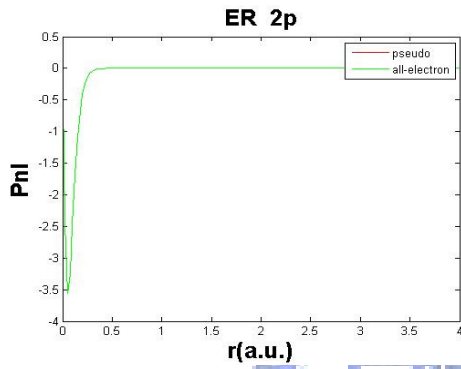


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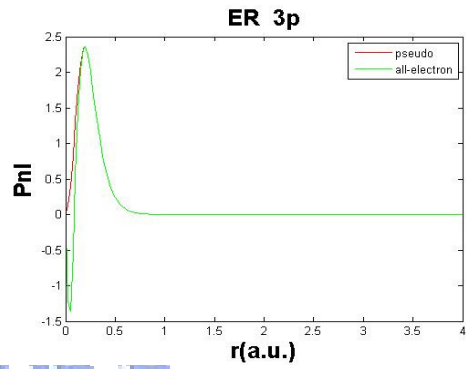


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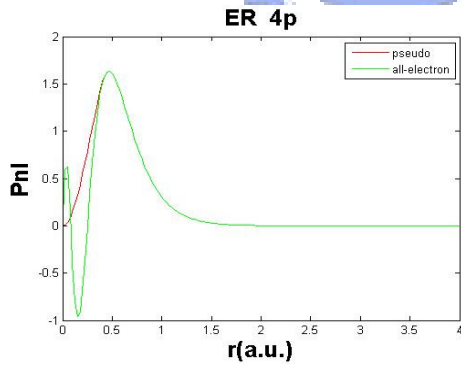


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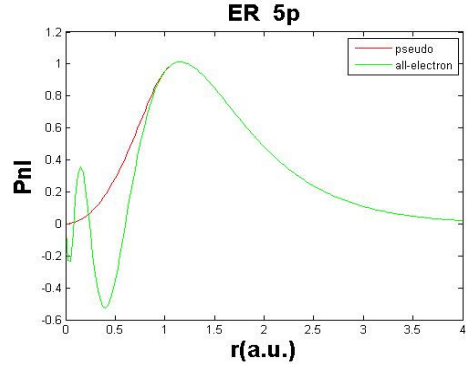


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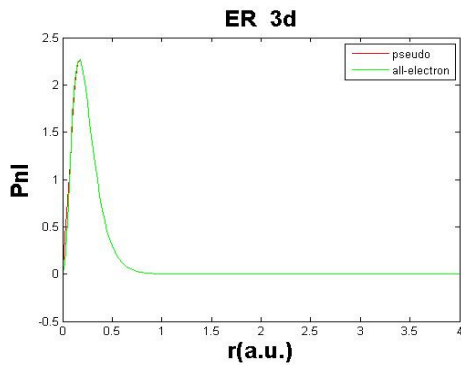


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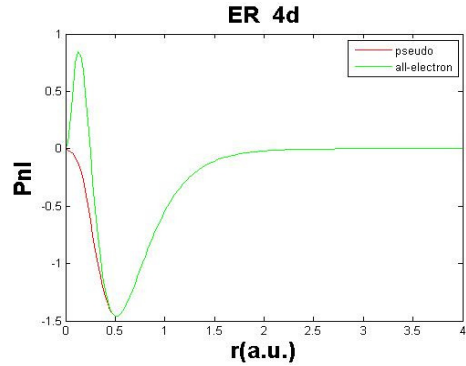


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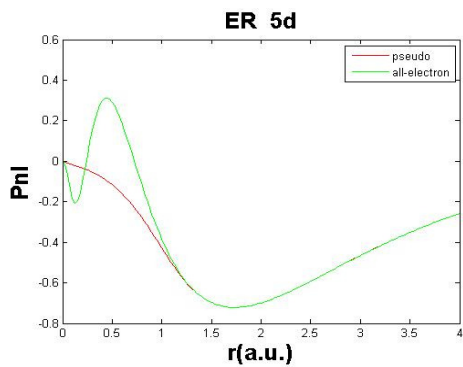


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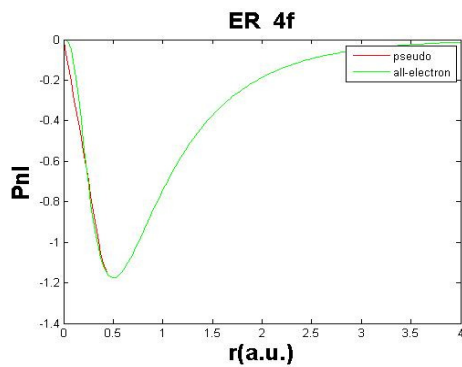


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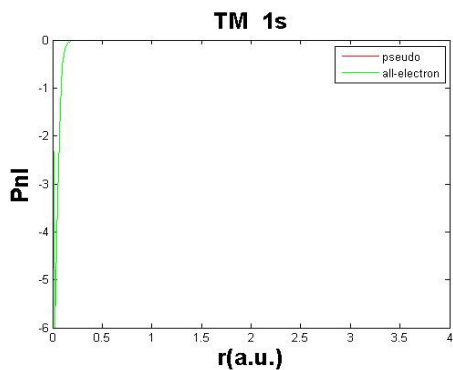


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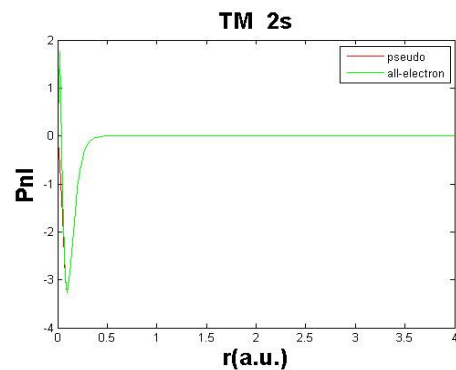


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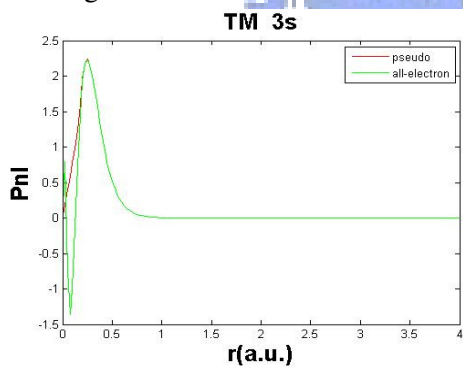


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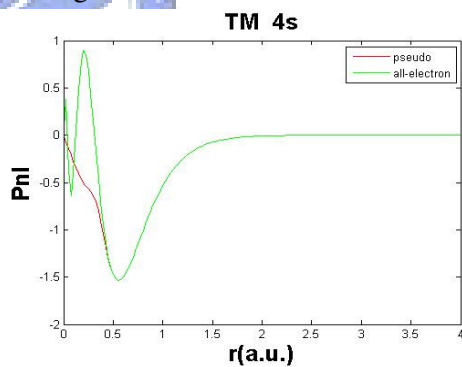


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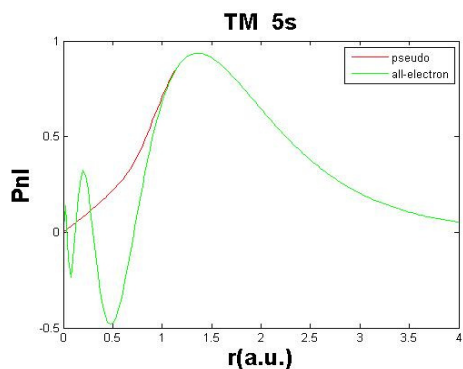


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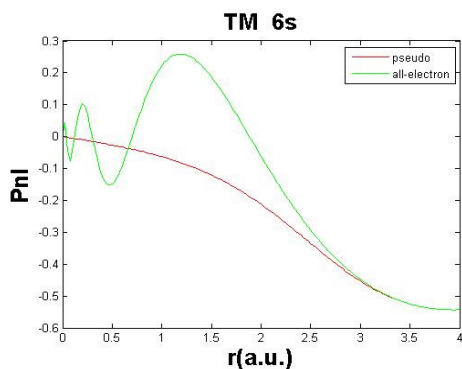


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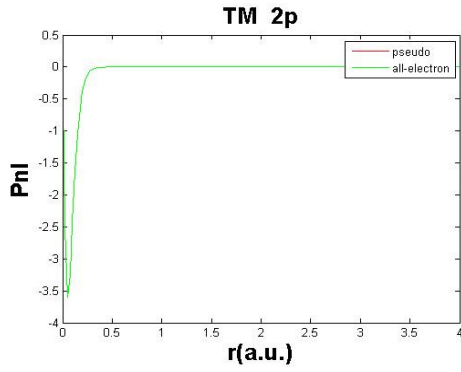


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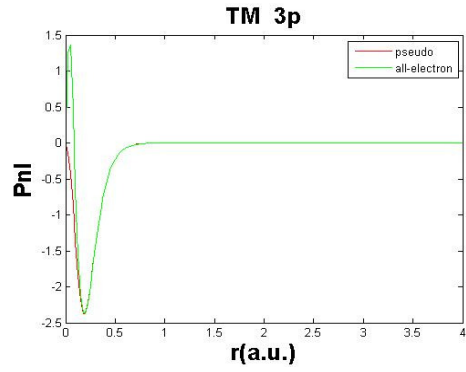


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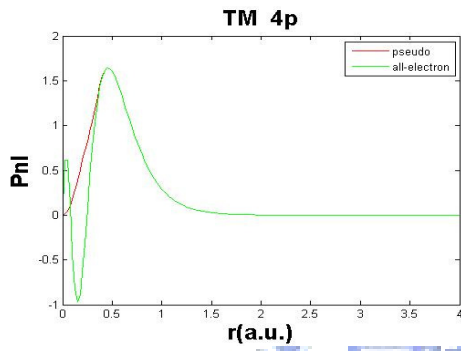


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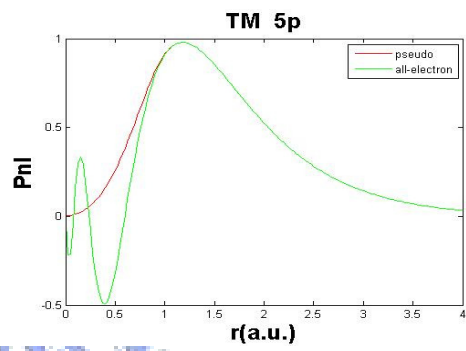


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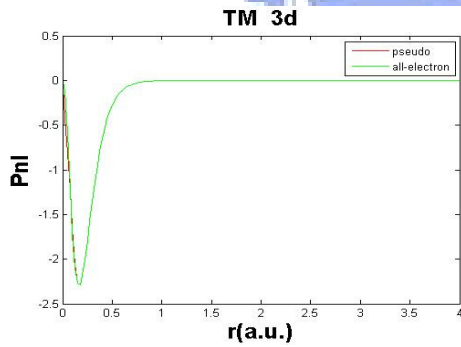


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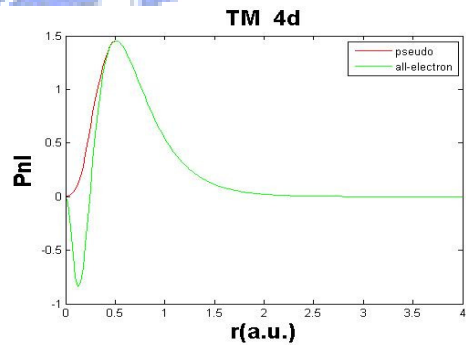


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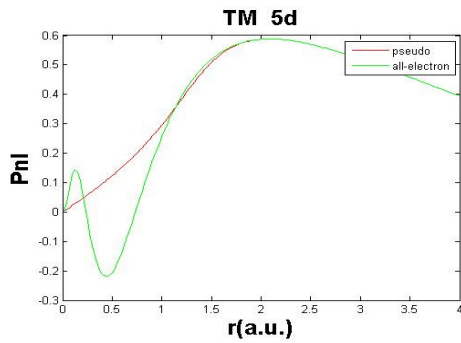


Figure 582.

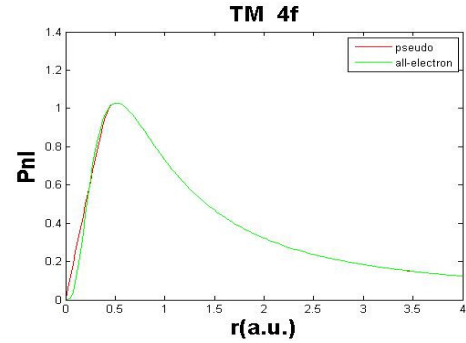


Figure 583.

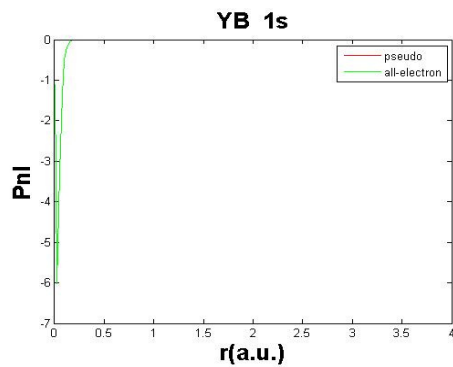


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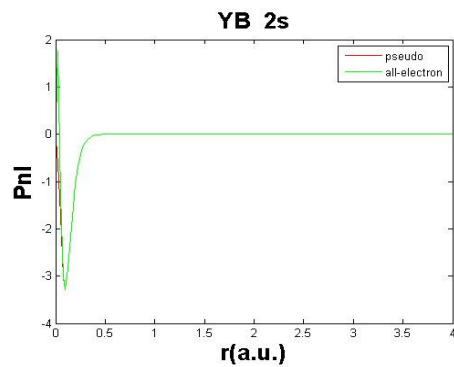


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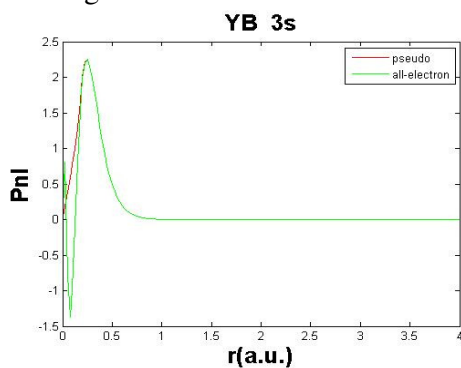


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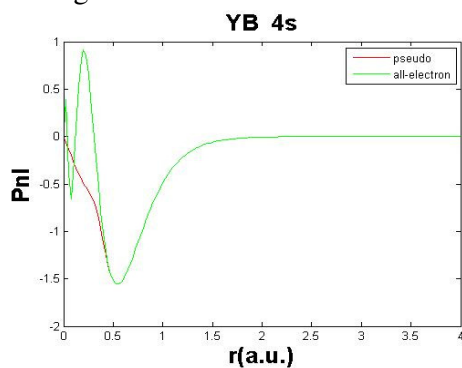


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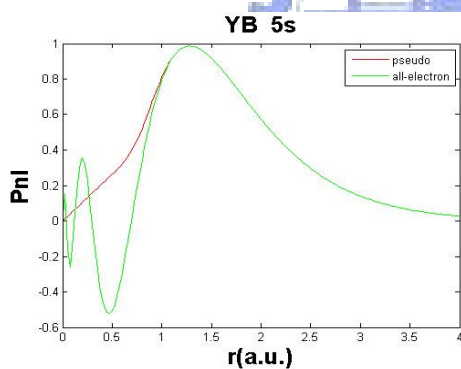


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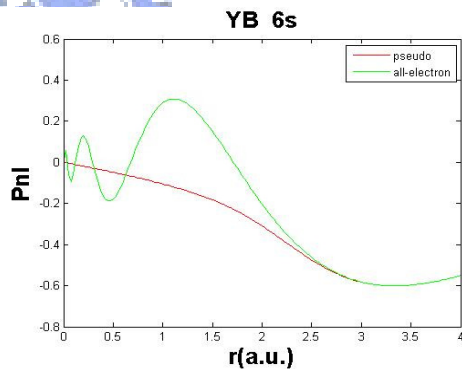


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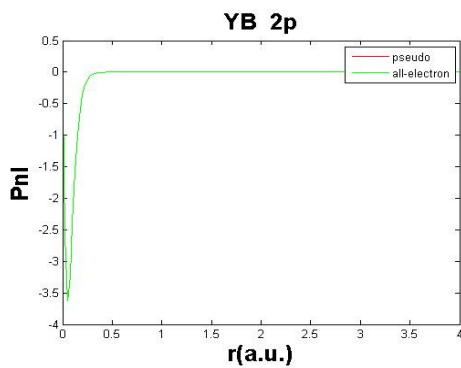


Figure 590.

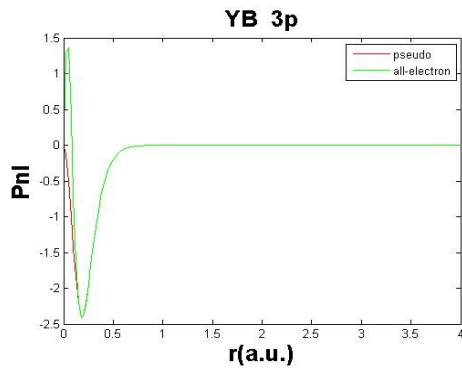


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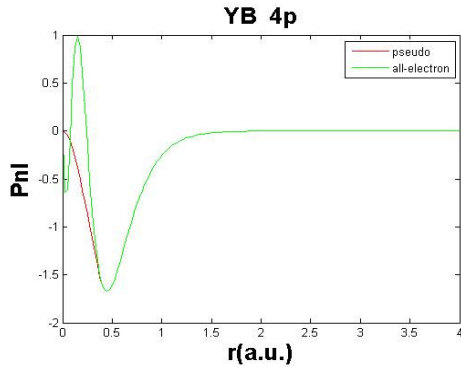


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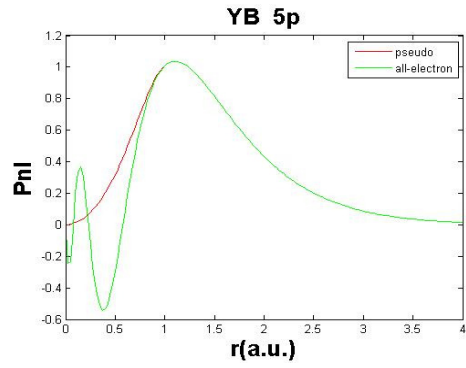


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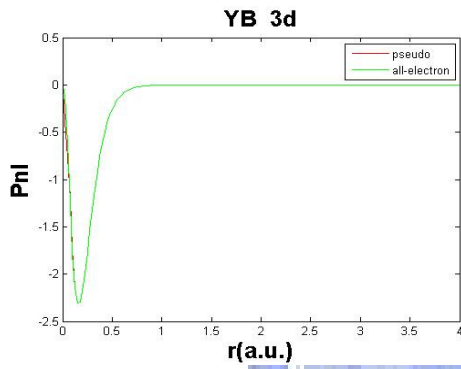


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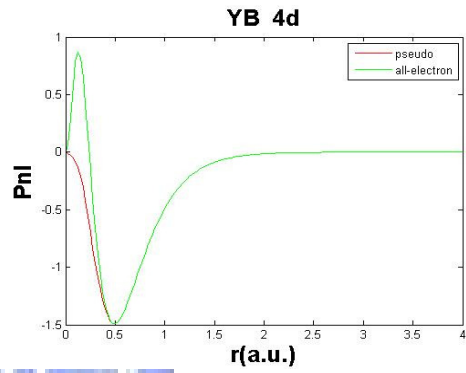


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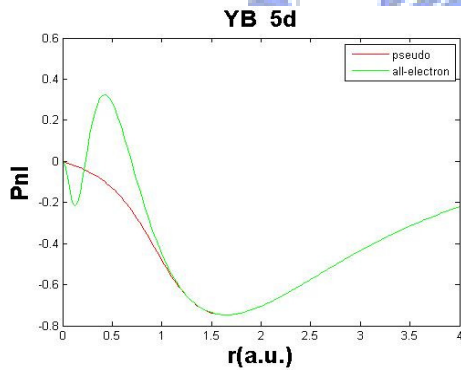


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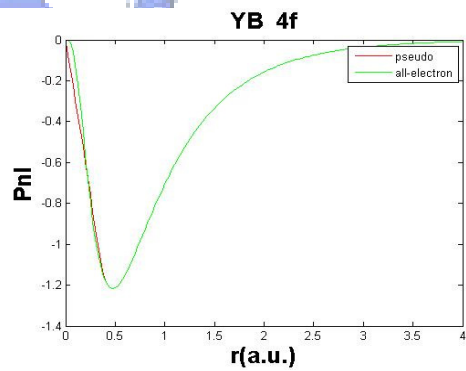


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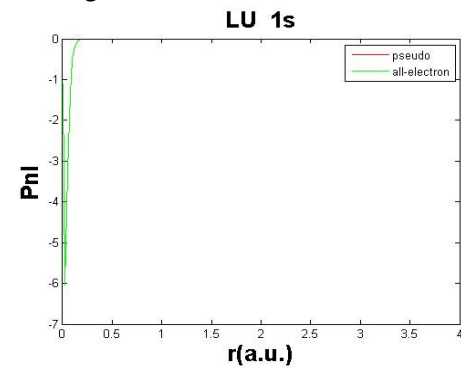


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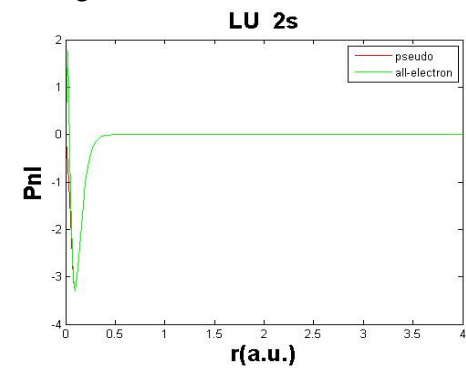


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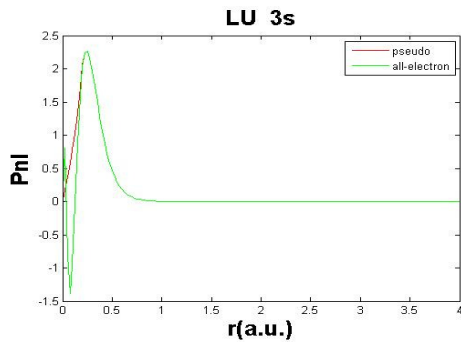


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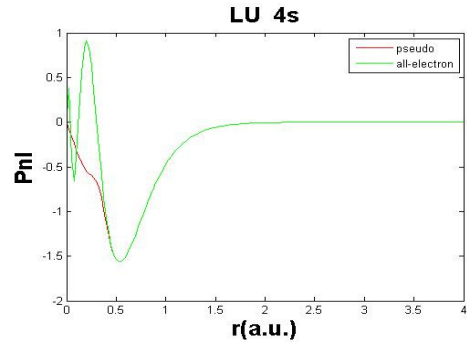


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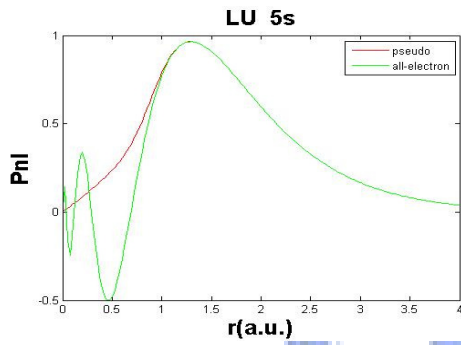


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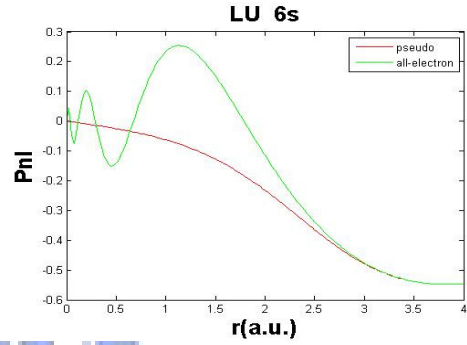


Figure 603.

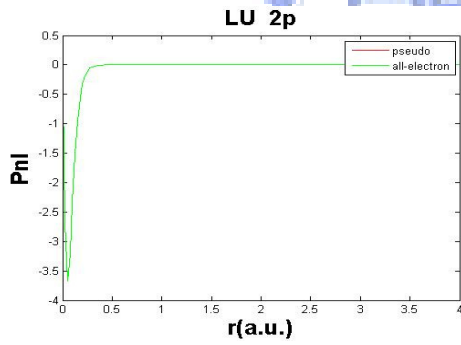


Figure 604.

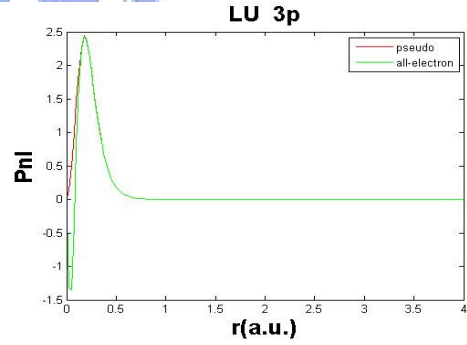


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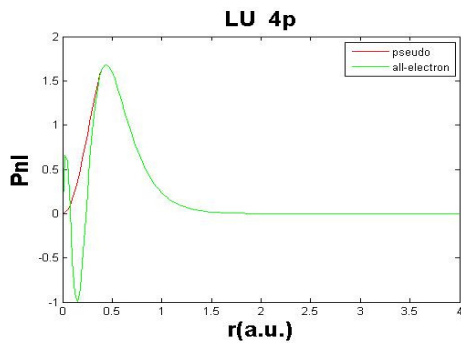


Figure 606.

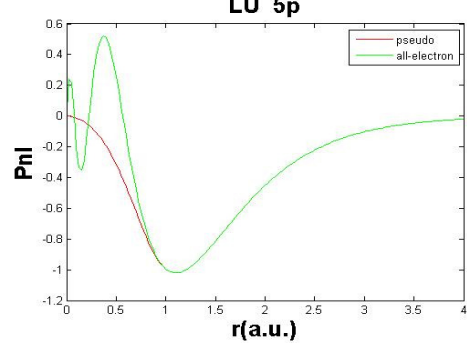


Figure 607.

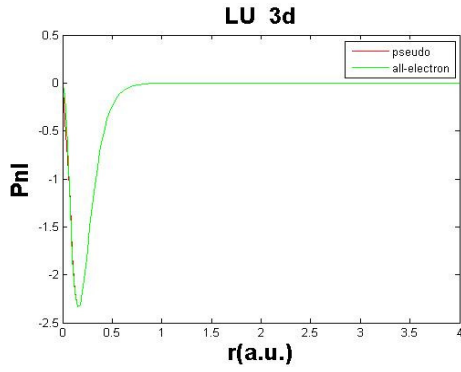


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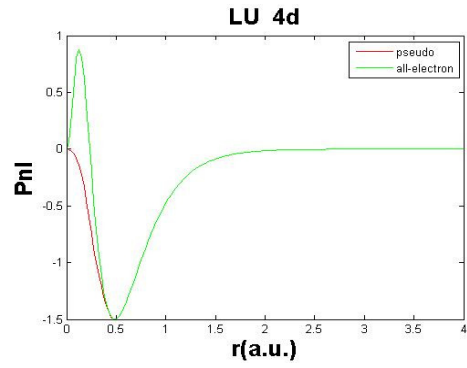


Figure 609.

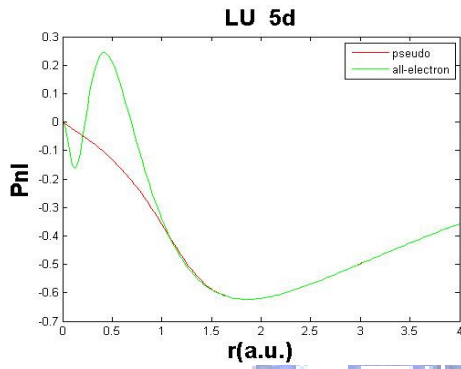


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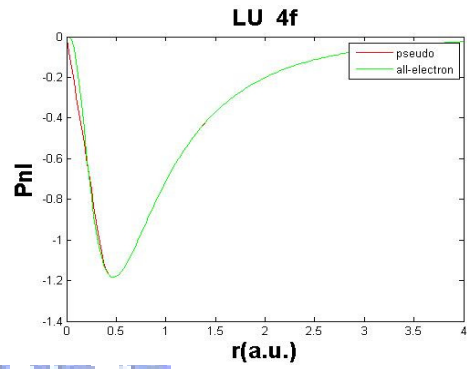


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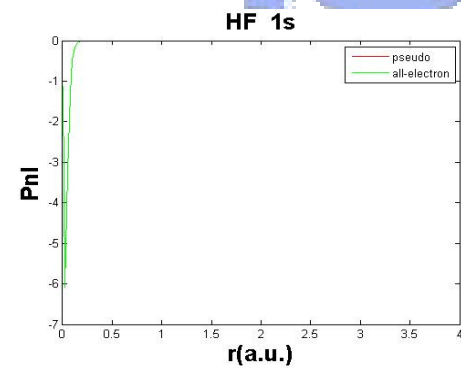


Figure 612.

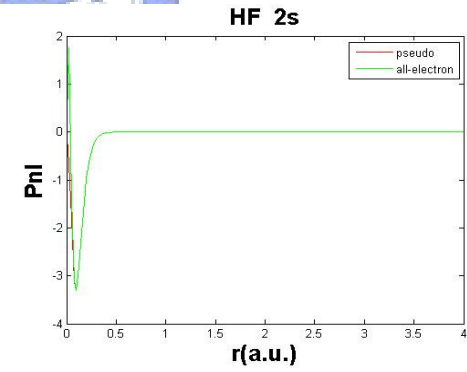


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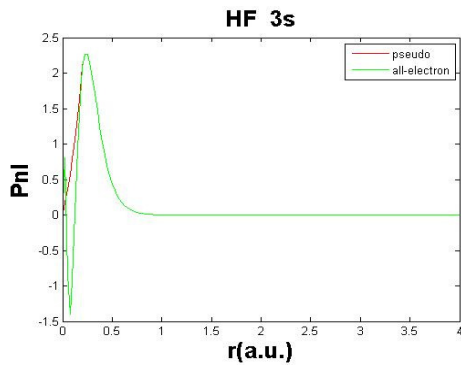


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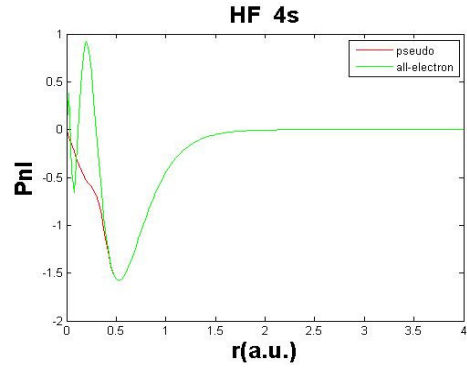


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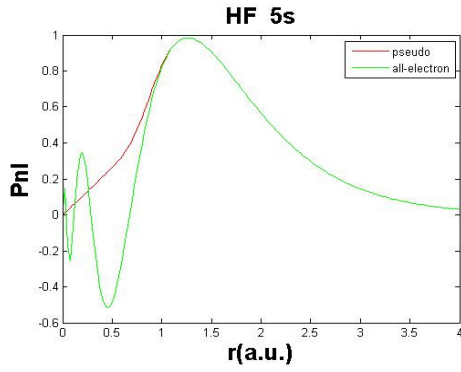


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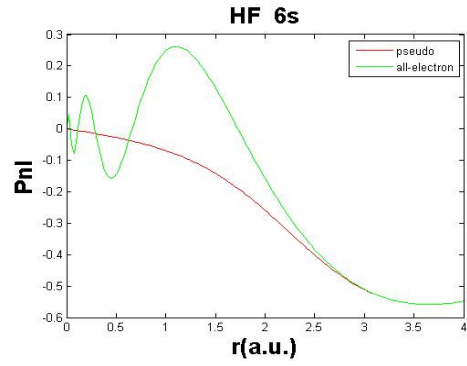


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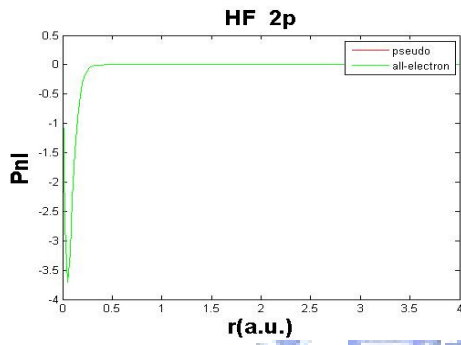


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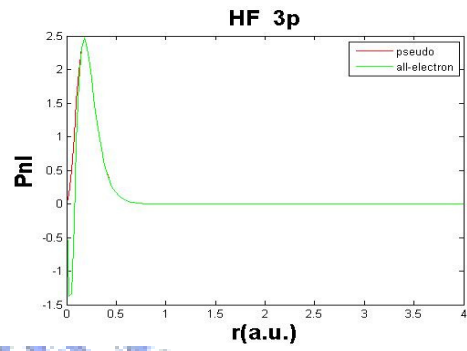


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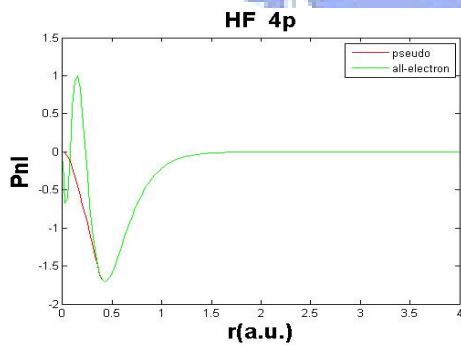


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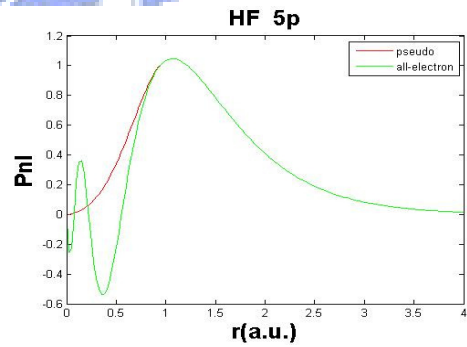


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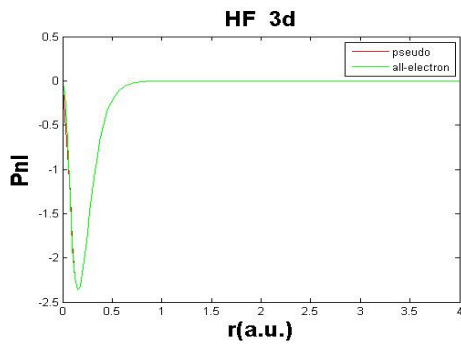


Figure 622.

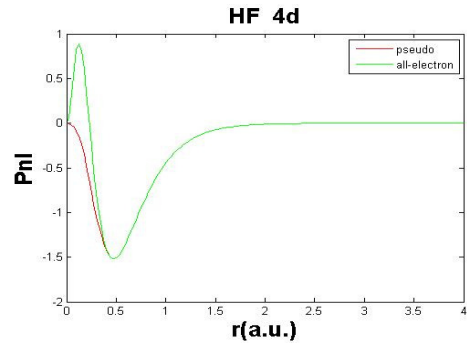


Figure 623.

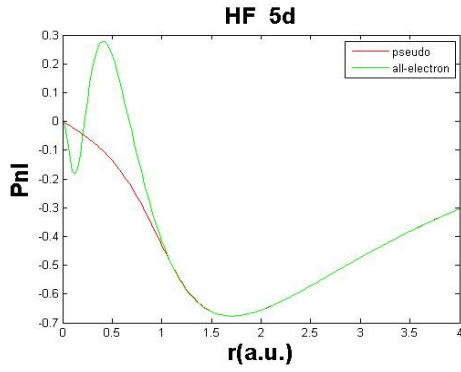


Figure 624.

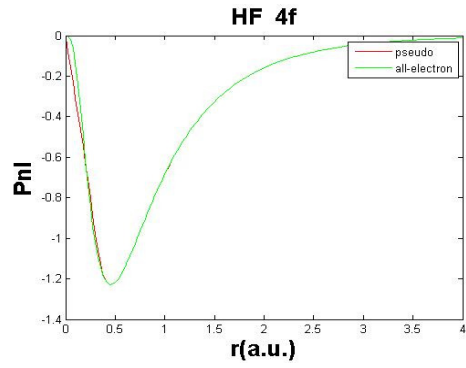


Figure 625.

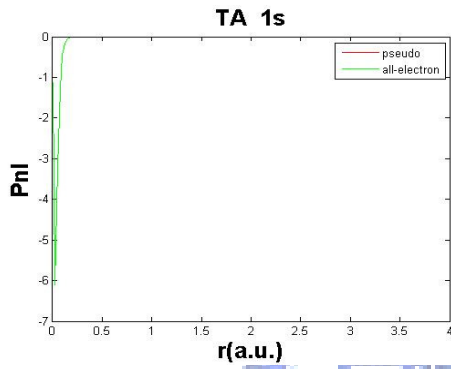


Figure 626.

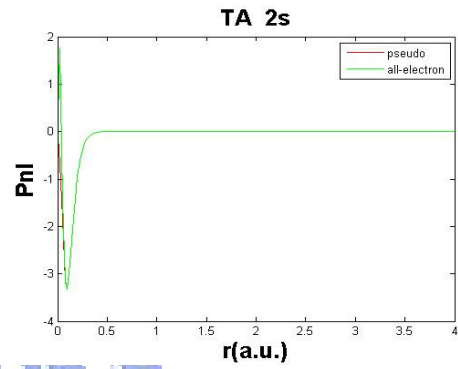


Figure 627.

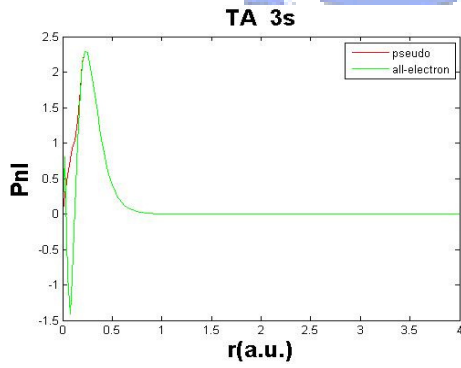


Figure 628.

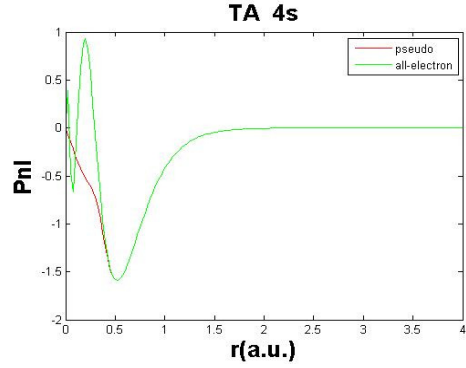


Figure 629.

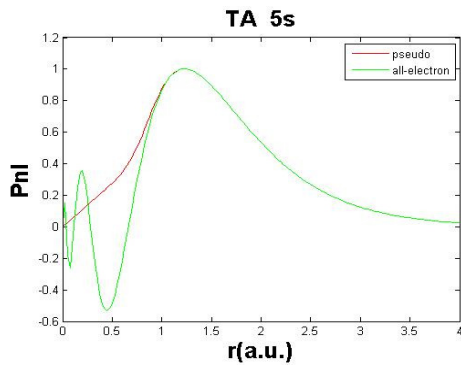


Figure 630.

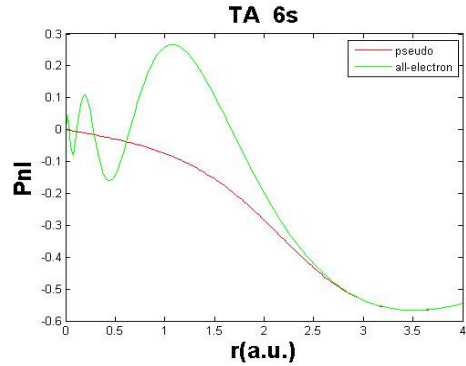


Figure 631.

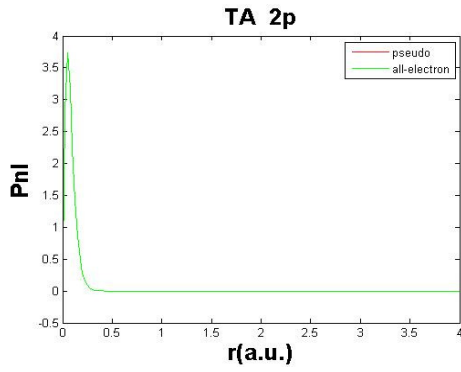


Figure 632.

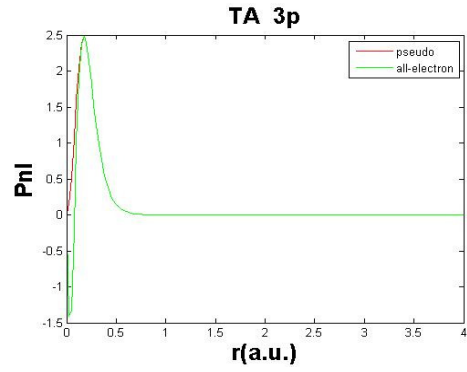


Figure 633.

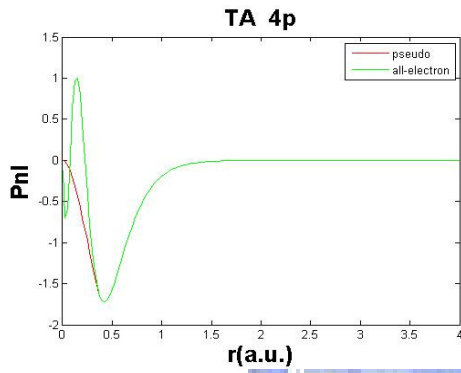


Figure 634.

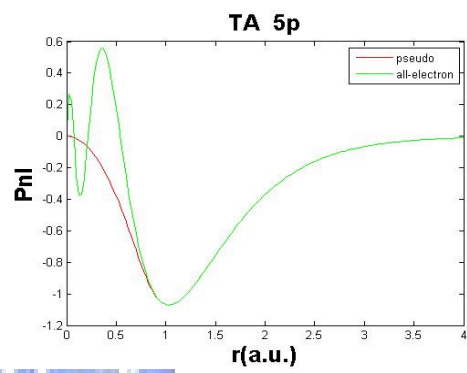


Figure 635.

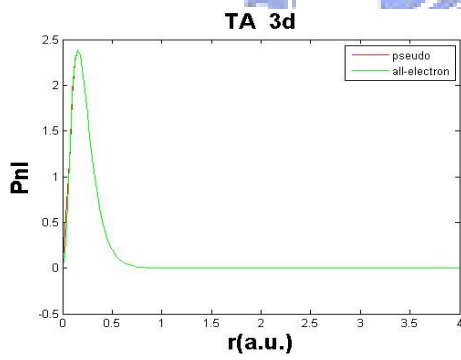


Figure 636.

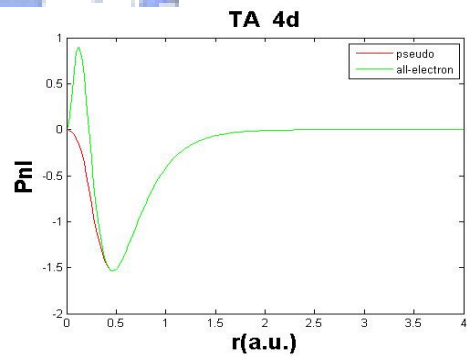


Figure 637.

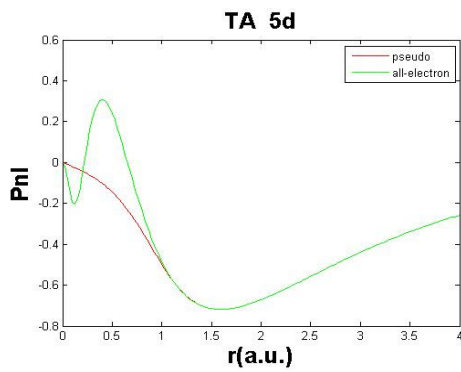


Figure 638.

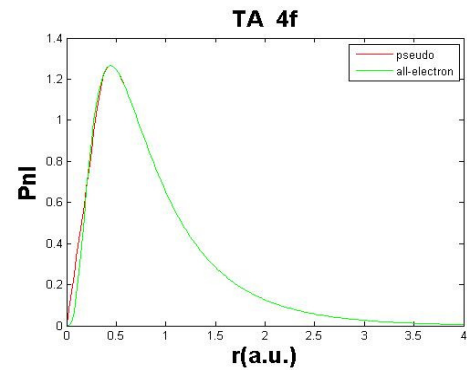


Figure 639.