



ELSEVIER

Available online at [www.sciencedirect.com](http://www.sciencedirect.com)

Journal of Magnetism and Magnetic Materials 310 (2007) e335–e336

[www.elsevier.com/locate/jmmm](http://www.elsevier.com/locate/jmmm)

# Fabrication and X-ray absorption spectroscopy in layered cobaltate $\text{Na}_x\text{CoO}_2$ thin films

W.J. Chang<sup>a,\*</sup>, J.-Y. Lin<sup>b</sup>, T.Y. Chung<sup>a</sup>, J.M. Chen<sup>c</sup>, C.-H. Hsu<sup>a</sup>, S.Y. Hsu<sup>c</sup>, T.M. Uen<sup>a</sup>,  
K.H. Wu<sup>a</sup>, Y.S. Gou<sup>a,d</sup>, J.Y. Juang<sup>a,d</sup>

<sup>a</sup>Department of Electrophysics, National Chiao Tung University, Hsinchu 30010, Taiwan

<sup>b</sup>Institute of Physics, National Chiao Tung University, Hsinchu 30010, Taiwan

<sup>c</sup>National Synchrotron Radiation Research Center (NSRRC), Hsinchu 30076, Taiwan

<sup>d</sup>Department of Physics, National Taiwan Normal University, Taipei 11677, Taiwan

Available online 3 November 2006

## Abstract

$\text{Na}_x\text{CoO}_2$  ( $x \sim 0.68$ ) thin films were fabricated on sapphire (0001) substrates via the lateral diffusion of sodium into  $\text{Co}_3\text{O}_4$  (111) films, which were grown by pulsed-laser deposition. From the results of X-ray diffraction and in-plane resistivity  $\rho_{ab}$ , the single phase and the metallic behaviors of these  $\text{Na}_x\text{CoO}_2$  films can be identified. For the same sodium content  $x$ ,  $\rho_{ab}$  is consistent with that of single crystals. In addition, the O 1s X-ray absorption near edge spectra of thin films are measured and compared with those of single crystals.

© 2006 Elsevier B.V. All rights reserved.

PACS: 81.15.-z; 71.27.+a; 78.70.Dm

Keywords:  $\text{Na}_x\text{CoO}_2$  thin film; Layered cobaltates; Lateral diffusion; X-ray diffraction; X-ray absorption spectroscopy

Recently, the layered cobaltate  $\text{Na}_x\text{CoO}_2$  has generated much interest of researchers due to its superconductivity in hydrated compounds at  $x = 0.35$  [1], a large thermoelectric power [2], a charge ordered insulator at  $x = 0.5$ , and the antiferromagnetic metal at  $x = 0.75$  [3]. Many theoretical and experimental works are focused on the fantastic ground states of  $\text{Na}_x\text{CoO}_2$  due to its two-dimensional triangular lattice. Because of the high vapor pressure of Na in these materials, it is difficult to grow single phase and high-quality epitaxial  $\text{Na}_x\text{CoO}_2$  thin films by pulsed-laser deposition (PLD). For this reason, reactive solid-phase growth via lateral diffusion of sodium was applied on the growth of the epitaxial  $\text{Na}_x\text{CoO}_2$  thin films [4].

First,  $\text{Co}_3\text{O}_4$  (111) thin films were grown on  $\text{Al}_2\text{O}_3$  (0001) substrates by PLD, KrF excimer laser (248 nm), with  $T_{\text{substrate}} = 680^\circ\text{C}$ ,  $P_{\text{oxygen}} = 0.2$  Torr, and  $d = 125$  nm. To keep the surface clean in the process of thermal diffusion,  $\text{Co}_3\text{O}_4$  (111) thin films were capped by  $\text{Al}_2\text{O}_3$  substrates when placing specimens in a ceramic boat with  $\text{Na}_2\text{CO}_3$

powder. They were then put on the ceramic boat into the quartz tube oven for thermal diffusion. The diffusion process was operated at  $700^\circ\text{C}$  for 10 h in air with the heating (and cooling) rate  $< 10^\circ\text{C/h}$ . After the lateral diffusion of sodium,  $\text{Co}_3\text{O}_4$  (111) thin films became  $\text{Na}_x\text{CoO}_2$  (001) epitaxial thin films with the thickness  $\sim 250$  nm.

To check the crystal structure of  $\text{Co}_3\text{O}_4$  and  $\text{Na}_x\text{CoO}_2$  thin films, we have performed the X-ray diffraction (XRD) measurements, as shown in Fig. 1. From the results of XRD,  $\text{Co}_3\text{O}_4$  (111) are epitaxial growth on sapphire (0001) substrates, which are similar to the results of Ohta *et al.* [4] and Venimadhav *et al.* [5]. The lateral-diffused  $\text{Na}_x\text{CoO}_2$  thin films also show single phase and are epitaxial to the substrates with (001) orientation. Because of the affiliated mutuality between the crystal structures and Na content of  $\text{Na}_x\text{CoO}_2$  materials, we judge the Na content by comparing the XRD results with the structure phase diagram [6]. From the  $c$ -axis lattice constant of  $\text{Na}_x\text{CoO}_2$  thin films,  $10.9328(8)\text{Å}$ , we infer  $x \sim 0.68$  in our specimens.

As shown in the inset (a) of Fig. 2, the roughness of layered cobaltate thin films was measured by the atomic force microscopy (AFM) after thermal-diffusion process.

\*Corresponding author. Tel.: +886 3 5731653; fax: +886 3 5720728.

E-mail address: [waithen.ep86@nctu.edu.tw](mailto:waithen.ep86@nctu.edu.tw) (W.J. Chang).

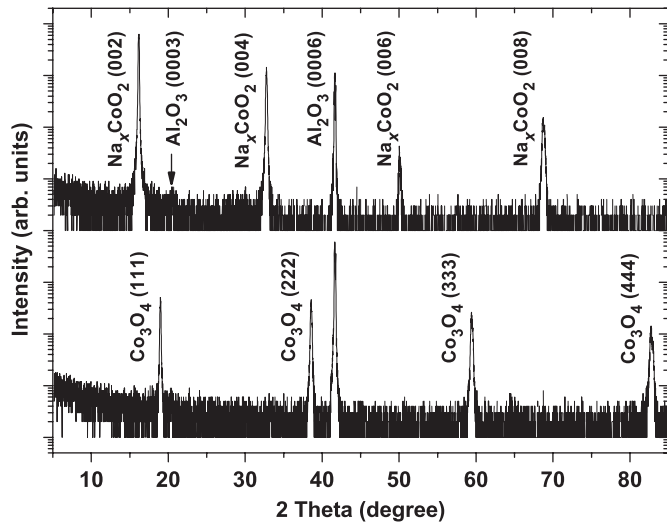


Fig. 1. X-ray diffraction of Na<sub>x</sub>CoO<sub>2</sub> (001) after sodium lateral diffusion (upper panel) and Co<sub>3</sub>O<sub>4</sub> (111) thin films on sapphire (0001) substrates (lower panel). The results indicate that the both of thin films are in single phase.

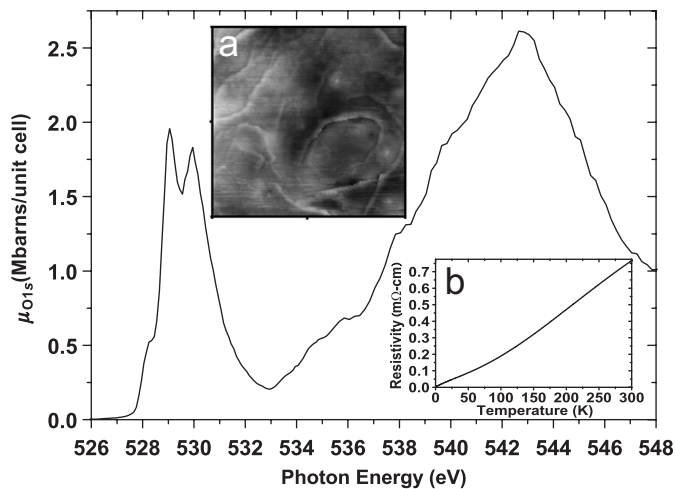


Fig. 2. Main panel: Isotropic O K-edge X-ray absorption spectrum,  $(I_{E//c} + I_{E\perp c})/2$ , of the Na<sub>x</sub>CoO<sub>2</sub> thin film at room temperature. Inset (a): AFM image ( $5\ \mu\text{m} \times 5\ \mu\text{m}$ ) of the Na<sub>x</sub>CoO<sub>2</sub> thin film shows the layered structure of this cobaltate. Inset (b): The  $\rho_{ab}$ - $T$  curve illustrates the high quality of this thin film.

The RMS roughness is about 1.67 nm. The  $\rho_{ab}$ - $T$  measurements have been done by PPMS of Quantum Design<sup>®</sup>. In Fig. 2 (b), the  $\rho_{ab}$ - $T$  curve of thin films is identical to that of single crystals [3] around  $x = 0.68$ , and with a much smaller residual resistivity,  $\rho(300\text{ K})/\rho(0.4\text{ K}) = 145$ , indicating that there are very few impurities or defects in the thin films.

Beside the electric transport measurements, X-ray absorption near edge spectroscopy (XANES) has been a powerful tool to investigate the electronic state of these compounds, as in the case of perovskite manganites [7]. The main panel of Fig. 2 illustrates the O K-edge fluorescent spectrum, which characterizes the hybridization of Co 3d and O 2p of the Na<sub>x</sub>CoO<sub>2</sub> (001) thin film at room temperature. At the pre-edge, the O K-edge XANES shows three absorption peaks of Co<sup>3+</sup> and Co<sup>4+</sup> 3d bands, in which the  $t_{2g}$  orbitals split into  $a_{1g}$  and  $e_g'$  orbitals under trigonal crystal field [8,9]. All the three pre-edge peaks resemble those of the  $x = 0.67$  single crystal with respect to the peak energy and the relative intensity [8]. These results suggest that the Na<sub>x</sub>CoO<sub>2</sub> (001) thin film has an electronic structure similar to that of the single crystal with the same Na content. Currently, the anisotropy of Na<sub>x</sub>CoO<sub>2</sub> from the orbital symmetry of  $a_{1g}$  and  $e_g'$  orbitals remains controversial. Further investigation of polarized-dependant XANES is certainly desirable.

In summary, we have successfully fabricated Na<sub>x</sub>CoO<sub>2</sub> thin films with  $x = 0.68$  via the lateral diffusion of sodium. Judged from the results of XRD and  $\rho_{ab}$ - $T$  curve, the thin films quality is as decent as that of single crystals. XANES of Na<sub>x</sub>CoO<sub>2</sub> thin film shows an electronic structure similar to that of the single crystal, too.

This work was supported by the National Science Council of Taiwan, under Grant Nos. NSC-94-2112-M-009-006, NSC-94-2112-M-009-007, and NSC-94-2112-M-009-008.

## References

- [1] K. Takada, H. Sakurai, E. Takayama-Muromachi, F. Izumi, R.A. Dilanian, T. Sasaki, Nature 422 (2003) 53.
- [2] I. Terasaki, Y. Sasago, K. Uchinokura, Phys. Rev. B 56 (1997) R12685.
- [3] M.L. Foo, Y. Wang, S. Watauchi, H.W. Zandbergen, T. He, R.J. Cava, N.P. Ong, Phys. Rev. Lett. 92 (2004) 247001.
- [4] H. Ohta, S.-W. Kim, S. Ohta, K. Koumoto, M. Hirano, H. Hosono, Cryst. Growth Design 5 (2005) 25.
- [5] A. Venimadhav, A. Soukiassian, D.A. Tenne, Q. Lia, X.X. Xi, D.G. Schlom, R. Arroyave, Z.K. Liu, H.P. Sun, X. Pan, M. Lee, N.P. Ong, Appl. Phys. Lett. 87 (2005) 172104.
- [6] Q. Huang, M.L. Foo, R.A. Pascal Jr., J.W. Lynn, B.H. Toby, T. He, H.W. Zandbergen, R.J. Cava, Phys. Rev. B 70 (2004) 184110.
- [7] W.J. Chang, J.Y. Tsai, H.-T. Jeng, J.-Y. Lin, K.Y.-J. Zhang, H.L. Liu, J.M. Lee, J.M. Chen, K.H. Wu, T.M. Uen, Y.S. Gou, J.Y. Juang, Phys. Rev. B 72 (2005) 132410.
- [8] W.B. Wu, D.J. Huang, J. Okamoto, A. Tanaka, H.-J. Lin, F.C. Chou, A. Fujimori, C.T. Chen, Phys. Rev. Lett. 94 (2005) 146402.
- [9] T. Mizokawa, L.H. Tjeng, H.-J. Lin, C.T. Chen, R. Kitawaki, I. Terasaki, S. Lambert, C. Michel, Phys. Rev. B 71 (2005) 193107.