

Fig. 23. The current of stacking and separation.

Separation micellar buffer, 180 mM SDS in 25 mM citric acid/disodium hydrogen phosphate buffer at pH 2.6; separation nonmicellar buffer, in 25 mM citric acid/disodium hydrogen phosphate buffer at pH 2.6; injection pressure, 5 p.s.i.; injection time, 7 min; stacking voltage, -20 kV; applied voltage, -30 kV; sample concentration, 50 μ g/ L, sample dissolved in water, effect length, 50 cm x 50 μ m I.D; detection wavelength, 210 nm; temperature, 25 $^{\circ}$.

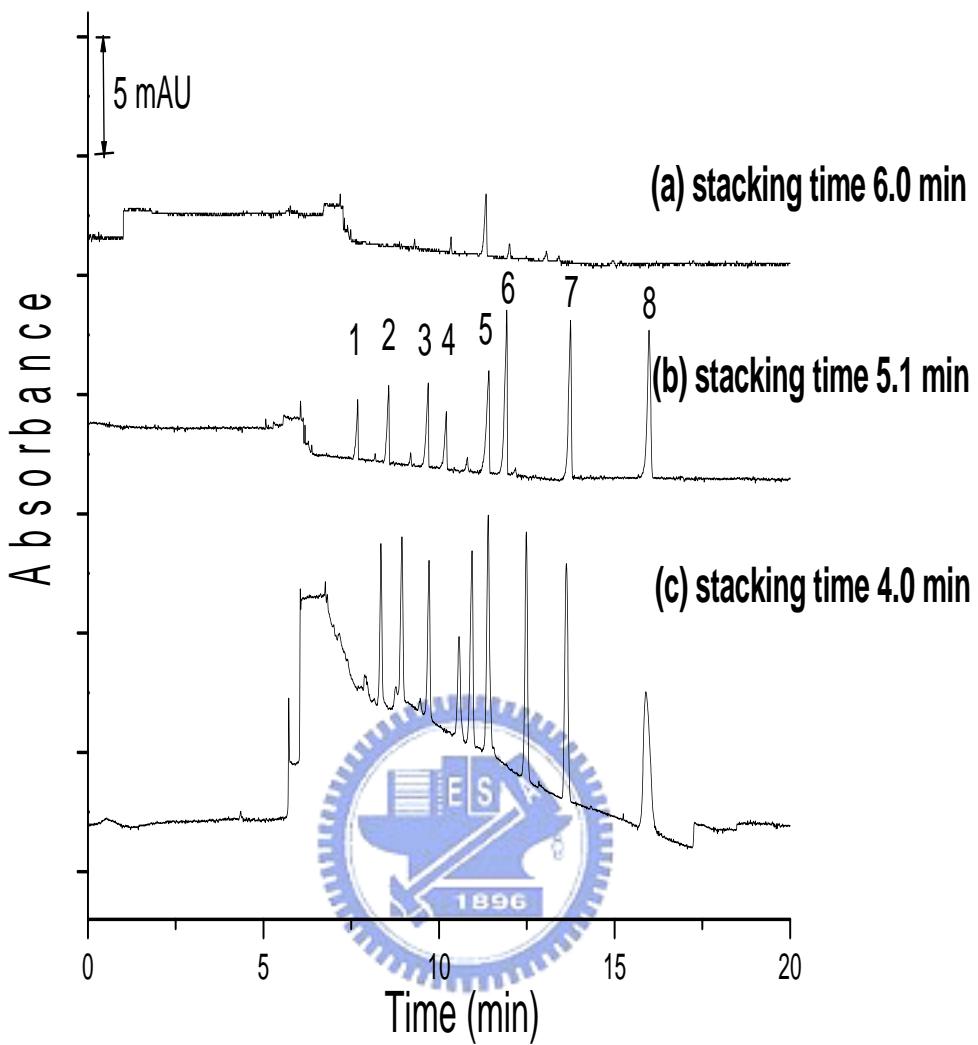


Fig. 24. Effect of the stacking time.

Separation micellar buffer, 180 mM SDS in 25 mM citric acid/disodium hydrogen phosphate buffer at pH 2.6; separation nonmicellar buffer, in 25 mM citric acid/disodium hydrogen phosphate buffer at 25 mM citric acid/disodium hydrogen phosphate buffer pH 2.6; injection pressure, 5 p.s.i.; injection time, 7 min; stacking voltage, -20 kV; applied voltage, -30 kV; sample concentration, 50 µg/ L, sample dissolved in water, effect length, 50 cm x 50 µm I.D; detection wavelength, 210 nm; temperature, 25 .

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|--------------------|-----------------------|-----------------|
| 1.sinapic acid | 2. ferulic acid | 3.coumaric acid |
| 4.chlorogenic acid | 5.caffeic acid | 6.syringic acid |
| 7.vanillic acid | 8.hydroxybenzoic acid | |

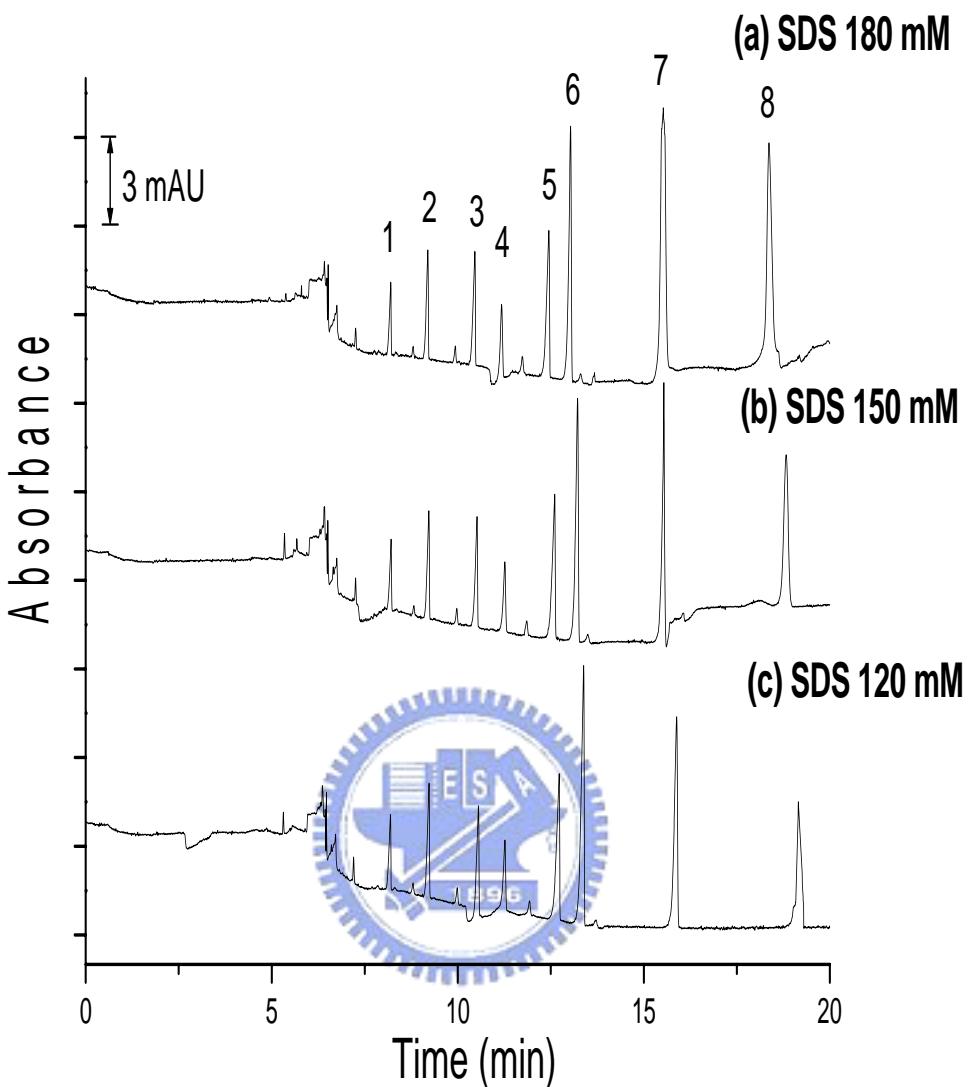


Fig. 25. Effect of the concentration of SDS in running buffer.

Separation micellar buffer, 25 mM citric acid/disodium hydrogen phosphate buffer at pH 2.6; separation nonmicellar buffer, 25 mM citric acid/disodium hydrogen phosphate buffer at pH 2.6; injection pressure, 5 p.s.i.; injection time, 7 min; stacking time, 5.1 min; stacking voltage, -20 kV; applied voltage, -30 kV; sample concentration, 50 µg/ L, sample dissolved in water, effect length, 50 cm x 50 µm I.D; detection wavelength, 210 nm; temperature, 25 .

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| 1.sinapic acid | 2. ferulic acid | 3.coumaric acid |
| 4.chlorogenic acid | 5.caffeic acid | 6.syringic acid |
| 7.vanillic acid | 8.hydroxybenzoic acid | |

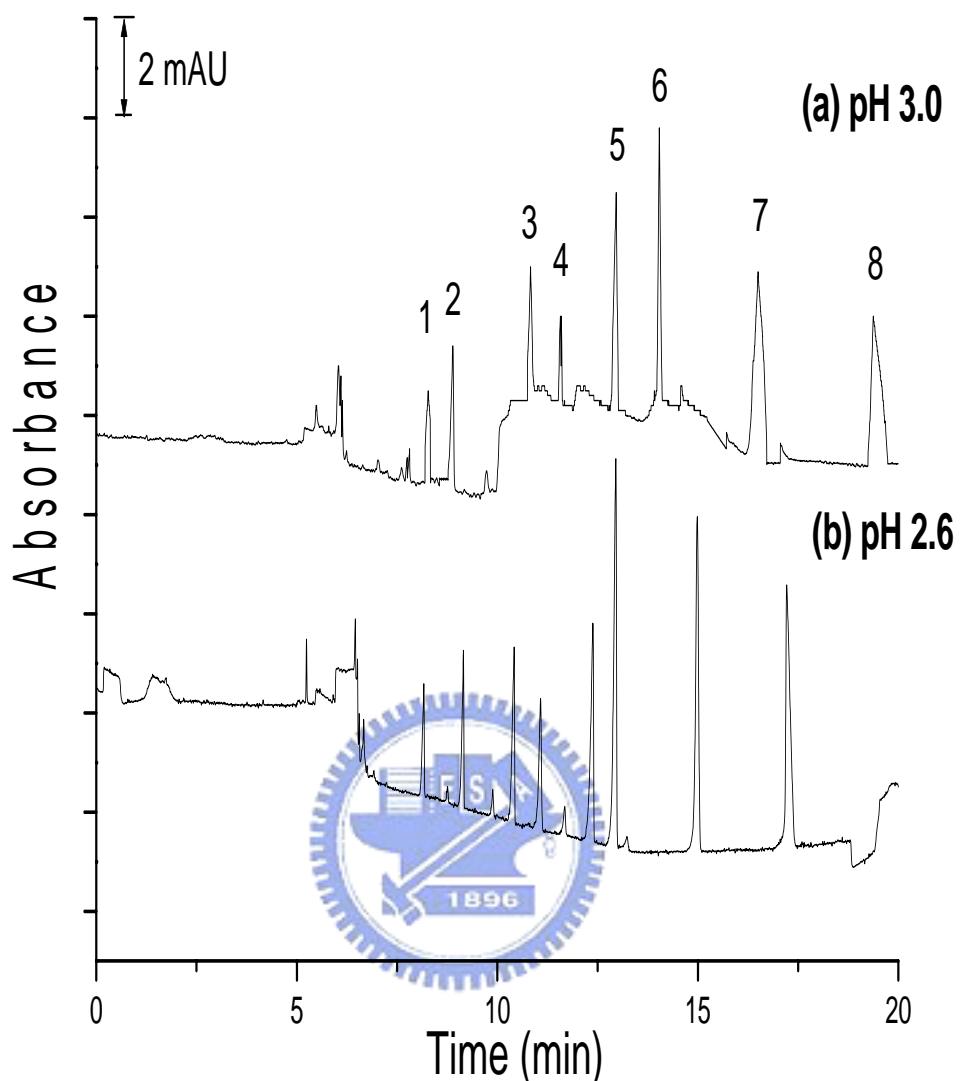


Fig. 26. Effect of the pH of micellar buffer and nonmicellar buffer. Separation micellar buffer, 180 mM SDS in 25 mM citric acid/disodium hydrogen phosphate buffer; separation nonmicellar buffer, 25 mM citric acid/disodium hydrogen phosphate buffer; injection pressure, 5 p.s.i.; injection time, 7 min; stacking time, 5.1 min; stacking voltage, -20 kV; applied voltage, -30 kV; sample concentration, 50 µg/ L, sample dissolved in water, effect length, 50 cm x 50 µm I.D; detection wavelength, 210 nm; temperature, 25 .

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| 1.sinapic acid | 2. ferulic acid | 3.coumaric acid |
| 4.chlorogenic acid | 5.caffeic acid | 6.syringic acid |
| 7.vanillic acid | 8.hydroxybenzoic acid | |

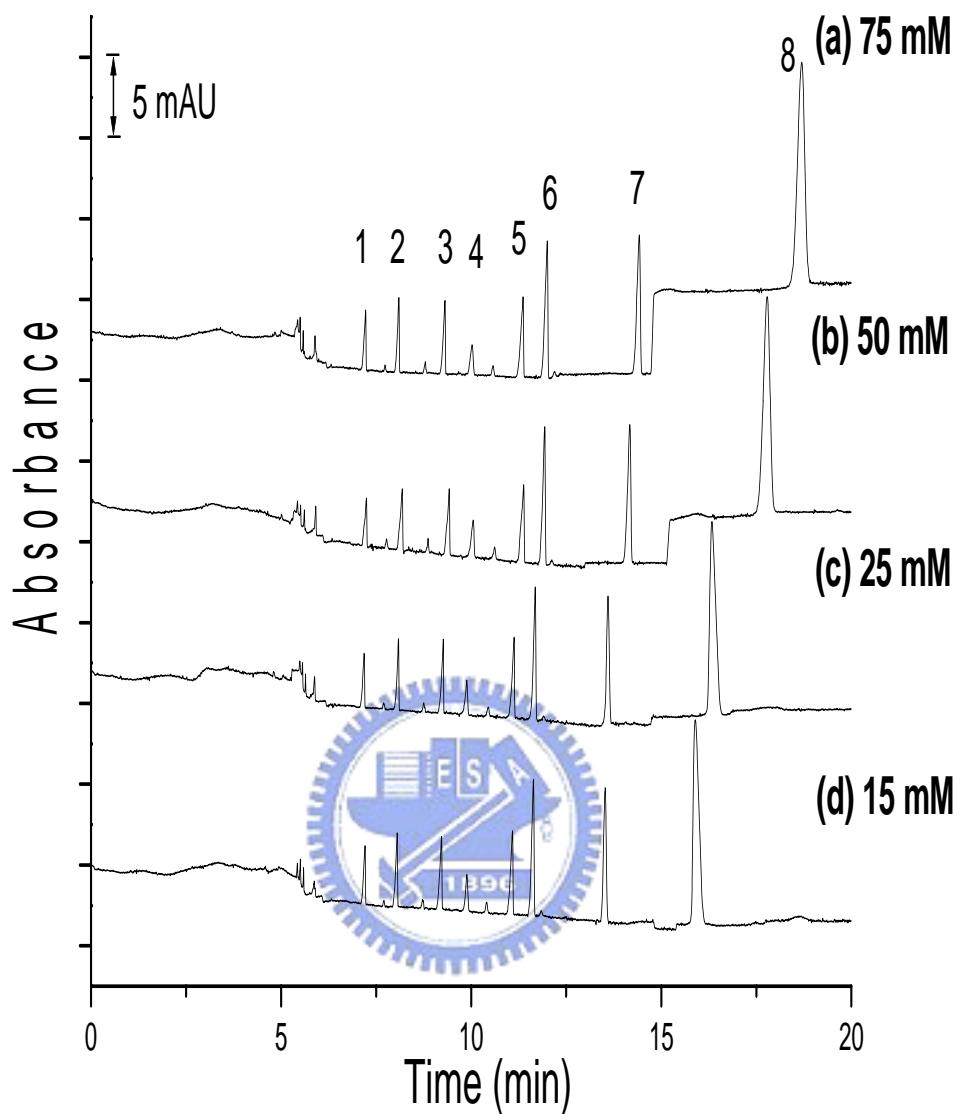


Fig. 27. Effect of the concentration of micellar buffer.

Separation micellar buffer, 180 mM SDS at pH 2.6; separation nonmicellar buffer, 25 mM citric acid/disodium hydrogen phosphate at pH 2.6; injection pressure, 5 p.s.i.; injection time, 7 min; stacking time, 5.1 min; stacking voltage, -20 kV; applied voltage, -30 kV; sample concentration, 50 µg/L, sample dissolved in water, effect length, 50 cm x 50 µm I.D; detection wavelength, 210 nm; temperature, 25 .

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| 1.sinapic acid | 2. ferulic acid | 3.coumaric acid |
| 4.chlorogenic acid | 5.caffeic acid | 6.syringic acid |
| 7.vanillic acid | 8.hydroxybenzoic acid | |

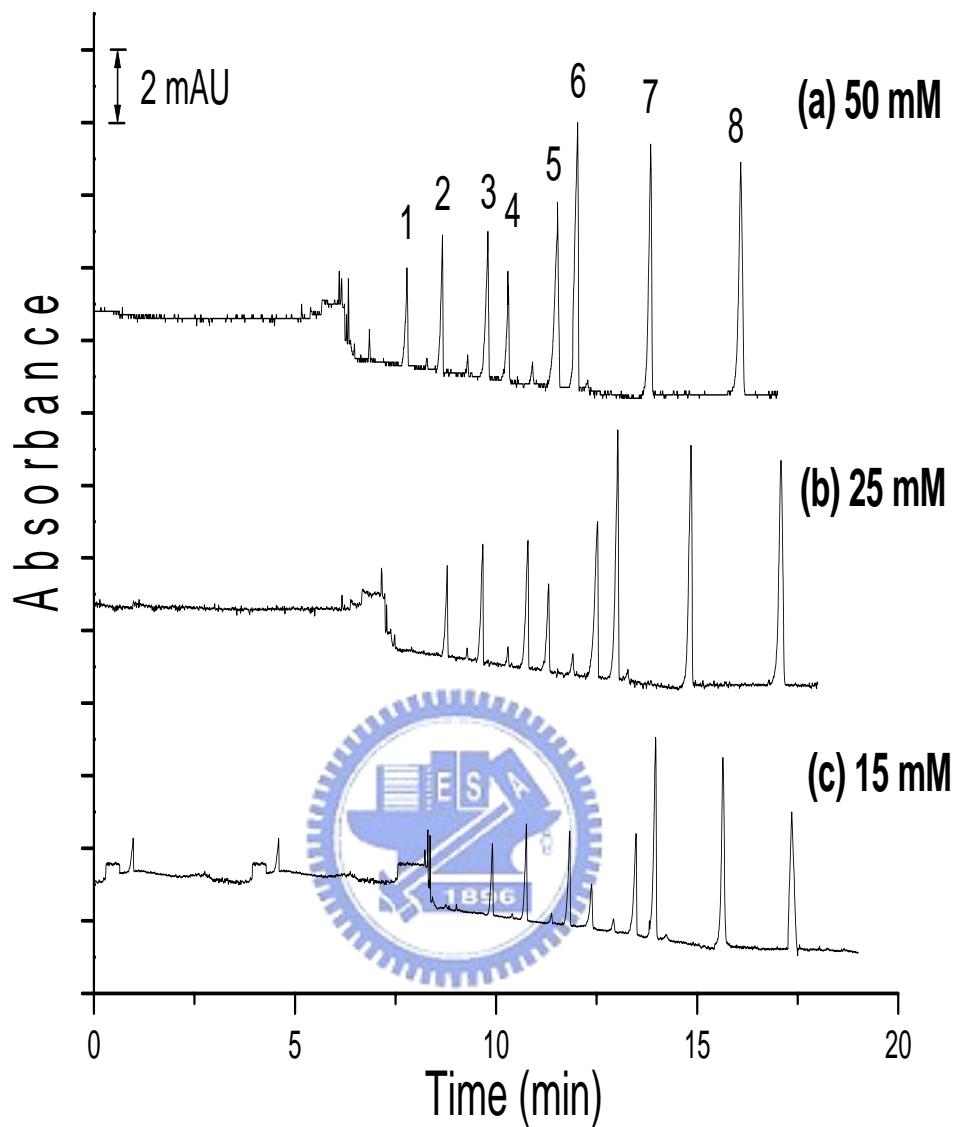


Fig. 28. Effect of the concentration of nonmicellar buffer.

Separation micellar buffer, 180 mM SDS in 15 mM citric acid/disodium hydrogen phosphate buffer at pH 2.6; separation nonmicellar buffer, at pH 2.6; injection pressure, 5 p.s.i.; injection time, 7 min; stacking time, 5.1 min; stacking voltage, -20 kV; applied voltage, -30 kV; sample concentration, 50 µg/L, sample dissolved in water, effect length, 50 cm x 50 µm I.D; detection wavelength, 210 nm; temperature, 25 °C.

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| 1.sinapic acid | 2. ferulic acid | 3.coumaric acid |
| 4.chlorogenic acid | 5.caffeic acid | 6.syringic acid |
| 7.vanillic acid | 8.hydroxybenzoic acid | |