

國立交通大學

電子工程學系 電子工程所碩士班

碩士論文

高介電常數材料(HfAlO)沉積前後表面處理之研究



Investigation of Surface Treatment Before and After
High-k(HfAlO) Dielectric Deposition

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中華民國九十六年七月

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Advisor：Jen-Chung Lou



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近年來很多研究致力於發展可替代二氧化矽介電層材料，用高介電常數取代二氧化矽的方法使得半導體的元件能夠持續發展下去，並且克服物理上縮小的限制，在高介電常數的材料中，二氧化鈣是較有潛力的高介電常數材料，它有較高的介電係數跟載子能障，且與矽反應所需能量較高，如果我們在沉積二氧化鈣時，參雜一些三氧化二鋁可以提高結晶溫度，增加熱穩定性，但由於鋁堆積在介電層與矽之間的介面，依然有固定負氧化物電荷的問題，使得遷移率下降。因此在介電層與矽基板之間的介面層就變

得很重要，它的品質會影響到元件的特性還有可靠度等。所以在本論文中，我們比較用二氧化氮和紫外光加臭氧表面處理形成的一層薄膜類似二氧化矽，看哪種表面處理對元件的特性有較好的影響。

在另一方面，經過表面處理可能會改變介電層的主要鍵結，減少在金屬與介電層介面或介電層中的缺陷，我們用 NH_3 電漿與紫外光加臭氧兩種表面處理伴隨之後適當快速熱退火溫度來改變介電層中的主要鍵結，看是否對電性能有正面的影響。從磁滯效應，漏電流，電容，可靠度，等效二氧化矽厚度等特性來看，紫外光加臭氧的前處理以及 NH_3 電漿伴隨 800°C 的後處理能夠達到較好的電性。




Investigation of Surface Treatment Before and After High- κ (HfAlO) Dielectric Deposition

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Abstract

The logo of National Chiao Tung University is a circular emblem with a gear-like border. Inside the circle, there is a stylized building and the letters 'ES' and 'A'. Below the building, the year '1896' is inscribed.

In recent years, a lot of research effort has been focused on developing alternative gate dielectric materials to replace SiO_2 . Using high dielectric constant provide the way to make the semiconductor roadmap go on, and overcome the present physical limits of semiconductor scaling down. Among High- κ materials, HfO_2 can be used due to its relatively high dielectric constant, its relatively high free energy of reaction with Si, and its relatively high band gap, and if we add Al during deposition process to form HfAlO can increase crystallization temperature, and have better thermal stability than HfO_2 , but HfAlO also introduces negative fixed oxide charges due to Al accumulation at the HfAlO-Si interface, resulting in mobility degradation, therefore the control of SiO_2 -like interface between high- κ dielectrics and silicon substrate pays more and more important, since the device performances and reliability characteristics are strongly affected by the interface quality. In this thesis we compare Nitridation of the Si surface using N_2O and Oxidation of the Si surface using UV ozone prior to the deposition of high- κ gate dielectrics to see

which is good for the electrical properties of devices. In the other hand the interface treatment can change the texture bonding, and reduce defects such as dangling bonds at the interface between metal gate and dielectric as well as incorporation of extra impurities. The changes of interface and incorporation will upset a balance within the primary interface, we use UV ozone and NH_3 . two treatment accompany with appropriate RTA temperature to change the texture bonding and see which improve the electrical properties effectively. In this thesis we find a suitable surface treatment before and after HfAlO deposition and appropriate annealing temperature to improve electrical characteristics. From the hysteresis , leakage current density, capacitance, reliability, EOT, etc. the UV ozone pre-treatment and NH_3 post treatment with 800°C RTA has excellent electrical characteristics.



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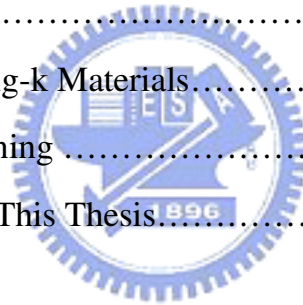
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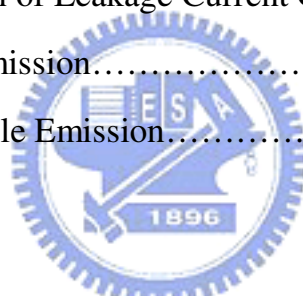
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Chapter 1

Introduction

1.1 General Background

The novel metal-oxide-semiconductor (MOS) technology has kept developing alone the moore's law "the device will scale down every eighteen month" By the improvement of semi-conductor technology, in order to lower costs, the devices in the integrated circuit must be increased, therefore the size of devices must be scaled down. In respect of dielectric, for maintaining the fixed power density, when the length of channel decreased, the thickness of dielectric also must be decreased, especially when the thickness of oxide dielectric is lower than 2nm. The viability of SiO₂ will face severe challenge such as increasing gate leakage has become a serious issue because of tunneling effect. Even if thermal oxide have several advantages , for example high quality interface, good thermal stability, being amorphous phase throughout the integration process and can have low leakage current, but the thickness of SiO₂ will reach the predicted limits of scaling, therefore high-k dielectric are possible replaced for SiO₂ because of its high dielectric constant.

In recent years, a lot of research effort has been focused on developing alternative gate dielectric materials to replace SiO₂. Using high dielectric constant provide the way to make the semiconductor roadmap go on, and overcome the present physical limits of semiconductor scaling down. Therefore the applications of high dielectric constant materials become a trend in semiconductor technology below the 100nm

dimensions. [1]~[4]

1.2 Motivation

With the improvement of the process technology, when the devices scale down, how to suppress the leakage current effectively is our target. The shrinkage of channel length is the most effective way to promote the driving current of a transistor. Another way to improve the I_{ds} is increasing the C_{ox} (eq-1) by scaling down oxide thickness (eq-2), therefore finding High-k materials, and making them have thicker true thickness but have thinner EOT (Equivalent Oxide Thickness), is one of our researching direction.


$$I_{ds} = 1/2 C_{ox} \mu_n (W/L) (V_{GS} - V_t) (V_{DS}) \quad [\text{eq-1-1}]$$

Where μ_n is the effective mobility, C_{ox} is the gate capacitance, W is the channel width, L is channel length. Obviously, drive current is inversely proportional to the channel length L . Therefore, the shrinkage of channel length is the most effective way to promote the driving current of a transistor.

$$C_{ox} = \frac{k_{ox} \epsilon_0}{t_{ox}} A \quad [\text{eq-1-2}]$$

Generally there are several point we should care about high-k materials (1) low

leakage current. (2) low interface trap density. (3) low bulk trap. (4) good thermal stability. (5) good reliability. But now the subject matter of high-k materials is the high interface trap density lead to high leakage current and not available to work on devices. Therefore, this thesis is researching different surface treatment after depositing high-k materials to lower interface trap and abate Fermi level pinning effect. We expect that it is useful to increase high-k materials feasibility and will replace SiO₂ in the future.

1.3 The Choice of High-K Materials

There are many factors we should think about when we choose high-k materials. First, the most of high-k materials don't have wide enough bandgap, therefore it is easy for both electron and hole through the dielectric, and lead to large leakage current. The bandgap of high-k materials are present in Fig1-1.

Second, materials with slightly high-k value than thermal oxide don't have low leakage enough to conform to our criteria, and large k value materials have poor thermal stability generally, and will suffer from larger fringing field and degrade short channel characteristics [5], the fringing effect will make extra field lines terminate in the S/D region and share the charge controllability with the gate. This will lower the potential in the channel for electron, and influence the gate control ability. Third, most of high-k would react with the Si, there are two possible reaction, one is [eq1], after the high temperature process, the metal in high-k materials would separate out, and the oxygen would react with the Si to form an interfacial layer between silicon substrate and high-k.



The other is the high-k materials react with Si to form silicide [eq4], the formation of the silicide would provide the transport path to carrier, and induce large leakage current.



The interfacial layers would lower dielectric constant, therefore the dimension size will hard to scale down.

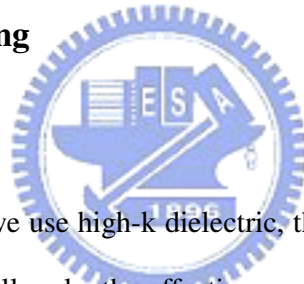
Fourth, the interface quality and morphology between high-k and Si substrate should be considered. The quality would affect the carrier mobility in the channel, and make mobility degradation, lower the device performance. Another aspect, the film morphology is also important, during the all process, the device has to go through many high temperature thermal treatment, including annealing, dopant activation annealing, the thermal budget will affect the film morphology, high temperature process would make high-k dielectric change phase, from the amorphous phase to the polycrystalline phase, lead to large grain boundaries which supply leakage path and induce large leakage current.

Considering all of above, we should choose the appropriate high-k material with more advantage and less drawback to device. Among High-k materials, HfO₂ can be used due to its relatively high dielectric constant(~25), its relatively high free energy of reaction with Si, and its relatively high band gap(5.68ev), the properties comparison of SiO₂ and HfO₂ is showed in table 1-1. If we add Al during deposition process to form HfAlO can increase crystallization temperature, and the concentration of Al is higher, the crystallization temperature is higher too . If the crystallization

temperature is low, grain boundaries in crystallized gate dielectric can be the fast paths for oxygen or dopants diffusion into gate dielectric and even into FET channel region in silicon substrate causing uncontrolled interfacial layer growth at the high-K/Si interface, threshold voltage instability and defect generation. Adding Al during deposition process can also prevent the formation of Hf silicide between dielectric and Si substrate.

Therefore, HfAlO can have better thermal stability than HfO₂, but HfAlO also introduces negative fixed oxide charges due to Al accumulation at the HfAlO-Si interface, resulting in mobility degradation.

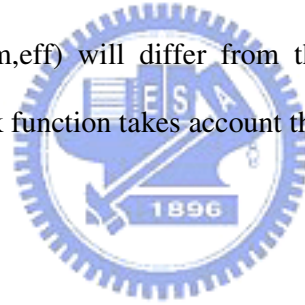
1.4 Fermi Level Pinning



There is a problem when we use high-k dielectric, the phenomenon is called Fermi level pinning (FLP) and it will make the effective work function of gate materials be controlled difficultly. It means that $\Phi_{m,eff}$ differs from $\Phi_{m,vac}$, therefore the flatband voltage is almost independent upon metal gates, FLP is a consequence of interfacial reaction or interfacial charge exchange between gate electrode and gate dielectric, the reasons of Fermi-level pinning is intrinsic charge exchange.

The MIGS (metal induced gap states) and VIGS (virtue gap states) theory are used to explain the intrinsic charge exchange, The origin of the VIGS is from the dangling bonds of uncoordinated surface atoms. These dangling bonds produce surface states which continuously disperse in the energy gap of dielectrics. The surface states are confined at the surface to exponentially decay into vacuum and into solid . MIGS assumes that the exponentially decaying bulk states are the main component of the

surface states of the dielectric and that the states of the metal in contact with the dielectric couple to its surface states exchanging in order to maintain local charge neutrality. Free electrons in the metal penetrate into the dielectric at the interface and cause the interface charges. The amount of transferred charges is dependent upon the charge neutrality level (ECNL) and metal Fermi level (EFm). ECNL is a branch point of surface energy level, which is usually near the mid-gap of dielectric energy band. No charge will be transferred across the interface when the EFm is coincided with ECNL. With the surface level above or below ECNL, the net charge of the surface is negative and positive on the dielectric side, respectively. In the case where EFm is above ECNL, the dipole layer created at the interface will be negative on the dielectric side. This dipoles will tend to drive the EFm close to ECNL, and hence the effective work function ($\Phi_{m,eff}$) will differ from the work function in vacuum ($\Phi_{m,vac}$). The effective work function takes account the effects of interfacial dipoles and is determined by



$$\Phi_{m,eff} = \Phi_{CNL} + S(\Phi_{m,vac} - \Phi_{CNL}) \quad [eq-1-4]$$

Where S is the dielectric pinning strength, it is the function of the surface density of states N, the surface states penetration depth δ , and the permittivity of the dielectric. (eq 1-5). And it also can be estimated from the empirical (eq 1-6)

$$S = (1 + e^2 N \delta / \epsilon_s) \quad [eq-1-5]$$

$$S = [1 + 0.1 (\epsilon_\infty - 1)^2]^{-1} \quad [eq-1-6]$$

where ϵ_{∞} [6] is the electronic component of the dielectric constant.

Figure 1-2 shows two critical conditions, Fermi Level Free $S=1$, and Fermi Level Fixed $S=0$. A high-k dielectric will lower the slope parameter S and highly pin the $E_{m,eff}$ in the ECNL since the electronic component of the dielectric constant is high. This indicates that high-k materials will suffer from serious Fermi-level pinning effect, and induce effective work function close to Φ_{CNL} . To obtain suitable effective work function, metal gates on high-k materials for pMOSFETs (nMOSFETs) should have higher (lower) work function to compensate the effects of Fermi-level pinning. Another theory is that Fermi level pinning in high work function materials is governed by the O vacancy generation and subsequent formation of interface dipole near gate electrodes due to the electron transfer, and it in low work function is governed by O interstitial formation, therefore the work function pinning free region appears due the difference in the mechanism of Fermi level pinning of high and low work function materials. And the type of metal silicide is also affect the pinning position, n-like metal silicides are usually located between the pinning position of n^+ and p^+ poly-Si gate, and those of p-like metal silicides are located near the pinning position of p^+ -poly-Si gates, the pinning position is shown in Figure1-3

1.5 Organization of This Thesis

.In this thesis have four chapters in this dissertation. In chapter 1, we make an introduction to describe the background of the Semiconductor technology and

discuss the possible issues that we may meet during the dimensions scaling down. In addition , we would talk about the hopeful solutions to overcome the physical limits. The possible solution is the alternative high dielectric constant materials. They would replace the traditional SiO_2 material in the gate materials. In chapter 2 , we will introduce the properties of the high dielectric constant material hafnium oxide (HfAlO), and describe the experimental procedures, electrical measurements, material analysis. In chapter 3 , we will investigate the electrical properties of HfAlO thin film including two surface treatment UV ozone treatment and N_2O treatment. before HfAlO deposition and two interface treatment UV ozone treatment and NH_3 treatment accompany with RTA treatment after HfAlO deposition We will discuss the experimental results. At the end of this thesis, conclusions are given in chapter 4.



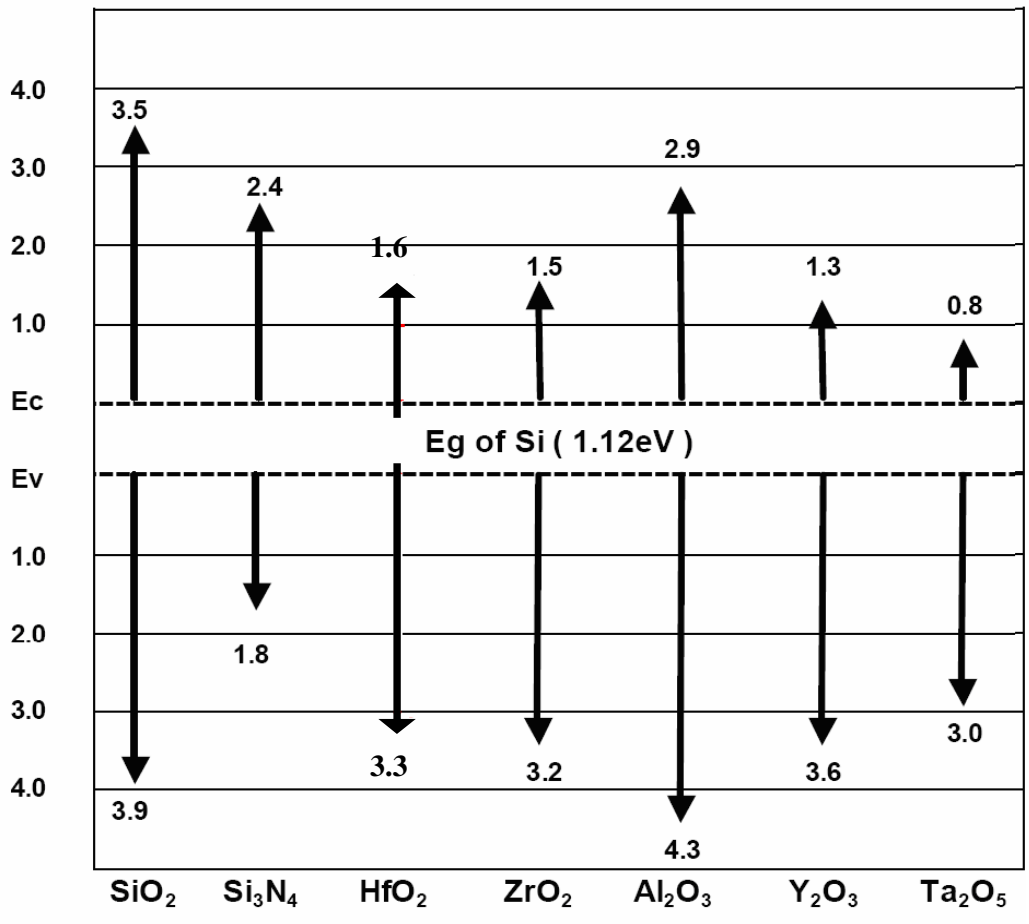


Fig 1-1 The bandgap of high-k materials

Physical parameters	SiO ₂	HfO ₂
Dielectric constant	3.9	21-25
High-k/Si barrier height	3.1-3.5	1.5
Energy band gap	9	6
Breakdown field(MV/cm)	10-15	3-6.7
Structure Type	Amorphous	Polycrystal (400~900°C)
Interface state density, Dit(cm ⁻² ev ⁻¹)	<10 ¹⁰	8X10 ¹⁰
J(A/cm ²) at 1V for EOT=1nm	>1	<1X10 ⁻⁴
Contact stability with Si(kcal/mol)	Stable	47.6

Table 1-1 properties comparison of SiO₂ and HfO₂

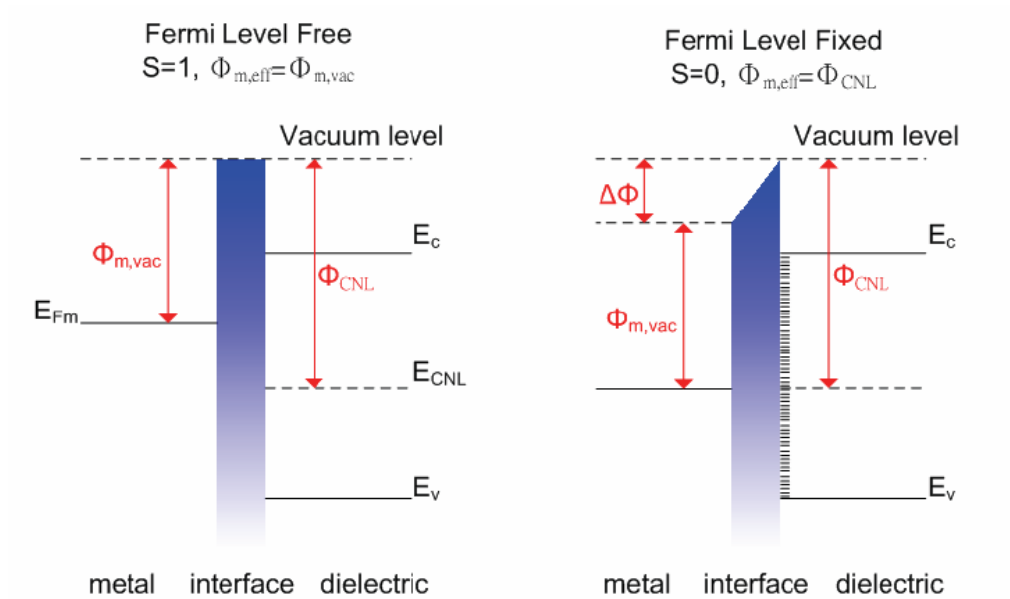


Figure 1-2 The critical Fermi level of metal gate on dielectric. the effective work function of metal on dielectric is the same as in vacuum while the interface is perfect. The work function is fixed to the dielectric-charge-neutral level while the interface-state density is high. the interface states are caused by both intrinsic states (MIGS, ViGS) and extrinsic states (defect levels).

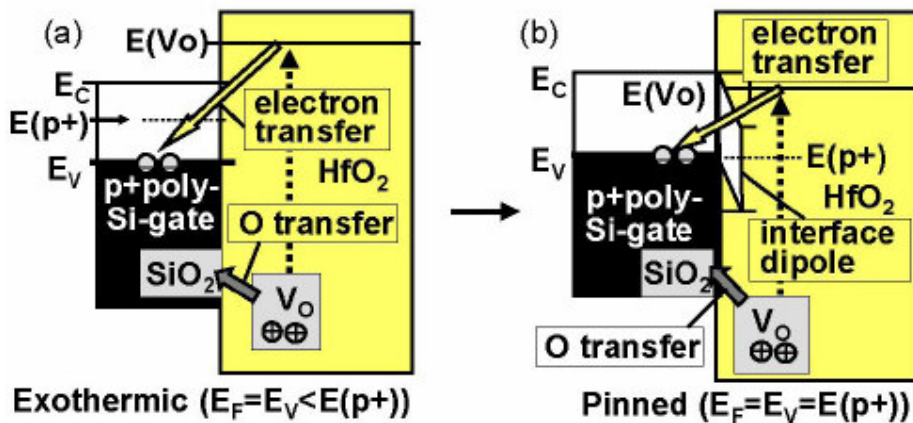
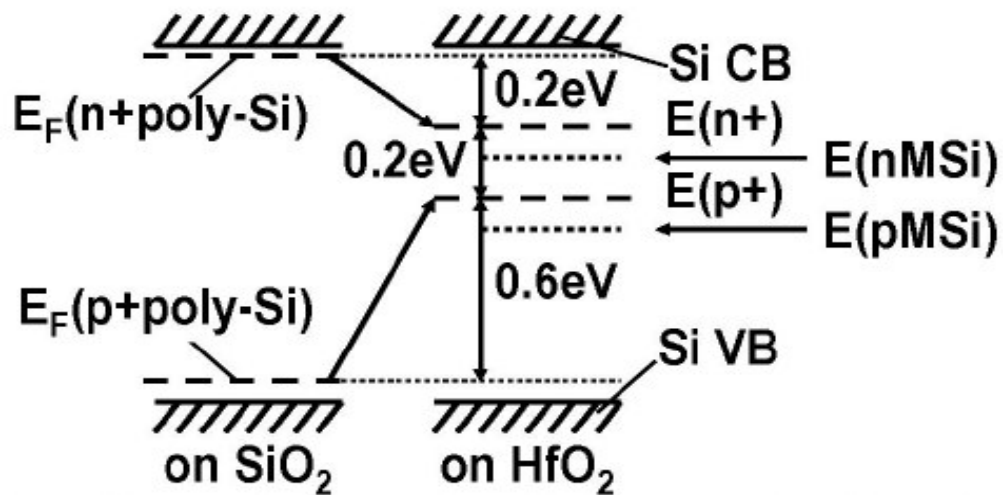


Fig 1-3 Schematic illustration of the Fermi level pinning observed in poly-Si and n-like, p-like MSi gates

Chapter 2

Experimental Procedures and measurements

2-1 Properties of Hafnium Aluminum Oxide

HfAlO film was deposited by electron-beam evaporation method. The background pressure of the main chamber was 10^{-6} – 10^{-7} Pa, and the substrate temperature was kept at 300 °C. High purity sintered HfO₂ and Al₂O₃ pellets were coevaporated at the rate of 0.2–0.5 Å/ s, We use Hafnium aluminum oxide (HfAlO) to replace Hafnium oxide (HfO₂), because of the some poor characteristics of HfO₂. [7] When the HfO₂ is thin, it exhibit poor thermal stability and tend to crystallize at low temperature of 400-500°C, this will cause significant increases in leakage current, therefore it can increase crystallization temperature by the incorporation of Al₂O₃ forming the HfAlO alloy [8]-[12]. The comparison of leakage current in different annealing temperature of this two materials is showed in Fig 2-1. we can observed the leakage current of HfO₂ at higher temperature is larger than HfAlO at same RTA temperature.

2-2 Experiment Process of PMOSFET with Treatment Before HfAlO Deposition

MOS transistors was fabricated on 4-inch p-type Si with (1 0 0) orientation. After removing the native oxide, RCA clean was performed with HF-dip last, followed by UV ozone and N₂O treatment. After surface treatments sample with a high temperature 900°C 60 seconds was used for density interface oxide. After

surface treatment HfAlO was deposited at room temperature and $7.6E-3$ torr by ion sputter system. Followed by a high temperature 900°C post deposition annealing (PDA) in the nitrogen ambient for 30 seconds in order to improve the film quality. A 200nm undoped polycrystalline silicon (poly-Si) layer was directly deposited by low pressure chemical vapor deposition (LPCVD) on top of HfAlO films After the gate electrode was patterned by lithography and etching processing. Then implant As 20 KeV dose $5E15 \text{ \#/cm}^2$ after that activation was done at 950°C by rapid thermal annealing (RTA) for 30seconds in the nitrogen ambient. After a 500nm SiO_2 passivation was done by PECVD. Define the contact hole by the lithography and etching passivation SiO_2 and high k layer in order to silicon contact whit metal. Al was deposited by thermal evaporation system. Metal pad is defined by lithography and etching. Backside contact was formed by thermal evaporation. Fig 2-2 , Fig2-3 show the cross section and the process flow of pMOS transistor.

2-3 Experiment Process of Capacitance with Treatment Before and After HfAlO Deposition

Wafers were cleaned by standard RCA processes with HF-last for the removal of the particles and native oxides. and UV ozone was used to grow an ultra-thin ozone oxide about $9\sim 11\text{\AA}$, after UV ozone surface treatments, a high temperature 900°C 60 seconds was used for densify ozone oxide. HfAlO was then deposited at room temperature and $7.6E-3$ torr by ion sputter system, followed by UV-ozone or plasma NH_3 treatment and then a different high temperature (as deposition, 800°C , 900°C , 1000°C) post deposition annealing (PDA) in the nitrogen ambient for 60 seconds were investigated, before Aluminum metal gate deposition which was created by a thermal evaporation system. After gate electrodes patterned and

backside contact was formed by thermal evaporation. The process flow and structure are shown in Fig 2-3, the area of capacitors is $4.5216 \times 10^{-4} \text{ cm}^2$

2-4 Interface Treatments Before HfAlO Deposition

The control of SiO₂-like interface between high- κ dielectrics and silicon substrate pays more and more important, since the device performances and reliability characteristics are strongly affected by the interface quality. Nitridation of the Si surface using N₂O prior to the deposition of high- κ gate dielectrics has been shown to be effective in achieving the low EOT (equivalent oxide thickness) and preventing boron penetration [13], [14]. However this technique results in higher interface charges [15], which leads to higher hysteresis and reduced channel mobility. Ozone-formed oxide (ozone oxide) has superior characteristics. Even when the formation temperature is less than 400°C, ozone oxide has a high film density comparable to that of the device-grade oxide film formed at higher temperature, a low interface trap density, and a much thinner structural transition layer near the SiO₂/Si interface [16]. The ozone surface treatment was employed to improve the interface quality between HfAlO and silicon substrate.

2-5 Interface Treatments After HfAlO Deposition

The interface treatment can be performed with the surface treatment before metal deposition and it can be plasma treatment or reactive gas annealing. The interface treatment can change the texture bonding, and reduce defects such as dangling bonds at the interface between metal gate and dielectric as well as incorporation of extra impurities. The changes of interface and incorporation will upset a balance within the primary interface. According the past research the direct contact of high-k materials

and Si-substrate will have many issues. We need high quality interface of dielectric/Si or metal/dielectric with low density of interface states ($D_{it} \sim 2 \times 10^{10}$ states/cm²) arising from unsaturated surface bonds and other electrically active imperfections. Interface states will lead to low on-current because of carrier mobility is limited by scattering at the interface with the vertical electric fields present in the channel. Therefore in this thesis we try two surface treatment including ozone plasma and NH₃ plasma after HfAlO deposition to make the higher quality interface between metal gate and high-k dielectric, or change the texture bonding of HfAlO.

2-6 Electrical Measurement

The Capacitance-Voltage (C-V) and Current-Voltage (I-V) characteristics were measured by Hp-4284 and Hp-4156. The dielectric constant (k-value) is then calculated from the measured capacitance at accumulation mode. The equivalent oxide thickness (EOT) was extracted by fitting the measured high-frequency (100 kHz) capacitance-voltage (C-V) data under accumulation condition. UCLA CVC simulation program was utilized to obtain the accurate flat band voltage (V_{FB}). The C-V hysteresis phenomenon was measured by sweeping the gate voltage from accumulation to inversion then back. The tunneling leakage current density-electric field ($J-E$) and the reliability characteristics of MOS capacitors were measured by semiconductor parameter analyzer HP 4156C.

2-7 Reliability Measurement

We observe the reliability from stress-induced leakage current (SILC), measurements. SILC were measured at room temperature, gate bias=-4 for RTA after

surface treatment and -3V for RTA before surface treatment, the J-E curves were measured at 10 sec 20sec 50sec 100sec 200sec 1000sec respectively during stress.



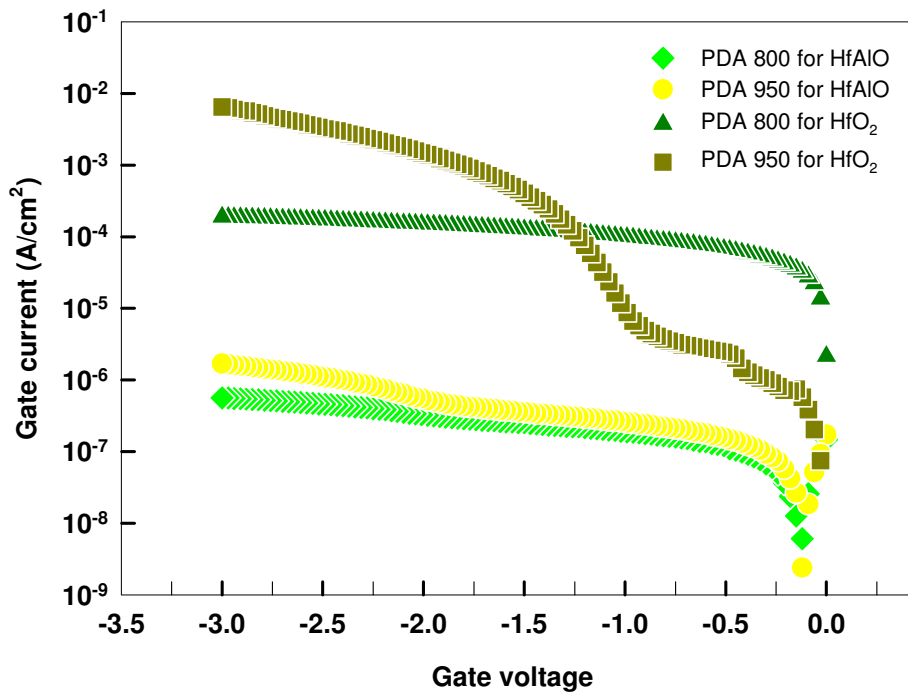
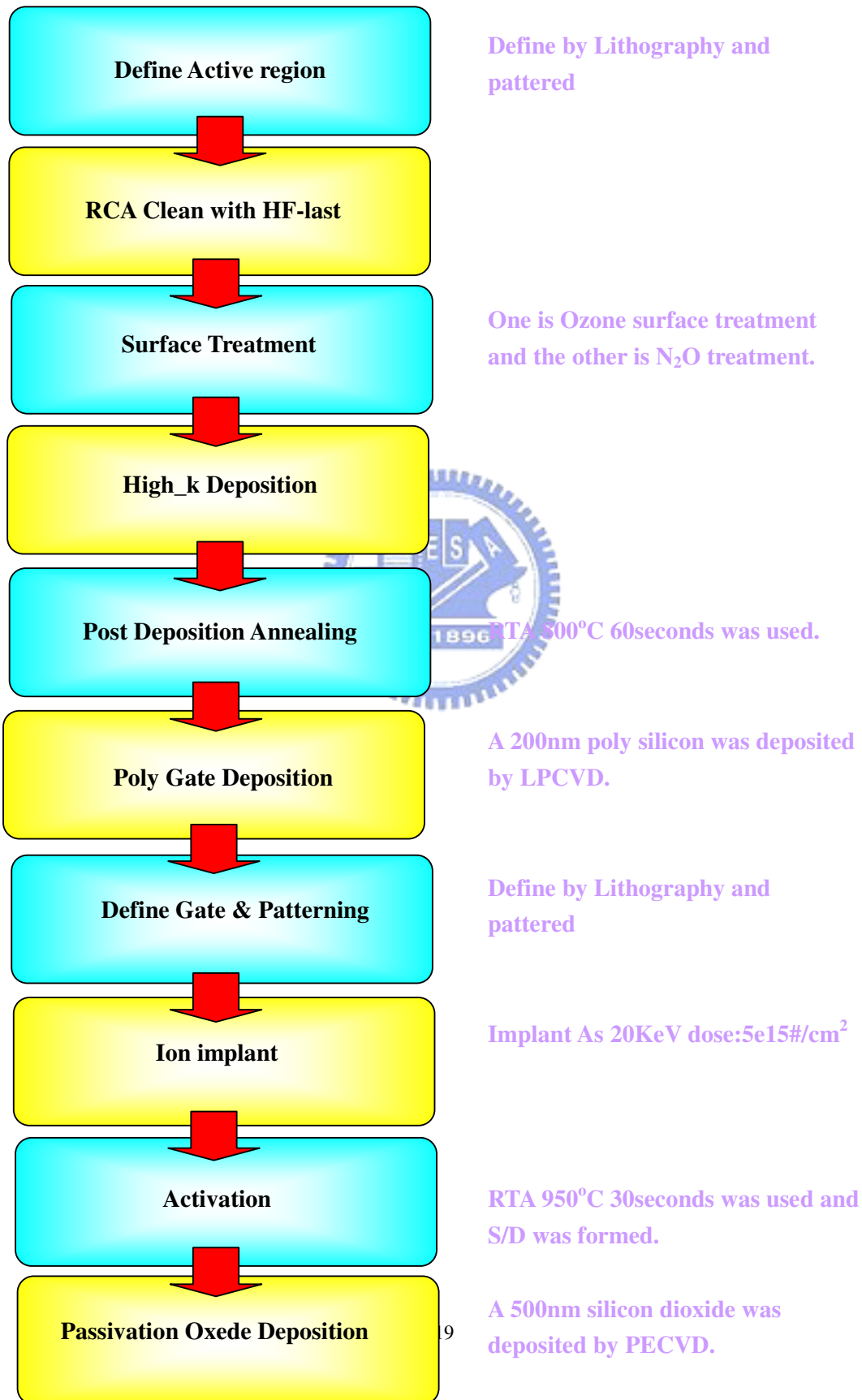


Fig 2-1 The leakage current of ALD 30nm of HfO₂ and HfAlO after rapid thermal annealing of 800°C and 950°C



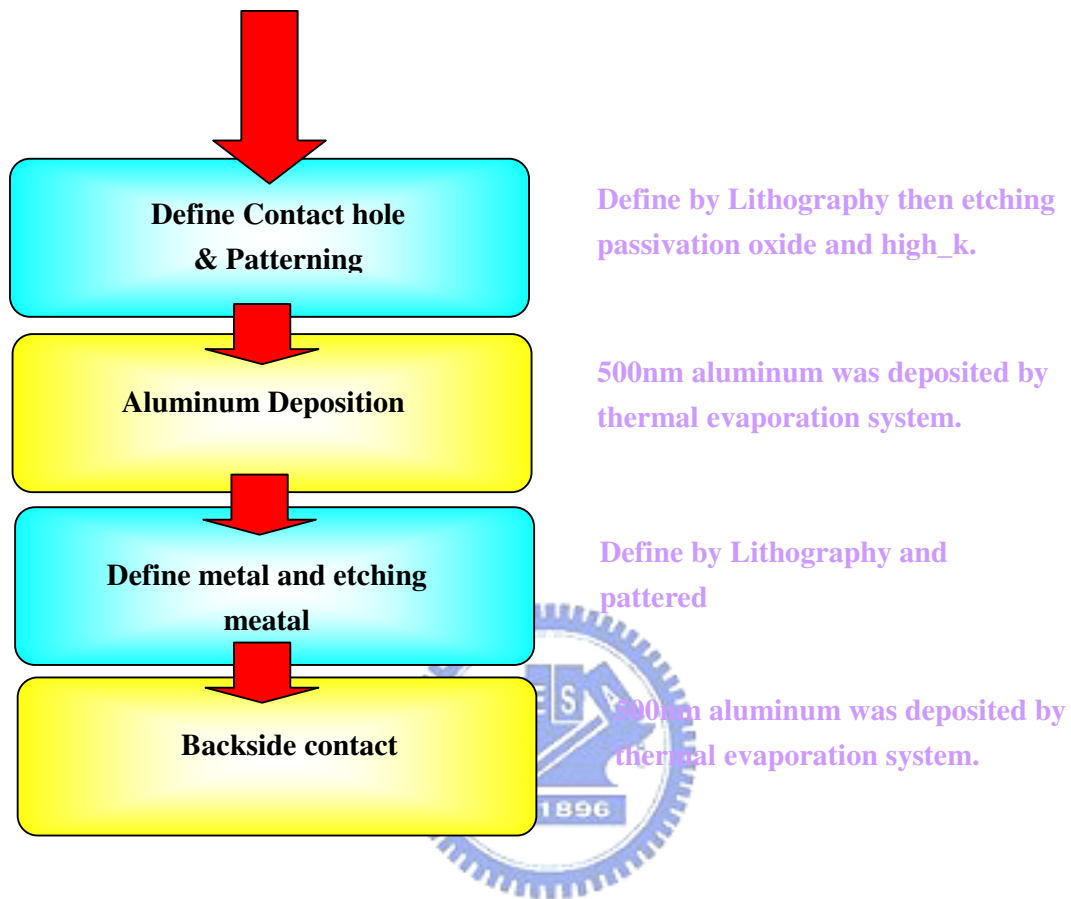


Fig 2-2 The diagram of process flow of pMOS transistor

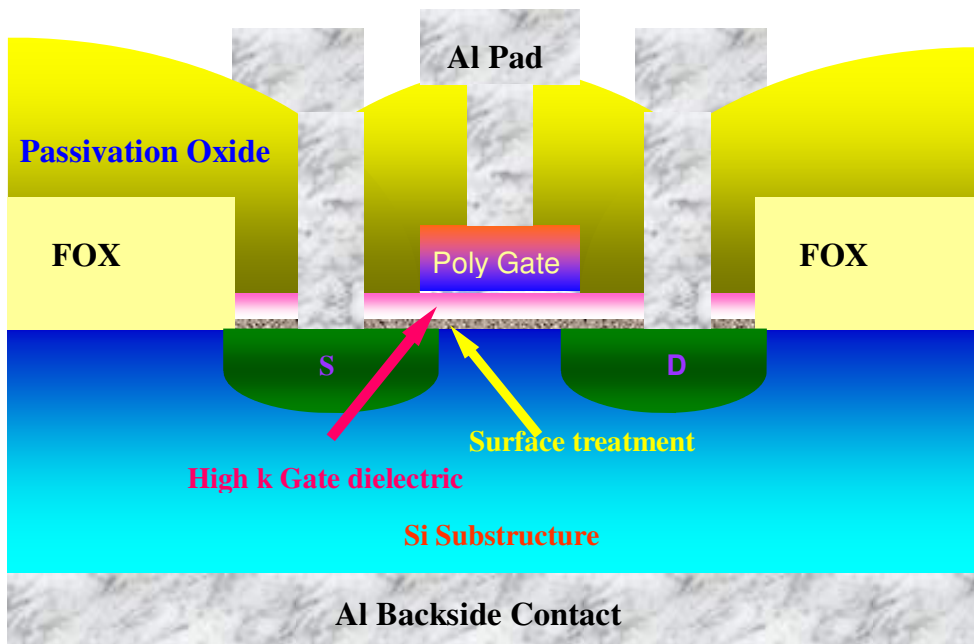


Figure 2-3 The cross section of pMOS transistor.

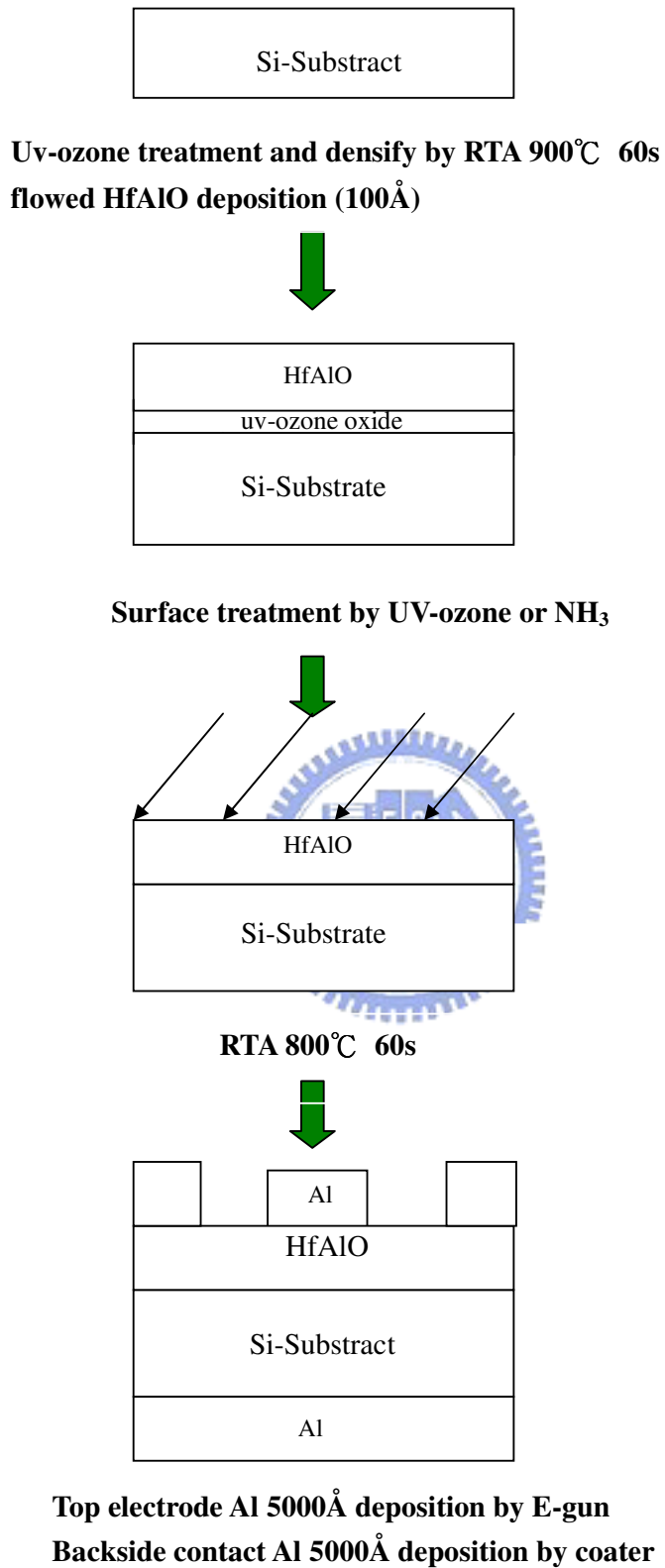


Fig 2-4 The diagram and structure of process flow of capacitance

Chapter 3

Result and Discussion

3-1 The Electrical Characteristics of Surface Treatment Before HfAlO Deposition

Nitridation of the Si surface using N_2O prior to the deposition of high- κ gate dielectrics has been shown to be effective in achieving the low EOT and preventing boron penetration [17], [18]. However this technique results in higher interface charges [19], which leads to higher hysteresis and reduced channel mobility. So we use UV ozone to oxidation, ozone and atomic oxygen, produced by exposure of atmospheric oxygen to ultraviolet radiation, Figure 3-1 showed the schematic diagram of UV ozone system. The ozone generator (AnserosPAP-2000) decomposed the oxygen molecular to generate ozone gas by high electrical field. The ozone gas was mixed with UV lamp in chamber. By changing the oxygen flow and ozone generation power, the ozone concentration in the chamber could be adjusted. Ozone-formed oxide (ozone oxide) has superior characteristics. Even when the formation temperature is less than 400°C , ozone oxide has a high film density comparable to that of the device-grade oxide film formed at higher temperature [20], a low interface trap density [21], and a much thinner structural transition layer near the SiO_2/Si interface [22]. The ozone surface treatment was employed to improve the interface quality between HfAlO and silicon substrate. Figure 3-2 shows the drain

current (I_d) versus the drain voltage (V_d) characteristic for UV ozone surface treatment and N_2O surface treatment. The drain current is larger for the UV ozone surface treatment. Fig 3-3 shows the transconductance characteristics, we observe that transconductance peak value of UV ozone treatment is higher than samples with N_2O surface treatment. Figure 3-4 and Figure 3-5 shows the drain current (I_d) versus the gate voltage (V_g) of pMOSFETs with and UV ozone treatment and N_2O treatment. We can see that the subthreshold swing (S.S.) of UV-ozone treatment is 75.26 mV/decade which is better than the 91.2mV/decade of N_2O treatment.

In this section, we also research the relationship between charge pumping and mobility through devices in different surface treatment. Unlike SiO_2 films, high-k films are more susceptible to charge trapping. Charge trapping is arguable one of the most important issue in CMOS devices with high-k gate dielectrics, because of the large amount of bulk traps present in the high k films [23]-[26], and they may cause mobility degradation [27]-[30] and Vth instability [31]-[40], it used to use charge pumping to measure interface state density in MOSFET devices by utilizing the exclusion of gate leakage to calculate interface state density in high-k dielectric. We consider an p-channel device of gate length L and width W Evaluate mobility by Spilt-CV technique. Figure 3-7 shows electron mobility with UV ozone and N_2O surface treatment measured by split-CV method. From the mobility we can find that UV ozone treatment have higher mobility, it may be due to the UV ozone treatment have lower I_{cp} and N_{it} shows in Fig 3-6 and from Fig 3-8 we can make a conclusion that high interface states lead to low mobility. . We compare to this two samples on electrical characteristic above. We found that UV ozone treatment effectively improve the electrical characteristics, such interface state density, electric mobility, transconductance and etc.

3-2 Electrical Properties with Surface Treatment After HfAlO Deposition

We use uv-ozone, NH_3 two surface treatment after HfAlO deposition. Figure 3-9 shows the comparison of the C-V curves of the samples with UV ozone or NH_3 surface treatment, both ozone and NH_3 surface treatment will reduce leakage current, but samples with NH_3 post treatment accompany annealing reduce more, in Figure 3-10, the comparison of J-V curve of the samples with UV ozone or NH_3 surface treatment, the NH_3 treatment samples have lower leakage with thinner EOT, to sum up, samples with NH_3 treatment have large capacitance and lower leakage current with thinner EOT. It may be due to RTA after NH_3 treatment will make N diffuss into the HfAlO and form HfAlO_xN and make partial change of the local coordination from O-Hf-O to O-Hf-N. The increase of O-Hf-N component drastically degraded the gate leakage current in $\text{HfAlO}_x(\text{N})$ film [41]. Fig 3-11 shows the leakage current after gate injection stress with UV ozone, and NH_3 treatment, trapping is observed obviously in UV ozone treatment samples. It may be due to there are large fixed charge and interface states at interface, therefore the trapping is observed after gate injection stress. We also look from the other side, Figure 3-12 shows the normalize C-V curve, The C-V curve of NH_3 surface treatment was similar to that of HfAlO, indicating that no additional fixed charge was generated.

3-3 Electrical Properties of Post Dielectric Annealing (PDA)

Fig 3-13 shows the capacitance-voltage curves of the as deposited HfAlO sample after NH_3 surface treatment. There is clear that the scan from inversion to accumulation has a little shift with the scan from accumulation to inversion scan, this suggest that the interface quality is poor, and the film may be a mixing phase of amorphous and polycrystalline at 400°C deposition, it may be have many imperfect bonds and defects due to incomplete crystal structure and it will lead to plentiful interface states exist at interface to be the trap centers and reduce the mobility. Therefore we try post dielectric annealing at different temperature to see weather can be improving the quality of interface and lower the interface states. Fig 3-14 shows the C-V curve of HfAlO samples with 800°C 900°C 1000°C PDA, and it is clear that the shift is reduced compare to the samples without PDA. PDA treatment at higher temperature after NH_3 treatment improved the flat-band voltage shift. Besides improve the quality of interface, we also observe the PDA 1000°C capacitance at accumulation is smaller than PDA 800°C , It suggests that the increasing thickness of interfacial layer would raise the CET and reduce the dielectric constant. Fig 3-15 shows the current density- voltage curve of the HfAlO samples with different PDA temperature 800°C 900°C 1000°C . Higher annealing temperature as 800°C 900°C lead to higher leakage due to high annealing temperature trigger small grains to merge into a large grain, and it will provide short leakage path with the boundaries around the large grains which lead the carries tunnel from top electrode to bottom electrode, Therefore the leakage current would increase. But for 1000°C the leakage current is fall away, it may be due to the thicker interfacial layer resist leakage current,

therefore the curve will drop. Fig 3-16 shows the current- electrical field curves of the HfAlO with no PDA, We can observe that samples with no post dielectric annealing have more trap and the interface quality is not good lead to the obvious different with first scan and second scan, and electron trapping is saturated after the second injection. Fig 3-17 , shows the current-electrical field curves of the HfAlO with different PDA temperature, 800°C 900°C 1000°C , it is clear that the result of first scan and second scan are almost the same, therefore trapping is eliminated after annealing, it may due to high temperature can reduce defects in high-k, so PDA can suppress the trapping.

3-4 The Trapping Phenomenon of Observation Under SILC Stress

For Figure 3-18 ~ Figure 3-20 shows the J-E curves of as deposited samples and with, O₃ and NH₃ surface treatment under SILC stress. The result we compare samples with treatment and without treatment, that the trapping phenomenon after surface treatment is not severe as the as deposited HfAlO, therefore surface treatment can reduce trap generation under stress .In the other aspect we observed NH₃ treatment is superior to UV ozone treatment in suppressing both electron, hole trappings, and interface trap creation under high-field stress. Interface hardness against hot-carrier bombardment and neutral electron trap generation are also improved.

3-5 Surface Composition Analysis by X-ray Photoelectron Spectrometer

We use X-ray photoelectron spectrometer (XPS) to analysis composition of surface films, prior to and after periods of UV/ozone treatment, Fig 3-21 shows the Si(2p) binding energies before and after exposure to UV ozone, and after surface exposure to UV ozone, the XPS Si(2p) binding energy shifted toward that of SiO₂ (103.6ev) [42], consistent with the formation of silicon that is coordinated to four oxygen atoms. The sampling depth of XPS, however, is only of the order of 7 Å and thus indicates near complete conversion to form a SiO₂ surface layer that is at least 7 Å thick, and Figure3-22, Figure 3-23 show the electron binding energies of O(1s) and C xps spectra, The main O 1s binding energy is the Si-O type with a binding energy of 534.1 eV we can see the reduction in the amount of carbon and an increase in the amount of oxygen within the converted surface film, the reason for reduction of carbon is may be due to UV/ozone treatment removes up to 89% of the carbon from the resultant surface film [43], leading to an overall stoichiometry close to that of SiO₂. Fig 3-26 shows the Hf (4f) binding energy with NH₃ treatment accompany with RTA and w/o NH₃ treatment, there is shift after NH₃ treatment. This result suggests a partial change of the average local coordination structure from O-Hf-O to O-Hf-N after high temperature annealing with NH₃ treatment because the electronegativity of nitrogen is relatively low compared to that of oxygen. The results of the N 1s and O 1s XPS spectra also support this interpretation, shown in Fig 3-24, Fig 3-25. The main O 1s binding energy is the Hf-O type with a binding energy of

523.3 eV, therefore the film may turn to the $\text{HfAlO}_x(\text{N})$ after NH_3 treatment and RTA.

3-6 The Instruction of Leakage Current Conduction Mechanism

For the micro electrical devices, the goal for using insulating thin film is hope this film can insulate completely and without any leakage current, including TiO_2 , HfO_2 , SiO_2 , etc has larger band gap and maintain the structure of amorphous or polycrystalline to reduce leakage current, but on reality there are many physical mechanism can make carriers move in the insulating dielectric and form the leakage current, especially in the very thin film or electrical field is large.

There are two classes of leakage current mechanisms in the insulating thin film, one is electrode-limited conduction mechanism, it is determined by the character of emission electrode, for example Schottky emission [44] [45] field emission, it is also called tunneling, and thermionic field emission. Among them, tunneling is divided into direct tunneling and Fowler-Nordheim tunneling (F-N tunneling). The other is Transport-limited conduction mechanism, it is determined by the character of material, for example Ohmic conduction, Frenkel-Poole emission [46][47], Hopping conduction and ionic conduction. In this thesis we discussion and analysis the Schottky emission and F-P emission.

3-6.1 Schottky Emission

Under the function with electric field, the electron of metal cross the potential energy barrier from metal electrode to the conduction band of insulator, it is called Schottky emission. Consider a electron at x place from metal surface, use the law of virtual image, it will produce equal positive charge at $-x$ place from metal surface,

the gravitation between electron and positive charge is called image force, the image force would lower energy barrier, it is Schottky effect Fig 3-27 Shows the diagram of Schottky emission, therefore electrons across the potential energy barrier easier via field-assisted and increase leakage current. The leakage current equation is showed in (eq.3-1)

$$J = A^* T^2 \exp\left(\frac{\beta_s E^{1/2} - \phi_s}{k_B T}\right) \quad [\text{eq } 3-1]$$

Where $\beta = (e^3 / 4\pi\epsilon_0\epsilon)^{1/2}$, A^* is effective Richardson constant, ϵ_0 is the permittivity of free space, ϵ is the high frequency relative dielectric constant, K_B is the Boltzmann constant (1.38×10^{-23} J/K), ϕ_s is the contact potential barrier. We can find the slope of the leakage current equation. (eq 3-2), so we can draw the $\log(J/T^2)$ versus $E^{1/2}$ curve to see whether can receive a straight line to judge whether is the Schottky emission mechanism.

$$\ln \frac{J}{T^2} = \left[\frac{\beta_s}{K_B T} \sqrt{E} \right] + \left[\ln A^* - \frac{\phi_s}{K_B T} \right] \quad [\text{eq } 3-2]$$

3-6.2 Frenkel-Poole Emission

The theorem of Frenkel-Poole effect is similar to Schottky-emission, the different is that electron is excited from trap center to the conduction band of dielectric by way of thermal excitation via field-assisted, Fig 3-28 shows the diagram of

Frenkel-Poole emission Usually the dielectric with bigger energy band gap have larger probability to happen Frenkel-Poole emission.

In Fig 3-29, Fig 3-20 the conduction mechanism of the UV ozone-treated and NH₃-treated HfAlO was extracted from fine Frenkel-Poole fitting. The current from the Frenkel-Poole

$$J = B \times E_{eff} \times \exp\left(\frac{-q(\phi_B - \sqrt{qE_{eff} / \pi\epsilon_{HfAlO}\epsilon_0})}{kT}\right); B = q\mu N_T$$

$$\Rightarrow \ln(J / E_{eff}) = \frac{q\sqrt{q / \pi\epsilon_{HfAlO}\epsilon_0}}{kT} \sqrt{E_{eff}} - \left[\frac{q\phi_B}{kT} + \ln(B) \right]$$

$$\text{or } \ln(J / E_{eff}) = E_{act} \times \left(\frac{q}{kT}\right) + \ln(aB); E_{act} = -(\phi_B - \sqrt{aqE_{eff} / \pi\epsilon_{HfO_2}\epsilon_0})$$

emission is of the form. where B is a constant related to the trapping density and carrier mobility in the HfAlO film, ϕ_B is the barrier height, E_{eff} is the effective electric field in the SiO₂ film, ϵ_0 is the free space permittivity, ϵ_{HfAlO} is the dielectric constant of HfAlO, k is the Boltzmann constant (1.38×10^{-23} J/K), E_{act} is a field-dependent effective activation, and T is the temperature (K)., From an Arrhenius plot of E_{act} [i.e., $\ln(J/E_{eff})$ vs. (q/kT)], we can obtain E_{act} and B . The barrier height ϕ_B and dielectric constant ϵ_{HfAlO} of HfAlO can then be calculated from the intercept of the y axis and the slope of the fitting curves in the plot of E_{act} vs. $\sqrt{E_{eff}}$, according to $E_{act} = q\sqrt{aq\pi\epsilon_k\epsilon_0}(\sqrt{E_{eff}}) - q\phi_B$. As indicated in Fig.3-29, Fig 3-30 we obtained excellent linearity for each current characteristic. This tendency indicates that the Frenkel-Poole conduction mechanism is dominant in the Ozone-treated and

NH₃-treated samples. It may be due to there are large tunneling effect in the interface.

We calculated the value the value of φ_B for electrons was 0.33 eV. and 0.46 eV

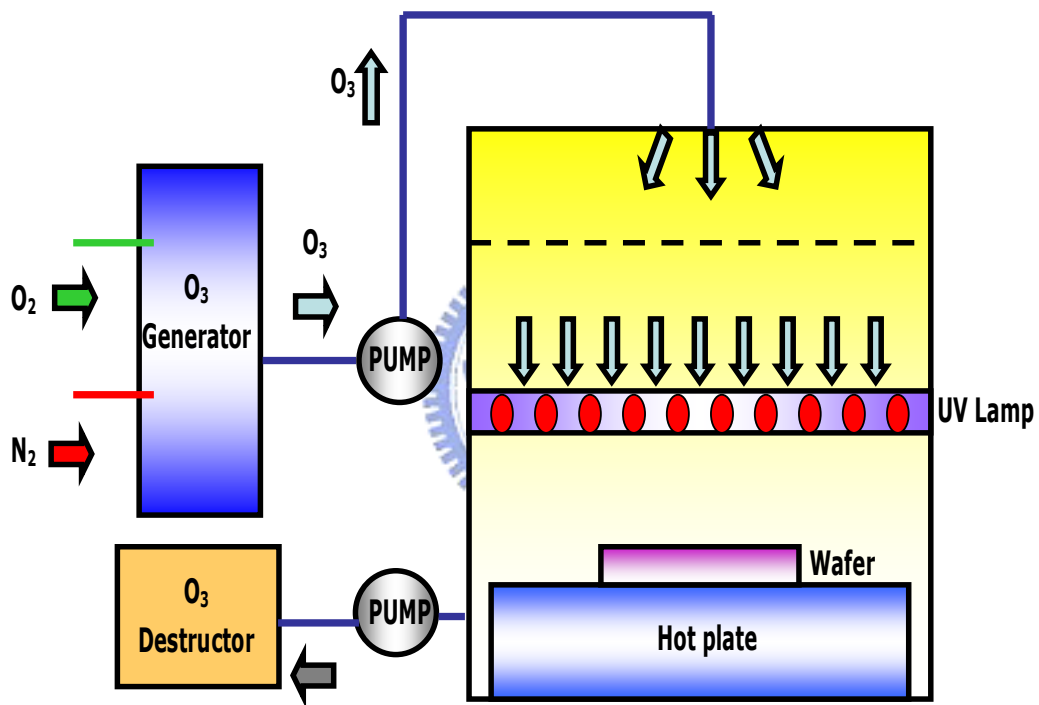


Figure 3-1 UV ozone system schematic diagram

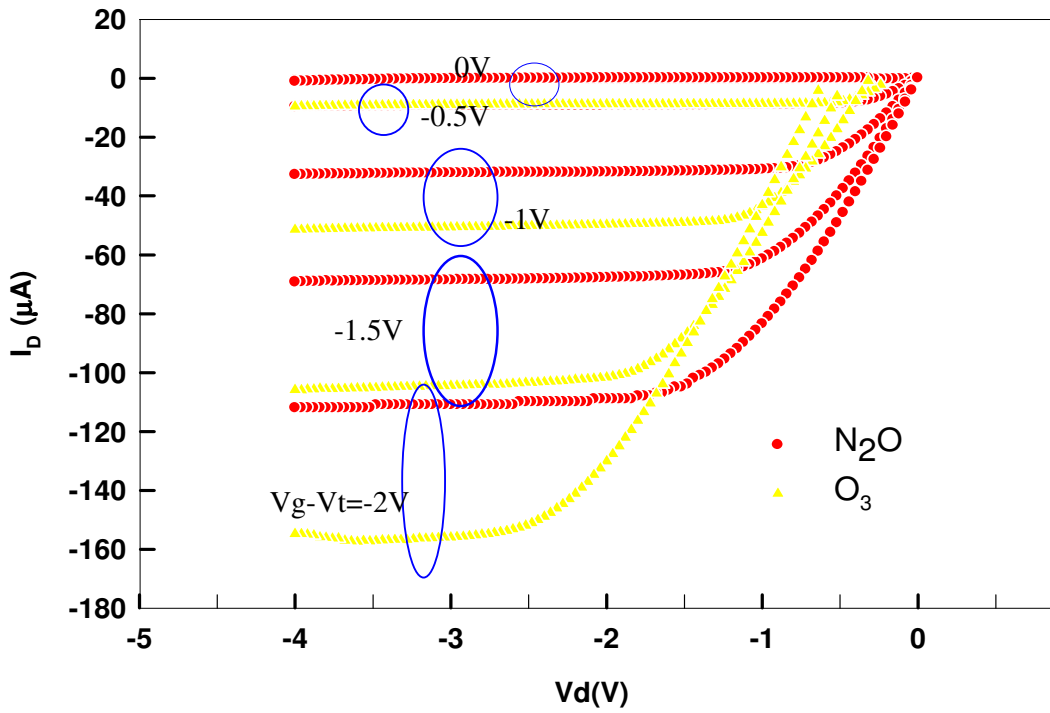


Fig 3-2 I_d - V_d characteristics with UV ozone and N_2O treatment. $V_g - V_{th} = 0 \sim -2$ V,

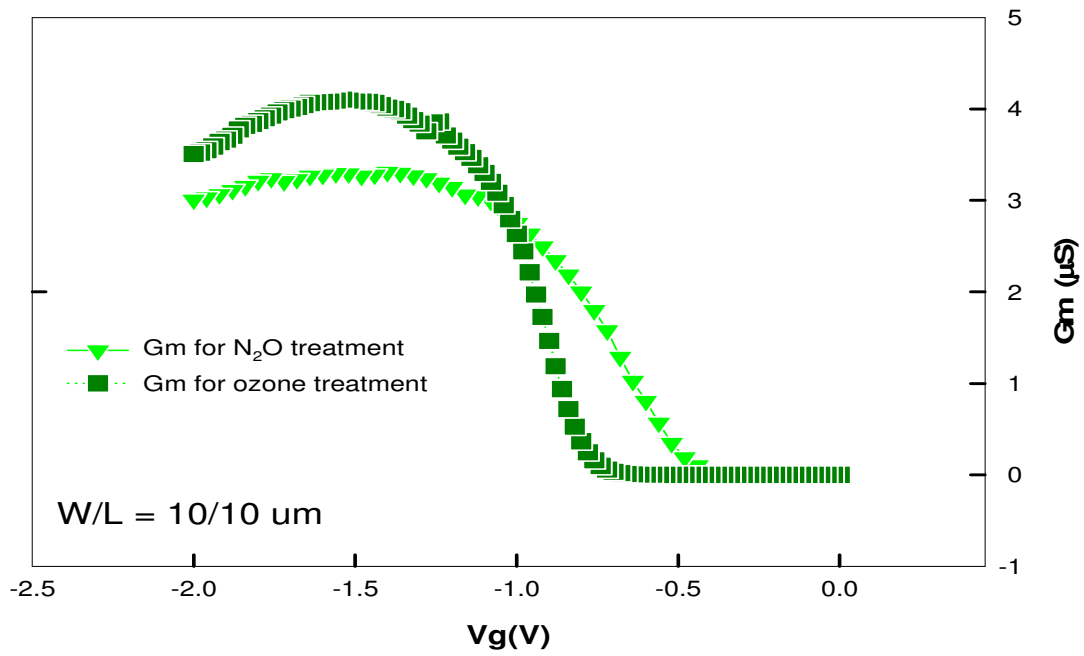


Fig 3-3 The transconductance characteristic of HfAlO samples with UV ozone

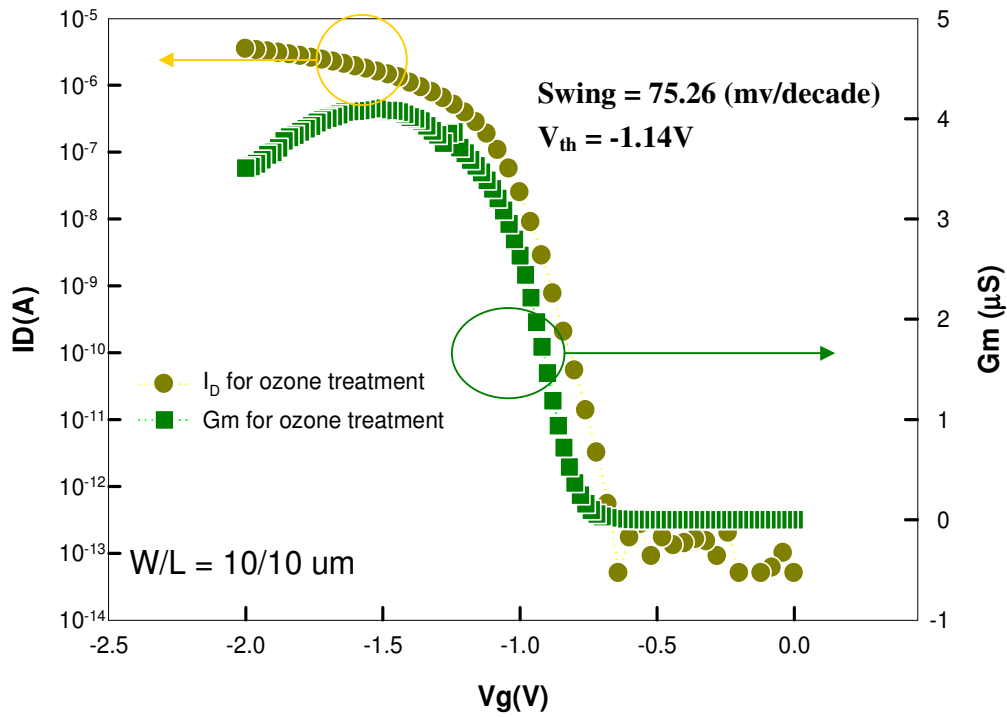


Fig 3-4 I_D - V_g characteristics of PMOSFET with UV ozone treatment.

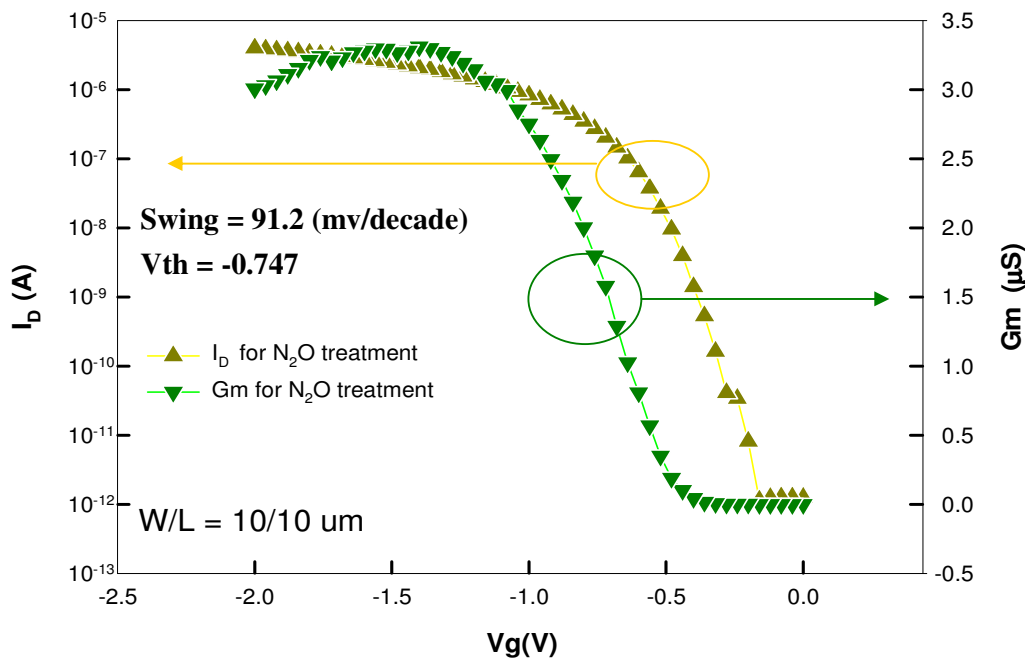


Fig 3-5 I_D - V_g characteristics of PMOSFET with N_2O treatment.

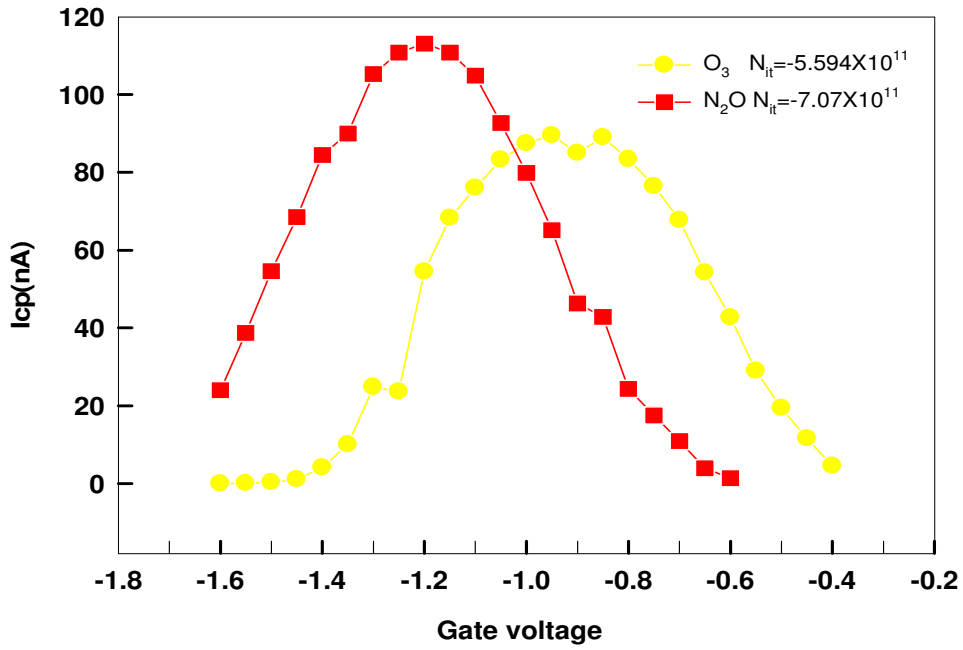


Fig 3-6 Charge pumping measurement I_{cp}

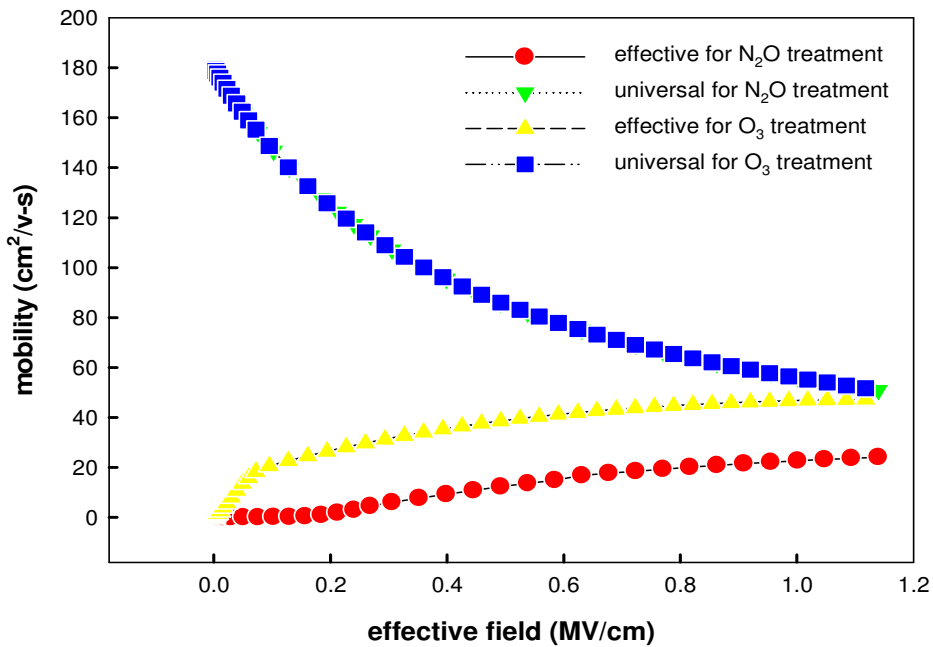


Fig 3-7 Mobility characteristic of PMOSFET with N_2O and UV ozone treatment

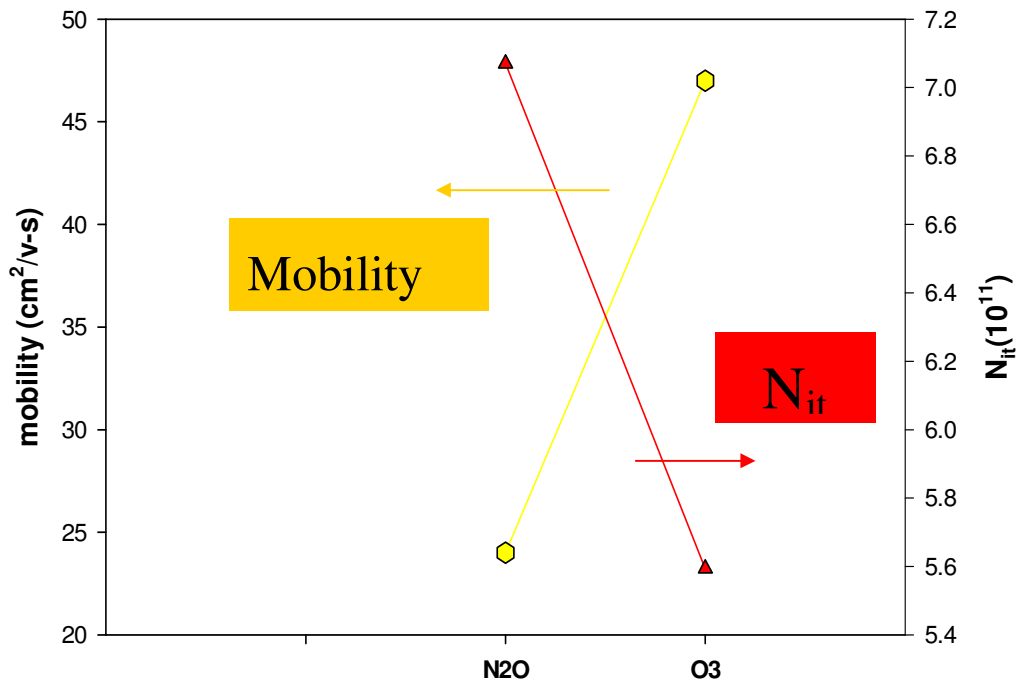


Fig 3-8 The relationship between mobility and Nit of HfAlO with UV ozone and N₂O surface treatment

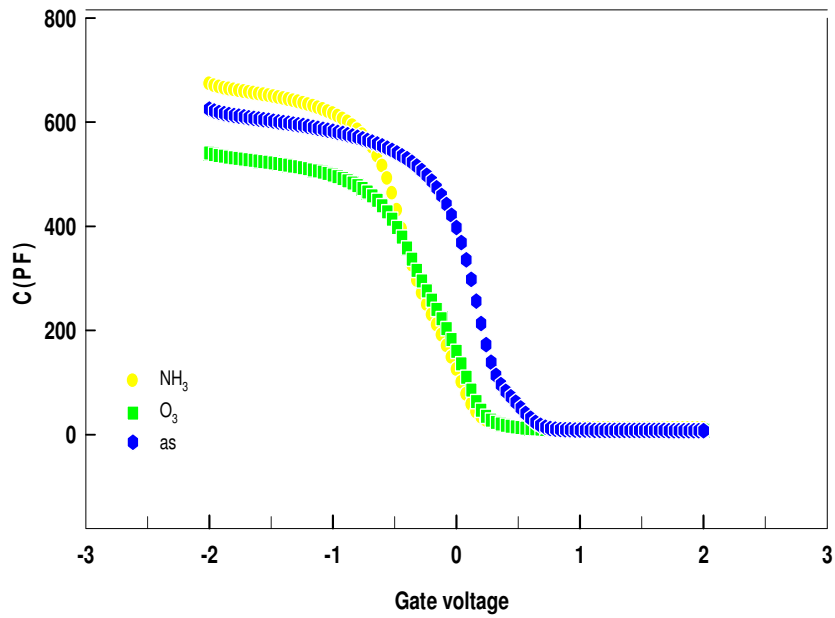


Fig 3-9 The C-V curves of O₃ and NH₃ treatment with 800°C RTA after HfAlO deposition

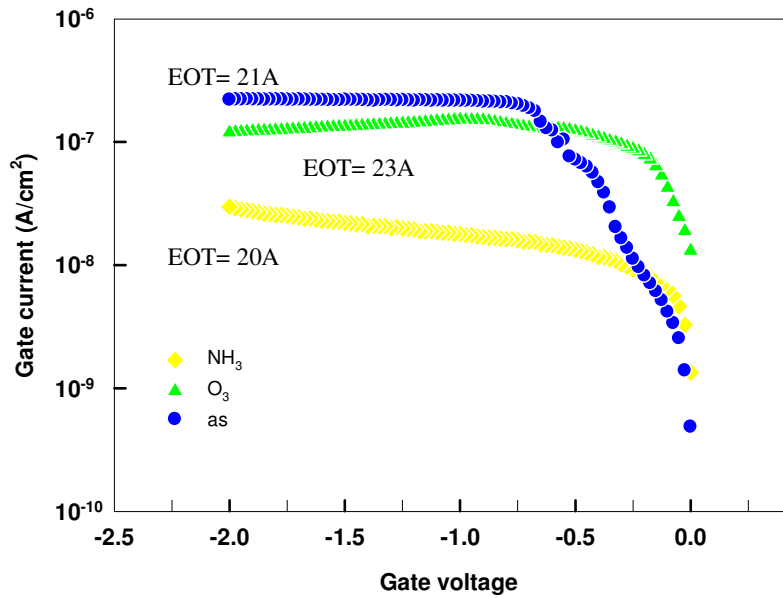


Fig 3-10 The J-V curves of O₃ and NH₃ treatment with 800°C RTA after HfAlO deposition

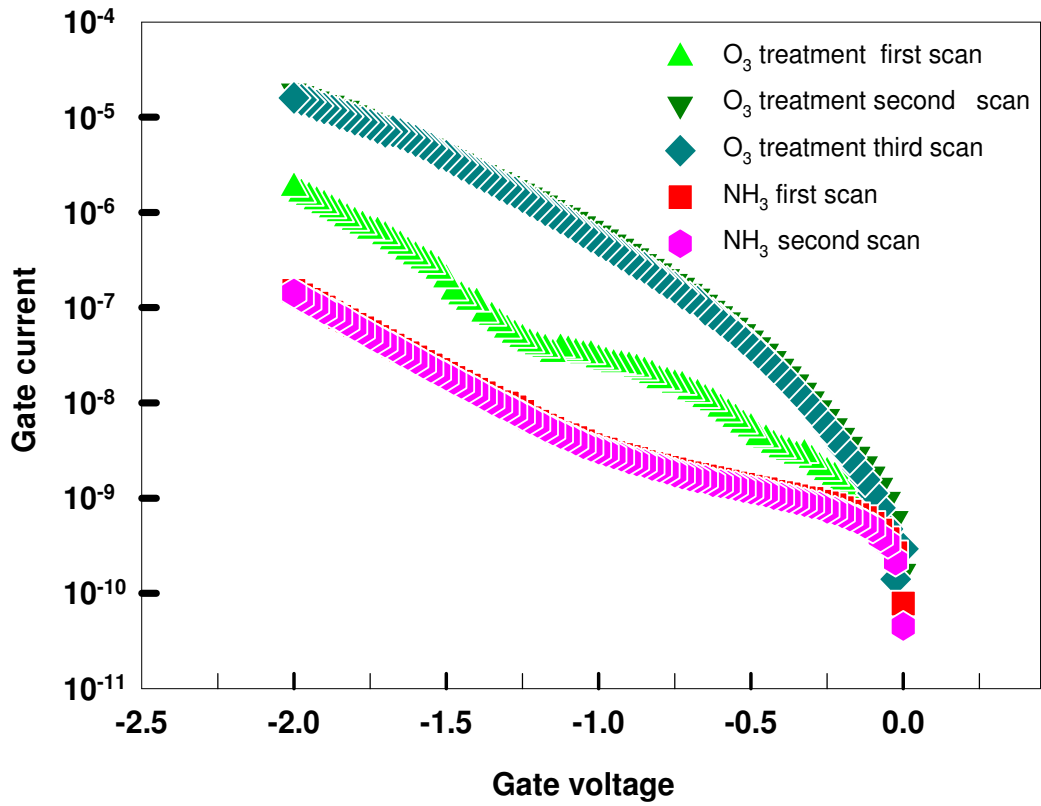


Fig 3-11 The leakage current after a gate injection stress of UV ozone and NH_3 treatment. Electron trapping is observed in ozone treat samples

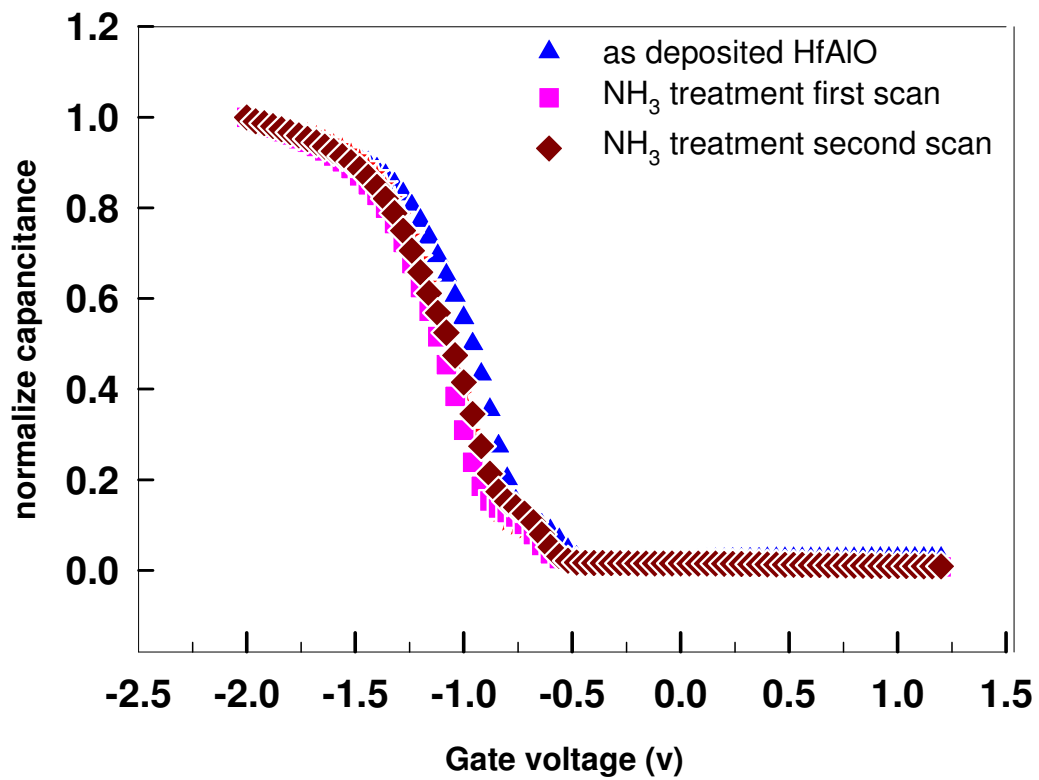


Fig 3-12 The normalize C-V curve of HfAlO with NH₃ treatment and without treatment samples

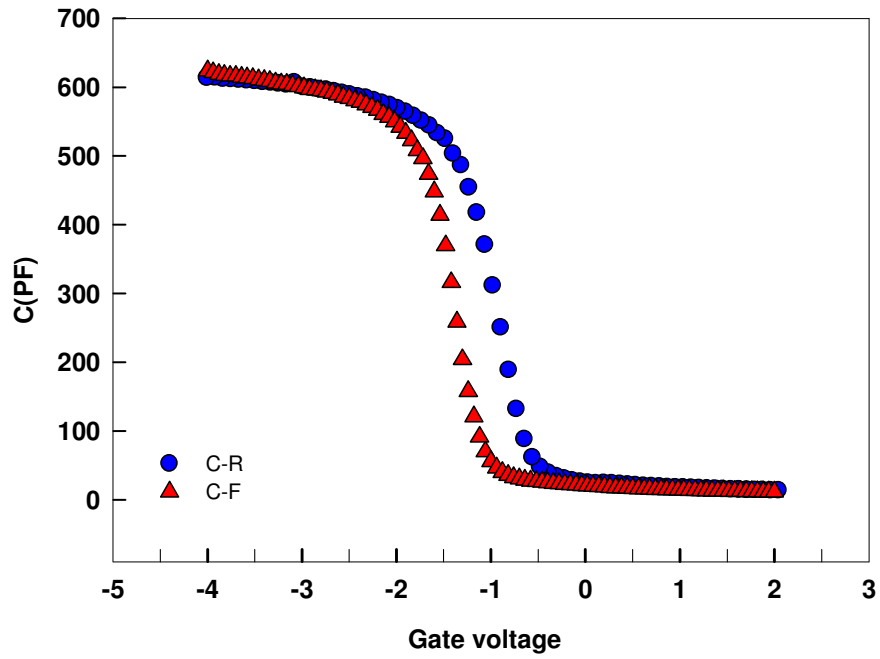


Fig 3-13 The C-V curves of as deposited HfAlO

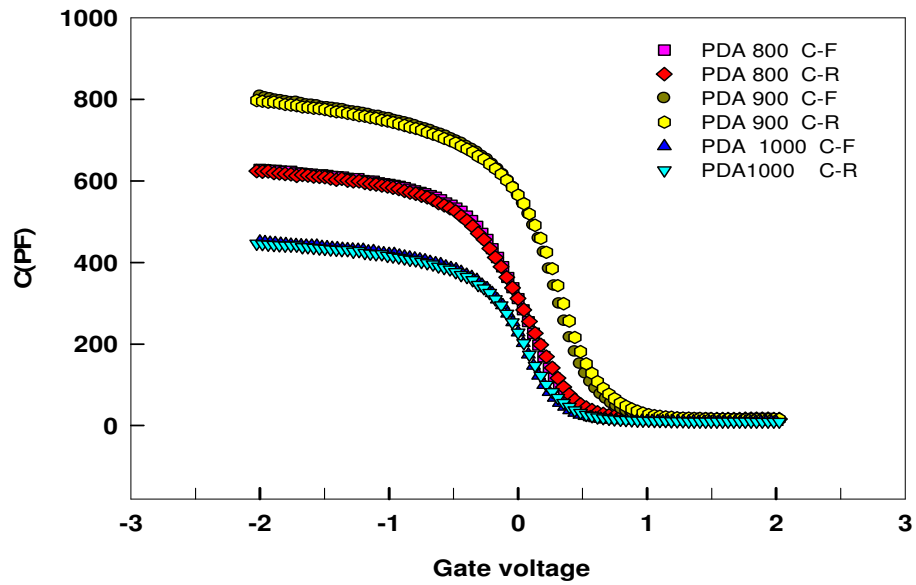


Fig 3-14 The C-V curves of HfAlO after PDA

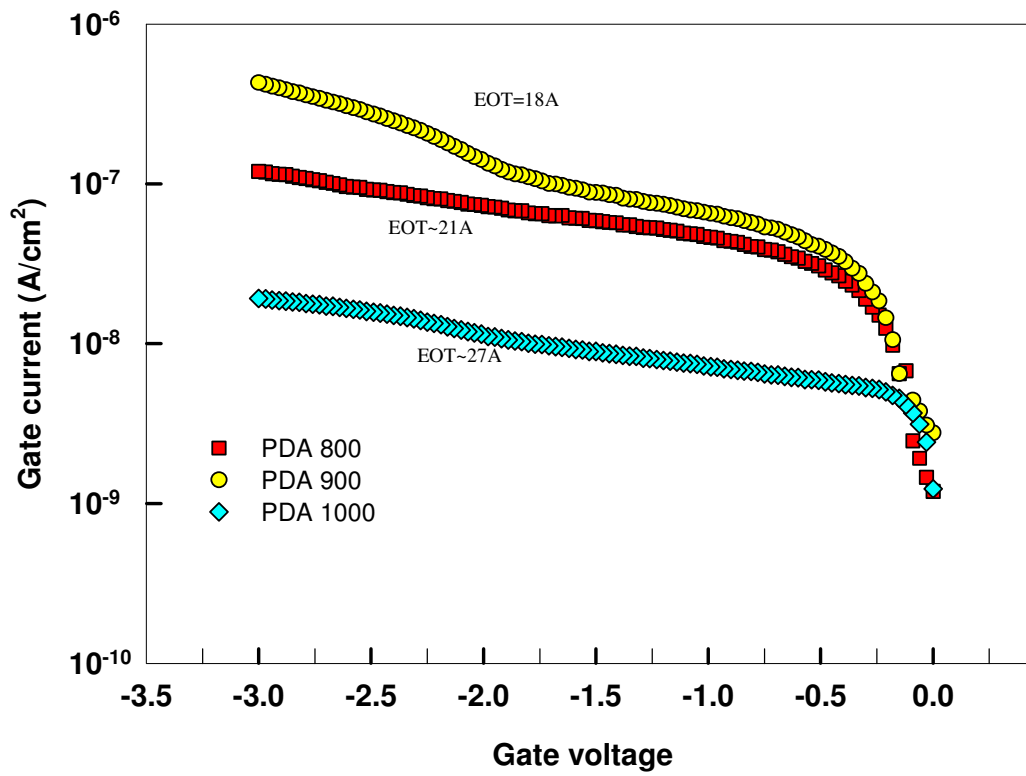


Fig 3-15 The J-V curves of HfAlO before PDA and after PDA

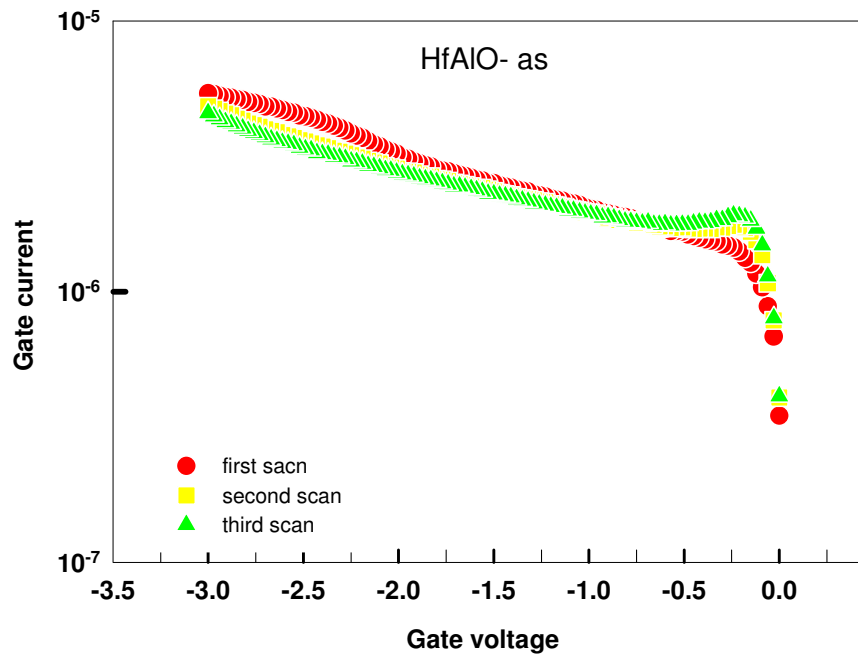


Fig 3-16 The leakage current after gate injection stress of as-deposited HfAlO

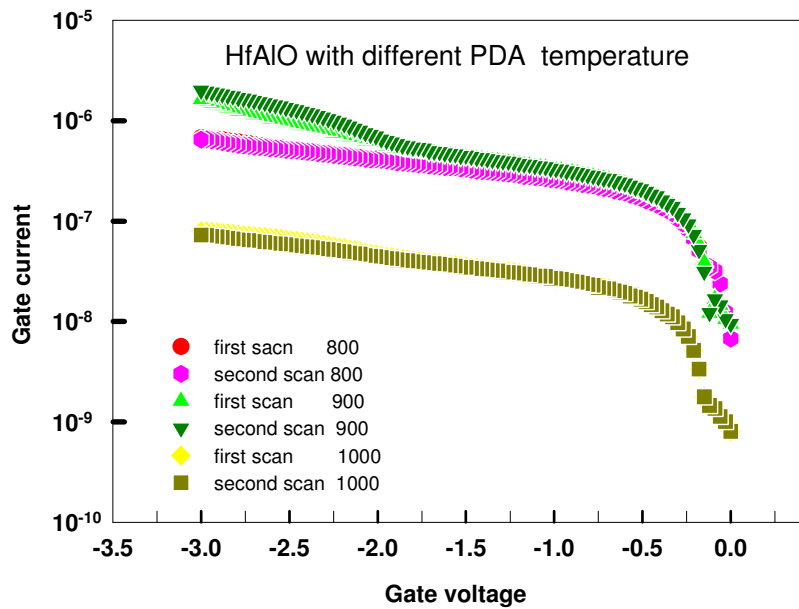


Fig 3-17 The leakage current after gate injection stress of HfAlO with different PDA temperature

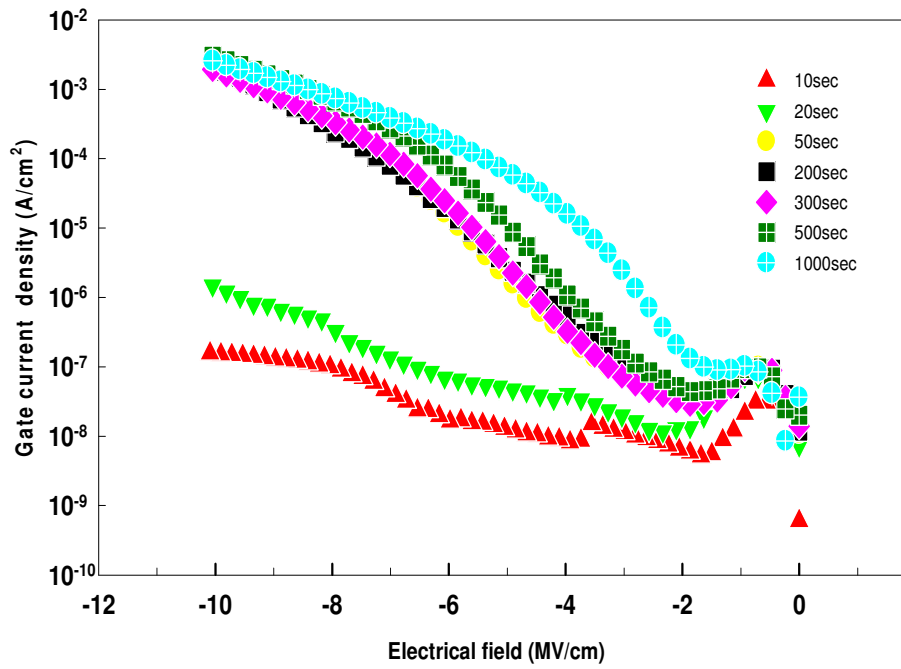


Fig 3-18 The J-E curves of as deposited HfAlO samples under SILC stress

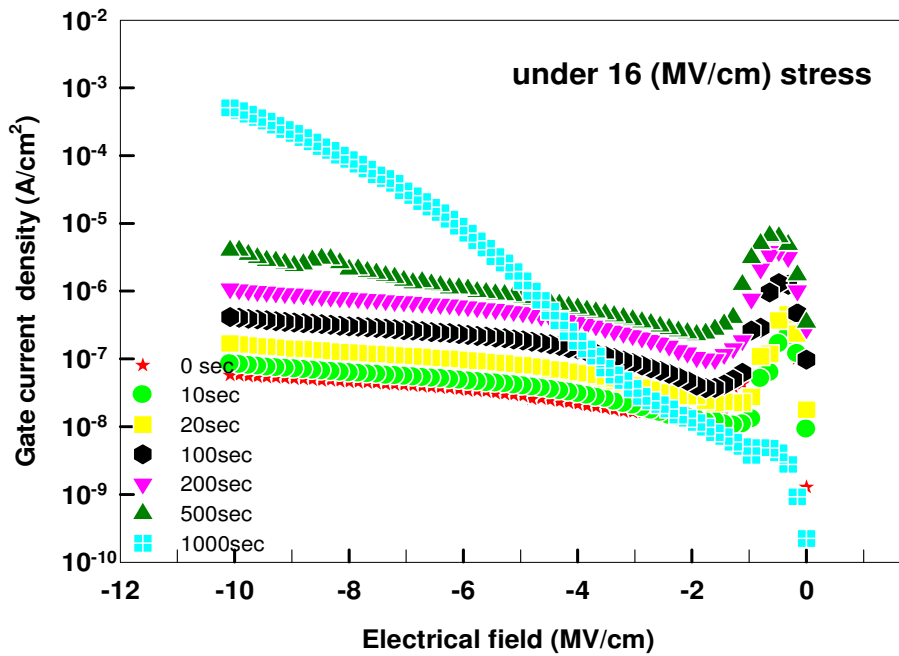


Fig 3-19 The J-E curves of HfAlO samples with O₃ surface treatment under SILC stress

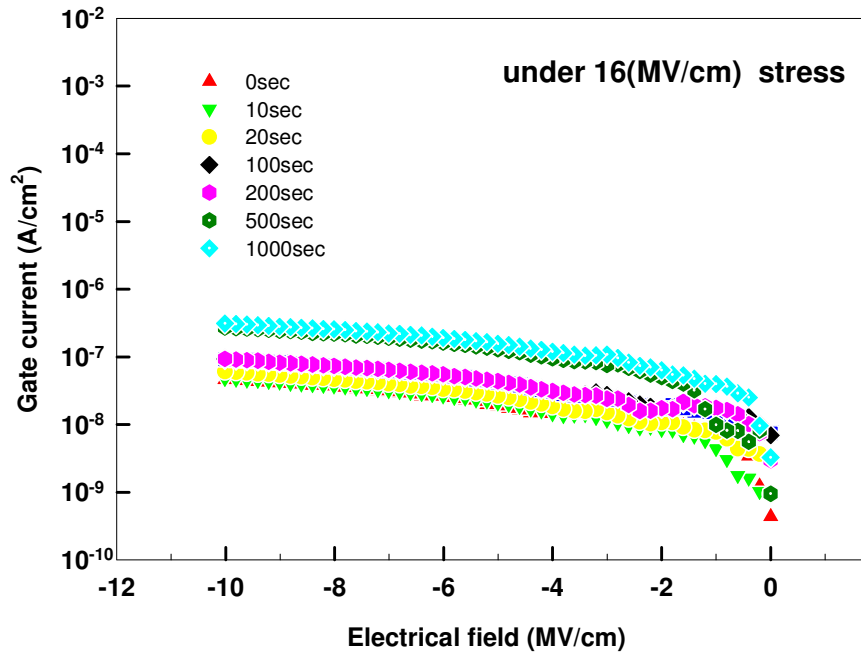


Fig 3-20 The J-E curves of HfAlO samples with NH₃ surface treatment under SILC stress

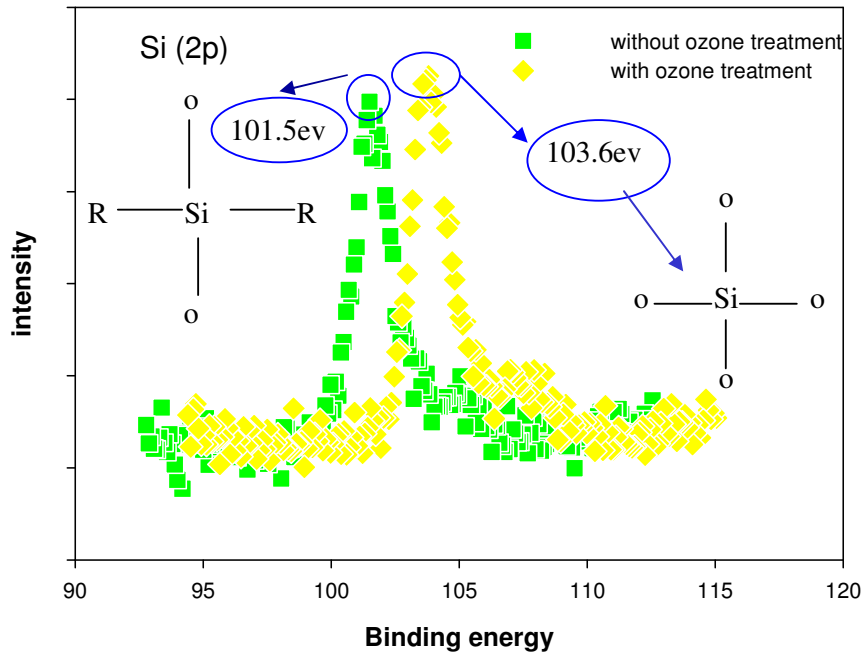


Fig 3-21 The electron binding energies of Si 2p orbital before and after UV ozone surface treatment

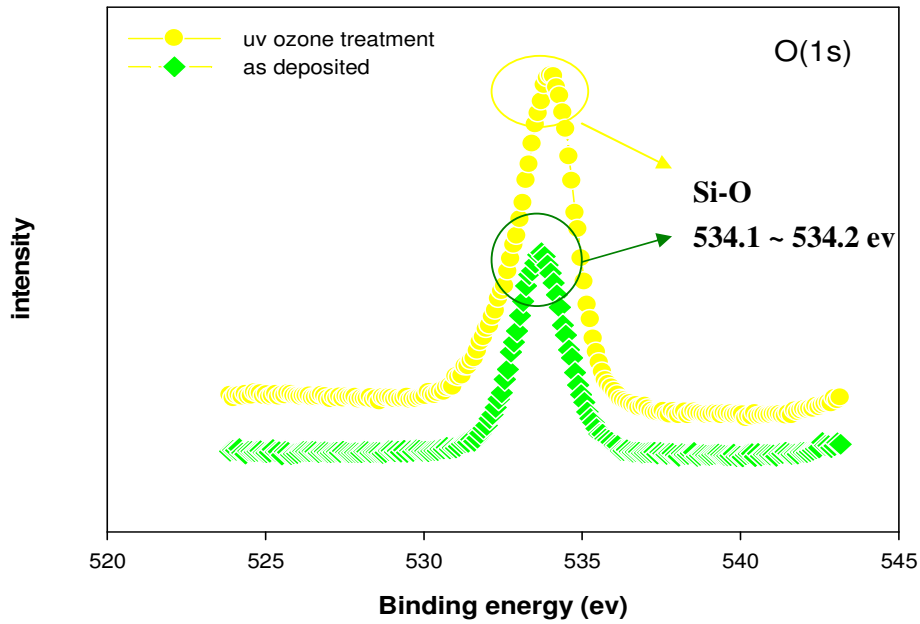


Fig 3-22 The electron binding energies of O (1s) orbital before and after UV ozone surface treatment

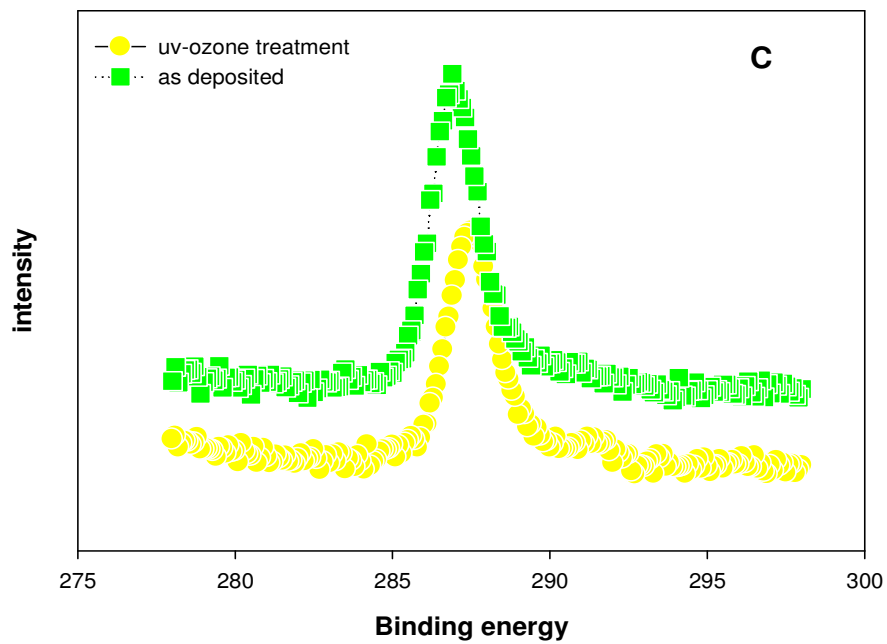


Fig 3-23 The electron binding energies of C orbital before and after UV ozone surface treatment

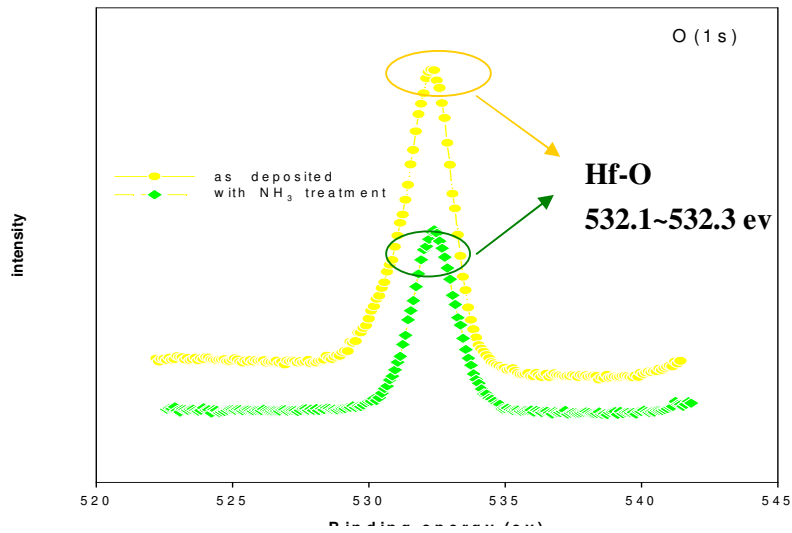


Fig 3-24 The electron binding energies of O (1s) orbital with NH₃ treatment with different RTA temperature

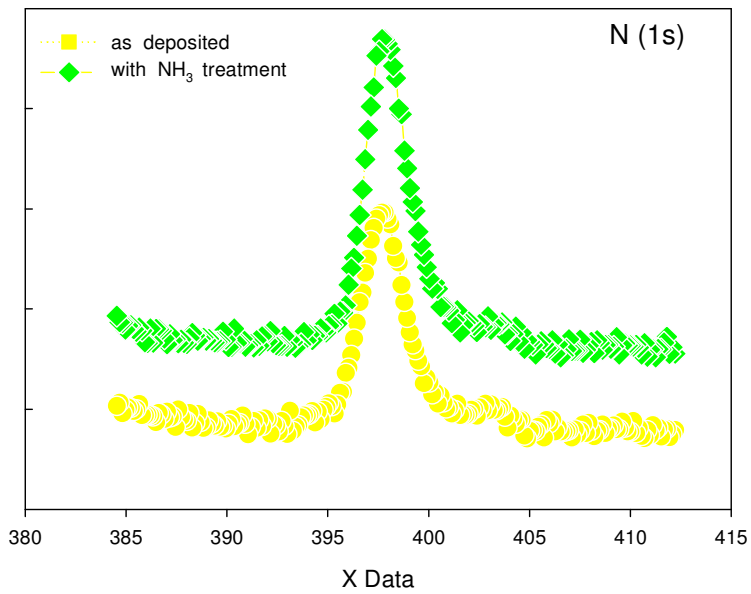


Fig 3-25 The electron binding energies of N(1s) orbital with NH₃ treatment with different RTA temperature

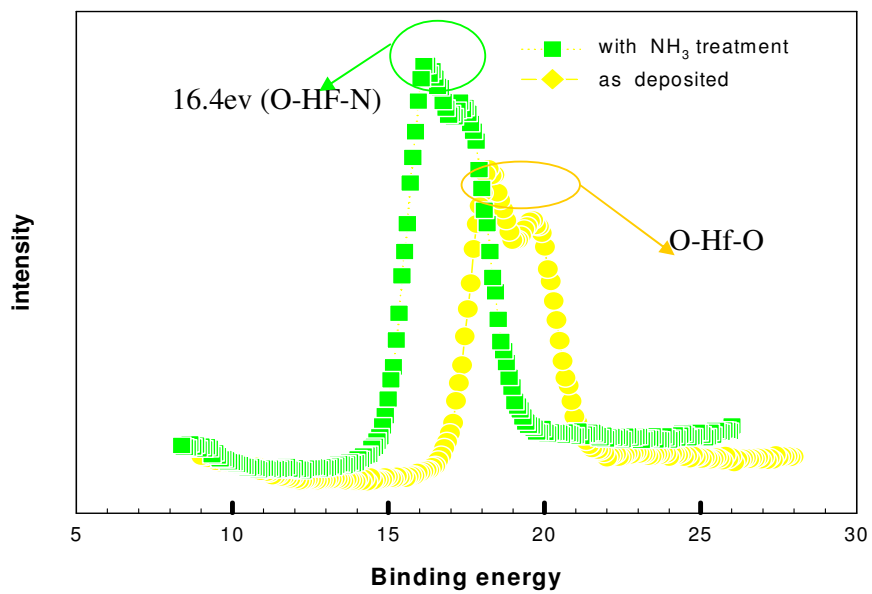


Fig 3-26 The electron binding energies of HF(4f) orbital with NH₃ treatment with different RTA temperature

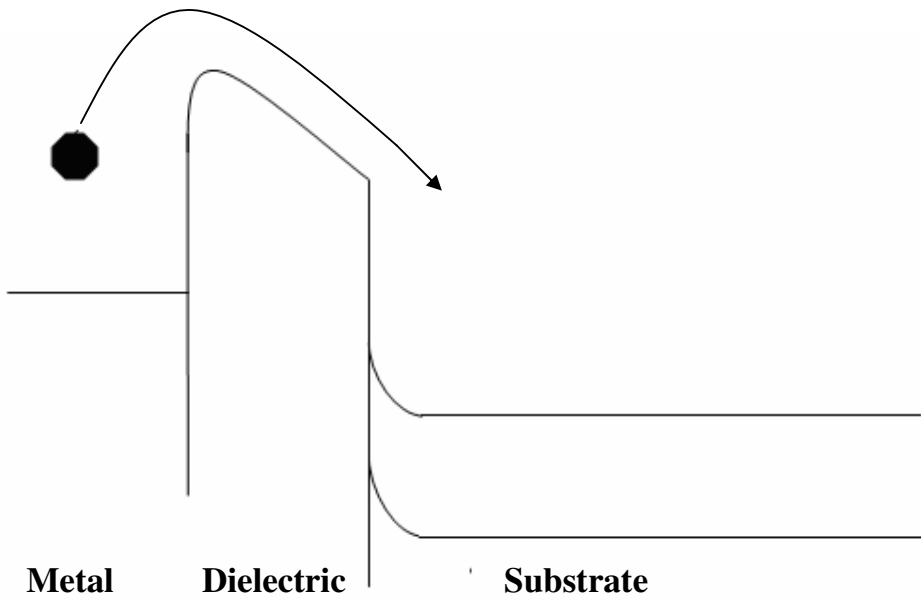


Fig 3-27 Schematic band diagram of Schottky emission

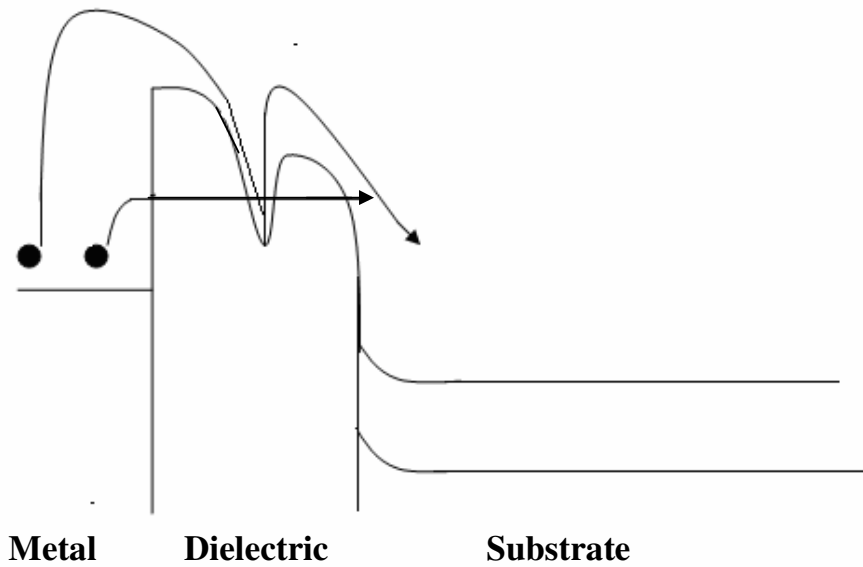


Figure 3-28 Schematic band diagram of Frenkel-Poole emission

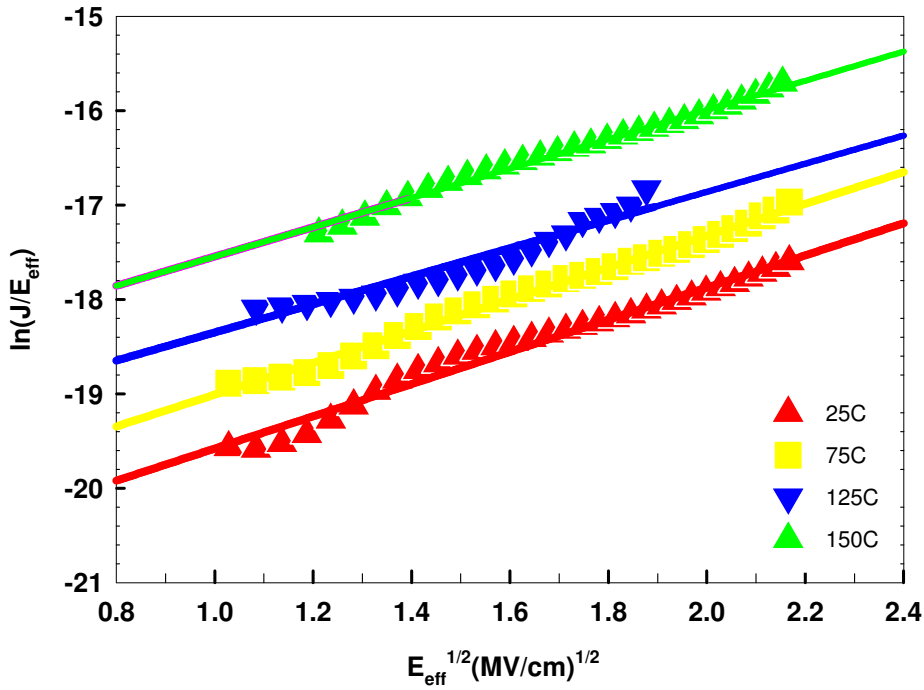


Fig 3-29 The conduction mechanism fitting of HfAlO samples with UV ozone treatment .

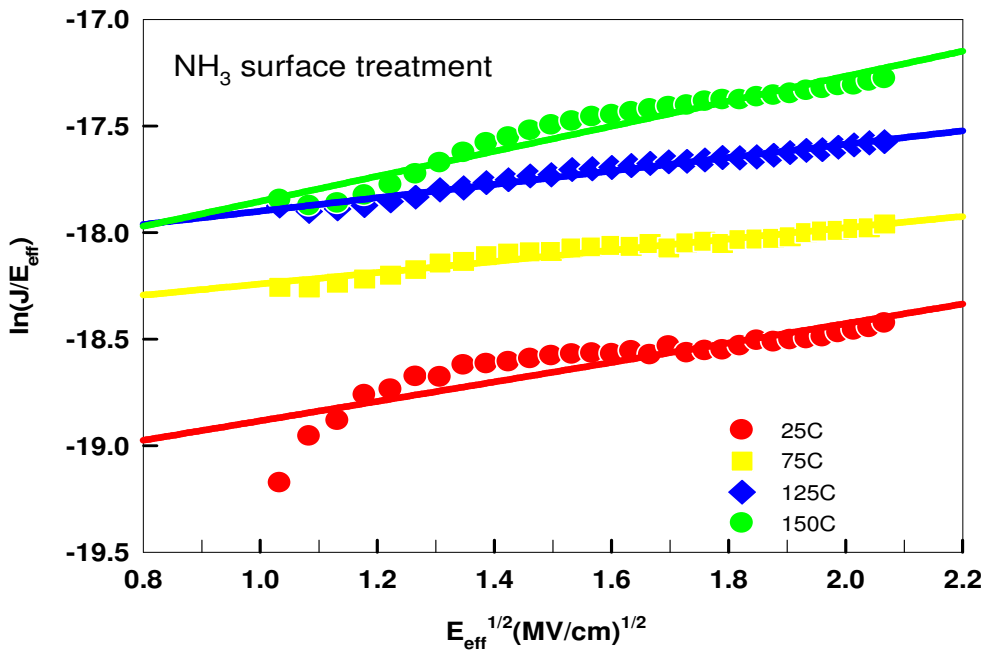


Fig 3-30 The conduction mechanism fitting of HfAlO samples with NH₃ treatment .

CHAPTER 4

Conclusions and Recommendations for Future Works

4-1 Conclusion

The high dielectric constant material HfAlO shows the property of low leakage with thinner EOT with UV ozone treatment before HfAlO deposition and NH₃ with 800°C PDA treatment after HfAlO deposition. Nitridation of the Si surface using N₂O prior to the deposition of high- κ gate dielectrics has been shown to be effective in achieving the low EOT and preventing boron penetration. However this technique results in higher interface charges, which leads to higher hysteresis and reduced channel mobility. So we use UV ozone treatment before HfAlO deposition. NH₃ treatment accompany RTA will make N diffuss into the HfAlO and form HfAlO_xN, it make partial change of the local coordination from O-Hf-O to O-Hf-N. The increase of O-Hf-N component drastically degraded the gate leakage current in HfAlO_x(N) film, and samples with NH₃ surface treatment accompany RTA treatment didn't shift V_t of HfAlO, it may be due to less additional fixed charge in the film, the V_t of UV ozone treatment have larger shift, it may be due to that O vacancy formation induced interface dipoles and subsequent electron transfer across the interface definitely causes substantial V_{th} shifts. NH₃ treatment, So we use NH₃ treatment accompany RTA treatment after HfAlO deposition. The chemical state of N incorporated HfAlO films was investigated and the physical properties were investigated. N was gradually added in the interfacial region at temperature below

800 °C, while it was significantly incorporated into the film at a temperature of 900 °C. Moreover, the N incorporated into the film is relatively more unstable than that in the interfacial region when the NH₃ is used in the nitridation process.

4.2 Suggestions of Future Work

- 1 More physical analysis to understand the properties of HfAlO film after NH₃ and RTA treatment
- 2 Another Metal gate replace with Al gate to study the capacitance and device characteristics
- 3 Comparing the properties the deposition of HfAlO_xN_y with the deposition of HfAlO with NH₃ and RTA treatment



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Investigation of Surface Treatment Before and After

High-k(HfAlO) Dielectric Deposition