# **MOCVD**

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### E-mail: tschao@cc.nctu.edu.tw

### **Introduction**

For standard  $SiO<sub>2</sub>$  or SiON gate insulators with thickness <1.5 nm, leakage currents will become prohibitive in a number of the applications of complementary metal-oxide semiconductor technology.<sup>1</sup> Thus, to improve reliability and reduce power consumption, there is a worldwide search for a replacement for these gate dielectrics having a higher dielectric constant, κ. Recent attention has focused on the silicates of those transition or rare -earth metals that are thermodynamically stable on silicon in the sense that their oxides will not be reduced by silicon to form metal or silicide layers. Zirconium silicate films have been shown to be promising candidates for a replacement gate insulator in deep-submicron CMOS technology but a production-worthy deposition technique still needs to be found. One such technique is metal-organic chemical vapour depositon (MOCVD). In MOCVD, alkoxide and ß-diketonate precursors are commonly used. However, the former are air- and moisture-sensitive. The ß-diketonates are less ambient sensitive, and have higher thermal stability requiring higher deposition temperatures. The films can exhibit severe carbon contamination or, for the ß-diketonates with fluorine-containing ligands, fluorine contamination. In order to combine the advantages of alkoxides and ß-diketonates, mixed ligand Zr and Si precursors were developed to lower the thermal stability but

maintain a higher ambient stability and higher volatility.

In this work, ultra-thin Zr silicate films were deposited on Si(100) using mixed alkoxide β-diketonate precursors<sup>2</sup>.  $Zr(O^i-Pr)_2(thd)_2$  and  $Si(O<sup>t</sup>-Bu)<sub>2</sub>(thd)<sub>2</sub>$  dissolved in octane were introduced into the reactor with a liquid -injection system.  $O_2$  or NO were introduced into the chamber separately to reduce C contamination in the films. The C concentration was further reduced by using a "pulse-mode" deposition in which the precursors and the oxidant were introduced during separate steps. Analysis by *X*-ray Photoelectron Spectroscopy (XPS) has shown that using this pulse-mode technique both NO and  $O_2$  reduce the carbon contamination in the films effectively.

The zirconium silicate films have been analyzed by high-resolution transmission electron microscopy (HRTEM), atomic force microscopy (AFM), *x*-ray photoelectron spectroscopy (XPS), and medium energy ion scattering (MEIS). The effects of post deposition anneals were investigated. The current density versus voltage (*J-V*) characteristics and capacitance versus voltage (*C-V*) characteristics were analysed for Al-gated capacitors.

### **Result and Discussion**

Fig. 1 shows the XPS Zr3d and C1s spectra for a  $Zr_{0.5}Si_{0.5}O_2$  film deposited using NO. No evidence of C-Si, Zr-C and Zr-Si bonds were found in the spectra of the optimized films. HRTEM images show that the as-deposited Zr silicate films and the interfacial layers are amorphous (Fig. 2).

 Nitric oxide forms an oxynitride buffer layer which retards oxygen diffusion to the Si interface, thus minimizing the thickness of the interfacial layer. The thinnest interfacial layers for the as-deposited films were 0.6 nm determined by HRTEM (see Fig. 2) or 0.3 nm determined by medium energy ion-scattering (MEIS). After post deposition anneals at 850 ºC Zr silicate films with the composition  $Zr_{0.5}Si_{0.5}O_2$  and their interfacial layers were still amorphous and no phase separation was observed. The Zr concentration in the interfacial layer was below the sensitivity of the MEIS technique  $($   $>$  5 at.%) before and after annealing and this has been confirmed by EELS analysis.

The films were annealed in  $O_2$  to remove surface carbon contamination from unreacted precursor on the surface of the film and improve the electrical quality of the interface layer. A "spike anneal in  $O_2$  at 850 °C increased the thickness of the interfacial layer.

Fig. 3a shows the MEIS spectrum for the as-deposited film. The C, N, and O peaks for pieces of the same wafer before and after a spike anneal in  $O_2$  at 850 °C is shown in Fig. 3b. The most important change in the MEIS spectra is observed in the broadening of the O peak on the low energy side. The simulation indicates that the interfacial layer increased to a thickness of 0.9 nm.

The Zr silicate films described in this work exhibit promising *C-V* (Fig. 4) and *J-V* (Fig. 5) characteristics. The hysteresis, fixed charge density and the effective density of trapped charge at the interface were improved significantly by anneals at 850 ºC (Fig. 6). Most of the frequency-dependent interface states, not passivated by the low temperature FGA, were eliminated after the 60 s nitrogen anneal at 850 ºC (Fig. 7). After a spike anneal in  $O_2$  at 850 °C and an anneal for 60 s in  $N_2$  at 850°C, a Zr silicate film with an equivalent oxide thickness (EOT) of 2.3 nm has exhibited a leakage current about 20 times lower than that of a SiO<sub>2</sub> film with the same EOT.

In this study ultra-thin Zr silicate films were deposited on Si(100) using  $Zr(O^i-Pr)_2(thd)_2$  and  $Si(O<sup>t</sup>-Bu)<sub>2</sub>(thd)<sub>2</sub>$  dissolved in octane. Either  $O<sub>2</sub>$  or NO was used as the oxidizing gas. One of the functions of the  $O_2$  anneal was to eliminate the C from precursor evidently left on the surface of the film. The fact that this remaining precursor was not oxidized during the final oxidation stage in the deposition chamber suggests that the decomposition of the precursor is a rather complicated process requiring both a separate oxidant and species produced during the succeeding precursor pulse. In the precursors used for the Zr silicate deposition the Zr and Si are co-ordinated to six O atoms, while only two are required to form the silicates. This leaves 4 oxygen atoms per molecule that may release active oxygen during the decomposition of the precursor. The ultra-thin Zr silicate films described in this work exhibit promising *C-V* and *J-V* characteristics.

#### **Conclusion**

These electrical properties demonstrate that a thin Zr silicate film formed by  $Zr(O^i-Pr)_2(thd)_2$ ,  $Si(O<sup>t</sup>-Bu)<sub>2</sub>(thd)<sub>2</sub>$  and nitric oxide might be a promising gate dielectric for deep submicron CMOS devices, if ways can be found to decrease the interface layer thickness.

### **References**

1. G.D. Wilk, R.M. Wallace and J.M. Anthony, *J. Appl. Phys.* **89**, 5243 (2001).

2. A.C. Jones, *Chem. Vap. Deposition*, **4**, 169 (1998).



Figure. 1. XPS spectra with background subtracted showing data (,,,), fitted peaks (- - -), and sum of fitted peaks (→) for the as-deposited  $Zr_{0.5}Si_{0.5}O_2$  film.



Figure. 2. HRTEM cross section of as-deposited  $Zr_{0.5}Si_{0.5}O_2$ film.



Figure3 (a) MEIS spectrum of 2.7 nm thick as-deposited Zr silicate film with a Si:Zr ratio of 1.3:1:  $(.,.)$  data,  $(\rightarrow)$ simulation. (b) C, N, and O peaks for the as-deposited Zr silicate film with a Si:Zr ratio of 1.3:1 of Fig. 3a before and after spike anneal in  $O_2$  at  $850 °C$ : (,,,) before anneal, (−) simulation, (777) after anneal, (-----) simulation.



Figure4. High frequency *C-V* characteristics of capacitors with Zr silicate gate dielectric obtained at 100 kHz. The samples were spike annealed in  $O_2$  at 850 °C (!!!), spike annealed in O<sub>2</sub> at 850 °C and FGA at 380 °C for 20 min  $(.,.)$ spike annealed in O<sub>2</sub> at 850 °C, then annealed in N<sub>2</sub> at 850 °C for 60 s and FGA at 380 ºC for 20 min(777). The equivalent oxide thicknesses are 2.3, 2.22, and 2.32 nm, respectively.





Figure 5. Current density as a function of gate voltage for the Zr silicate film of Fig. 6 with for substrate injection: as-deposited (777), after a spike anneal at 850 °C in  $O_2$  and a FGA at  $380 °C$  for  $20$  min (  $\qquad$  ), after only a spike anneal at 850 °C in  $O_2$  (  $\phantom{O_2}$ ), after a spike anneal at 850 °C in  $O_2$ , followed by a nitrogen anneal at 850 ºC for 60 s and a FGA at 380 ºC for 20 min (!!!).

Figure7. Frequency dependence of the *C*-V characteristics of the Zr silicate film which has an EOT of 2.32 nm after a spike anneal at 850 °C in O<sub>2</sub>, followed by a nitrogen anneal at 850 ºC for 60 s and a FGA at 380 ºC for 20 min: 10 kHz (!!!), 100 kHz (,,,), and 1 MHz (777). A slight frequency dispersion is evident below midgap, indicating the existence of donor-like interface states. The inset shows the frequency dispersion of the sample after a spike anneal at  $850 °C$  in  $O<sub>2</sub>$  and a FGA at 380 ºC for 20 min.



Figure 6. The effect of different post deposition anneals for Zr silicate films on (a) the flatband voltage shift,  $V_{FB}$ , (b) the fixed charge density, *Nf* , and the effective density of trapped charge at the interface  $N_t$ . The spike anneal was done in  $O_2$  at 850 ºC. The FGA is a forming gas anneal at 380 ºC for 20 min. The  $N_2$  anneal was done at 850 °C for 60 s.